# Cross metathesis of allyl alcohols 

how to suppress and how to promote double bond isomerization

## Suggested citation referring to the original publication:

Organic \& biomolecular chemistry 11 (2013), pp. 4194-4206
DOI http://dx.doi.org/10.1039/c3ob40167g

[^0]Cite this: Org. Biomol. Chem., 2013, 11, 4194

Received 25th January 2013,
Accepted 30th April 2013
DOI: 10.1039/c3ob40167g
www.rsc.org/obc

# Cross metathesis of allyl alcohols: how to suppress and how to promote double bond isomerization $\dagger$ 

Bernd Schmidt* and Sylvia Hauke


#### Abstract

Under standard conditions the cross metathesis of allyl alcohols and methyl acrylate is accompanied by the formation of ketones, resulting from uncontrolled and undesired double bond isomerization. By conducting the CM in the presence of phenol, the catalyst loading and the reaction time required for quantiative conversion can be reduced, and isomerization can be suppressed. On the other hand, consecutive isomerization can be deliberately promoted by evaporating excess methyl acrylate after completing cross metathesis and by adding a base or silane as chemical triggers.


## Introduction

De novo synthesis is a useful method to obtain rare carbohydrates from non-carbohydrate precursors. ${ }^{1-21}$ Our contributions to the field have focused on the application of the assisted tandem RCM-isomerization sequence ${ }^{22-24}$ to the synthesis of deoxy glycals. ${ }^{25-27}$ This synthetic method relies on the in situ conversion of the Ru-carbene into a Ru-hydride by the addition of a "chemical trigger" ${ }^{28}$ after completion of the metathesis step. The Ru-hydride then catalyzes a subsequent double bond isomerization. We and others have proposed several additives to trigger the isomerization reaction, such as a diluted hydrogen atmosphere, ${ }^{29}$ inorganic hydrides, ${ }^{30} \mathrm{NaOH}$ and 2-propanol, ${ }^{31}$ vinyl ethers ${ }^{32,33}$ or silanes. ${ }^{32}$ In particular, with second generation catalysts, uncontrolled isomerization reactions may occur even in the absence of isomerization inducing additives, because the propagating Ru-NHC-methylidene species undergoes a bimolecular decomposition into a Ruhydride at elevated temperatures. ${ }^{34,35}$ In these cases isomerization is often an undesired side reaction and measures to prevent double bond migration by trapping the Ru-hydride, $e . g$. with benzoquinone, have been devised. ${ }^{36}$

We encountered such an undesired isomerization side reaction when investigating a new synthesis of the highly deoxygenated carbohydrate amicetose, based on the cross metathesis ${ }^{37,38}$ reaction of allyl alcohols 2 (derived from $S$-ethyl lactate) ${ }^{26,39}$ and methyl acrylate (3) (Scheme 1).

In spite of a literature precedent describing the successful cross metathesis of various allyl alcohols and methyl acrylate, ${ }^{40-45}$

[^1]

Scheme 1 Envisaged synthesis of a protected amicetose.
we found that under standard reaction conditions the cross metathesis reaction leading to $\mathbf{1}$ is accompanied by an isomerization of the double bond to furnish a 1,4-dicarbonyl compound. Although the so-called Ru-catalyzed "redox isomerization" of allyl alcohols is well documented ${ }^{46-50}$ and has also been reported as a side reaction of cross metathesis steps involving allyl alcohols, ${ }^{51,52}$ we were surprised by the extent of this problem in our case. We expected that the tendency of CM products such as $\mathbf{1}$ to isomerize to ketones should be comparatively low, because the C-C-double bond to be isomerized is electron deficient. In addition, it has been reported that acrylates inhibit olefin isomerization, presumably by trapping Ru-hydride impurities. ${ }^{53,54}$

In this contribution we report conditions that allow the complete suppression of redox isomerization of cross metathesis products of allyl alcohols, and conditions which promote a subsequent double bond isomerization, leading selectively to 1,4-dicarbonyl compounds from allyl alcohols and methyl acrylate.

## Results and discussion

Optimization of cross metathesis conditions
Initially, we investigated the cross metathesis of $\mathbf{4 a}$ and methyl acrylate (3) using ten equivalents of the CM partner and

Table 1 Optimization of CM conditions


| Entry | Cat. <br> loading | $c / \mathrm{mol} \mathrm{L}^{-1}$ | Phenol/ equiv. | T/ ${ }^{\circ} \mathrm{C}$ | $t / \mathrm{h}$ | Yield of $5 a^{a}$ | Yield of $\mathbf{6 a}{ }^{a, b}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | $5 \mathrm{~mol} \%$ | 1.0 | - | 110 | 12 | 62\% | 28\% |
| 2 | $5 \mathrm{~mol} \%$ | 0.5 | - | 110 | 2.5 | 80\% | 12\% |
| 3 | $5 \mathrm{~mol} \%$ | 0.5 | - | 110 | 0.5 | 83\% | n.d. |
| 4 | $5 \mathrm{~mol} \%$ | 0.5 | - | 80 | 0.5 | 73\% | n.d. |
| 5 | $5 \mathrm{~mol} \%$ | 1.0 | - | 110 | 0.5 | 80\% | n.d. |
| 6 | $5 \mathrm{~mol} \%$ | Neat | - | 80 | 0.5 | 83\% | n.d. |
| 7 | $2.5 \mathrm{~mol} \%$ | 0.5 | - | 110 | 0.5 | 75\% | n.d. |
| 8 | $2.5 \mathrm{~mol} \%$ | 0.5 | 0.5 | 110 | 0.5 | 86\% | n.d. |
| 9 | 2.5 mol\% | 1.0 | 0.5 | 110 | 0.5 | 98\% | $n . d$. |
| 10 | $2.5 \mathrm{~mol} \%$ | Neat | 0.5 | 80 | 0.5 | 83\% | n.d. |
| 11 | $5 \mathrm{~mol} \%$ | 0.5 | 0.5 | 110 | 2.5 | 77\% | 17\% |
| 12 | $2.5 \mathrm{~mol} \%$ | 1.0 | 0.5 | 110 | 12 | 48\% | 20\% |

${ }^{a}$ Yields of isolated and purified products. ${ }^{b}$ n.d. = not determined.
second generation catalyst $\mathbf{A}^{55}$ in refluxing toluene for 12 hours. Under these conditions, $\mathbf{5 a}$ and its isomerization product 6a were obtained in a $2: 1$ ratio (Table 1, entry 1 ). Reducing the reaction time to 2.5 hours (entry 2 ) and then to 0.5 hours (entry 3 ) led to a significant improvement. Upon lowering the temperature to $80{ }^{\circ} \mathrm{C}$ (entry 4) the conversion remained incomplete and the yield of $5 a$ was only $73 \%$. Comparison of entries 2,5 and 6 suggests that a further improvement by the variation of the initial substrate concentration is not possible. Reducing the catalyst loading to $2.5 \mathrm{~mol} \%$ (entry 7) results in a lower yield. Based on these results, we thought that the key to an isomerization-free cross metathesis could be a combination of reduced reaction temperatures, short reaction times and lower catalyst loadings, because - as mentioned in the Introduction - the isomerization is caused by a Ruhydride resulting from a thermally induced bimolecular decomposition of the Ru-NHC-methylidene species. ${ }^{34}$ Considering the fact that reducing the reaction temperature to $80^{\circ} \mathrm{C}$ resulted in a decrease of the yield by $10 \%$ (compare entries 3 and 4) we decided to maintain a reaction temperature of $110{ }^{\circ} \mathrm{C}$ and find other means to accelerate the CM reaction while reducing the catalyst loading at the same time. A very simple yet effective method to improve olefin metathesis reactions has been devised by Forman et al., who described a beneficial effect of the added phenol. ${ }^{56,57}$ Presumably, phenol coordinates to the catalytically active 14 -electron species, leading to a retarded catalyst decomposition. Indeed, the addition of phenol to the reaction mixture led to a significant improvement ( $86 \%$ compared to $75 \%$, entries 7 and 8) under otherwise identical conditions. The yield could be further improved to nearly quantitative by raising the initial substrate
concentration to 1.0 M (entry 9), but conducting the reaction without any solvent (entry 10) led to a significantly lower yield, which might, however, be attributed to the lower reaction temperature. In entries 11 and 12 the results for CM reactions in the presence of phenol at longer reaction times are listed. From these experiments it can be concluded that phenol is not an efficient isomerization inhibitor itself, but that the reduced extent of isomerization can be mainly attributed to the short reaction time, which was accomplished by phenol accelerated cross metathesis.

## Optimization of cross metathesis-isomerization conditions

Next, we set out to find the conditions that would allow the selective synthesis of the CM-isomerization product 6a. The results gathered during the optimization of the cross metathesis reaction suggest that the dicarbonyl compound $\mathbf{6 a}$ is not accessible in synthetically useful yields simply by heating the reaction mixture for a prolonged period of time. These experiments are listed in Table 1 as entries 1 and 12, and are repeated for comparison in Table 2 as entries 1 and 3. We suspected that excess methyl acrylate inhibits the isomerization to a certain extent and tested therefore a modification of the CM protocol which involves removal of the unreacted methyl acrylate by evaporation after 0.5 h , re-dissolving the mixture in toluene and heating to induce the isomerization step (Table 2, entries 2 and 4). If phenol is present in the reaction mixture, the product distribution is virtually unaffected by this measure; however, without added phenol the evaporation of excess methyl acrylate prior to inducing the isomerization leads to the formation of a substantial amount of $\mathbf{6 a}$ (entry 4). We then tested various additives to promote the Ru-carbene to Ru-hydride conversion. Surprisingly, ethyl vinyl ether ${ }^{32,58}$ proved to be ineffective, as only cross metathesis product $5 \mathbf{5}$ was observed. NaOH did promote the isomerization, but synthetically useful yields could only be obtained in the absence of phenol (entries 7-10), because removal of the phenol turned out to be laborious. We then tested triethyl silane (entries 11 and 12) and the cheaper alternative polymethylhydrosiloxane (PMHS) ${ }^{59,60}$ (entry 13) and found incomplete isomerization if phenol was present. However, if the CM reaction was conducted without phenol, the isomerized product 6a could be isolated in a synthetically useful yield of 76\% (entry 14). $\ddagger$

Trost and Kulawiec proposed a mechanism for Ru-catalyzed redox isomerizations which relies on the initial coordination of the allylic hydroxy group to the Ru centre, followed by $\beta$-hydride elimination to furnish an enone bound to a Ruhydride. This complex reacts further in a migratory insertion to a Ru-enolate, which then undergoes protonation by the allylic alcohol, liberating the ketone and a Ru-alkoxide. ${ }^{49}$ In our case, however, a slightly modified scenario might be possible, which is depicted in Scheme 2: after the formation of the Ru-hydride $\mathbf{B}$ from the reaction of the metathesis catalyst and

[^2]Table 2 Optimization of CM-isomerization conditions

|  |  |  | A (5 mol-\%); toluene (1 M); 3 <br> (10 equiv.); phenol ( 0.5 equiv.); <br> 5a <br> $110^{\circ} \mathrm{C} ; 0.5 \mathrm{~h}$; then evaporate; re-dissolve in toluene ( 0.2 M ); additive; $110^{\circ} \mathrm{C} ; \mathrm{t}(\mathrm{h})$ |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Entry | Cat. loading | Phenol added? | $t / \mathrm{h}$ | Excess 3 evaporated? | Additive $^{c}$ (equiv.) | Yield of $5 \mathbf{a}^{a, b}$ | Yield of $\mathbf{6 a}{ }^{a, b}$ |
| 1 | $2.5 \mathrm{~mol} \%$ | Yes | 12 | No | - | 48\% | 20\% |
| 2 | $2.5 \mathrm{~mol} \%$ | Yes | 12 | Yes | - | 55\% | 21\% |
| 3 | $5 \mathrm{~mol} \%$ | No | 12 | No | - | 62\% | 28\% |
| 4 | $5 \mathrm{~mol} \%$ | No | 12 | Yes | - | 26\% | 56\% |
| 5 | $5 \mathrm{~mol} \%$ | Yes | 12 | Yes | EVE (5) | n.d. | <5\% |
| 6 | $5 \mathrm{~mol} \%$ | No | 12 | Yes | EVE (5) | n.d. | <5\% |
| 7 | $5 \mathrm{~mol} \%$ | Yes | 3 | Yes | NaOH (0.5) | n.d. | 29\% |
| 8 | $5 \mathrm{~mol} \%$ | No | 3 | Yes | $\mathrm{NaOH}(0.5)$ | n.d. | 65\% |
| 9 | $5 \mathrm{~mol} \%$ | Yes | 3 | Yes | NaOH (1.5) | n.d. | 70\% |
| 10 | $5 \mathrm{~mol} \%$ | No | 3 | Yes | NaOH (1.5) | n.d. | 70\% |
| 11 | $5 \mathrm{~mol} \%$ | Yes | 12 | Yes | TESH (0.5) | 23\% | 45\% |
| 12 | $5 \mathrm{~mol} \%$ | No | 12 | Yes | TESH (0.5) | n.d. | 67\% |
| 13 | $5 \mathrm{~mol} \%$ | Yes | 12 | Yes | PMHS (0.2) ${ }^{d}$ | 29\% | 39\% |
| 14 | $5 \mathrm{~mol} \%$ | No | 12 | Yes | PMHS (0.2) ${ }^{\text {d }}$ | n.d. | 76\% |

${ }^{a}$ Yields of isolated and purified products. ${ }^{b}$ n.d. $=$ not determined. ${ }^{c}$ Abbreviations for additives: EVE = ethyl vinyl ether; TESH $=$ triethylsilane; PMHS = polymethylhydrosiloxane. ${ }^{d}$ Equivalents of PMHS were calculated based on an effective mass of 60 g mol ${ }^{-1}$ per hydride. ${ }^{59}$


Scheme 2 Mechanistic rationale for the isomerization step.
the isomerization inducing additive, a conjugate addition to the cross metathesis product 5 occurs, yielding a Ru-enolate $\mathbf{C}$. Precedence for the formation of such Ru-enolates ${ }^{61}$ and their intermediacy in catalytic cycles ${ }^{62,63}$ exists. In this particular case, the enolate may undergo cyclization to $\mathbf{D}$, which reacts in an inter- or intramolecular proton transfer to furnish Ru-alkoxide E. From this intermediate, the Ru-hydride can be regenerated via $\beta$-hydride elimination and formation of the isomerized product 6. In this scenario, the inhibitory effect of excess methyl acrylate ${ }^{53,54}$ can be explained by a conjugate addition of the Ru-hydride to methyl acrylate (3), giving the

Ru-enolate $\mathbf{F}$. Due to the large excess of acrylate, the equilibrium is shifted to the enolate, thereby removing major amounts of the isomerization catalyst from the reaction mixture.

## Application of isomerization-free CM and CM-isomerization conditions to other allyl alcohols

The optimization studies revealed that the best conditions for a cross metathesis-isomerization sequence of allyl alcohols appear to be the use of a higher catalyst loading of $5 \mathrm{~mol} \%$, absence of phenol, removal of excess acrylate after the completion of the CM step, and addition of either 1.5 equiv. of NaOH or 0.2 equiv. of PMHS to trigger the formation of the isomerization catalyst. On the other hand, for isomerizationfree cross metathesis, a reduced catalyst loading of $2.5 \mathrm{~mol} \%$, addition of phenol as a rate accelerating agent and short reaction times of 0.5 h at $110{ }^{\circ} \mathrm{C}$ appear to be beneficial. Having established the optimum conditions for isomerization-free cross metathesis and for the CM-isomerization sequence of allyl alcohol 4a, we applied these protocols to several other allyl alcohols $\mathbf{4 b} \mathbf{- 4 p}$ (Table 3). In general, the CM products 5 were obtained in high yields of $c a .90 \%$. For the 1,4 -dicarbonyl compounds $\mathbf{6}$, yields of approximately $70 \%$ could be obtained. Although the yields obtained with NaOH are somewhat lower for most examples, reaction times for the isomerization step are significantly shorter (ca. 3 h ) compared to those with PMHS as an additive ( $c a .12 \mathrm{~h}$ ). Notably, isomerization induced by PMHS as a chemical trigger appears to be significantly faster than reduction of the C -C-double bond, which
Table 3 Scope of isomerization-free CM and CM-isomerization sequence
Entry
Table 3 (Contd.)
Entry
 add PMHS ( 0.2 equiv.); $110^{\circ} \mathrm{C}, 12 \mathrm{~h} .{ }^{d}$ A complex mixture of products. ${ }^{e} 10 \mathrm{~mol} \%$ of $\mathbf{A}$ and 20 equiv. of methyl acrylate were used.
has also been observed with silanes in the presence of Rucarbene complexes. ${ }^{64,65}$

## Synthesis of BOM-protected amicetose

Hydrogenation of $\mathbf{5 a}$ using commercial $\mathrm{Pd} / \mathrm{C}$ as a catalyst resulted in spontaneous cyclization to the $\gamma$-butyrolactone 7 . For these reasons we thought it might be advantageous to mask the $\mathrm{C} 4-\mathrm{OH}$ group prior to hydrogenation with a protecting group orthogonal to TBS. Initially, the MOM-group was chosen, and the corresponding MOM-ether 8a could be isolated in $81 \%$ yield. Hydrogenation was again accomplished in quantitative yield using hydrogen and Pd/C. Unfortunately, this method turned out to be very unreliable, because when we changed the sample of Pd/C the reaction failed completely. Reproducibility problems associated with Pd/C have been known for quite some time ${ }^{66}$ and the in situ preparation of such catalysts from $\mathrm{Pd}(\mathrm{OAc})_{2}$ and activated carbon has been suggested as an alternative. ${ }^{67,68}$ Considering that the hydrogenation in question involves a double bond in conjugation with an ester group, we thought that Lipshutz' variant ${ }^{60}$ of Stryker's reagent ${ }^{69}$ should work well. Indeed, the desired ester 9a was reliably and reproducibly obtained in nearly quantitative yield with $[(\mathrm{BDP}) \mathrm{CuH}]$ as a catalyst, which was formed in situ from 1,2-bis(diphenyl-phosphino)benzene (BDP), Cu-(II)acetate and polymethylhydro-siloxane (PMHS) as a reducing agent (Scheme 3).

Unfortunately, it turned out at this point that the MOMprotecting group was rather impractical, because after desilylation it became virtually impossible to control the progress of any reaction steps via TLC. For this reason, we replaced the MOM-group by the UV-active benzyloxymethyl-(BOM)-protecting group. The analogous compound $\mathbf{8 b}$ was obtained in somewhat better yield than $\mathbf{8 a}$, and the subsequent conjugate reduction with the modified Stryker's reagent gave 9b in quantitative yield. Desilylation and lactonization led to compound 10, which was finally reduced with DIBAl-H to the desired BOM-protected amicetose 11 (Scheme 4).


Scheme 3 Towards MOM-protected amicetose.


Scheme 4 Synthesis of BOM-protected amicetose.

## Conclusions

In summary, we have described a protocol for the rapid, iso-merization-free cross metathesis of allyl alcohols and methyl acrylate using phenol as an efficient rate-accelerating reagent. These conditions were applied to an allylic alcohol derived from $S$-ethyl lactate, and the resulting cross metathesis product could be elaborated in a few steps to the 2,3,6-trideoxy sugar amicetose as a BOM-ether. The isomerization-free CM conditions were complemented by the development of a protocol which allows the synthesis of 1,4-dicarbonyl compounds from allyl alcohols through an assisted tandem CM-isomerization sequence, using a single precatalyst and base or a silane as a chemical trigger for inducing the conversion of the Rucarbene catalyst into a Ru-hydride.

## Experimental

## General remarks

All experiments were conducted in dry reaction vessels in an atmosphere of dry nitrogen. Solvents were purified by standard procedures. ${ }^{1} \mathrm{H}$ NMR spectra were obtained at 300 MHz or at 600 MHz in $\mathrm{CDCl}_{3}$ with $\mathrm{CHCl}_{3}(\delta=7.24 \mathrm{ppm})$ as an internal standard. Coupling constants ( $J$ ) are given in Hz. ${ }^{13} \mathrm{C}$ NMR spectra were recorded at 75 MHz or at 150 MHz in $\mathrm{CDCl}_{3}$ with $\mathrm{CDCl}_{3}(\delta=77.0 \mathrm{ppm})$ as an internal standard. The number of coupled protons was analyzed by APT-experiments and is denoted by the number in parentheses following the chemical shift value. IR spectra were recorded neat on NaCl or KBr plates. Wavenumbers $(\nu)$ are given in $\mathrm{cm}^{-1}$. The peak intensities are defined as strong ( s ), medium ( m ) or weak ( w ). Mass spectra were obtained at 70 eV .

## General procedure for isomerization-free cross metathesis

To a solution of the corresponding allyl alcohol ( 1.0 mmol ), methyl acrylate ( $0.9 \mathrm{~mL}, 10.0 \mathrm{mmol}$ ) and phenol ( 47 mg ,
0.5 mmol ) in dry toluene ( 1.0 mL ) was added Ru-catalyst A ( $21.2 \mathrm{mg}, 2.5 \mathrm{~mol} \%$ ). The solution was heated to $110{ }^{\circ} \mathrm{C}$ for 0.5 h , all volatiles were removed in vacuo and the residue was purified by column chromatography on silica using hexaneMTBE mixtures of increasing polarity.
(4R, 5S, E)-Methyl-5-(tert-butyldimethylsilyloxy)-4-hydroxy-hex-2-enoate (5a). Following the general procedure, 5a was obtained from 4a ( $200 \mathrm{mg}, 0.92 \mathrm{mmol}$ ) as a colourless oil (249 mg, 98\%). $[\alpha]_{\mathrm{D}}^{27}=+21.3\left(c=0.61, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 6.89(\mathrm{dd}, J=15.7,4.7,1 \mathrm{H}), 6.10(\mathrm{dd}, J=$ $15.7,1.8,1 \mathrm{H}), 4.22$ (ddd, $J=8.6,4.0,1.8,1 \mathrm{H}), 3.91(\mathrm{qd}, J=6.3$, $3.9,1 \mathrm{H}), 3.73(\mathrm{~s}, 3 \mathrm{H}), 2.57(\mathrm{~d}, J=4.0,1 \mathrm{H}), 1.07(\mathrm{~d}, J=6.3,3 \mathrm{H})$, $0.89(\mathrm{~s}, 9 \mathrm{H}), 0.07(\mathrm{~s}, 3 \mathrm{H}), 0.07(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 75 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 167.2$ (0), 146.4 (1), 121.7 (1), 75.2 (1), 71.1 (1), 51.9 (3), 26.1 (1), 18.4 (0), 18.2 (1), -4.7 (3), -4.8 (3); IR (neat): $\nu$ 3481 (w), 2930 (m), 2857 (m), 1726 (s), 1093 (s), 834 (s), 775 (s); MS (EI): $m / z 275$ (23), 257 (100), 213 (34), 143 (11), 111 (90); HRMS (ESI) calcd for $\mathrm{C}_{13} \mathrm{H}_{27} \mathrm{O}_{4} \mathrm{Si}[\mathrm{M}+\mathrm{H}]^{+}: 275.1679$, found: 275.1694; Anal. calcd for $\mathrm{C}_{13} \mathrm{H}_{26} \mathrm{O}_{4} \mathrm{Si}$ : C, 56.9; H, 9.6; found: C, 56.9; H, 9.7.
( $E$ )-Methyl-4-hydroxy-4-phenylbut-2-enoate (5b). Following the general procedure, $\mathbf{5 b}$ was obtained from $\mathbf{4 b}(134 \mathrm{mg}$, 1.0 mmol ) as a colourless liquid ( $174 \mathrm{mg}, 91 \%$ ). ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.43-7.22(5 \mathrm{H}), 7.02(\mathrm{~d}, J=15.6,1 \mathrm{H}), 6.14$ (d, $J=15.6,1 \mathrm{H}), 5.31(\mathrm{bs}, 1 \mathrm{H}), 3.70(\mathrm{~s}, 3 \mathrm{H}), 2.62(\mathrm{bs}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 167.0$ (0), 148.9 (1), 140.8 (0), 128.8 (1), 128.3 (1), 126.5 (1), 119.6 (1), 73.4 (1), 51.6 (3); IR (neat): $\nu$ 3430 (w), 2952 (w), 1705 (s), 1657 (m), 1436 (m), 1271 (s), 1167 (s), 981 (s), 698 (s); MS (EI): m/z 192 (10, [M + H $]^{+}$), 174 (30), 163 (100), 131 (100), 115 (41), 105 (74), 87 (60), 77 (85), 55 (33); HRMS (EI): calcd for $\mathrm{C}_{11} \mathrm{H}_{12} \mathrm{O}_{3}[\mathrm{M}]^{+}$: 192.0786, found: 192.0776; Anal. calcd for $\mathrm{C}_{11} \mathrm{H}_{12} \mathrm{O}_{3}: \mathrm{C}, 68.7 ; \mathrm{H}, 6.3$, found: C, 68.2; H, 6.5 .
( $E$ )-Methyl-4-(4-bromophenyl)-4-hydroxy-but-2-enoate (5c). Following the general procedure, $5 \mathbf{c}$ was obtained from $4 \mathbf{c}$ $(213 \mathrm{mg}, 1.0 \mathrm{mmol})$ as a colourless liquid ( $254 \mathrm{mg}, 94 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.46$ (ddd, $J=8.4,2 \mathrm{H}$ ), 7.18 (ddd, $J=$ $8.5,2 \mathrm{H}), 6.95(\mathrm{dd}, J=15.6,4.9,1 \mathrm{H}), 6.10(\mathrm{dd}, J=15.6,1.6,1 \mathrm{H})$, $5.26(\mathrm{dm}, J=4.0,1 \mathrm{H}), 3.70(\mathrm{~s}, 3 \mathrm{H}), 3.01(\mathrm{bs}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 166.8$ (0), 148.4 (1), 139.8 (0), 131.8 (1), 128.2 (1), 122.1 (0), 120.1 (1), 72.6 (1), 51.7 (3); IR (neat): $\nu 3426$ (m), 2951 (w), 1703 (s), 1436 (m), 1275 (s), 1167 (s), 1010 (s), 827 (s); MS (EI): m/z 270 (8, [M + H] ${ }^{+}$), 255 (18), 254 (28), 243 (79), 241 (100), 213 (25), 211 (65), 185 (57), 183 (62), 131 (47), 87 (91), 77 (54), 55 (28); HRMS (ESI): calcd for $\mathrm{C}_{11} \mathrm{H}_{12} \mathrm{O}_{3} \mathrm{Br}$ $[\mathrm{M}+\mathrm{H}]^{+}: 270.9970$, found: 270.9994.
(E)-Methyl-4-hydroxy-4-(4-methoxyphenyl)-but-2-enoate (5d). Following the general procedure, $\mathbf{5 d}$ was obtained from $\mathbf{4 d}$ $(164 \mathrm{mg}, 1.0 \mathrm{mmol})$ as a colourless liquid (203 mg, 91\%). ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.24(\mathrm{dm}, J=8.5,2 \mathrm{H}), 7.02(\mathrm{ddd}, J=$ $15.6,4.7,0.5,1 \mathrm{H}), 6.87(\mathrm{dm}, J=8.6,2 \mathrm{H}), 6.14$ (ddd, $J=15.6$, $0.8,0.7,1 \mathrm{H}), 5.28(\mathrm{~d}, J=4.6,1 \mathrm{H}), 3.79(\mathrm{~s}, 3 \mathrm{H}), 3.72(\mathrm{~s}, 3 \mathrm{H}), 2.37$ (bs, 1H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 167.0$ (0), 159.6 (0), 149.1 (1), 133.0 (0), 128.0 (1), 119.5 (1), 114.2 (1), 73.0 (1), 55.3 (3), 51.6 (3); IR (neat): $\nu 3443$ (m), 2953 (w), 1716 (s), 1511 (s), 1247 (s), 1168 (s), 1030 (s), $833(\mathrm{~m})$; MS (EI): m/z 222 (41, [M] ${ }^{+}$), 193
(28), 161 (75), 145 (41), 135 (100), 121 (43), 91 (37), 77 (55), 55 (31); HRMS (EI): calcd for $\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{O}_{4}[\mathrm{M}]^{+}: 222.0892$, found: 222.0874; Anal. calcd for $\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{O}_{4}$ : C, 64.9; H, 6.4, found: C, 64.6; H, 6.5.
(E)-Methyl 4-hydroxy-4-(3-methoxy-4-(methoxymethoxy)-phenyl)-but-2-enoate (5e). Following the general procedure, 5e was obtained from $4 \mathbf{e}(224 \mathrm{mg}, 1.0 \mathrm{mmol})$ as a colourless liquid ( $239 \mathrm{mg}, 85 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.09$ (d, $J=$ $8.2,1 \mathrm{H}$ ), 7.01 (dd, $J=15.6,4.8,1 \mathrm{H}), 6.88$ (d, $J=2.0,1 \mathrm{H}), 6.82$ (dd, $J=8.3,2.0,1 \mathrm{H}), 6.13$ (dd, $J=15.6,1.7,1 \mathrm{H}), 5.28(\mathrm{~m}, 1 \mathrm{H})$, $5.19(\mathrm{~s}, 2 \mathrm{H}), 3.85(\mathrm{~s}, 3 \mathrm{H}), 3.71(\mathrm{~s}, 3 \mathrm{H}), 3.48(\mathrm{~s}, 3 \mathrm{H}), 2.53(\mathrm{bs}$, 1H); ${ }^{13}$ C NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 166.7$ (0), 148.8 (1), 135.3 (0), 119.6 (1), 119.0 (1), 116.5 (1), 110.0 (1), 95.4 (2), 73.2 (1), 56.1 (3), 55.9 (3), 51.6 (3); IR (neat): $\nu 3454$ (w), 2952 (w), 1720 (m), 1508 (m), 1261 (s), 1152 (s), 979 (s); MS (EI): m/z 282 (87, [M] ${ }^{+}$), 265 (27), 233 (30), 220 (29), 45 (100); HRMS (ESI): calcd for $\mathrm{C}_{14} \mathrm{H}_{19} \mathrm{O}_{6}[\mathrm{M}+\mathrm{H}]^{+}$: 283.1182, found: 283.1198; Anal. calcd for $\mathrm{C}_{14} \mathrm{H}_{18} \mathrm{O}_{6}$ : C, $59.6 ; \mathrm{H}, 6.4$; found: C, 59.4; H, 6.6 .
(E)-Methyl-4-(4-(tert-butyldimethylsilyloxy)phenyl)-4-hydro-xybut-2-enoate (5f). Following the general procedure, $5 \mathbf{f}$ was obtained from $\mathbf{4 f}$ ( $264 \mathrm{mg}, 1.0 \mathrm{mmol}$ ) as a colourless liquid ( $295 \mathrm{mg}, 92 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.18$ (d, $J=7.4$, $2 \mathrm{H}), 7.03(\mathrm{dm}, J=15.6,1 \mathrm{H}), 6.81(\mathrm{~d}, J=7.4,2 \mathrm{H}), 6.14(\mathrm{~d}, J=$ $15.6,1 \mathrm{H}), 5.28(\mathrm{bs}, 1 \mathrm{H}), 3.73(\mathrm{~s}, 3 \mathrm{H}), 2.30(\mathrm{bs}, 1 \mathrm{H}), 0.97(\mathrm{~s}, 9 \mathrm{H})$, 0.18 (s, 6H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 167.0$ (0), 155.8 (0), 149.1 (1), 133.6 (0), 127.9 (1), 120.4 (1), 119.5 (1), 73.1 (1), 51.6 (3), 25.7 (3), 18.2 (0), -4.5 (3); IR (neat): $\nu 3431(\mathrm{~m}), 2932(\mathrm{w})$, 1723 (m), 1508 (s), 1255 ( s), 1165 (s), 910 (s), 837 (s), 780 (s); MS (EI): m/z 322 (38, [M] ${ }^{+}$), 291 (31), 265 (100), 235 (50), 209 (49), 179 (30), 135 (25), 89 (50), 55 (20); HRMS (EI): calcd for $\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{O}_{4} \mathrm{Si}[\mathrm{M}]^{+}: 322.1600$, found: 322.1628; Anal. calcd for $\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{O}_{4} \mathrm{Si}: \mathrm{C}, 63.3$; H, 8.1, found: C, 63.2; H, 8.4.
( $E$ )-Methyl-4-(4-benzyloxyphenyl)-4-hydroxybut-2-enoate (5g). Following the general procedure, $5 \mathbf{g}$ was obtained from $\mathbf{4 g}$ ( $240 \mathrm{mg}, 1.0 \mathrm{mmol}$ ) as a colourless solid ( $263 \mathrm{mg}, 88 \%$ ), mp $69-70{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.49-7.30(5 \mathrm{H}), 7.26(\mathrm{~d}$, $J=6.8,2 \mathrm{H}), 7.05(\mathrm{dm}, J=15.7,1 \mathrm{H}), 6.97(\mathrm{~d}, J=6.9,2 \mathrm{H}), 6.16$ (d, $J=15.6,1 \mathrm{H}), 5.30(\mathrm{bs}, 1 \mathrm{H}), 5.07(\mathrm{~s}, 2 \mathrm{H}), 3.74(\mathrm{~s}, 3 \mathrm{H}), 3.72(\mathrm{~s}$, $3 \mathrm{H}), 2.36$ (bs, 1H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 167.0$ (0), 159.6 (0), 149.1 (1), 133.0 (0), 128.0 (1), 119.5 (1), 114.2 (1), 73.0 (1), 55.3 (3), 51.6 (3); IR (neat): ע 3436 (m), 3031 (w), 1707 (s), 1509 (s), 1238 (s), 1168 (s), 1016 (s), 739 (s); MS (EI): m/z 298 (20, [M] ${ }^{+}$), 281 (7), 91 (100), 65 (7); HRMS (EI): calcd for $\mathrm{C}_{18} \mathrm{H}_{18} \mathrm{O}_{4}$ $[\mathrm{M}]^{+}: 298.1205$, found: 298.1206; Anal. calcd for $\mathrm{C}_{18} \mathrm{H}_{18} \mathrm{O}_{4}: \mathrm{C}$, 72.5; H, 6.1, found: C, 72.5; H, 6.2.
(E)-Methyl-4-(2-fluorophenyl)-4-hydroxybut-2-enoate (5h). Following the general procedure, $\mathbf{5 h}$ was obtained from $\mathbf{4 h}$ ( $152 \mathrm{mg}, 1.0 \mathrm{mmol}$ ) as a colourless liquid ( $198 \mathrm{mg}, 94 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.40(\mathrm{t}, J=7.2,1 \mathrm{H}), 7.27(\mathrm{~m}, 1 \mathrm{H})$, $7.14(\mathrm{~m}, 1 \mathrm{H}), 7.09-6.95(2 \mathrm{H}), 6.15(\mathrm{~d}, J=15.6,1 \mathrm{H}), 5.66(\mathrm{bs}$, $1 \mathrm{H}), 3.71(\mathrm{~s}, 3 \mathrm{H}), 3.06(\mathrm{~s}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 167.0 ( 0 ), 159.7 (d, $J=246.9,0$ ), 147.8 (1), 129.7 (d, $J=8.3,1$ ), 128.0 (d, $J=13.4,0), 127.7$ (d, $J=3.8,1$ ), 124.5 (d, $J=3.5,1$ ), 120.0 (1), 115.5 (d, $J=21.5,1$ ), 67.0 (d, $J=3.4,1$ ), 51.7 (3); IR (neat): $\nu 3436(\mathrm{~m}), 2953(\mathrm{w}), 1705(\mathrm{~s}), 1274(\mathrm{~s}), 1171(\mathrm{~s}), 759(\mathrm{~s}) ;$ MS (EI): m/z 210 (3, [M] $]^{+}$), 181 (100), 149 (77), 123 (56), 87 (56),

55 (31); HRMS (EI): calcd for $\mathrm{C}_{11} \mathrm{H}_{11} \mathrm{O}_{3} \mathrm{~F}[\mathrm{M}]^{+}: 210.0692$, found: 210.0685.
( $E$ )-Methyl-4-hydroxy-5-phenylpent-2-enoate (5i). Following the general procedure, $5 \mathbf{i}$ was obtained from $4 \mathbf{i}(148 \mathrm{mg}$, 1.0 mmol ) as a colourless liquid ( $182 \mathrm{mg}, 88 \%$ ). ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.42-7.17(5 \mathrm{H}), 7.02(\mathrm{dm}, J=15.7,1 \mathrm{H})$, 6.06 (d, $J=15.6,1 \mathrm{H}$ ), $4.52(\mathrm{bs}, 1 \mathrm{H}), 3.74(\mathrm{~s}, 3 \mathrm{H}), 3.02-2.72$ (2H), 2.09 (bs, 1H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 167.0$ (0), 149.2 (1), 136.8 (0), 129.4 (1), 128.6 (1), 126.9 (1), 120.1 (1), 71.7 (1), 51.6 (3), 43.2 (2); IR (neat): $\boldsymbol{\nu} 3435$ (m), 2950 (w), 1707 (s), 1437 (m), 1275 (s), 1168 (s), 702 (s); MS (EI): m/z 206 (1, [M] ${ }^{+}$), 115 (18), 91 (100), 83 (76); HRMS (EI): calcd for $\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{O}_{3}[\mathrm{M}]^{+}$: 206.0943, found: 206.0954; Anal. calcd for $\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{O}_{3}$ : C, 69.9; H, 6.8, found: C, 69.4; H, 7.1.
(E)-Methyl-4-hydroxy-6-phenylhex-2-enoate (5j). Following the general procedure, $5 \mathbf{j}$ was obtained from $\mathbf{4 j}$ ( 162 mg , 1.0 mmol ) as a colourless solid ( $197 \mathrm{mg}, 90 \%$ ), mp 29-31 ${ }^{\circ} \mathrm{C}$. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.36-7.16(5 \mathrm{H}), 6.97(\mathrm{dd}, J=15.7$, $4.9,1 \mathrm{H}), 6.07$ (dd, $J=15.6,1.7,1 \mathrm{H}), 4.32(\mathrm{~m}, 1 \mathrm{H}), 3.75(\mathrm{~s}, 3 \mathrm{H})$, 2.85-2.66 (2H), $2.41(\mathrm{bs}, 1 \mathrm{H}), 1.98-1.81(2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 167.0$ (0), 150.3 (1), 141.2 (0), 128.4 (1), 128.4 (1), 126.0 (1), 119.9 (1), 70.2 (1), 51.6 (3), 37.9 (2), 31.3 (2); IR (neat): $\nu 3436$ (m), 2949 (w), 1705 (s), 1275 (s), 1169 (m), 700 (s); MS (EI): m/z 221 (18, [M + H] ${ }^{+}$), 220 (7, [M] ${ }^{+}$), 202 (18), 142 (30), 128 (30), 105 (67), 91 (100), 87 (46); HRMS (EI): calcd for $\mathrm{C}_{13} \mathrm{H}_{16} \mathrm{O}_{3}[\mathrm{M}]^{+}: 220.1099$, found: 220.1111; Anal. calcd for $\mathrm{C}_{13} \mathrm{H}_{16} \mathrm{O}_{3}$ : C, 70.9; H, 7.3, found: C, 70.4; H, 7.2.
(E)-Methyl-4-hydroxy-6-methylhept-2-enoate (5k). Following the general procedure, $5 \mathbf{k}$ was obtained from $\mathbf{4 k}$ ( 114 mg , 1.0 mmol ) as a colourless liquid ( $162 \mathrm{mg}, 94 \%$ ). ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 6.93(\mathrm{dm}, J=15.7,1 \mathrm{H}), 6.02(\mathrm{~d}, J=15.6$, 1H), 4.34 (bs, 1H), 3.72 (s, 3H), 2.10 (bs, 1H), 1.78 (m, 1H), 1.57-1.28 (2H), $0.92(\mathrm{~d}, J=5.2,6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 167.1$ (0), 151.0 (1), 119.3 (1), 69.3 (1), 51.6 (3), 45.6 (2), 24.4 (1), 23.0 (3), 22.0 (3); IR (neat): $\nu 3432$ (m), 2956 (m), 1706 (s), 1275 (s), 1170 (s), 938 (s); MS (EI): m/z 143 (31), 115 (42), 87 (100), 55 (38), 41 (25); HRMS (ESI): calcd for $\mathrm{C}_{9} \mathrm{H}_{17} \mathrm{O}_{3}[\mathrm{M}+\mathrm{H}]^{+}$: 173.1178, found: 173.1192; Anal. calcd for $\mathrm{C}_{9} \mathrm{H}_{16} \mathrm{O}_{3}$ : $\mathrm{C}, 62.8 ; \mathrm{H}$, 9.4, found: C, 62.8; H, 9.6.
(E)-Methyl-4-hydroxy-4-(( $R$ )-1,4-dioxaspiro[4.5]decan-2-yl)-but-2-enoate (51). Following the general procedure, $5 \mathbf{l}$ was obtained from $41(198 \mathrm{mg}, 1.0 \mathrm{mmol})$ as a colourless liquid ( $233 \mathrm{mg}, 91 \%$ ) as an inseparable mixture of diastereomers. Analytical data were obtained from the mixture. ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 6.90(\mathrm{dd}, J=15.8,4.1,0.5 \mathrm{H}), 6.85(\mathrm{dd}, J=$ 15.7, 4.7, 0.5H), 6.16 (dd, $J=15.7,2.6,0.5 \mathrm{H}), 6.14$ (dd, $J=15.7$, $2.6,0.5 \mathrm{H}), 4.46(\mathrm{~m}, 0.5 \mathrm{H}), 4.21(\mathrm{~m}, 0.5 \mathrm{H}), 4.17-3.78(3 \mathrm{H}), 3.73$ (s, 3H), 2.68 (bs, 0.5 H ), 2.57 (bs, 0.5 H ), 1.69-1.47 (8H), 1.47-1.30 (2H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 166.6$ (0), 166.5 (0), 145.4 (1), 145.2 (1), 122.3 (1), 121.6 (1), 110.6 ( 0 ), 110.3 ( 0 ), 77.5 (1), 77.0 (1), 71.8 (1), 70.5 (1), 65.3 (2), 64.3 (2), 51.6 (3), 36.2 (2), 36.1 (2), 34.6 (2), 34.5 (2), 25.0 (2), 25.0 (2), 23.9 (2), 23.9 (2), 23.6 (2); IR (neat): $\nu 3457$ (m), 2936 (w), 1722 (s), 1274 (s), 1165 (s), 1097 (s), 927 (s); MS (EI): m/z 256 (22, [M] ${ }^{+}$), 227 (23), 213 (83), 141 (100), 109 (35), 81 (31), 55 (42); HRMS (EI): calcd for $\mathrm{C}_{13} \mathrm{H}_{20} \mathrm{O}_{5}[\mathrm{M}]^{+}$: 256.1311, found: 256.1313;

Anal. calcd for $\mathrm{C}_{13} \mathrm{H}_{20} \mathrm{O}_{5}$ : C, 60.9; H, 7.9, found: C, 60.7; H, 7.9.
( $E$ )-Methyl-4-hydroxynon-2-enoate (5m). Following the general procedure, 5 m was obtained from $\mathbf{4 m}(128 \mathrm{mg}$, 1.0 mmol ) ( $172 \mathrm{mg}, 93 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 6.93$ $(\mathrm{dm}, J=15.7,1 \mathrm{H}), 6.00(\mathrm{~d}, J=15.7,1 \mathrm{H}), 4.27(\mathrm{bs}, 1 \mathrm{H}), 3.71(\mathrm{~s}$, 3H), 2.22 (bs, 1H), 1.63-1.47 (2H), 1.47-1.18 (6H), 0.92-0.78 (3H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 167.1$ (0), 150.7 (1), 119.5 (1), 71.0 (1), 51.6 (3), 36.5 (2), 31.6 (2), 22.5 (2), 13.9 (3); IR (neat): $\nu$ 3437 (m), 2930 (m), 1723 (s), 1437 (m), 1274 (s), 1169 (s); MS (EI): $m / z 157$ (39), 115 (46), 87 (100), 55 (32), 43 (20); HRMS (ESI): calcd for $\mathrm{C}_{10} \mathrm{H}_{19} \mathrm{O}_{3}[\mathrm{M}+\mathrm{H}]^{+}$: 187.1334, found: 187.1327; Anal. calcd for $\mathrm{C}_{10} \mathrm{H}_{18} \mathrm{O}_{3}$ : C, 64.5; H, 9.7, found: C, $64.4 ; \mathrm{H}$, 10.1.
( $E$ )-Methyl-4-hydroxyhept-2-enoate (5n). Following the general procedure, 5 n was obtained from $4 n(100 \mathrm{mg}$, 1.0 mmol ) as a colourless liquid ( $138 \mathrm{mg}, 87 \%$ ). ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 6.93$ (dd, $\left.J=15.7,4.9,1 \mathrm{H}\right), 6.01(\mathrm{dd}, J=$ 15.7, 1.7, 1H), 4.29 (ddm, $J=6.2,1.6,1 \mathrm{H}), 3.72$ (s, 3H), 2.09 (bs, 1H), 1.60-1.48 (2H), 1.48-1.30 (2H), $0.92(\mathrm{t}, \mathrm{J}=7.3,3 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 150.6$ (0), 119.6 (1), 70.8 (1), 51.5 (3), 38.7 (2), 18.4 (2), 13.8 (3); IR (neat): $\nu 3441$ (w), 2958 (m), 1724 (s), 1705 (s), 1659 (m), 1436 (m), 1274 (s), 1169 (s), 981 (s); MS (EI): m/z 129 (44), 115 (50), 87 (100), 83 (37), 71 (25), 55 (35); HRMS (ESI): calcd for $\mathrm{C}_{8} \mathrm{H}_{15} \mathrm{O}_{3}[\mathrm{M}+\mathrm{H}]^{+}: 159.1021$, found: 159.1015 Anal. calcd for $\mathrm{C}_{8} \mathrm{H}_{14} \mathrm{O}_{3}$ : C, 60.7; H, 8.9, found: C, 60.8; H, 8.9.
(2E, $\quad 2^{\prime} E$ )-Dimethyl-4,4'-(1,4-phenylene)bis(4-hydroxybut-2enoate) (50). Following the general procedure, 50 was obtained from $4 \mathbf{0}(190 \mathrm{mg}, 1.0 \mathrm{mmol})$ as a colourless solid ( $254 \mathrm{mg}, 83 \%$ ), mp $98-99{ }^{\circ} \mathrm{C}$, as a mixture of diastereomers. The general procedure was modified by using increased amounts of catalyst A ( $85 \mathrm{mg}, 10 \mathrm{~mol} \%$ ) and methyl acrylate $(1.80 \mathrm{~mL}, 20.0 \mathrm{mmol})$. Analytical data were obtained from the mixture. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.30-7.22(4 \mathrm{H}), 6.94$ (ddm, $J=15.6,4.8,2 \mathrm{H}$ ), 6.14-6.01 (ddm, $J=15.6,1.6,2 \mathrm{H}), 5.28$ (bs, 2H), 3.65 (s, 6H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 166.9$ (0), 149.0 (1), 141.1 (0), 126.8 (1), 119.5 (1), 72.7 (1), 51.4 (3); IR (neat): $\nu 3424$ (m), 2953 (w), 1702 (s), 1437 (m), 1274 (s), 1168 (s), 983 (m); MS (EI): m/z 277 (91), 245 (100), 219 (49), 159 (30), 131 (52), 115 (58), 87 (77), 55 (36); HRMS (ESI): calcd for $\mathrm{C}_{16} \mathrm{H}_{19} \mathrm{O}_{6}[\mathrm{M}+\mathrm{H}]^{+}$: 307.1182, found: 307.1178; Anal. calcd for $\mathrm{C}_{16} \mathrm{H}_{18} \mathrm{O}_{6}$ : C, 62.7; H, 5.9; found: C, 62.7; H, 5.8.
(2E, 9E)-Dimethyl-4,8-dihydroxyundeca-2,9-dienedioate (5p). Following the general procedure, $\mathbf{5 p}$ was obtained from $\mathbf{4 p}$ ( $156 \mathrm{mg}, 1.0 \mathrm{mmol}$ ) as a colourless solid ( $176 \mathrm{mg}, 82 \%$ ), mp $82-84{ }^{\circ} \mathrm{C}$, as a mixture of diastereomers. The general procedure was modified by using increased amounts of catalyst A ( $85 \mathrm{mg}, 10 \mathrm{~mol} \%$ ) and methyl acrylate ( $1.80 \mathrm{~mL}, 20.0 \mathrm{mmol}$ ). Analytical data were obtained from the mixture. ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 6.92(\mathrm{dm}, J=15.6,2 \mathrm{H}), 6.02(\mathrm{~d}, J=15.6$, $2 \mathrm{H}), 4.30(2 \mathrm{H}), 3.72(\mathrm{~s}, 6 \mathrm{H}), 2.81(\mathrm{bs}, 2 \mathrm{H}), 1.75-1.40(6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 167.1$ (0), 150.4 (1), 119.7 (1), 70.7 (1), 70.6 (1), 51.7 (3), 36.0 (2), 36.0 (2), 20.9 (2), 20.9 (2); IR (neat): $\nu$ 3427 (m), 2950 (w), 1706 (s), 1276 (s), 1171 (s), 982 (s); MS (EI): m/z 255 (16), 223 (54), 191 (54), 162 (43), 140 (100), 111 (76), 83
(94), 55 (74); HRMS (ESI): calcd for $\mathrm{C}_{13} \mathrm{H}_{20} \mathrm{O}_{6} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$: 295.1158, found: 295.1164; Anal. calcd for $\mathrm{C}_{13} \mathrm{H}_{20} \mathrm{O}_{6}$ : C, 57.3; H, 7.4, found: C, 57.2; H, 7.4.

## General procedure for tandem cross metathesis-isomerization

To a solution of the corresponding allyl alcohol ( 1.0 mmol ) and methyl acrylate ( $0.9 \mathrm{~mL}, 10.0 \mathrm{mmol}$ ) in dry toluene $(1.0 \mathrm{~mL})$ was added Ru-catalyst A $(42.5 \mathrm{mg}, 5 \mathrm{~mol} \%)$. The solution was heated to $110{ }^{\circ} \mathrm{C}$ for 0.5 h , cooled to ambient temperature, and excess methyl acrylate was removed in vacuo. The residue was re-dissolved in toluene ( 5.0 mL ), and either NaOH ( $60 \mathrm{mg}, 1.5 \mathrm{mmol}$ ) was added and the mixture was heated to reflux (oil bath temperature $130{ }^{\circ} \mathrm{C}$ ) for $3 \mathrm{~h}(\operatorname{method} \mathrm{~A})$, or PMHS ( $16 \mu \mathrm{~L}, 0.2 \mathrm{mmol}$ ) was added and the mixture was heated to reflux (oil bath temperature $=130{ }^{\circ} \mathrm{C}$ ) for 12 h (method B).

Work-up procedure for method A: The reaction mixture was hydrolyzed, and the aqueous layer was extracted twice with MTBE. The combined organic layers were washed with brine, dried with $\mathrm{MgSO}_{4}$, filtered and all volatiles were removed in vacuo. The residue was purified by column chromatography on silica, using hexane-MTBE mixtures of increasing polarity.

Work-up procedure for method B: All volatiles were removed in vacuo and the residue was purified by column chromatography, on silica, using hexane-MTBE mixtures of increasing polarity.
(S)-Methyl-5-(tert-butyldimethylsilyloxy)-4-oxohexanoate (6a). Following the general procedure, $\mathbf{6 a}$ was obtained from $\mathbf{4 a}$ ( $200 \mathrm{mg}, 0.92 \mathrm{mmol}$ ) via method A ( $178 \mathrm{mg}, 70 \%$ ) or via method $\mathrm{B}(193 \mathrm{mg}, 76 \%)$ as a colourless liquid. $[\alpha]_{\mathrm{D}}^{26}=-8.2(c$ $=0.63, \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ); ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 4.18$ ( $\mathrm{q}, J=6.8$, 1H), 3.66 (s, 3H), 2.97 (ddd, $J=19.1,6.8,6.6,1 \mathrm{H}$ ), 2.83 (ddd, $J$ $=19.1,6.8,5.9,1 \mathrm{H}), 2.57$ (ddd, $J=6.8,5.9,2.5,1 \mathrm{H}), 2.56$ (ddd, $J=6.7,6.6,3.1,1 \mathrm{H}), 1.29(\mathrm{~d}, J=6.8,3 \mathrm{H}), 0.91(\mathrm{~s}, 9 \mathrm{H}), 0.08(\mathrm{~s}$, 3H), $0.08(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 212.8$ (0), 173.6 (0), 75.2 (1), 52.1 (3), 32.4 (2), 27.7 (2), 26.1 (3), 21.2 (3), 18.4 (0), -4.3 (3), -4.7 (3); IR (neat): $\nu 2930$ (m), 2857 (m), 1741 (s), 1721 (s), 1116 (s), 832 (s), 776 (s); MS (EI) m/z 275 (19), 259 (21), 243 (100), 213 (9), 185 (11), 143 (82), 111 (29); HRMS (ESI) calcd for $\mathrm{C}_{13} \mathrm{H}_{27} \mathrm{O}_{4} \mathrm{Si}\left(\mathrm{M}+\mathrm{H}^{+}\right)$: 275.1679, found: 275.1653; Anal. calcd for $\mathrm{C}_{13} \mathrm{H}_{26} \mathrm{O}_{4} \mathrm{Si}$ : C, 56.9; H, 9.6; found: C, 57.1; H: 9.6.

Methyl 4-oxo-4-phenylbutanoate (6b). Following the general procedure, $\mathbf{6 b}$ was obtained from $\mathbf{4 b}(134 \mathrm{mg}, 1.0 \mathrm{mmol})$ via method A (160 mg, 83\%) or via method B ( $136 \mathrm{mg}, 71 \%$ ) as a colourless liquid. ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.98$ (ddd, $J=$ $7.1,1.5,1.0,2 \mathrm{H}), 7.56$ (tt, $J=7.3,1.3,1 \mathrm{H}), 7.45$ (ddd, $J=7.6$, $7.3,1.3,2 \mathrm{H}), 3.70(\mathrm{~s}, 3 \mathrm{H}), 3.32(\mathrm{t}, J=6.6,2 \mathrm{H}), 2.76(\mathrm{t}, J=6.6$, 2H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 197.9$ (0), 173.2 (0), 136.6 (0), 133.1 (1), 128.6 (1), 128.0 (1), 51.7 (3), 33.4 (2), 28.0 (2); IR (neat): $\nu 2952$ (w), 1734 (s), 1684 (s), 1218 (s), 1162 (s), 751 (s), 690 (s); MS (EI): m/z 192 (10, [M + H] ${ }^{+}$), 185 (30), 159 (52), 105 (100), 77 (40), 73 (55), 45 (80); HRMS (EI): calcd for $\mathrm{C}_{11} \mathrm{H}_{12} \mathrm{O}_{3}$ $[M]^{+}: 192.0786$, found: 192.0783; Anal. calcd for $\mathrm{C}_{11} \mathrm{H}_{12} \mathrm{O}_{3}$ : C, 68.7; H, 6.3, found: C, $68.4 ;$ H, 6.8 .

Methyl-4-(4-bromophenyl)-4-oxobutanoate (6c). Following the general procedure, $\mathbf{6 c}$ was obtained from $\mathbf{4 c}(213 \mathrm{mg}$, 1.0 mmol ) via method A ( $170 \mathrm{mg}, 63 \%$ ) or via method B (182 mg, 67\%) as a colourless solid, mp 45-47 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.84(\mathrm{ddd}, J=8.7,2 \mathrm{H}), 7.61$ (ddd, $J=8.7$, $2 \mathrm{H}), 3.70(\mathrm{~s}, 3 \mathrm{H}), 3.27(\mathrm{t}, J=6.5,2 \mathrm{H}), 2.76(\mathrm{t}, J=6.6,2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 197.0$ (0), 173.2 (0), 135.3 (0), 131.9 (1), 129.5 (1), 128.4 (0), 51.8 (3), 33.3 (2), 27.9 (2); IR (neat): $\nu$ 2952 (w), 1734 (s), 1686 (s), 1585 (s), 1218 (s), 1167 (s), 1070 (s), 816 (s); MS (EI): m/z 272 (13), 270 (15, [M] ${ }^{+}$), 241 (12), 239 (15), 185 (90), 183 (100), 157 (20), 155 (22), 76 (17), 75 (15); HRMS (EI): calcd for $\mathrm{C}_{11} \mathrm{H}_{11} \mathrm{O}_{3} \mathrm{Br}[\mathrm{M}]^{+}: 269.9892$, found: 269.9884; Anal. calcd for $\mathrm{C}_{11} \mathrm{H}_{11} \mathrm{O}_{3} \mathrm{Br}$ : C, 48.7; H , 4.1, found: C, 48.5 ; H , 4.1.

Methyl-4-(4-methoxyphenyl)-4-oxobutanoate (6d). Following the general procedure, $\mathbf{6 d}$ was obtained from $\mathbf{4 d}(164 \mathrm{mg}$, 1.0 mmol ) via method A ( $99 \mathrm{mg}, 45 \%$ ) or via method B ( $160 \mathrm{mg}, 72 \%$ ) as a colourless solid, $\mathrm{mp} 43-45{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.92(\mathrm{ddd}, J=8.9,2 \mathrm{H}), 6.90(\mathrm{ddd}, J=8.9$, $2 \mathrm{H}), 3.82(\mathrm{~s}, 3 \mathrm{H}), 3.66(\mathrm{~s}, 3 \mathrm{H}), 3.23(\mathrm{t}, J=6.7,2 \mathrm{H}), 2.71(\mathrm{t}, J=$ $6.7,2 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 196.4$ (0), 173.3 (0), 163.5 (0), 131.9 (1), 130.1 (1), 129.6 (0), 113.6 (1), 55.3 (3), 51.6 (3), 32.8 (2), 28.0 (2); IR (neat): $\nu 2984$ (m), 1738 (s), 1678 (s), 1602 (s), 1257 (s), 1167 (s), 1030 (s), 833 (m); MS (EI): m/z 223 (5, $\left.[\mathrm{M}+\mathrm{H}]^{+}\right), 222\left(17,[\mathrm{M}]^{+}\right), 191(15), 177$ (7), 136 (10), 135 (100), 107 (8), 92 (8), 77 (10); HRMS (EI): calcd for $\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{O}_{4}$ $[M]^{+}: 222.0892$, found: 222.0894; Anal. calcd for $\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{O}_{4}$ : C, 64.9 ; H, 6.4, found: C, $65.0 ; \mathrm{H}, 6.5$.

Methyl-4-(3-methoxy-4-(methoxymethoxy)-phenyl)-4-oxobutanoate (6e). Following the general procedure, $\mathbf{6 e}$ was obtained from $4 \mathrm{e}(224 \mathrm{mg}, 1.0 \mathrm{mmol})$ via method A ( $183 \mathrm{mg}, 65 \%$ ) or via method $\mathrm{B}(193 \mathrm{mg}, 68 \%)$ as a colourless solid, mp $74-75{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.56(\mathrm{dd}, J=8.4,2.0$, $1 \mathrm{H}), 7.53$ (d, $J=2.0,1 \mathrm{H}), 7.16$ (d, $J=8.4,1 \mathrm{H}), 5.28(\mathrm{~s}, 2 \mathrm{H})$, $3.91(\mathrm{~s}, 3 \mathrm{H}), 3.68(\mathrm{~s}, 3 \mathrm{H}), 3.49(\mathrm{~s}, 3 \mathrm{H}), 3.27(\mathrm{t}, J=6.6,2 \mathrm{H})$, 2.73 ( $\mathrm{t}, J=6.6,2 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 196.7, $173.4,150.8,149.4,130.8,122.3,114.3,110.5,94.9,56.4$, 55.9, 51.8, 32.9, 28.1; IR (neat): $\nu 2953$ (w), 1735 (m), 1676 (m), 1511 (m), 1417 (m), 1262 (s), 1155 (s), 1138 (s), 1080 (s), 980 (s); MS (EI): m/z 283 (32, [M + H] ${ }^{+}$), 282 (57, [M] ${ }^{+}$), 251 (45), 181 (38), 165 (30), 151 (10), 137 (18), 115 (23), 45 (100); HRMS (EI): calcd for $\mathrm{C}_{14} \mathrm{H}_{18} \mathrm{O}_{6} \quad[\mathrm{M}]^{+}$: 282.1103, found: 282.1103; Anal. calcd for $\mathrm{C}_{14} \mathrm{H}_{18} \mathrm{O}_{6}$ : C, $59.6 ; \mathrm{H}, 6.4$, found: C, 59.5; H, 6.7.

Methyl-4-(4-(tert-butyldimethylsilyloxy)phenyl)-4-oxobutanoate (6f). Following the general procedure, $6 \mathbf{f}$ was obtained from 4 ( $264 \mathrm{mg}, 1.0 \mathrm{mmol}$ ) via method A ( 84 mg , $27 \%$ ) or via method B ( $225 \mathrm{mg}, 70 \%$ ) as a colourless liquid. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.90(\mathrm{ddd}, J=8.7,2 \mathrm{H}), 6.87(\mathrm{ddd}, J=$ $8.7,2 \mathrm{H}), 3.69(\mathrm{~s}, 3 \mathrm{H}), 3.26(\mathrm{t}, J=6.7,2 \mathrm{H}), 2.74(\mathrm{t}, J=6.7,2 \mathrm{H})$, $0.97(\mathrm{~s}, 9 \mathrm{H}), 0.22(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 196.6$ (0), 173.4 (0), 160.4 (0), 130.2 (1), 119.9 (1), 51.7 (3), 33.0 (2), 28.1 (2), 25.5 (3), 18.2 (0), -4.4 (3); IR (neat): $\nu 2932$ (w), 1739 (s), 1680 ( s), 1597 ( s), 1255 (s), 1161 (s), 906 (s), 832 (s), 781 (s); MS (EI): m/z 322 (38, [M] $]^{+}$), 291 (31), 265 (100), 235 (50), 209 (49), 179 (30), 135 (25), 89 (50), 55 (20); HRMS (EI): calcd for
$\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{O}_{4} \mathrm{Si}[\mathrm{M}]^{+}: 322.1600$, found: 322.1600; Anal. calcd for $\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{O}_{4} \mathrm{Si}: \mathrm{C}, 63.3$; H, 8.1, found: C, 63.3; H, 8.0.

Methyl-4-(4-benzyloxyphenyl)-4-oxobutanoate (6g). Following the general procedure, $\mathbf{6 g}$ was obtained from $\mathbf{4 g}(240 \mathrm{mg}$, 1.0 mmol ) via method B ( $212 \mathrm{mg}, 71 \%$ ) as a colourless solid, $\mathrm{mp} 70-71{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.96(\mathrm{dm}, J=8.9$, $2 \mathrm{H}), 7.46-7.30(5 \mathrm{H}), 7.01(\mathrm{dm}, J=8.9,2 \mathrm{H}), 5.13(\mathrm{~s}, 2 \mathrm{H}), 3.70(\mathrm{~s}$, $3 \mathrm{H}), 3.27$ (t, $J=6.7,2 \mathrm{H}), 2.75(\mathrm{t}, J=6.7,2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 196.4(0), 173.4(0), 162.7$ (0), 136.2 (0), 130.3 (1), 129.9 (0), 128.6 (1), 128.2 (1), 127.4 (1), 114.6 (1), 70.1 (2), 51.7 (3), 33.0 (2), 28.1 (2); IR (neat): $~ 3021$ (w), 1734 (m), 1676 (m), 1599 (m), 1218 (s), 1165 (s), 746 (s); MS (EI): m/z 299 (13, [M + H] ${ }^{+}$), 298 (18, [M] ${ }^{+}$), 267 (6), 211 (5), 115 (15), 91 (100); HRMS (EI): calcd for $\mathrm{C}_{18} \mathrm{H}_{18} \mathrm{O}_{4}[\mathrm{M}]^{+}: 298.1205$, found: 298.1205; Anal. calcd for $\mathrm{C}_{18} \mathrm{H}_{18} \mathrm{O}_{4}$ : C, $72.5 ; \mathrm{H}, 6.1$, found: C, 72.0; H, 6.2.

Methyl-4-(2-fluorophenyl)-4-oxobutanoate (6h). Following the general procedure, $\mathbf{6 h}$ was obtained from $\mathbf{4 h}(152 \mathrm{mg}$, 1.0 mmol ) via method A ( $143 \mathrm{mg}, 68 \%$ ) or via method B ( $159 \mathrm{mg}, 76 \%$ ) as a colourless liquid. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 7.87(\mathrm{td}, J=7.7,1.9,1 \mathrm{H}), 7.50(\mathrm{~m}, 1 \mathrm{H}), 7.21(\mathrm{ddd}, J=$ $8.4,7.9,1.1,1 \mathrm{H}), 7.12$ (ddd, $J=11.3,8.3,0.9,1 \mathrm{H}), 3.69(\mathrm{~s}, 3 \mathrm{H})$, $3.30(\mathrm{td}, J=6.6,3.3,2 \mathrm{H}), 2.73(\mathrm{t}, J=6.5,2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 196.2(\mathrm{~d}, J=4.1,0), 173.2(0), 162.1(\mathrm{~d}, J=$ $254.8,0), 134.7$ (d, $J=8.8,1$ ), 130.6 (d, $J=2.5,1), 125.1(\mathrm{~d}, J=$ $3.2,0), 124.4(\mathrm{~d}, J=3.4,1), 116.6(\mathrm{~d}, J=23.8,1), 51.7$ (3), 38.2 (d, $J=8.4,2$ ), 28.0 (d, $J=2.3,2$ ); IR (neat): $\nu 2953(\mathrm{w}), 1734(\mathrm{~s})$, 1686 (s), 1609 (s), 1452 (s), 1212 (s), 1166 (s), 1152 (s), 762 (s); MS (EI): $m / z 210$ ( $8,[\mathrm{M}]^{+}$), 179 (8), 135 (30), 123 (100), 95 (15), 75 (11); HRMS (EI): calcd for $\mathrm{C}_{11} \mathrm{H}_{11} \mathrm{O}_{3} \mathrm{~F}[\mathrm{M}]^{+}: 210.0692$, found: 210.0700.

Methyl-4-oxo-5-phenylpentanoate (6i). Following the general procedure, $6 \mathbf{i}$ was obtained from $4 \mathbf{i}(148 \mathrm{mg}, 1.0 \mathrm{mmol})$ via method A (136 mg, 66\%) or via method B (132 mg, 64\%) as a colourless liquid. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.37-7.17$ (5H), $3.74(\mathrm{~s}, 2 \mathrm{H}), 3.65(\mathrm{~s}, 3 \mathrm{H}), 2.76(\mathrm{t}, J=6.6,2 \mathrm{H}), 2.56(\mathrm{t}, J=6.6$, 2H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 206.2$ (0), 173.0 (0), 134.0 (0), 129.4 (1), 128.7 (1), 127.0 (1), 51.6 (3), 50.0 (2), 36.4 (2), 27.8 (2); IR (neat): $\nu 2952$ (w), 1716 (s), 1357 (m), 1200 (s), 1174 (s), 700 (s); MS (EI): m/z 206 (15, [M] ${ }^{+}$), 175 (20), 115 (100), 91 (63), 55 (26); HRMS (EI): calcd for $\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{O}_{3}[\mathrm{M}]^{+}: 206.0943$, found: 206.0930; Anal. calcd for $\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{O}_{3}$ : C, 69.9; H, 6.8, found: C, 69.4; H, 7.0.

Methyl-4-oxo-6-phenylhexanoate (6j). Following the general procedure, $\mathbf{6 j}$ was obtained from $\mathbf{4 j}(162 \mathrm{mg}, 1.0 \mathrm{mmol})$ via method A ( $175 \mathrm{mg}, 80 \%$ ) or via method B ( $160 \mathrm{mg}, 73 \%$ ) as a colourless liquid. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.35-7.15(5 \mathrm{H})$, $3.67(\mathrm{~s}, 3 \mathrm{H}), 2.92(\mathrm{td}, J=6.9,2.4,2 \mathrm{H}), 2.79(\mathrm{td}, J=7.1,2.3,2 \mathrm{H})$, $2.71(\mathrm{t}, J=6.6,2 \mathrm{H}), 2.59(\mathrm{t}, J=6.6,2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 75 MHz , $\mathrm{CDCl}_{3}$ ) $\delta 207.8$ (0), 173.1 (0), 140.9 (0), 128.4 (1), 128.2 (1), 126.1 (1), 51.7 (3), 44.2 (2), 37.2 (2), 29.6 (2), 27.6 (2); IR (neat): $\nu 2951$ (w), 1735 (s), 1713 (s), 1362 (m), 1204 (s), 1174 (s), 699 (s); MS (EI): m/z 221 (18, [M + H] $]^{+}$), 220 ( $\left.55,[\mathrm{M}]^{+}\right), 188$ ( 60 ), 105 (75), 91 (100), 55 (18); HRMS (EI): calcd for $\mathrm{C}_{13} \mathrm{H}_{16} \mathrm{O}_{3}[\mathrm{M}]^{+}$: 220.1099, found: 220.1100; Anal. calcd for $\mathrm{C}_{13} \mathrm{H}_{16} \mathrm{O}_{3}$ : C, 70.9; H, 7.3, found: C, 70.6; H, 7.6.

Methyl-6-methyl-4-oxoheptanoate (6k). Following the general procedure, $\mathbf{6 k}$ was obtained from $\mathbf{4 k}$ (114 mg, 1.0 mmol ) via method A ( $76 \mathrm{mg}, 44 \%$ ) or via method B ( $105 \mathrm{mg}, 61 \%$ ) as a colourless liquid. ${ }^{1} \mathrm{H}$ NMR $(300 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta 3.62(\mathrm{~s}, 3 \mathrm{H}), 2.66(\mathrm{t}, J=6.5,2 \mathrm{H}), 2.53(\mathrm{t}, J=6.5,2 \mathrm{H})$, $2.28(\mathrm{~d}, J=6.8,2 \mathrm{H}), 2.12(\mathrm{tq}, J=6.6,1.2,1 \mathrm{H}), 0.88(\mathrm{~d}, J=1.2$, $3 \mathrm{H}), 0.86(\mathrm{~d}, J=1.2,3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 208.5$ (0), 173.1 (0), 51.6 (3), 51.6 (2), 37.5 (2), 27.6 (2), 24.6 (1), 22.4 (3); IR (neat): $\nu 2956$ (m), 1738 (s), 1712 (s), 1363 (s), 1206 (s), 1167 (s); MS (EI): m/z 134 (24), 115 (26), 98 (38), 85 (42), 71 (43), 57 (100), 43 (63); HRMS (ESI): calcd for $\mathrm{C}_{9} \mathrm{H}_{17} \mathrm{O}_{3}[\mathrm{M}+\mathrm{H}]^{+}$: 173.1178, found: 173.1194; Anal. calcd for $\mathrm{C}_{9} \mathrm{H}_{16} \mathrm{O}_{3}: \mathrm{C}, 62.8 ; \mathrm{H}$, 9.4, found: C, 62.6; H, 9.6.
( $R$ )-Methyl-4-oxo-4-(1,4-dioxaspiro[4.5]decan-2-yl)-butanoate (61). Following the general procedure, 61 was obtained from 41 ( $198 \mathrm{mg}, 1.0 \mathrm{mmol}$ ) via method A ( $105 \mathrm{mg}, 41 \%$ ) or via method $\mathrm{B}(171 \mathrm{mg}, 67 \%)$ as a colourless liquid. $[\alpha]_{24}^{\mathrm{D}}=+13.9(c$ $0.37, \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ); ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 4.45(\mathrm{dd}, J=7.7$, $5.7,1 \mathrm{H}), 4.18(\mathrm{dd}, J=8.7,7.7,1 \mathrm{H}), 4.01(\mathrm{dd}, J=8.7,5.7,1 \mathrm{H})$, $3.66(\mathrm{~s}, 3 \mathrm{H}), 3.04-2.81(2 \mathrm{H}), 2.64-2.55(2 \mathrm{H}), 1.76-1.50(8 \mathrm{H})$, 1.50-1.34 (2H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 209.3$ (0), 173.0 (0), 111.7 (0), 79.9 (1), 66.1 (2), 51.7 (3), 35.6 (2), 34.5 (2), 33.4 (2), 27.1 (2), 25.0 (2), 23.9 (2), 23.7 (2); IR (neat): $\nu 2931$ (w), $1253(\mathrm{~m}), 1090(\mathrm{~m}), 832(\mathrm{~s}), 775(\mathrm{~s})$; MS (EI): m/z 256 (15, [M] ${ }^{+}$), 213 (55), 141 (100), 115 (15), 83 (17), 55 (32); HRMS (EI): calcd for $\mathrm{C}_{13} \mathrm{H}_{20} \mathrm{O}_{5}[\mathrm{M}]^{+}$: 256.1311, found: 256.1309; Anal. calcd for $\mathrm{C}_{13} \mathrm{H}_{20} \mathrm{O}_{5}$ : C, 60.9; H, 7.9, found: C, 60.9; H, 8.0.
Methyl-4-oxononanoate ( 6 m ). Following the general procedure, $\mathbf{6 m}$ was obtained from $\mathbf{4 m}(128 \mathrm{mg}, 1.0 \mathrm{mmol})$ via method A ( $79 \mathrm{mg}, 42 \%$ ) or via method B ( $122 \mathrm{mg}, 65 \%$ ) as a colourless liquid. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 3.63(\mathrm{~s}, 3 \mathrm{H})$, $2.68(\mathrm{t}, J=6.3,2 \mathrm{H}), 2.54(\mathrm{t}, J=6.3,2 \mathrm{H}), 2.40(\mathrm{t}, J=7.5,2 \mathrm{H})$, 1.62-1.47 (2H), 1.35-1.15 (4H), $0.85(\mathrm{t}, J=6.9,3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 208.9$ (0), 173.2 (0), 51.6 (3), 42.6 (2), 31.3 (2), 27.6 (2), 23.4 (2), 22.3 (2), 13.8 (3); IR (neat): $\nu 2930$ (m), 1738 (s), 1715 (s), 1361 (m), 1198 (m), 1166 (m); MS (EI): m/z 155 (23), 130 (39), 115 (42), 98 (95), 71 (70), 57 (89), 43 (100); HRMS (ESI): calcd for $\mathrm{C}_{10} \mathrm{H}_{19} \mathrm{O}_{3}[\mathrm{M}+\mathrm{H}]^{+}$: 187.1334, found: 187.1334; Anal. calcd for $\mathrm{C}_{10} \mathrm{H}_{18} \mathrm{O}_{3}: \mathrm{C}, 64.5 ; \mathrm{H}, 9.7$, found: C, 64.2; H, 9.8.

Methyl-4-oxoheptanoate (6n). Following the general procedure, $6 \mathbf{n}$ was obtained from $\mathbf{4 n}(100 \mathrm{mg}, 1.0 \mathrm{mmol})$ via method A (89 mg, 56\%) or via method B ( $110 \mathrm{mg}, 70 \%$ ) as a colourless liquid. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 3.63(\mathrm{~s}, 3 \mathrm{H})$, $2.68(\mathrm{t}, J=6.3,2 \mathrm{H}), 2.54(\mathrm{t}, J=6.3,2 \mathrm{H}), 2.40(\mathrm{t}, J=7.4,2 \mathrm{H})$, $1.58(\mathrm{tq}, J=7.4,7.4,2 \mathrm{H}), 0.88(\mathrm{t}, J=7.4,3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 208.8$ (0), 173.2 (0), 51.6 (3), 44.6 (2), 36.9 (2), 27.6 (2), 17.2 (2), 13.6 (3); IR (neat): 2960 (w), 1737 (s), 1713 (s), 1362 (m), 1205 (s), 1165 (s); MS (EI): m/z 148 (23), 134 (33), 115 (55), 98 (48), 71 (88), 57 (77), 43 (100); HRMS (ESI): calcd for $\mathrm{C}_{8} \mathrm{H}_{15} \mathrm{O}_{3}[\mathrm{M}+\mathrm{H}]^{+}$: 159.1021, found: 159.1029; Anal. calcd for $\mathrm{C}_{8} \mathrm{H}_{14} \mathrm{O}_{3}: \mathrm{C}, 60.7 ; \mathrm{H}, 8.9$, found: C, 60.5; H, 9.2.

Dimethyl-4,4'-(1,4-phenylen)bis(4-oxobutanoate) (60). Following the general procedure, 60 was obtained from 40 ( $190 \mathrm{mg}, 1.0 \mathrm{mmol}$ ) via method A ( $167 \mathrm{mg}, 55 \%$ ) or via method B (114 mg, 37\%) as a colourless solid, mp $116-117^{\circ} \mathrm{C}$.

The general procedure was modified by using increased amounts of catalyst A ( $85 \mathrm{mg}, 10 \mathrm{~mol} \%$ ) and methyl acrylate $(1.80 \mathrm{~mL}, 20.0 \mathrm{mmol}) .{ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.06$ ( $\mathrm{s}, 4 \mathrm{H}$ ), $3.71(\mathrm{~s}, 6 \mathrm{H}), 3.33(\mathrm{t}, J=6.5,4 \mathrm{H}), 2.78(\mathrm{t}, J=6.5,4 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 197.5$ (0), 173.1 (0), 139.8 (0), 128.3 (1), 51.9 (3), 33.8 (2), 27.9 (2); IR (neat): ע 2952 (w), 1722 (s), 1682 (s), 1213 (s), 1162 (s), 754 (s); MS (EI): m/z 306 (7, [M] ${ }^{+}$), 275 (20), 219 (100), 159 (17), 104 (15), 55 (6); HRMS (EI): calcd for $\mathrm{C}_{16} \mathrm{H}_{18} \mathrm{O}_{6}[\mathrm{M}]^{+}: 306.1103$, found: 306.1098; Anal. calcd for $\mathrm{C}_{16} \mathrm{H}_{18} \mathrm{O}_{6}$ : C, 62.7 ; $\mathrm{H}, 5.9$, found: C, 62.7; H, 5.9.

Dimethyl-4,8-dioxoundecandioate (6p). Following the general procedure, $\mathbf{6 p}$ was obtained from $\mathbf{4 p}$ (156 mg, 1.0 mmol ) via method B ( $116 \mathrm{mg}, 43 \%$ ) as a colourless liquid. The general procedure was modified by using increased amounts of catalyst $\mathbf{A}(85 \mathrm{mg}, 10 \mathrm{~mol} \%)$ and methyl acrylate $(1.80 \mathrm{~mL}, 20.0 \mathrm{mmol}) .{ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 3.65(\mathrm{~s}$, $6 \mathrm{H}), 2.68(\mathrm{t}, J=6.8,4 \mathrm{H}), 2.56(\mathrm{t}, J=6.6,4 \mathrm{H}), 2.48(\mathrm{t}, J=7.1$, 4 H ), 1.86 (quint, $J=7.0,2 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 208.3 (0), 173.1 (0), 51.7 (3), 41.4 (2), 37.0 (2), 27.7 (2), 17.1 (2); IR (neat): $\nu 2953$ (w), 1733 (s), 1712 (s), 1362 (m), 1198 (s), 1171 (s); MS (EI): m/z 273 (10, [M + H] ${ }^{+}$), 255 (41), 223 (41), 167 (42), 153 (72), 125 (54), 115 (100), 97 (46), 55 (67); HRMS (ESI): calcd for $\mathrm{C}_{13} \mathrm{H}_{20} \mathrm{O}_{6} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}: 295.1158$, found: 295.1153; Anal. calcd for $\mathrm{C}_{13} \mathrm{H}_{20} \mathrm{O}_{6}$ : C, 57.3; H, 7.4, found: C, $56.8 ; \mathrm{H}, 7.7$.

## Synthesis of protected l-amicetose

(R)-5-((S)-1-(tert-Butyldimethylsilyloxy)ethyl)-dihydro-furan$2(3 \mathbf{H})$-one (7). To a solution of $5 \mathrm{a}(300 \mathrm{mg}, 1.1 \mathrm{mmol})$ in dry and degassed methanol ( 20 mL ) was added Pd/C ( $30 \mathrm{mg}, 10 \mathrm{wt} \%$ ). The suspension was stirred for 12 h in an atmosphere of hydrogen (1 bar). The mixture was filtered through a pad of Celite $®$, and the pad was washed three times with MTBE. Evaporation of the solvents furnished $7(235 \mathrm{mg}, 88 \%)$ as a colourless liquid. $[\alpha]_{\mathrm{D}}^{26}=+13.3\left(c \quad 0.55, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 4.33$ (ddd, $\left.J=8.1,5.5,3.2,1 \mathrm{H}\right), 4.06(\mathrm{qd}, J$ $=6.4,3.2,1 \mathrm{H}), 2.55(\mathrm{ddd}, J=17.7,10.2,7.0,1 \mathrm{H}), 2.44(\mathrm{ddd}, J=$ 17.7, 10.0, 6.6, 1H), 2.27 (dddd, $J=12.6,10.1,6.7,5.6,1 \mathrm{H}$ ), $2.22(\mathrm{~m}, 1 \mathrm{H}), 1.13(\mathrm{~d}, J=6.5,3 \mathrm{H}), 0.87(\mathrm{~s}, 9 \mathrm{H}), 0.07(\mathrm{~s}, 3 \mathrm{H})$, $0.05(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 177.4$ (0), 83.5 (1), 68.8 (1), 28.5 (2), 25.8 (3), 23.6 (0), 20.9 (2), 19.5 (3), -4.8 (3), -4.9 (3); IR (neat): $\nu 2930$ (m), 2857 (m), 1776 (s), 1082 (m), 834 (s), 776 (s); MS (EI): $m / z 245$ (35, [M + H] $]^{+}$), 242 (69), 227 (37), 185 (17), 113 (100); HRMS (ESI): calcd for $\mathrm{C}_{12} \mathrm{H}_{25} \mathrm{O}_{3} \mathrm{Si}$ [M $+\mathrm{H}]^{+}: 245.1573$, found: 245.1593; Anal. calcd for $\mathrm{C}_{12} \mathrm{H}_{24} \mathrm{O}_{3} \mathrm{Si}$ : C, 59.0; H, 9.9, found: C, 58.9 ; H, 10.2.
(4R, 5S, E)-Methyl-5-(tert-butyldimethylsilyloxy)-4-(methoxy-methoxy)-hex-2-enoate (8a). To a solution of $5 \mathbf{5 a}(300 \mathrm{mg}$, $1.1 \mathrm{mmol})$ in dry dichloromethane ( 20 mL ) were added ${ }^{\mathrm{i}} \mathrm{Pr}_{2} \mathrm{EtN}$ $(0.8 \mathrm{~mL}, 4.4 \mathrm{mmol})$ and MOM-bromide $(0.3 \mathrm{~mL}, 3.2 \mathrm{mmol}$, $90 \%$ ). The solution was heated to reflux for 12 h , cooled to ambient temperature and hydrolyzed. The aqueous layer was separated and extracted twice with MTBE. The combined organic layers were dried with $\mathrm{MgSO}_{4}$, filtered and all volatiles were removed in vacuo. After purification by column chromatography on silica, using hexane-MTBE mixtures as the eluent,

8a was obtained as a colourless liquid ( $280 \mathrm{mg}, 81 \%$ ). $[\alpha]_{22}^{\mathrm{D}}$ $-38.0\left(c 0.50, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ;{ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 6.86(\mathrm{dd}, J$ $=15.8,6.3,1 \mathrm{H}), 6.01(\mathrm{dd}, J=15.9,1.3,1 \mathrm{H}), 4.63(\mathrm{~s}, 2 \mathrm{H}), 4.04$ (ddd, $J=6.3,5.0,1.3,1 \mathrm{H}), 3.83(\mathrm{qd}, J=6.2,5.1,1 \mathrm{H}), 3.74(\mathrm{~s}$, $3 \mathrm{H}), 3.37(\mathrm{~s}, 3 \mathrm{H}), 1.17(\mathrm{~d}, J=6.2,3 \mathrm{H}), 0.87(\mathrm{~s}, 9 \mathrm{H}), 0.05(\mathrm{~s}, 3 \mathrm{H})$, 0.03 ( $\mathrm{s}, 3 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 166.8$ (0), 146.3 (1), 123.3 (1), 95.5 (2), 80.3 (1), 71.1 (1), 56.1 (3), 51.8 (3), 26.1 (3), 18.4 (0), -4.3 (3), -4.5 (3); IR (neat): $\nu 2953$ (w), 2932 (w), 1728 (s), 1102 (s), 1032 (s), 832 (s), 775 (s); MS (EI): m/z 159 (34), 89 (29), 73 (74), 45 (100); HRMS (ESI): calcd for $\mathrm{C}_{15} \mathrm{H}_{31} \mathrm{O}_{5} \mathrm{Si}[\mathrm{M}+$ $\mathrm{H}]^{+}$: 319.1941, found: 319.1938; Anal. calcd for $\mathrm{C}_{15} \mathrm{H}_{30} \mathrm{O}_{5} \mathrm{Si}$ : C, 56.6; H, 9.5, found: C, 56.3; H, 9.6.
(4R, 5S)-Methyl-5-(tert-butyldimethylsilyloxy)-4-(methoxy-methoxy)-hex-2-anoate (9a)

Hydrogenation catalyzed by $P d / C$. To a solution of $8 \mathbf{8}$ ( $150 \mathrm{mg}, 0.47 \mathrm{mmol}$ ) in dry and degassed methanol ( 10 mL ) was added Pd/C ( 15 mg , $10 \mathrm{wt} \%$ ). The suspension was stirred for 12 h in an atmosphere of hydrogen ( 1 bar ). The mixture was filtered through a pad of Celite $®$, and the pad was washed three times with MTBE. Evaporation of the solvents gave $9 \mathbf{a}$ ( 151 mg , quantitative) as a colourless liquid.
Reduction with modified Stryker's reagent. $\mathrm{Cu}(\mathrm{OAc})_{2} \cdot \mathrm{H}_{2} \mathrm{O}$ ( $10 \mathrm{mg}, 5 \mathrm{~mol} \%$ ) and BDP ( $4.5 \mathrm{mg}, 10 \mathrm{~mol} \%$ ) were dissolved in dry and degassed toluene ( 2.0 mL ) and tert-butanol ( 1.5 mL ). The mixture was stirred for 10 min at ambient temperature, and PMHS ( $134 \mu \mathrm{~L}, 2.0 \mathrm{mmol}$ ) was added. Stirring was continued for 0.5 h , after which time the colour changed from blue to green. A solution of $8 \mathrm{a}(318 \mathrm{mg}, 1.0 \mathrm{mmol})$ in toluene $(1.0 \mathrm{~mL})$ was added and the solution was stirred at ambient temperature for 12 h . The reaction mixture was diluted with MTBE, and washed with a saturated aqueous $\mathrm{NaHCO}_{3}$ solution, followed by aqueous $\mathrm{HCl}(1 \mathrm{M})$. The aqueous layer was extracted twice with MTBE, and the combined organic layers were dried with $\mathrm{MgSO}_{4}$. Purification by column chromatography on silica, using hexane-MTBE as an eluent, furnished 9a ( $266 \mathrm{mg}, 96 \%$ ) as a colourless liquid. $[\alpha]_{22}^{\mathrm{D}}=+22.1$ (c 0.66, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ); ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 4.75(\mathrm{~d}, J=6.7,1 \mathrm{H}), 4.62$ $(\mathrm{d}, J=6.7,1 \mathrm{H}), 3.80(\mathrm{qd}, J=6.3,3.7,1 \mathrm{H}), 3.66(\mathrm{~s}, 3 \mathrm{H}), 3.44(\mathrm{dt}$, $J=3.9,3.9,1 \mathrm{H}), 3.38(\mathrm{~s}, 3 \mathrm{H}), 2.55-2.32(2 \mathrm{H}), 1.90-1.67(2 \mathrm{H})$, $1.12(\mathrm{~d}, J=6.3,3 \mathrm{H}), 0.87(\mathrm{~s}, 9 \mathrm{H}), 0.04(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 174.1$ (0), 96.5 (2), 80.9 (1), 70.4 (1), 56.1 (3), 55.8 (3), 51.8 (3), 30.3 (2), 25.9 (2), 25.8 (3), 19.1 (3), 18.0 (0), -4.5 (3), -4.9 (3); IR (neat): $\nu 2953$ (w), 2932 (w), 1740 (s), 1253 (s), 1100 (s), 1032 (s), 832 (s); MS (EI): m/z 289 (7), 233 (40), 201 (100), 187 (19), 159 (84), 113 (27), 89 (32), 73 (53), 45 (41); HRMS (ESI): calcd for $\mathrm{C}_{15} \mathrm{H}_{33} \mathrm{O}_{5} \mathrm{Si}[\mathrm{M}+\mathrm{H}]^{+}: 321.2097$, found: 321.2124; Anal. calcd for $\mathrm{C}_{15} \mathrm{H}_{32} \mathrm{O}_{5} \mathrm{Si}$ : C, $56.2 ; \mathrm{H}, 10.1$, found: C, 56.1; H, 9.9.
(4R, 5S, E)-Methyl-4-(benzyloxymethoxy)-5-(tert-butyldi-methylsilyloxy)-hex-2-enoate (8b). To a solution of $5 \mathbf{5}(1.30 \mathrm{~g}$, $4.70 \mathrm{mmol})$ in dry dichloromethane ( 20 mL ) were added ${ }^{\mathrm{i}} \mathrm{Pr}_{2} \mathrm{EtN}(1.3 \mathrm{~mL}, 7.05 \mathrm{mmol})$ and BOM-chloride $(0.8 \mathrm{~mL}$, 6.10 mmol ), and the solution was heated to reflux for 12 h . The reaction mixture was hydrolyzed and the aqueous layer was extracted twice with MTBE. The combined organic layers were dried with $\mathrm{MgSO}_{4}$, filtered and all volatiles were removed
in vacuo. After purification by column chromatography on silica, using hexane-MTBE as an eluent, $\mathbf{8 b}$ was obtained as a colourless liquid ( $1.72 \mathrm{~g}, 93 \%$ ). $[\alpha]_{24}^{\mathrm{D}}=-39.4\left(c 0.73, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$; ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.38-7.27(5 \mathrm{H}), 6.90(\mathrm{dd}, J=15.8$, $6.5,1 \mathrm{H}), 6.04(\mathrm{dd}, J=15.8,1.3,1 \mathrm{H}), 4.79(\mathrm{~d}, J=6.9,1 \mathrm{H}), 4.75$ $(\mathrm{d}, J=6.9,1 \mathrm{H}), 4.67(\mathrm{~d}, J=11.8,1 \mathrm{H}), 4.58(\mathrm{~d}, J=11.8,1 \mathrm{H}), 4.12$ (ddd, $J=6.3,5.0,1.2,1 \mathrm{H}), 3.85(\mathrm{qd}, J=6.2,5.1,1 \mathrm{H}), 3.75(\mathrm{~s}$, $3 \mathrm{H}), 1.19(\mathrm{~d}, J=6.2,3 \mathrm{H}), 0.88(\mathrm{~s}, 9 \mathrm{H}), 0.06(\mathrm{~s}, 3 \mathrm{H}), 0.04(\mathrm{~s}, 3 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 166.4$ (0), 145.9 (1), 137.6 (0), 128.4 (1), 127.8 (1), 127.7 (1), 123.0 (1), 93.0 (2), 80.0 (1), 70.6 (1), 69.7 (2), 51.5 (3), 25.7 (3), 19.9 (3), 18.0 (0), -4.6 (3), -4.8 (3); IR (neat): $\nu 2930$ (w), 1724 (m), 1034 (m), 753 (s); MS (EI): $m / z 307$ (8), 257 (19), 159 (48), 91 (100), 73 (31); HRMS (ESI): calcd for $\mathrm{C}_{21} \mathrm{H}_{34} \mathrm{O}_{5} \mathrm{SiNa}[\mathrm{M}+\mathrm{Na}]^{+}: 417.2073$, found: 417.2083; Anal. calcd for $\mathrm{C}_{21} \mathrm{H}_{34} \mathrm{O}_{5} \mathrm{Si}$ : C, 63.9; H, 8.7, found: C, 63.8; H, 8.9.
(4R, 5S)-Methyl-4-(benzyloxymethoxy)-5-(tert-butyldimethyl-silyloxy)-hex-2-anoate (9b). Following the procedure stated above for $\mathbf{9 a}$, the title compound $\mathbf{9 b}$ was obtained from $\mathbf{8 b}$ ( 395 mg , 1.0 mmol ) as a colourless liquid ( $375 \mathrm{mg}, 95 \%$ ). $[\alpha]_{24}^{\mathrm{D}}$ $=+36.9\left(c \quad 0.52, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $7.39-7.27(5 \mathrm{H}), 4.88(\mathrm{~d}, J=6.8,1 \mathrm{H}), 4.80(\mathrm{~d}, J=6.8,1 \mathrm{H}), 4.70$ $(\mathrm{d}, J=11.9,1 \mathrm{H}), 4.59(\mathrm{~d}, J=11.9,1 \mathrm{H}), 3.85(\mathrm{qd}, J=6.3,3.7$, $1 \mathrm{H}), 3.65(\mathrm{~s}, 3 \mathrm{H}), 3.54(\mathrm{dt}, J=3.9,3.9,1 \mathrm{H}), 2.59-2.34(2 \mathrm{H})$, $1.95-1.75(2 \mathrm{H}), 1.15(\mathrm{~d}, J=6.3,3 \mathrm{H}), 0.89(\mathrm{~s}, 9 \mathrm{H}), 0.06(\mathrm{~s}, 6 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 174.1$ (0), 137.9 (0), 128.4 (1), 127.7 (1), 127.6 (1), 94.4 (2), 80.9 (1), 70.2 (1), 69.8 (2), 51.4 (3), 30.2 (2), 25.8 (3), 25.7 (2), 19.1 (3), 18.0 (0), -4.5 (3), -4.9 (3); IR (neat): $\nu 2954$ (w), 2858 (w), 1740 ( s$), 1255$ ( s$), 1104$ (s), 1037 (s), 834 (s); MS (EI): m/z 309 (12), 201 (22), 159 (27), 91 (100), 73 (20); HRMS (ESI): calcd for $\mathrm{C}_{21} \mathrm{H}_{37} \mathrm{O}_{5} \mathrm{Si}[\mathrm{M}+\mathrm{H}]^{+}$: 397.2410, found: 397.2415; Anal. calcd for $\mathrm{C}_{21} \mathrm{H}_{36} \mathrm{O}_{5} \mathrm{Si}$ : $\mathrm{C}, 63.6 ; \mathrm{H}, 9.2$, found: C, 63.1; H, 9.3.
(5R, 6S)-5-(Benzyloxymethoxy)-6-methyl-tetrahydropyran-2one (10). To a solution of 9b ( $250 \mathrm{mg}, 0.63 \mathrm{mmol}$ ) in THF $(10 \mathrm{~mL})$ was added TBAF ( $248 \mathrm{mg}, 0.95 \mathrm{mmol}$ ). The reaction mixture was stirred at ambient temperature for 12 h . Then aqueous $\mathrm{NaOH}(5 \%, 10 \mathrm{~mL})$ was added and the solution was stirred for 2 h at $40^{\circ} \mathrm{C}$. After neutralization with aqueous HCl ( $1 \mathrm{M}, 10 \mathrm{~mL}$ ) the aqueous layer was extracted several times with ethyl acetate and the organic layer was dried with $\mathrm{MgSO}_{4}$. Purification by column chromatography on silica, using hexane-ethyl acetate as an eluent, furnished 10 ( $107 \mathrm{mg}, 68 \%$ ) as a colourless liquid. $[\alpha]_{24}^{\mathrm{D}}=-64.1\left(c 0.49, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.39-7.27(5 \mathrm{H}), 4.85(\mathrm{~d}, J=7.2,1 \mathrm{H}), 4.81$ (d, $J=7.2,1 \mathrm{H}), 4.64(\mathrm{~d}, J=11.9,1 \mathrm{H}), 4.60(\mathrm{~d}, J=11.9,1 \mathrm{H}), 4.45$ (qd, $J=6.3,6.2,1 \mathrm{H}), 3.72(\mathrm{ddm}, J=5.6,4.4,1 \mathrm{H}), 2.70(\mathrm{ddd}, J=$ $17.6,9.5,7.2,1 \mathrm{H}$ ), 2.48 (ddd, $J=17.6,6.4,5.3,1 \mathrm{H}$ ), 2.10 (dddd, $J=13.9,10.7,6.5,4.3,1 \mathrm{H}), 2.01-1.87(1 \mathrm{H}), 1.38(\mathrm{~d}, J=6.5,3 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 170.5$ (0), 137.3 (0), 128.4 (1), 127.8 (1), 127.7 (1), 93.2 (2), 78.3 (1), 72.9 (1), 70.0 (2), 26.4 (2), 23.2 (2), 19.3 (3); IR (neat): $\nu 2941$ (w), 1731 (s), 1030 (s), 741 (m); MS (EI): m/z 114 (18), 107 (15), 91 (100), 85 (65), 65 (14), 43 (20); HRMS (ESI): calcd for $\mathrm{C}_{14} \mathrm{H}_{18} \mathrm{O}_{4} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$: 273.1103, found: 273.1100; Anal. calcd for $\mathrm{C}_{14} \mathrm{H}_{18} \mathrm{O}_{4}$ : C, 67.2; H, 7.3, found: C, 67.1; H, 7.2.
(5R, 6S)-5-(Benzyloxymethoxy)-6-methyl-tetrahydro-2H-pyran-2-ol (11). To a solution of $\mathbf{1 0}(66 \mathrm{mg}, 0.26 \mathrm{mmol})$ in dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1.3 \mathrm{~mL})$ was added DIBAL-H ( $0.77 \mathrm{~mL}, 1.02 \mathrm{M}$ solution in cyclohexane, 0.78 mmol ) dropwise at $-78^{\circ} \mathrm{C}$. The solution was stirred for 0.25 h and then quenched at this temperature by the addition of $\mathrm{MeOH}(1.3 \mathrm{~mL})$. A saturated aqueous solution of $\mathrm{Na}^{+} / \mathrm{K}^{+}$-tartrate ( 2 mL ) was added at ambient temperature and the mixture was repeatedly extracted with MTBE. The combined organic layers were dried with $\mathrm{MgSO}_{4}$ and concentrated under reduced pressure to furnish $11(57 \mathrm{mg}, 86 \%$, a 1:1 mixture of anomeric lactols in $\mathrm{CHCl}_{3}$ ) as a colourless liquid. $[\alpha]_{\mathrm{D}}^{22}=-59.0\left(c \quad 0.53, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ;{ }^{1} \mathrm{H}$ NMR $(300 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta 7.40-7.27(5 \mathrm{H}), 4.21(\mathrm{dm}, J=2.6,0.5 \mathrm{H}), 4.89(\mathrm{~d}, J=$ $7.1,0.5 \mathrm{H}), 4.85(\mathrm{~d}, J=7.0,0.5 \mathrm{H}), 4.82-4.76(0.5 \mathrm{H}), 4.77(\mathrm{~d}, J=$ $3.3,0.5 \mathrm{H}), 4.74(\mathrm{~d}, J=3.2,0.5 \mathrm{H}), 4.63(\mathrm{~s}, 1 \mathrm{H}), 4.62(\mathrm{~s}, 1 \mathrm{H}), 3.98$ (qd, $J=6.3,2.9,0.5 \mathrm{H}), 3.48(\mathrm{qd}, J=6.1,2.9,0.5 \mathrm{H}), 3.38-3.23$ (1H), 2.25-2.16 (0.5H), 2.03-1.64 (2.5H), 1.60-1.45 (1H), 1.32 (d, $J=6.1,1.5 \mathrm{H}), 1.26(\mathrm{~d}, J=6.2,1.5 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 75 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 137.7$ (0), 137.6 (0), 128.4 (1), 128.4 (1), 127.8 (1), 127.8 (1), 127.7 (1), 127.7 (1), 95.7 (1), 93.3 (2), 93.0 (2), 90.7 (1), 77.2 (1), 76.5 (1), 74.7 (1), 69.6 (2), 69.5 (2), 67.9 (1), 31.8 (2), 29.3 (2), 28.1 (2), 23.9 (2), 18.3 (3), 18.2 (3); IR (neat): $\nu$ 3405 (m), 2936 (m), 1453 (w), 1032 (s), 989 (s); MS (EI): m/z 235 (7), 190 (5), 129 (12), 91 (100), 87 (41), 69 (13); HRMS (ESI): calcd for $\mathrm{C}_{14} \mathrm{H}_{20} \mathrm{O}_{4} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$: 275.1259 , found: 275.1269.

## Acknowledgements

This work was generously supported by the Deutsche Forschungsgemeinschaft (DFG grant Schm 1095/6-2). We thank Evonik Oxeno for generous donation of solvents.

## Notes and references

1 D. Leonori and P. H. Seeberger, Org. Lett., 2012, 14, 4954-4957.
2 M. Brasholz and H.-U. Reissig, Angew. Chem., Int. Ed., 2007, 46, 1634-1637.
3 O. Calin, R. Pragani and P. H. Seeberger, J. Org. Chem., 2012, 77, 870-877.
4 X. Yu, M. Li and G. A. O’Doherty, Heterocycles, 2011, 82, 1577-1584.
5 H.-Y. L. Wang and G. A. O'Doherty, Chem. Commun., 2011, 47, 10251-10253.
6 L. F. Tietze, S. Dietz, N. Böhnke, M. A. Düfert, I. Objartel and D. Stalke, Eur. J. Org. Chem., 2011, 6574-6580.
7 S. K. Bagal, S. G. Davies, A. M. Fletcher, J. A. Lee, P. M. Roberts, P. M. Scott and J. E. Thomson, Tetrahedron Lett., 2011, 52, 2216-2220.
8 M. Shan, Y. Xing and G. A. O'Doherty, J. Org. Chem., 2009, 74, 5961-5966.
9 L. F. Tietze, N. Böhnke and S. Dietz, Org. Lett., 2009, 11, 2948-2950.
10 M. Brasholz and H.-U. Reißig, Eur. J. Org. Chem., 2009, 3595-3604.

11 M. Niggemann, A. Jelonek, N. Biber, M. Wuchrer and B. Plietker, J. Org. Chem., 2008, 73, 7028-7036.

12 J. A. Bodkin, G. B. Bacskay and M. D. McLeod, Org. Biomol. Chem., 2008, 6, 2544-2553.
13 P. A. Wade, S. G. D'Ambrosio, J. A. Rao, S. Shah-Patel, D. T. Cole, J. K. Murray and P. J. Carroll, J. Org. Chem., 1997, 62, 3671-3677.
14 F. E. McDonald and H. Y. H. Zhu, J. Am. Chem. Soc., 1998, 120, 4246-4247.
15 F. E. McDonald, K. S. Reddy and Y. Díaz, J. Am. Chem. Soc., 2000, 122, 4304-4309.
16 A. Kirschning, U. Hary and M. Ries, Tetrahedron, 1995, 51, 2297-2304.
17 A. Kirschning, M. Jesberger and K.-U. Schöning, Synthesis, 2001, 507-540.
18 M. W. Peczuh and N. L. Snyder, Tetrahedron Lett., 2003, 44, 4057-4061.
19 S. Castro, M. Duff, N. L. Snyder, M. Morton, C. V. Kumar and M. W. Peczuh, Org. Biomol. Chem., 2005, 3, 3869-3872.
20 S. Castro and M. W. Peczuh, J. Org. Chem., 2005, 70, 3312-3315.
21 J. Saha and M. W. Peczuh, Chem.-Eur. J., 2011, 17, 7357-7365.
22 B. Schmidt, Eur. J. Org. Chem., 2004, 1865-1880.
23 B. Schmidt, Pure Appl. Chem., 2006, 78, 469-476.
24 B. Schmidt, J. Mol. Catal. A: Chem., 2006, 254, 53-57.
25 B. Schmidt and A. Biernat, Org. Lett., 2008, 10, 105-108.
26 B. Schmidt and A. Biernat, Chem.-Eur. J., 2008, 14, 6135-6141.
27 B. Schmidt and A. Biernat, Eur. J. Org. Chem., 2008, 5764-5769.
28 D. E. Fogg and E. N. dos Santos, Coord. Chem. Rev., 2004, 248, 2365-2379.
29 A. E. Sutton, B. A. Seigal, D. F. Finnegan and M. L. Snapper, J. Am. Chem. Soc., 2002, 124, 13390-13391.
30 B. Schmidt, Eur. J. Org. Chem., 2003, 816-819.
31 B. Schmidt, Chem. Commun., 2004, 742-743.
32 B. Schmidt, J. Org. Chem., 2004, 69, 7672-7687.
33 M. Arisawa, Y. Terada, M. Nakagawa and A. Nishida, Angew. Chem., Int. Ed., 2002, 41, 4732-4734.
34 S. H. Hong, M. W. Day and R. H. Grubbs, J. Am. Chem. Soc., 2004, 126, 7414-7415.
35 S. Fustero, M. Sánchez-Roselló, D. Jiménez, J. F. SanzCervera, C. del Pozo and J. L. Aceña, J. Org. Chem., 2006, 71, 2706-2714.
36 S. H. Hong, D. P. Sanders, C. W. Lee and R. H. Grubbs, J. Am. Chem. Soc., 2005, 127, 17160-17161.

37 S. J. Connon and S. Blechert, Angew. Chem., Int. Ed., 2003, 42, 1900-1923.
38 S. J. Connon and S. Blechert, Top. Organomet. Chem., 2004, 11, 93-124.
39 S. V. Ley, A. Armstrong, D. Díez-Martín, M. J. Ford, P. Grice, J. G. Knight, H. C. Kolb, A. Madin, C. A. Marby, S. Mukherjee, A. N. Shaw, A. M. Z. Slawin, S. Vile, A. D. White, D. J. Williams and M. Woods, J. Chem. Soc., Perkin Trans. 1, 1991, 667-692.

40 M. J. Comin, D. A. Parrish, J. R. Deschamps and V. E. Marquez, Org. Lett., 2006, 8, 705-708.

41 J. S. Yadav, T. Swamy and B. V. Reddy, Synlett, 2008, 2773-2776.
42 B. M. Trost, A. Aponick and B. N. Stanzl, Chem.-Eur. J., 2007, 13, 9547-9560.
43 H. Fuwa, H. Yamaguchi and M. Sasaki, Org. Lett., 2010, 12, 1848-1851.
44 H. Fuwa, H. Yamaguchi and M. Sasaki, Tetrahedron, 2010, 66, 7492-7503.
45 J. S. Yadav, G. M. Reddy, T. S. Rao, B. V. S. Reddy and A. Al Khazim Al Ghamdi, Synthesis, 2012, 44, 783-787.

46 A. Bouziane, B. Carboni, C. Bruneau, F. Carreaux and J.-L. Renaud, Tetrahedron, 2008, 64, 11745-11751.

47 V. Cadierno, P. Crochet and J. Gimeno, Synlett, 2008, 1105-1124.
48 R. García-Álvarez, F. J. Suárez, J. Díez, P. Crochet, V. Cadierno, A. Antinolo, R. Fernández-Galán and F. Carrillo-Hermosilla, Organometallics, 2012, 31, 8301-8311.
49 B. M. Trost and R. J. Kulawiec, J. Am. Chem. Soc., 1993, 115, 2027-2036.
50 B. M. Trost, A. C. Gutierrez and R. C. Livingston, Org. Lett., 2009, 11, 2539-2542.
51 E. S. Greenwood, P. J. Parsons and M. J. Young, Synth. Coттип., 2003, 33, 223-228.
52 J. P. Jordan and R. H. Grubbs, Angew. Chem., Int. Ed., 2007, 46, 5152-5155.
53 G. B. Djigoué and M. A. R. Meier, Appl. Catal., A, 2009, 368, 158-162.
54 B. Schmidt and D. Geißler, ChemCatChem, 2010, 2, 423-429.

55 M. Scholl, S. Ding, C. W. Lee and R. H. Grubbs, Org. Lett., 1999, 1, 953-956.
56 G. S. Forman, A. E. McConnell, R. P. Tooze, W. J. van Rensburg, W. H. Meyer, M. M. Kirk, C. L. Dwyer and D. W. Serfontein, Organometallics, 2005, 24, 4528-4542.

57 G. S. Forman and R. P. Tooze, J. Organomet. Chem., 2005, 690, 5863-5866.
58 J. Louie and R. H. Grubbs, Organometallics, 2002, 21, 2153-2164.
59 N. J. Lawrence, M. D. Drew and S. M. Bushell, J. Chem. Soc., Perkin Trans. 1, 1999, 3381-3391.
60 B. A. Baker, Z. V. Boskovic and B. H. Lipshutz, Org. Lett., 2008, 10, 289-292.
61 B. T. Rasley, M. Rapta and R. J. Kulawiec, Organometallics, 1996, 15, 2852-2854.
62 T. Doi, T. Fukuyama, J. Horiguchi, T. Okamura and I. Ryu, Synlett, 2006, 721-724.
63 T. Doi, T. Fukuyama, S. Minamino and R. Ilhyong, Synlett, 2006, 3013-3016.
64 C. Menozzi, P. I. Dalko and J. Cossy, Synlett, 2005, 2449-2452.
65 B. Schmidt and O. Kunz, Eur. J. Org. Chem., 2012, 1008-1018.
66 J. G. Young, W. H. Hartung and H. H. Daniels, J. Org. Chem., 1953, 18, 229-234.
67 F.-X. Felpin, E. Fouquet and C. Zakri, Adv. Synth. Catal., 2009, 351, 649-655.
68 F.-X. Felpin and E. Fouquet, Chem.-Eur. J., 2010, 16, 12440-12445.
69 W. S. Mahoney, D. M. Brestensky and J. M. Stryker, J. Am. Chem. Soc., 1988, 110, 291-293.


[^0]:    Postprint archived at the Institutional Repository of the Potsdam University in:
    Postprints der Universität Potsdam
    Mathematisch-Naturwissenschaftliche Reihe ; 241
    ISSN 1866-8372
    http://nbn-resolving.de/urn:nbn:de:kobv:517-opus4-95037

[^1]:    Universitaet Potsdam, Institut fuer Chemie, Organische Synthesechemie, Karl-Liebknecht-Straße 24-25, D-14476 Potsdam-Golm, Germany.
    E-mail: bernd.schmidt@uni-potsdam.de
    $\dagger$ Electronic supplementary information (ESI) available. See DOI: 10.1039/c3ob40167g

[^2]:    $\ddagger$ When the isolated and purified CM product $5 \mathbf{5}$ was treated with $5 \mathrm{~mol} \%$ of catalyst $\mathbf{A}$ and isomerization inducing additives such as ethyl vinyl ether or silanes isomerization product $\mathbf{6 a}$ was obtained in similar yields.

