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## PRE-COLLISIONAL ACCRETION AND EXHUMATION ALONG THE SOUTHERN LAURASIAN ACTIVE MARGIN, CENTRAL PONTIDES, TURKEY

Ph.D. THESIS
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LAVRASYA'NIN GÜNEY AKTİF KENARI BOYUNCA ÇARPIŞMA ÖCESİ EKLEMLENME VE YÜZEYLENME, ORTA PONTİDLER, TÜRKİYE

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To my family,

## FOREWORD

Well, this thesis took more time than I expected. However, I feel satisfied and comfortable because I enjoyed almost every moment of this passage of time by learning new things and improving my knowledge. In fact, I realized that the point is not about simply knowing things but rather how did you get that information. I think, that is the essence of science or scientific research. Beyond satisfying my personal curiosity, however, I also hope that this thesis will be useful for other researchers and make some contribution to the geology of Turkey and some similar geological settings.
I am very happy that this PhD thesis is done under the supervision of Prof. Aral Okay. Without his huge motivating, improving and constructive efforts, I wouldn't be able to finish this thesis. I am grateful to Prof. Aral Okay for his supervision, from field work to petrographic microscope studies. The submitted manuscripts and the thesis itself have largely benefited from his comments, suggestions and corrections. I am also grateful to my co-advisor Prof. em. Dr. Roland Oberhänsli. He kindly provided me opportunity to work in University of Potsdam. During my stay in Potsdam, I used many of the laboratory facilities and without these analytical data this thesis definitely wouldn't have finished. But more importantly, I discussed and learned a lot during my stay in Potsdam! Mostly with Prof. Oberhänsli (sometimes half a day!) but also with almost all of the other members of Mineralogy and Petrology groups. I am particularly grateful to Dr. Matthias Konrad-Schmolke for discussions on thermodynamics and Theriak-Domino software.
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## TABLE OF CONTENTS

Page
FOREWORD ..... ix
TABLE OF CONTENTS ..... xi
ABBREVIATIONS ..... xv
LIST OF TABLES ..... xvii
LIST OF FIGURES ..... xix
SUMMARY ..... xxvii
ÖZET ..... xxxi

1. INTRODUCTION ..... 1
1.1 Regional Geology ..... 3
1.2 Geology of the Central Pontides ..... 5
1.3 Purpose of Thesis ..... 9
1.4 Methodology ..... 9
1.4.1 PT calculations ..... 10
1.4.1.1 THERMOCALC ..... 10
1.4.1.2 Theriak-Domino ..... 10
1.4.1.3 Raman spectra of carbonaceous material (RSCM) thermometers ..... 10
1.4.2 Geochemistry ..... 12
1.4.3 Geochronology ..... 12
1.4.3.1 ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ white mica geochronology ..... 12
1.4.3.2 U/Pb zircon geochronology. ..... 14
2. GEOLOGY ..... 15
2.1 Mesozoic Subduction-Accretionary Complexes ..... 15
2.1.1 Esenler Unit: Accreted distal turbidites ..... 17
2.1.2 Domuzdağ Complex: Oceanic metabasalts and metasediments ..... 18
2.1.2.1 Extensional shear zone ..... 20
2.1.3 Kunduz Metamorphics ..... 21
2.1.4 Kirazbaşı Complex: Forearc basin and mélanges ..... 22
2.1.5 İkiçam Formation: Alkaline volcanic rocks and turbidites ..... 23
2.2 Accreted Arc Sequence ..... 24
2.2.1 Kösdağ Formation ..... 25
2.2.2 Dikmen Formation ..... 27
2.2.3 Ophiolitic mélange (İAES) ..... 29
2.3 Cover Units ..... 30
2.4. Structural Analysis of the Wedge. ..... 32
3. PETROGRAPHY AND MINERAL CHEMISTRY ..... 35
3.1 Esenler Unit. ..... 35
3.2 Domuzdağ Complex ..... 40
3.2.1 Metabasites ..... 40
3.2.1.1 Lawsonite-blueschist (sample 128) ..... 41
3.2.1.2 Garnet-epidote-blueschist (sample 702A) ..... 42
3.2.2 Micaschists ..... 44
3.2.2.1 Garnet-chloritoid-micaschist (sample 775) ..... 46
3.2.2.2 Chloritoid-micaschist (sample 753A) ..... 48
3.2.2.3 Chloritoid-micaschist (sample 178). ..... 49
3.2.2.4 Micaschist with albite porphyroblasts(sample 392) ..... 52
3.3 Kunduz Metamorphics ..... 52
3.4 Kösdağ Formation ..... 54
3.4.1 Felsic volcanic rocks ..... 54
3.4.2 Basaltic andesite/andsesite ..... 56
4. RAMAN MICROSPECTROSCOPY OF CARBONACEOUS MATERIAL. 59
4.1 Esenler Unit ..... 59
4.2 Domuzdağ Complex ..... 62
4.3 Concluding Remarks ..... 63
5. METAMORPHIC CONDITIONS ..... 65
5.1 Esenler Unit ..... 65
5.1.1 Metabasite within the Esenler Unit ..... 66
5.1.2 Peak metamorphic temperatures of the slate/phyllites ..... 67
5.2 Domuzdağ Complex ..... 68
5.2.1 Lawsonite-blueschist ..... 69
5.2.2 Garnet-blueschist ..... 70
5.2.3 Chloritoid-micaschists ..... 71
5.2.3.1 Sample 775 ..... 71
5.2.3.2 Sample 753A ..... 73
5.2.3.3 Sample 178 ..... 74
5.3 Concluding Remarks ..... 76
5.3.1 Esenler Unit ..... 76
5.3.2 Domuzdağ Complex ..... 77
6. GEOCHRONOLOGY ..... 81
$6.1{ }^{40} \mathrm{Ar} /{ }^{39} \mathrm{Ar}$ Geochronology ..... 81
6.1.1 Esenler Unit ..... 81
6.1.2 Domuzdağ Complex ..... 83
6.1.3 Kunduz Metamorphics ..... 84
6.1.4 Metamorphic age of the Kösdağ Formation ..... 86
6.2 U/Pb Zircon Geochronology ..... 87
6.2.1 Crystallization age of the Kösdağ Formation ..... 87
6.3 Concluding Remarks ..... 89
6.3.1 Accretionary wedge ..... 89
6.3.2 Arc sequence ..... 90
7. GEOCHEMISTRY ..... 91
7.1 Geochemistry of the Kösdağ Formation ..... 91
7.1.1 Major element compositions ..... 91
7.1.2 Trace and rare earth element compositions ..... 93
7.2 Interpretation of the Analitical Data ..... 95
7.3 Concluding Remarks ..... 95
8. GEODYNAMICS ..... 99
8.1 Tectonic Thickening of the Albian-Turonian Accretionary Wedge ..... 100
8.2 Exhumation of HP/LT Metamorphic Rocks ..... 105
8.3 Comparative Regional Geology ..... 108
8.3.1 Late Cretaceous intra-oceanic Kösdağ arc and supra-subduction ophiolites109
9. CONCLUSIONS ..... 113
REFERENCES ..... 117
APPENDICES ..... 141
CURRICULUM VITAE ..... 205

## ABBREVIATIONS

| Al | : Aluminium |
| :--- | :--- |
| Appx | : Appendix |
| Ar | : Argon |
| B | : Boron |
| BA1 | : Basaltic andesite 1 |
| BA2 | : Basaltic andesite 2 |
| ca. | : circa |
| Ca | : Calcium |
| CaF | : Calcium fluoride |
| cm | : centimeter |
| CO | : Carbon dioxide |
| c.p.f.u | : cation per formula unit |
| CPS | : Central Pontide Supercomplex |
| e.g. | : for example |
| EMPA | : Electron microprobe analysis |
| et al. | : and others |
| GPS | : Global Positioning System |
| HFSE | : High Field Strenght Elements |
| Hg | : mercury |
| HP/LT | : high pressure/low temperature |
| HREE | : Heavy Rare Earth Elements |
| ICP-AES | : Inductively Coupled Plasma Atomic Emission Spectrometry |
| İAES | : İmir-Ankara-Erzincan Suture |
| Kbar | : kilobar |
| keV | : kilo-electronvolt |
| K | : Potassium |
| K2SO | : Potassium sulfate |
| LA-ICP-MS | : Laser ablation inductively coupled plasma mass spetrometry |
| LILE | : Large Ion Lithophile Elements |
| LREE | : Light Rare Earth Elements |
| m | : meter |
| Ma | : megaannus |
| Mg | : magnesium |
| MORB | : Mid-Ocean Ridge Basalt |
| mm | : millimeter |
| Mm | : micrometer |
| nA | : nanoampere |
| NAF | : North Anatolian Fault |
| nm | : nanometer |
| Pb | : lead |
| REE | : Rare Earth Elements |
| RSCM | : Raman spectra of carbonaceous material |
|  |  |

$\mathbf{S}$
SPT
U
W
XRF
${ }^{\circ} \mathrm{C}$
: second
: Sodium polytungstate
: uranium
: watt
: X-ray fluorescence
: degree Celsius

## LIST OF TABLES

Page
Table 4.1 : Parameters of decomposed Raman spectra of carbonaceous material from the Esenler Unit ..... 61
Table 4.2 : Parameters of decomposed Raman spectra of carbonaceous material from the Domuzdağ Complex ..... 62
Table 5.1 : Raman parameters and the calculated peak metamorphic temperatures for the Esenler Unit ..... 67
Table 5.2 : Raman parameters and the calculated peak metamorphic temperatures for the chloritoid micaschists. ..... 73
Table 6.1 : LA-ICP-MS U-Pb data of the measured zircons from the metarhyolites. ..... 89
Table 7.1 : Major, trace and rare earth element analysis of the metavolcanic rocks. 92
Table A. 1 : Mineral chemistry of the measured amphiboles. Mineral formulacalculations are based on 23 oxygens147
Table A. 2 : Mineral chemistry of the measured phengitic white micas. Mineral formula calculations are based on 11 oxygens ..... 154
Table A. 3 : Mineral chemistry of the measured paragonites. Mineral formula calculations are based on 11 oxygens ..... 163
Table A. 4 : Mineral chemistry of the measured garnets. Mineral formula calculations are based on 12 oxygens ..... 165
Table A. 5 : Mineral chemistry of the measured chloritoids. Mineral formula calculations are based on 12 oxygens. ..... 169
Table A. 6 : Mineral chemistry of the measured lawsonites. Mineral formula calculations are based on eight oxygens. ..... 179
Table A. 7 : Mineral chemistry of the measured sodic-pyroxenes. Mineral formula calculations are based on four cations ..... 180
Table A. 8 : Mineral chemistry of the measured epidotes. Mineral formula calculations are based on 12.5 oxygens. ..... 181
Table A. 9 : Mineral chemistry of the measured chlorites. Mineral formula calculations are based on 14 oxygens ..... 186
Table A.10: Mineral chemistry of the measured plagioclases. Mineral formula calculations are based on eight oxygens. ..... 192
Table A.11 : Bulk rock compositions and calculated element mol percentages used in the pseudosection calculations. ..... 195
Table A. 12 : UTM coordinates (European 1979 datum) of the metapelitic samples of the Esenler Unit that were analyzed by Raman Spectroscopy ..... 196
Table A.13 : ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ analytical data of the dated samples. ..... 197

## LIST OF FIGURES

## Page

Figure 1.1 : Main components of an accretionary-type orogeny including oceanward tectonic growth of subduction-accretionary wedge and addition of juvenile material to the crust. Modified from Barr et al. (1999)
.2
Figure 1.2 : Tectonic map of Turkey and surrounding regions (modified from Okay and Tüysüz, 1999). Green and pale blue colors highlight the Laurasia and Gondawana-derived terranes, respectively. CPS: Central Pontide Supercomplex, CTF: Lower Cretaceous submarine turbidite fan. North of the Black Sea, "V" marks the Albian volcanic arc modified from Nikishin et al. (2015)
Figure 1.3: Geological map of the Central Pontides modified from Tüysüz (1990), Uğuz et al. (2002), Okay et al. $(2013,2014)$ and this study
Figure 1.4 : A photomicrograph in plane polarized light showing carbonaceous material (CM) within albite and quartz. $\mathrm{Ab}=$ albite, $\mathrm{Qtz}=$ quartz, Phe=phengite11

Figure 2.1 : Geological map of the area studied across the Central Pontide Supercomplex between Kastamonu-Tosya. For the location see Figure 1.3

Figure 2.2 : Geological cross sections. See Fig. 2.1 for their locations. a) A-A' section shows the initial structural relations between the Çangaldag Complex, the Esenler Unit and the Domuzdağ Complex. b) B-B' section shows the post-metamorphic faulting and folding between the Esenler Unit, Domuzdağ Complex and the Kunduz Metamorphics17

Figure 2.3 : Field photos of the Esenler Unit. a) Sheared and deformed slate and metasandstone intercalation. b) Phyllite exposed close to the contact c) A debris flow level consisting of various sized marble olistoliths. d) A serpentinite slice within the phyllite along the main Kastamonu-Tosya road
Figure 2.4 : Field photos of the Domuzdağ Complex. a) Micaschist with its typical dark grey outcrop color. b) Well-foliated epidote-blueschist
Figure 2.5 : Field photos of the extensional shear zone. a) General view of the shear zone with blocks of (from S to N ) marbles and a retrogressed metabasite within a grey cataclasite. b) A micaschist block within the cataclasite with NW sense of shear (pencil // lineation)
Figure 2.6 : Field photos of the Kunduz Metamorphics. a) General view of the multicolored exposures of marble and metabasite/metatuff intercalation. b) A closer view of the same outcrop. Marbles form strained pods within the metatuff. c) Thinly bedded sodic-amphibole-bearing metatuff and marble intercalation. d) Light colored metatuff, metachert and metabasite interbeds

Figure 2.7 : Field photos of the Kirazbaşı Complex. a) Sheared slate-greywacke type mélange. b) Hemipelagic volcanoclastic sedimentary rocks. c) Discoidal shape pillow lavas. d) Sheared sedimentary mélange/debris flow with blocks of sandstone, siltstone, limestone and serpentinite.
Figure 2.8 : Field photos of the İkiçam Formation. (a) \& (b) Pyroclastic rocks of andesitic composition. c) Leucite-bearing basalt. d) Volcanoclastic dark grey turbidites. This part mainly consists of sandstone and siltstone .... 24
Figure 2.9 : Geological map of the studied section south of Tosya, along the İzmir-Ankara-Erzincan suture with a cross section between A and A'. For location see Fig. 1.3
Figure 2.10 : Field photos of the Kösdağ Formation. All rocks have undergone a low-grade metamorphism but are referred here with their protolith names. a) Basaltic andesite and rhyolite. b) Thin chert interbeds within the acidic and basic rocks. Pyroclastic rocks of acidic (c) and andesitic (d) compositions. e) Red pelagic limestone/chert interbeds with basaltic andesitic rocks. f) A close look to the same outcrop in (e). Folded bright cherts occur within the red pelagic limestone. 26
Figure 2.11 : Field photos of the Dikmen Formation. a) Contact between rhyolite and overlying recrystallized reddish micritic limestone of Dikmen Formation in the upper limb of the overturned anticline. b) Yellowish and reddish slate and recrystallized limestone exposed along the transional section between Kösdağ and Dikmen formations. c\&d) Recrystallized calciturbidite consisting intercalation of moderately bedded dark levels consisting of transported coarse grains and thinly bedded pinkish pelagic levels

27
Figure 2.12 : Measured stratigraphic section illustrating the transitional character between the Kösdağ Formation and the overlying Dikmen Formation. The section is about 250 m and is measured along the main road to Yukarıdikmen village. For location, see Fig. 2.9.................................. 28
Figure 2.13 : Field photos of the ophiolitic mélange. a) Serpentinite (green) that is overthrusted by recrystallized limestone (pinkish) of Dikmen Formation. b) Pillow basalts. 29
Figure 2.14 : Maastrichtian limestone with Orbitoides (a) and Helonocyclina (b) type foraminifera................................................................................. 30
Figure 2.15 : Field photos of the Lower-Middle Eocene limestone (a) and turbidite (b). c) Eocene limestone and overlaying basalt cover. d) Vesicular basalts of probable Eocene age. 31
Figure 2.16 : a) A general view of the Tosya Basin between the Kunduz Metamorphics to the north and hills made up from the metavolcanic rocks of the Kösdağ Formation to the south. b \& c) Field photos of the NAF-related unconsolidated Neogene gravelstone and sandstone of the Tosya Basin. 32
Figure 2.17 : Lower hemisphere, equal area projections of the foliations and the lineations. a) Foliation and b) lineation data from the Esenler Unit. c) Foliation and d) lineation measurements of the northern part of the Domuzdag Complex. Lineations are mostly orthogonal to the NW dipping foliation planes and plunge to the NW related exhumation. e) Foliation data from the southern part of the Domuzdağ Complex and f) the Kunduz Metamorphics. Contour interval is 10................................ 33

Figure 2.18 : Lower hemisphere, equal area projections of the foliations of the Kösdağ Formation. Contour interval is 10

34
Figure 3.1 : Representative photomicrographs of the metapelitic rocks from the Esenler Unit (plane polarized light). a \& b) Fine-grained slates (samples 847 and 853). Foliation is defined by white mica and partly recrystallized detrital quartz. c \& d) Coarser grained phyllites with spaced foliation (samples 31 and 879). $\mathrm{Ab}=$ albite, $\mathrm{Qtz}=\mathrm{quartz}$, Phe=phengite, $\mathrm{Cc}=$ calcite.36

Figure 3.2 : Photomicrographs of a greywake-type metasandstone from the Esenler Unit under plane (left) and cross (right) polarized light. Majority of the clasts are quartz ( Qtz ) with minor feldspar $(\mathrm{Pl})$ and white mica (Ms).. 36
Figure 3.3 : A marble with elongated fibrous calcite............................................... 37
Figure 3.4 : Plane-polarized photomicrographs of incipient blueschist facies metabasites from the Esenler Unit. a) A foliated blueschist. Folation is defined by sodic-amphibole ( $\mathrm{Na}-\mathrm{amph}$ ). b) An unfoliated metabasite with randomly distributed sodic-amphiboles (sample 211B). Cpx= augite relicts; $\mathrm{Ep}=$ epidote; $\mathrm{Ch}=$ chlorite; $\mathrm{Ab}=$ albite37
Figure 3.5 : Compositional range of amphibole. Act= actinolite, Gln= glaucophane.38
Figure 3.6 : Compositional range of sodic-pyroxene. ..... 39
Figure 3.7 : Compositional range of epidotes ..... 39
Figure 3.8 : Compositional range of chlorites ..... 39
Figure 3.9 : Plane-polarized photomicrographs from the Domuzdağ Complex of a)epidote-blueschist and b) albite-chlorite fels that represents a stronglyretrogressed blueschist. $\mathrm{Na}-\mathrm{amph}=$ sodic-amphbole, $\mathrm{Chl}=$ chlorite, $\mathrm{Ab}=$albite, $\mathrm{Ep}=$ epidote, $\mathrm{Cc}=$ calcite40

Figure 3.10 : Microphotograph of a lawsonite blueschist (sample 128) from the Domuzdağ Complex. Na-amph= sodic-amphbole, Na-cpx=sodicpyroxene, Lws= lawsonite, Ttn= titanite
Figure 3.11 : Compositional range of phengites. Al vs Si diagram showing the celadonite exchage. Decreasing Si content mainly depicts the effect of the retrogression42

Figure 3.12 : a) BSE image of the garnet-blueschist (702A). Garnet (Grt) exhibit pressure shadow consisting of glaucophane (Gln) + phengite (Phe) +epidote (Ep). Calcic-amphibole (Ca-amph), chlorite (Chl) and albite $(\mathrm{Ab})$ are secondary and replace glaucophane. b) Element mapping of the garnet in Fig. 3.9a with typical growth zoning. Ca and Mg are almost stable. Mn decreases rimward while Fe increases. Ttn=titanite, Aug=augite.

43
Figure 3.13 : Compositional range of garnets.......................................................... 44
Figure 3.14 : Microphotograph of a retrogressed micaschists from the Domuzdağ Complex with syn-kinematic albite porphyroblasts under plane (left) and cross (right) polarized light. Qtz= quartz, $\mathrm{Ph}=$ phengite, $\mathrm{Chl}=$ chlorite, $\mathrm{Ab}=$ albite

44
Figure 3.15 : Photomicrographs of the chloritoid-micaschists from the Domuzdağ Complex (plane polarized light). a) Sample 775, a garnet-chloritoid micaschist. b) Sample 753A, a chloritoid-micaschist. c) Syn-kinematic pseudomorphs after lawsonite (sample 775). Former lawsonite was replaced by epidote (Ep), paragonite ( Pa ) and quartz $(\mathrm{Qtz})$. The synkinematic character of the pseudomorphs suggests that they were stable
during the peak metamorphic conditions. d) Pseudomorphs after glaucophane (sample 753A). They are replaced by chlorite (Chl), phengite (Phe), and quartz (Qtz). Cld=chloritoid, Grt= garnet. ............ 45
Figure 3.16 : BSE images of the sample 775. Epidote (Ep) + paragonite (Pa) + quartz (Qtz) replaced former lawsonite and chlorite (Chl) + phengite (Phe) + quartz (Qtz) replaced former glaucophane. Albite ( Ab ) is secondary. Quartz (Qtz) inclusions in garnet (Grt) do not show any rotational fabric. $\mathrm{Cld}=$ chloritoid. 46
Figure 3.17 : A photomicrograph showing the partly preserved glaucophane (Gln) relicts in the sample 775. The majority of the glaucophanes are replaced by chlorite, phengite and quartz (see Fig 3.16) $\mathrm{Qtz}=$ quartz, Phe= phengite

47
Figure 3.18 : Element mapping of the post-kinematic garnet in sample 775. Dark areas within the garnet are quartz inclusions. Notice the Ca-rich outermost rim possibly depicting increasing rate of lawsonite breakdown

Figure 3.19 : Compositional range of the chloritoids. (a) Ternary compositional diagram. While in sample 178, main exchange occur between Fe and Mg , in the samples 775 and 753A Mg mainly substitute with Mn. (b) XMg vs XFe diagram 48
Figure 3.20 : BSE images of the sample 753A. Chlorite(Chl) + phengite (Phe) + quartz (Qtz) replaced former glaucophane. The matrix chlorite penetrates into former glaucophane as secondary thin stripes. Paragonite also possibly replaced former glaucophane along its rims. $\mathrm{Apa}=$ apatite.
Figure 3.21 : BSE images of the sample 178. a). Chloritoid (Cld) agregates together with a rectangular pseudomorph at the center of image composed of albite ( Ab ). Tiny inclusions of phengites within the pseudomorph exhibit rotational fabrics. Foliation also slightly bends around the pseudomorphs indicating its syn-kinematic origin. It is interpreted as former jadeite. b) A pseudomorph after chloritoid (Cld) within the quartz rich domain. It is replaced by chlorite (Chl) + quartz (Qtz) + phengite (Phe). c) The same pseudomorph in "b" under microscope (plane polarized light). $\mathrm{Pa}=$ paragonite
Figure 3.22: Microphotographs from the Kunduz Metamorphics. a) A greenschist consisting of calcic-amphibole (Ca-amph), chlorite (Chl), albite (Ab) and titanite (Ttn). b) A blueschist consisting of sodic amphibole (Naamph), epidote (Ep), albite ( Ab ) and chlorite (Chl). c) a phyllitic metatuff consisting of phengite (Phe), quartz (Qtz), calcic-amphibole (Ca-amph), chlorite (Chl), stilpnomelane (Stp) with relict clinopyroxene (Cpx). d) A weakly metamorphosed metabasite form the lower part of the Kunduz Metmorphics. Cpx= magmatic pyroxene, Chl= chlorite, $\mathrm{Al}=$ albite.
Figure 3.23 : Microphotographs (plane polarized in the left side and cross polarized in the right) of the acidic rocks of the Kösdağ Formation. a \& b) SiO 2 saturated metarhyolite with quartz ( Qtz ) and feldspar ( Pl ) phenocrysts. c \& d) A cognate mafic xenolith within a metarhyolite (1213). It consists of plagioclase ( Pl ), clinopyroxene, $\mathrm{Fe}-\mathrm{Ti}$ oxide ( Ilm ), and apatite. e \& f) A cognate xenolith with cumulate-like plagioclases ( Pl ) with interstitial clinopyroxene (Cpx) in a metarhyolite. Apatite (Apa) and Fe-Ti oxides
also occur in the xenolith. g \& h) A cumulate-like plagioclase cluster without mafic minerals within a metarhyolite (1304B).
Figure 3.24 : Microphotographs (plane polarized in the left side and cross polarized in the right) of the basaltic andesite/andesite of the Kösdağ Formation. a \&b) Glomeroporphyritic cluster within basaltic andesites, sample 1287 ( $\mathrm{a} \& \mathrm{~b}$ ) and sample 1288 (c\&d). e \& f) A cognate xenolith consisting of clinopyroxene, plagioclase and apatite in an andesite (1286). Apatite is also found as inclusion within the clinopyroxene. $\mathrm{g} \& \mathrm{~h}$ ) Metabasite with augite relicts. Ca -amphibole replaces the augite along their rims. 56
Figure 4.1 : Representative Raman spectra of carbonaceous material of the measured samples from the Esenler Unit. The RSCM forms two clusters corresponding texturally to the slates and phyllites. The sample numbers are given at the right side of the each spectrum.

60
Figure 4.2 : Representative examples of fitting of the measured Raman spectra. a) Low-temperature slates that are characterized by undifferentiating of graphite (G) and D2 defect bands and occurrence of broad D3 and D4 defect bands (sample 847). b) Phyllite showing slightly pronounced G band with D2 occurring on its shoulder (sample 230).61

Figure 4.3 : Representative RSCM of the metapelitic rocks of the Domuzdağ Complex.

63
Figure 5.1 : PT conditions of an incipient blueschist facies metabasite within the Esenler Unit. Stabilities of the amphiboles are from Otsuki and Banno (1990) and a petrogenetic grid in NCMASH system for low-grade metabasite is from Schiffman and Day (1999) 66
Figure 5.2 : Diagram showing RSCM temperatures of the metapelitic rocks of the Esenler Unit. The temperatures are obtained using calibrations of Beyssac et al. (2002a) and Rahl et al. (2005) 68
Figure 5.3 : Pseudosection produced by the Theriak-Domino for the sample 128. Thick blue line limits the stability field of lawsonite, thick dashed line garnet and thin dashed line of paragonite in this specific rock composition. According to XJd (red) of sodic-pyroxene and Si isopleths (grey) of phengite metamorphic conditions of the sample are constrained to $14 \pm 2$ kbar and $370-440^{\circ} \mathrm{C}$ (the shaded area)
Figure 5.4 : Pseudosection produced by the Theriak-Domino for the sample 702A. For this specific rock composition, thick black line limits stability field of epidote and dashed line of garnet. Stability field of lawsonite is marked by thick blue line. Isopleths; grey, Si content of phengite; green and red, XGrs and XAlm, respectively; dashed isopleth, XJd. Metamorphic conditions constrained as $17 \pm 1 \mathrm{kbar}$ and $500 \pm 40^{\circ} \mathrm{C}$ (the shaded area)

70
Figure 5.5 : a) Pseudosection modelling for the sample 775. Stability field of garnet is limited by thick black dashed line, of lawsonite by thick red line, of chloritoid by thick green dashed line, of glaucophane by blue line, of paragonite by thin black dashed line for this specific bulk rock composition. b) A detailed part of the pseudosection. Isopleths; grey $=\mathrm{Si}$ content of phengite; black and orange are XAlm and XGrs of garnet, respectively, green $=\mathrm{XMg}$ of chloritoid. Vertical grey strip indicates temperature values obtained by the RSCM thermometer based on Rahl et al. (2005) calibration. Metamorphic conditions constrained to $17.5 \pm 1$ kbar and $390-450^{\circ} \mathrm{C}$ (the shaded area in orange)................................. 72

Figure 5.6 : Pseudosection produced by the Theriak-Domino for the sample 753A. Stability field of garnet is limited by thick black dashed line, of chloritoid by thick green dashed line, of glaucophane by blue line, of paragonite by thin black dashed line for this specific bulk rock composition. Isopleths; grey $=\mathrm{Si}$ content of phengite, green $=\mathrm{XMg}$ of chloritoid. Vertical grey strip indicates temperature values obtained by the RSCM thermometer based on Rahl et al. (2005) calibration. Metamorphic conditions constrained to $16-18$ kbar and $475 \pm 40^{\circ} \mathrm{C}$ (the shaded area in orange).
Figure 5.7 : Pseudosection modelling for the sample 178. Stability field of garnet is limited by thick black dashed line, of chloritoid by thick green dashed line, of glaucophane by blue lines, of paragonite by red dashed line for the specific bulk rock composition. Isopleths; grey= Si content of phengite, green $=\mathrm{XMg}$ of chloritoid Vertical grey strip indicates temperature values obtained by the RSCM thermometer based on Rahl et al. (2005) calibration. Metamorphic conditions constrained to 22-25 kbar and $440 \pm 30^{\circ} \mathrm{C}$ (the shaded area in orange)................................ 75
Figure 5.8 : Thermal structure of the accreted distal turbidites. The temperature values are from the calibration of Rahl et al. (2005) due to the fact that it is essentially calibrated for low-grade metamorphic rocks. Relatively high temperature phyllites form a sliver within the low temperature slates. A sliver of Na-amphibole-bearing metabasite is possibly associated with the phyllites
Figure 5.9 : Compilation of the estimated PT data from the Domuzdağ Complex including chloritoid-micaschists, a lawsonite- and a granet-blueschist. PT conditions of the Elekdağ eclogites are from Okay et al. (2006a). Possible PT paths for the sample 178 and 775 are indicated by thick grey lines. The dashed parts are stand for zero strain decompressional exhumation. Metamorphic facies are modified from Evans (1990). Facies: LBS: lawsonite-blueschist, EBS: epidote-blueschist, EC: eclogite, AEA: albite-epidote amphibolite, GS: greenschist facies. The compilation suggest that Domuzdağ Complex consists of metamorphic rocks that were metamorphosed under distinct PT conditions .............. 79
Figure 6.1 : Microphotographs of the samples from the Esenler Unit dated by ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ phengite method. $\left.\mathrm{a} \& b\right)$ Sample 31 and $\left.\mathrm{c} \& d\right)$ sample 230 are phyllites that consist of quartz (Qtz), phengite (Phe), albite (Ab), chlorite (Chl). ...................................................................................... 82
Figure 6.2: ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ age spectra of the analyzed phengites from the (a) sample 31 and (b) sample 230 from the Esenler Unit.82

Figure 6.3 : Microphotographs of the samples from the Domuzdağ Complex dated by ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ phengite method. a\&b) Sample 392 consists of quartz (Qtz), phengite (Phe), albite (Ab), chlorite (Chl). c\&d) Sample 753A consists of chloritoid (Cld), quartz (Qtz), phengite (Phe), chlorite (Chl).
$\qquad$
Figure 6.4 : ${ }^{40} \mathrm{Ar} /{ }^{39} \mathrm{Ar}$ age spectra of the analyzed phengites from the Domuzdağ Complex. a) Sample 392, a micaschist. b) Sample 753A, a chloritoidmicaschist.
Figure 6.5 : Microphotographs of the samples from the Kunduz Metamorphics dated by ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ phengite method. a \& b) Sample 626A is a quartzmicaschist consisting of quartz (Qtz), phengite (Phe), hematite (Hem),
epidote (Ep). c \& d) Sample 1290 is a phyllite consisting of quartz (Qtz), phengite (Phe), chlorite (Chl), calcic-amphibole (Amph)85

Figure $6.6:{ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ age spectra of the analyzed phengites from the Kunduz Metamorphics. a) sample 626, a quartz-micaschist. b) Sample 1290, a phyllitic metatuff.
Figure 6.7 : a) Microphotograph of the dated sample 1214 from the Kösdağ Formation. The rock consists of quartz (Qtz), plagioclase ( Pl ), epidote (Ep) and muscovite (Ms). b) ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ age spectrum diagram obtained from the muscovites of this sample by stepwise heating method
Figure 6.8 : Cathodoluminescence images of the dated zircons from the sample 1214 (a) and 1215 (b).................................................................................... 88

Figure 6.9 : Zircon U/Pb concordia diagram for the sample 1214 (a) and 1215 (b) and their weighted mean age values ( c and d).88

Figure 7.1 : Total alkali $v s$ silica classification diagram (Le Bas et al., 1986) of the metavolcanic rocks. Symbols are in accordance with REE diagram in Fig. 7.3a.
Figure 7.2 : a) AFM (Irvine and Barager, 1971) and b) SiO 2 vs $\mathrm{FeO} / \mathrm{MgO}$ (Miyashiro, 1974) diagrams for the basaltic andesite/andesite of the Kösdağ Formation 93
Figure 7.3 : a) Chondrite normalized REE diagram for the metavolcanic rocks according to normalization values of Boynton (1984). b) NMORB normalized multi-element spider diagram of the metavolcanic rocks normalized to Sun and McDonough (1989) values

94
Figure 7.4 : a) Cartoon model showing the Late Cretaceous tectonic configuration of the Central Pontides. a) The Kösdağ Arc formed at ca. 95 Ma in an intra-oceanic setting located south of the Laurasian active margin. b) Slab rollback controlled arc migration leading to the termination of volcanism in the Kösdağ Arc
Figure 8.1 : A model showing a possible initial situation of the phyllites and the slates within the Albian accretionary wedge. While the slates represent the offscraped distal turbidites, the phyllite possibly represent underplated metasediments of the turbidite fan. Out-of-sequence thrusting (dashed thick lines) is proposed for uplift and tectonic emplacement of the phyllites. The model is modified after Moore et al. (2001)

Figure 8.2 : Model for the tectonic evolution of the Albian-Turonian wedge with a possible tectonic thickening and subsequent exhumation mechanism of the deep seated subduction-accretion complexes. During Albian, the wedge was dominated by accretion of clastic detritus. Lawsonite blueschists possibly formed during initial low shear stress Albian subduction. Late Albian to Turonian evolution of the wedge was mainly controlled by slab rollback, which creates the necessary space for deep level underplating by décollement propagation. Synchronous extension of the wedge by slab rollback, subsequently exhumes the deep-seated metamorphic sequence.
Figure 8.3 : An exhumation model for the deeply underplated oceanic HP/LT metamorphic sequence by decompression of the wedge along a retreating slab. Zero strain domains are characterized by partly preservation of the peak metamorphic assemblages. Retrogression is manifested by pseudomorphs formed through trapped fluids. B-rich
fluids are generated from the newly underplating oceanic metabasalts and metasediments through dehydration reactions. The fluids then were channelized and migrated upward along the high-strain shear zones. . 107
Figure 8.4 : Paleogeographic sketch map at ca. 95 Ma showing the dual subduction during Tethyan consumption (modified from Van der Voo et al. (1999)). Dashed line indicates location of the N-S section along the Central Pontides in Figure 7.4. ATB: Anatolide-Tauride Block.
Figure A. 1 : Geological map of the area studied across the Central Pontide Supercomplex between Kastamonu-Tosya.
Figure A.2 : Geological map of the studied section south of Tosya, along the İzmir-Ankara-Erzincan suture with a cross section between A and A' 145

# PRE-COLLISIONAL ACCRETION AND EXHUMATION ALONG THE SOUTHERN LAURASIAN ACTIVE MARGIN, CENTRAL PONTIDES, TURKEY 

## SUMMARY

The Central Pontides is an accretionary-type orogenic area within the AlpineHimalayan orogenic belt characterized by pre-collisional tectonic continental growth. The region comprises Mesozoic subduction-accretionary complexes and an accreted intra-oceanic arc that are sandwiched between the Laurasian active continental margin and Gondwana-derived the Kırşehir Block. The subduction-accretion complexes mainly consist of an Albian-Turonian accretionary wedge representing the Laurasian active continental margin. To the north, the wedge consists of slate/phyllite and metasandstone intercalation with recrystallized limestone, Na-amphibole-bearing metabasite ( $\mathrm{PT}=7-12$ kbar and $400 \pm 70^{\circ} \mathrm{C}$ ) and tectonic slices of serpentinite representing accreted distal part of a large Lower Cretaceous submarine turbidite fan deposited on the Laurasian active continental margin that was subsequently accreted and metamorphosed. Raman spectra of carbonaceous material (RSCM) of the metapelitic rocks revealed that the metaflysch sequence consists of metamorphic packets with distinct peak metamorphic temperatures. The majority of the metapelites are low-temperature (ca. $330^{\circ} \mathrm{C}$ ) slates characterized by lack of differentiation of the graphite ( G ) and D2 defect bands. They possibly represent offscraped distal turbidites along the toe of the Albian accretionary wedge. The rest are phyllites that are characterized by slightly pronounced G band with D2 defect band occurring on its shoulder. Peak metamorphic temperatures of these phyllites are constrained to $370-385^{\circ} \mathrm{C}$. The phyllites are associated with a strip of incipient blueschist facies metabasites which are found as slivers within the offscraped distal turbidites. They possibly represent underplated continental metasediments together with oceanic crustal basalt along the basal décollement. Tectonic emplacement of the underplated rocks into the offscraped distal turbidites was possibly achieved by out-of-sequence thrusting causing tectonic thickening and uplift of the wedge. ${ }^{40} \mathrm{Ar} /{ }^{39} \mathrm{Ar}$ phengite ages from the phyllites are ca. 100 Ma , indicating Albian subduction and regional HP metamorphism.

The accreted continental metasediments are underlain by HP/LT metamorphic rocks of oceanic origin along an extensional shear zone. The oceanic metamorphic sequence mainly comprises tectonically thickened deep-seated eclogite to blueschist facies metabasites and micaschists. In the studied area, metabasites are epidoteblueschists locally with garnet ( $\mathrm{PT}=17 \pm 1 \mathrm{kbar}$ and $500 \pm 40^{\circ} \mathrm{C}$ ). Lawsoniteblueschists are exposed as blocks along the extensional shear zone (PT= $14 \pm 2 \mathrm{kbar}$ and $370-440^{\circ} \mathrm{C}$ ). They are possibly associated with low shear stress regime of the initial stage of convergence. Close to the shear zone, the footwall micaschists consist of quartz, phengite, paragonite, chlorite, rutile with syn-kinematic albite porphyroblast formed by pervasive shearing during exhumation. These types of
micaschists are tourmaline-bearing and their retrograde nature suggests high-fluid flux along shear zones. Peak metamorphic mineral assemblages are partly preserved in the chloritoid-micaschist farther away from the shear zone representing the zero strain domains during exhumation. Three peak metamorphic assemblages are identified and their PT conditions are constrained by pseudosections produced by Theriak-Domino and by Raman spectra of carbonaceous material: 1) garnet-chloritoid-glaucophane with lawsonite pseudomorphs ( $\mathrm{P}=17.5 \pm 1 \mathrm{kbar}, \mathrm{T}: 390-450$
${ }^{\circ} \mathrm{C}$ ) 2) chloritoid with glaucophane pseudomorphs ( $\mathrm{P}=16-18 \mathrm{kbar}, \mathrm{T}: 475 \pm 40^{\circ} \mathrm{C}$ ) and 3) relatively high-Mg chloritoid ( $17 \%$ ) with jadeite pseudomorphs ( $\mathrm{P}=22-25$ kbar; T: $440 \pm 30^{\circ} \mathrm{C}$ ) in addition to phengite, paragonite, quartz, chlorite, rutile and apatite. The last mineral assemblage is interpreted as transformation of the chloritoid + glaucophane assemblage to chloritoid + jadeite paragenesis with increasing pressure. Absence of tourmaline suggests that the chloritoid-micaschist did not interact with B-rich fluids during zero strain exhumation. ${ }^{40} \mathrm{Ar} /{ }^{39} \mathrm{Ar}$ phengite age of a pervasively sheared footwall micaschist is constrained to $100.6 \pm 1.3 \mathrm{Ma}$ and that of a chloritoid-micaschist is constrained to $91.8 \pm 1.8$ Ma suggesting exhumation during on-going subduction with a southward younging of the basal accretion and the regional metamorphism. To the south, accretionary wedge consists of blueschist and greenschist facies metabasite, marble and volcanogenic metasediment intercalation. ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ phengite dating reveals that this part of the wedge is of Middle Jurassic age partly overprinted during the Albian. Emplacement of the Middle Jurassic subduction-accretion complexes is possibly associated with obliquity of the Albian convergence.
Peak metamorphic assemblages and PT estimates of the deep-seated oceanic metamorphic sequence suggest tectonic stacking within wedge with different depths of burial. Coupling and exhumation of the distinct metamorphic slices are controlled by decompression of the wedge possibly along a retreating slab. Structurally, decompression of the wedge is evident by an extensional shear zone and the footwall micaschists with syn-kinematic albite porphyroblasts. Post-kinematic garnets with increasing grossular content and pseudomorphing minerals within the chloritoidmicaschists also support decompression model without an extra heating.

Thickening of subduction-accretionary complexes is attributed to i) significant amount of clastic sediment supply from the overriding continental domain and ii) deep level basal underplating by propagation of the décollement along a retreating slab. Underplating by basal décollement propagation and subsequent exhumation of the deep-seated subduction-accretion complexes are connected and controlled by slab rollback creating a necessary space for progressive basal accretion along the plate interface and extension of the wedge above for exhumation of the tectonically thickened metamorphic sequences. This might be the most common mechanism of the tectonic thickening and subsequent exhumation of deep-seated HP/LT subduction-accretion complexes.
To the south, the Albian-Turonian accretionary wedge structurally overlies a lowgrade volcanic arc sequence consisting of low-grade metavolcanic rocks and overlying metasedimentary succession is exposed north of the İzmir-AnkaraErzincan suture (İAES), separating Laurasia from Gondwana-derived terranes. The metavolcanic rocks mainly consist of basaltic andesite/andesite and mafic cognate xenolith-bearing rhyolite with their pyroclastic equivalents, which are interbedded with recrystallized pelagic limestone and chert. The metavolcanic rocks are stratigraphically overlain by recrystallized micritic limestone with rare volcanogenic
metaclastic rocks. Two groups can be identified based on trace and rare earth element characteristics. The first group consists of basaltic andesite/andesite (BA1) and rhyolite with abundant cognate gabbroic xenoliths. It is characterized by relative enrichment of LREE with respect to HREE. The rocks are enriched in fluid mobile LILE, and strongly depleted in Ti and P reflecting fractionation of $\mathrm{Fe}-\mathrm{Ti}$ oxides and apatite, which are found in the mafic cognate xenoliths. Abundant cognate gabbroic xenoliths and identical trace and rare earth elements compositions suggest that rhyolites and basaltic andesites/andesites (BA1) are cogenetic and felsic rocks were derived from a common mafic parental magma by fractional crystallization and accumulation processes. The second group consists only of basaltic andesites (BA2) with flat REE pattern resembling island arc tholeiites. Although enriched in LILE, this group is not depleted in Ti or P.
Geochemistry of the metavolcanic rocks indicates supra-subduction volcanism evidenced by depletion of HFSE and enrichment of LILE. The arc sequence is sandwiched between an Albian-Turonian subduction-accretionary complex representing the Laurasian active margin and an ophiolitic mélange. Absence of continent derived detritus in the arc sequence and its tectonic setting in a wide Cretaceous accretionary complex suggest that the Kösdağ Arc was intra-oceanic. This is in accordance with basaltic andesites (BA2) with island arc tholeiite REE pattern.

Zircons from two metarhyolite samples give Late Cretaceous (93.8 $\pm 1.9$ and $94.4 \pm$ $1.9 \mathrm{Ma}) \mathrm{U} / \mathrm{Pb}$ ages. Low-grade regional metamorphism of the intra-oceanic arc sequence is constrained $69.9 \pm 0.4 \mathrm{Ma}$ by ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ dating on metamorphic muscovite from a metarhyolite indicating that the arc sequence became part of a wide Tethyan Cretaceous accretionary complex by the latest Cretaceous. The youngest ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ phengite age from the overlying subduction-accretion complexes is 92 Ma confirming southward younging of an accretionary-type orogenic belt. Hence, the arc sequence represents an intra-oceanic paleo-arc that formed above the sinking Tethyan slab and finally accreted to Laurasian active continental margin. Abrupt non-collisional termination of arc volcanism was possibly associated with southward migration of the arc volcanism similar to the Izu-Bonin-Mariana arc system.

The intra-oceanic Kösdağ Arc is coeval with the obducted supra-subduction ophiolites in NW Turkey suggesting that it represents part of the presumed but missing incipient intra-oceanic arc associated with the generation of the regional supra-subduction ophiolites. Remnants of a Late Cretaceous intra-oceanic paleo-arc and supra-subduction ophiolites can be traced eastward within the Alp-Himalayan orogenic belt. This reveals that Late Cretaceous intra-oceanic subduction occurred as connected event above the sinking Tethyan slab. It resulted as arc accretion to Laurasian active margin and supra-subduction ophiolite obduction on Gondwanaderived terranes.

# LAVRASYA'NIN GÜNEY AKTİF KENARI BOYUNCA ÇARPIŞMA ÖCESİ EKLEMLENME VE YÜZEYLENME, ORTA PONTİDLER, TÜRKİYE 

## ÖZET

Orta Pontidler Alp-Himalaya orojenik kuşağı içindeki çarpışa öncesi tektonik kıtasal büyüme ile karakterize olan eklemlenmeli-tipteki bir orojenik alanı temsil eder. Bölge, Lavrasya aktif kıta kenarı ile Gondwana kökenli Kırşehir bloğu arasına sıkıştırılmış Mesozoik yaşlı dalma-batma ve eklemlenme karmaşıkları ile okyanusiçi bir yayı içerir. Dalma-batma ve eklemlenlenme karmaşıkları temel olarak Lavrasya aktif kıta kenarını temsil eden Albiyen-Turoniyen eklemlenme kamasından oluşur. Kuzey kesimlerde, bu kama Lavrasya aktif kıta kenarında çökelmiş ve takiben eklenerek başkalaşıma uğramış Alt Kretase yaşlı geniş bir denizaltı türbidit yelpazesinin raksak kesimlerini temsil eden sleyt/fillat ve metakumtaşı ardalanması ile rekristallize kireçtaşı, sodik amfibolllü metabazit (PT=7-12 kbar and $400 \pm 70$ ${ }^{\circ} \mathrm{C}$ ) ve serpantinit tektonik dilimlerini içerir. Bu istife ait metapelitik kayaçların organik malzeme Raman spektrumları (RSCM), istifin farklı metamorfik sıcaklıklara sahip metamorfik paketler içerdiğini göstermektedir. Metapelitik kayaçların çoğunluğu düşük sıcaklıklarda (ca. $330^{\circ} \mathrm{C}$ ) oluşmuş sleytler olup grafit (G) ve D2 kusur bantlarının ayrışmaması ile karakterize olur. Bu kayaçlar muhtemelen Albiyen eklemlenme kamasının ucu boyunca kazınmış ıraksak türbiditleri temsil etmektedir. Geri kalan metapelitik kayaçlar ise omuz kesiminde D2 bandı yer alan hafifçe belirginleşmiş G bantları ile karaktarize olan fillatlardır. Bu fillatların metamorfizma sıcaklıkları 370-385 ${ }^{\circ} \mathrm{C}$ olarak sınırlandırıldı. Fillitik kayaçlar başlangıç mavişist fasiyesli metabazit şeridi ile ilişkilidir ve kazınmış raksak türbiditler içinde kıymık halinde bulunmaktadır. Bu kayaçlar muhtemelen bazal sıyrılma boyunca kama-altısıvanmış kıtasal metasedimentleri ile okyanus kabuğu basaltlarını temsil eder. Bu kama-altı-sıvanmış kayaçların, kazınmıș raksak türbiditler içine tektonik olarak yerleşmesi muhtemelen kamanın tektonik olarak kalınlaşmasına ve yükselmesine neden olan dizi-dışı bindirmelerce sağlanmıştır. Fillatların $\left.{ }^{40} \mathrm{Ar}\right)^{\beta 9} \mathrm{Ar}$ fengit yaşları yaklaşık 100 Ma olup Albiyen yaşlı bir dalma-batmaya ve bölgesel yüksek basınç başkalaşımına işaret eder.
Eklenmiş kıtasal metasedimentler genişlemeli bir makaslama zonu boyunca okyanusal yüksek basınç/düşük sıcaklık metamorfik kayaçlarının üzerine gelir. Okyanusal metamorfik dizi baskın olarak tektonik olarak kalınlaşmış derin-oluşumlu eklojit ve mavişist fasiyesli metabazitleri ve mikaşistleri kapsar. Çalışma alanında, metabazitler yersel olarak granatl epidot-mavişistlerdir ( $\mathrm{PT}=17 \pm 1 \mathrm{kbar}$ and $500 \pm$ $40^{\circ} \mathrm{C}$ ). Lavsonit-mavişistler makaslama zonu boyunca bloklar halinde yüzeyler ( $\mathrm{PT}=$ $14 \pm 2$ kbar and $370-440{ }^{\circ} \mathrm{C}$ ). Bu kayaçlar muhtemelen yitimin başlangıç aşamasındaki zayıf makaslama stres rejimi ile ilişkilidir. Makaslama zonu yakınında, yüzeyleme boyunca yaygın bir biçimde makaslanan taban mikaşistleri kuvars, fengit, paragonit, klorit, rutil, ile kinematik ile eș anlı gelişen albit porfiroblastı içerir. Bu tip mikaşistler turmalinli olup, bunların gerileme tabiatları makaslama zonları boyunca yüksek akışkan akısı olduğunu akla getirir. Başkalaşımın doruk koşullarına ait mineral toplulukları, makaslama zonunun uzağındaki, yüzeyleme esnasında
yamulmaya maruz kalmamış alanları temsil eden kloritoyid-mikaşistlerde kısmen korunmuştur. Kloritoyid-mikaşistlerde üç farklı metamorfik birliktelik tanımlandı ve bunların sıcaklık-basınç koşulları Theriak-Domino tarafından hesaplanan yalancı kesitler ile içerdikleri organik malzemenin Raman spektrumları kullanılarak sınırlandırıldı: Fengit-paragonit-kuvars-klorit-rutil ve apatite ek olarak 1) granat-kloritoyid-glokofan ile lavsonit psödomorfları ( $\mathrm{P}=17.5 \pm 1 \mathrm{kbar}, \mathrm{T}: 390-450^{\circ} \mathrm{C}$ ), 2) kloritoyid ile glokofan psödomorfları ( $\mathrm{P}=16-18 \mathrm{kbar}, \mathrm{T}: 475 \pm 40^{\circ} \mathrm{C}$ ) ve 3 ) görece yüksek Mg'lu kloritoyid (\%17) ile jadeyit psödomorfları ( $\mathrm{P}=22-25 \mathrm{kbar}$; T: $440 \pm 30$ ${ }^{\circ} \mathrm{C}$ ). Üçüncü mineral topluluğu kloritoyid-glokofan birlikteliğinin yükselen basınca bağlı olarak kloritoyid-jadeyit parajenezine dönüşümü olarak yorumlandı. Turmalinin bulunmaması, kloritoyid-mikaşistlerin yamulma olmadan yüzeylerken B'ca zengin akışkanlarla etkileşmediğini işaret eder. Yaygın olarak makaslanmış bir taban mikaşistinin ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ fengit yaşı $100.6 \pm 1.3 \mathrm{Ma}$ ve bir kloritoyid-mikaşistin ki ise $91.8 \pm 1.8 \mathrm{Ma}$ olarak sınırlandırıldı. Bu yaşlar yüzeylemenin yitim devam ederken gerçekleştiğini ve bazal eklemlenme ile bölgesel metamorfizmanın güneye doğru gençleştiğini gösterir. Güney kesimlerde eklemlenme kaması mavişist ve yeşilşist fasiyesli metabazit, mermer ve volkanojenik metasediment ardalanmasından oluşur. $\left.{ }^{40} \mathrm{Ar}\right)^{39} \mathrm{Ar}$ fengit yaşları eklemlenme kamasının bu kesimlerinin Orta Jura yaşlı olduğunu ve Albiyen döneminde kısmen tekrar metamorfizmaya uğradığını göstermektedir. Orta Jura yaşlı dalma-batma ve eklemlenme karmaşıklarının tektonik yerleşmesi muhtemelen Albiyen yitiminin verev olmasıyla ilintilidir.
Derin-oluşumlu okyanusal metamorfik diziye ait metamorfizmanın doruk koşulları temsil eden mineral toplulukları ve sıcaklık-basınç hesapları, kama içerisinde farklı derinliklerde gelişen bir tektonik istiflenmeye işaret eder. Farklı metamorfik dilimlerin yüzeylemesi ve yan yana gelmesi, muhtemelen yiten okyanusal levhanın geriye göçü neticesinde kamayı dekomprese etmesi tarafindan denetlenmektedir. Yapısal olarak, kamanın dekomprese olması genişlemeli makaslama zonu ve kinematik ile eş anlı albit porfiroblastlı taban mikaşistleri ile belirgindir. Kloritoyidmikaşistler içindeki kinematik sonrası granatların grosular bileşimindeki artış ve psödomorf içindeki mineraller de ilave isı artşı olmadan, dekompresyon nedeni ile yüzeyleme modelini destekler.

Dalma-batma ve eklemlenme karmaşıklarının kalınlaşması aşağıdaki iki hususa yoruldu: 1) üzerleyen kıtasal alandan önemli miktarda sediment akışının olması ve 2) geriye göç eden yiten okyanusal levha boyunca bazal sıyrılmanın yayılarak derin seviyelerde gerçekleşen bazal kama-altı-sıvanması. Derin-oluşumlu eklemlenme karmaşıklarının bazal sıyrılma yayılmasına bağlı kama-altı-sıvanması ve takibeden yüzeylemesi birbiriyle bağlantılı olup, levha gerilemesi tarafindan kontrol edilmektedir. Levha gerilemesi, plaka arayüzeyi boyunca gelișen bazal eklemlenme için gerekli mekânı yaratırken, kama üzerinde genişlemeye neden olarak tektonik olarak kalınlaşan bu eklemlenme karmaşıklarının yüzeylemesine olanak sağlar. Bu derin-oluşumlu dalma-batma ve eklemlenme karmaşıklarının tektonik kalınlaşmasının ve takibeden yüzeylemesinin en genel mekanizması olabilir.
Albiyen-Turoniyen eklemlenme kaması, güneyde, Lavrasya'yı Gondwana-türevli kıtasal bloklardan ayıran İzmir-Ankara-Erzincan kenedinin boyunca yüzeyleyen ve düşük-dereceli metavolkantiler ile bunları stratigrafik olarak üzerleyen metasedimeter kayaçlardan oluşan bir volkanik yay istifini yapısal olarak üzerler. Metavolkanik kayaçlar baskın olarak bazaltik andezit/andezit ve mafik ksenolit içeren riyolit ile pelajik kireçtaşı ve çört arakatkılı piroklastik eşleniklerini içerir. Metavolkanitler stratigrafik olarak volkanojenik metasediment arakatklı rekristalize
mikritik kireçtaşları tarafından üzerlenir. Volkanik kayaçlar nadir toprak ve iz elementlerinin özelliklerine göre iki gruba ayrılır. Birinci grup bazaltik andezit/andezit (BA1) ile jenetik olarak ilintili yaygın gabroyik ksenolitler barındıran riyolitleri kapsar. Bu grup hafif nadir toprak elementlerinin ağır olanlara göre zenginleşmesiyle karakterize olur. Bu gruba ait kayaçlar akışkanlarca kolayca taşınan geniş iyonlu elementlerce (LILE) zenginken, $\mathrm{Fe}-\mathrm{Ti}$ oksitlerin ve apatit fraksiyonlaşmasını yansıtan Ti ve P'ca fakirleşmiştir ki bu mineraller jenetik olarak ilintili gabroik ksenolitler içinde de bulunurlar. Riyolitlerde ve andezitlerde yaygın gabroyik ksenolitlerin oluşu ve benzer nadir toprak ve iz element içerikleri, riyolitler ile bazaltik andezit/andezitlerin (BA1) jenetik olarak ilintili olduğunu ve felsik kayaçların fraksiyonel kristallenme ve birikme süreçleriyle ortak bir mafik mağmadan türevlendiğini işaret eder. İkinci grup ise sadece bazaltik andezitleri içerir (BA2). Bu kayaçların nadir toprak element desenleri düz olup ada-yayı toleyitlerine benzemektedir. LILE'lerce zenginleşmesine rağmen, bu grup Ti ve P 'ca fakirleşmemiştir.
Metavolkanik kayaçların jeokimyası, HFSE fakirleşmesi ve LILE zenginleşmesi ile belirginleşen yitim-üstü volkanizmasını işaret eder. Yay dizisi Lavrasya aktif kıta kenarını temsil eden Albiyen-Turoniyen yaşlı dalma-batma/eklemlenme karmaşığı ile ofiyolitk melanj arasında sıkıştırılmıştır. Yay dizisi içerisinde kıta türevli krıntıların bulunmayışı ve geniş bir eklemlenme karmaşığı içerisindeki tektonik konumu, bu Kösdağ Yayı'nın okyanus-içi bir yay olduğuna iaşret eder. Bu ada yayı toleyitlerine benzer düz nadir toprak element desenli bazaltik andezitlerin (BA2) varlığıyla uyumludur.

Metariyolitlerden ayrılan zirkonlar Geç Kretase ( $93.8 \pm 1.9$ and $94.4 \pm 1.9 \mathrm{Ma}$ ) U/Pb yaşı vermektedir. Okyanus-içi yay dizisinin düşük dereceli bölgesel metamorfizması bir metariyolit örneğindeki metamorfik muskovitler üzerinde yapılan ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ yaşlandırmasıyla $69.9 \pm 0.4 \mathrm{Ma}$ olarak sınırlandırıldı. Bu yaş, yay dizisinin en geç Kretase'de geniş bir Kretase yaşlı Tetis eklemlenme karmaşığıın parçası haline geldiğine işaret eder. Üzerleyen dalma-batma ve eklemlenmme karmaşıklarına ait en genç ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ fengit yaşı 92 Ma olup, eklemlenmeli orojenezin güneye doğru gençleştiğini doğrulamaktadır. Bu durumda, Geç Kretase yaşlı yay dizisi yiten Tetis okyanusal levhasının üzerinde gelişmiş ve en sonunda Lavrasya aktif kıta kenarına yamanmış bir okyanus-içi eski volkanik yayı temsil etmektedir. Mağmatizmanın çarpışma öncesi ani bir şekilde kesilmesi yay volkanizmasının Izu-Bonin-Mariana yay sistemine benzer şekilde muhtemelen güneye doğru göç etmesi ile ilintilidir.
Okyanus-içi Kösdağ Yayının, KB Türkiye'de yüzeyleyen yitim-üstü ofiyolitleri ile eș yaşlı olması, onun bölgesel yitim-üstü ofiyolit oluşumuyla ilintili varolduğu tahmin edilen fakat tanımlanamayan okyanus-içi yayın bir parçasını temsil ettiğini akla getirmektedir. Geç Kretase yaşlı okyanus-içi eski yay kalıntıları ve bunlarla ilintili yitim-üstü ofiyolitler doğuya doğru Alp-Himlaya dağkuşağı boyunca izlenebilir. Bu durum Geç Kretase okyanus-içi yitiminin yiten Tetis okyanusal levhası üzerinde birbiriyle bağlantılı olarak gerçekleştiğini gösterir. Bu yitim, Lavrasya aktif kıta kenarına yay eklemlenmesi ve Gondwana-türevli kıtasal blokların üzerine ise yitim-üstü ofiyolitlerinin bindirmesi şeklinde sonuç vermiştir.

## 1. INTRODUCTION

The term of accretionary, Pacific- or Turkic-type orogeny has been used to emphasize the pre-collisional growth of continental crust by subduction and accretion processes (Şengör et al., 1993; Şengör and Natal'in, 1996; Maruyama, 1997; Dickinson, 2008; Cawood et al., 2009). This includes addition of juvenile material to the crust in the magmatic arcs and ocean-ward tectonic growth of the associated accretionary prism (Fig. 1.1). Additionally, oceanic crustal edifices like seamounts, oceanic plateaus and intra-oceanic arcs are favored to accrete rather than recycling into the mantle due to their relatively high buoyancy (Cloos, 1993). An accretionary orogeny is best recognized by tectonically thickened subductionaccretionary complexes composed of trench turbidites, high pressure/low temperature (HP/LT) metamorphic rocks, ophiolitic fragments and mélanges. There are two modes of accretion proposed: i) frontal accretion through offscraping mainly of trench-fill turbidites and oceanic pelagic sediments as wedge shape packages (e.g. Seely et al., 1974; Karig and Sharman, 1975; Moore et al., 1982; Ujiie 1997) and ii) underplating as duplexes beneath the offscraped part of the prism (Fig. 1.1, e.g. Platt et al., 1985; Sample and Fischer, 1986; Kimura et al., 1996). These two types of accretions were also simulated by analog sandbox experiments (e.g. Gutscher et al., 1998; Malavieille, 2010). A significant portion of accretion occurs at deeper part of the accretionary wedge ( $>30 \mathrm{~km}$ ) as revealed by seismic studies on modern subduction zones (Moore et al., 1991; Ye et al., 1997; Calvert, 2004; Kimura et al., 2010). This type of deep-level underplating is evidenced by eclogite and blueschist facies metamorphic rocks of the exhumed subduction-accretionary complexes (e.g. Takasu et al., 1994; Barr et al., 1999; Agard et al., 2001; Okay et al., 2002, 2006a; Tsujimori et al., 2006a). Tectonic thickening of these deep-seated HP/LT metamorphic rocks, however, remains ambiguous due to the necessary space problem for basal accretion along the plate interface.

Accretionary orogenies have been defined in many regions including the Altaids (Şengör et al., 1993; Şengör and Natal'in, 1996), the Circum-Pacific realm (e.g.

Isozaki, 1996; Maruyama, 1997; Dickinson, 2008; Fuis et al., 2008), and the Lachlan orogeny of eastern Australia (Foster and Gray, 2000; Spaggiari et al., 2002). Their geographical distribution and age ranges suggest that they play a major role for continental crustal growth (Cawood et al., 2009). In the Alpine-Himalayan orogenic belt, however, accretionary-type continental growth is apparently rare. Accretionary complexes or mélanges in the Tethyan belt are generally regarded as suture zones separating the main continental domains (Şengör and Natal'in, 1996) and the AlpineHimalayan orogeny is interpreted as a collisional rather than accretional mountain belt.


Figure 1.1 : Main components of an accretionary-type orogeny including oceanward tectonic growth of subduction-accretionary wedge and addition of juvenile material to the crust. Modified from Barr et al. (1999).

However, in the central part of the Pontides, an Alpine-Himalayan mountain chain along northern Turkey, Mesozoic subduction-accretionary complexes crop out over large areas suggesting a major contribution of pre-collisional accretionary processes along the southern Laurasian active continental margin. Accretionary units comprise Middle Jurassic and Cretaceous subduction-accretionary complexes called as Central Pontide Supercomplex (CPS) (Okay et al., 2013). In the CPS, Cretaceous accretionary units consist of HP/LT metamorphic units of continental and oceanic origin, forearc deposits and mélanges accreted to the Laurasian active continental margin. The Cretaceous accretionary complexes include low-grade distal turbidites exposed to the north. They are underlain by oceanic crust derived deep-seated HP/LT metamorphic rocks along an extensional shear zone. Furthermore, Middle Jurassic
subduction-accretionary complexes are found within the Cretaceous accretionary wedge as tectonic slices. To the south, the CPS structurally overlies arc-related metavolcanic rocks which are thrusted over the ophiolitic rocks of the İzmir-AnkaraErzincan Suture (İAES), separating the Laurasia from Gondwana-derived continental blocks.

Without an understanding of this accretionary-type orogenic period in the Central Pontides, attempts of tectonic reconstructions within the Tethyan realm will inevitably be incomplete. Moreover, the accretionary units that are sourced from both continental and oceanic crusts provide a natural laboratory for a better understanding of accretionary wedge tectonics worldwide.

### 1.1 Regional Geology

The Pontides are a mountain chain between the Black Sea and the İzmir-AnkaraErzincan suture, which separates Laurasia form Gondwana-derived terranes, Anatolide-Taurides and Kırşehir Massif (Fig. 1.2). It represents part of the Mesozoic active continental margin of Laurasia (Okay and Nikishin, 2015; Meijers et al., 2010a) and was rifted from the Eurasian mainland by opening of the Black Sea as back-arc basin during Late Cretaceous (Okay et al., 1994; Okay and Tüysüz, 1999). The Pontides comprise three tectonic units: the Strandja Massif, the İstanbul and Sakarya zones (Okay, 1989; Fig. 1.2). Contact relations between these tectonic units are still controversial. However, the most prominent structure is the east-west striking Intra-Pontide suture zone, separating the İstanbul Zone from the Sakarya Zone (Okay and Tüysüz, 1999; Robertson and Ustaömer, 2004; Akbayram et al., 2013; Göncüoğlu et al., 2014).


Figure 1.2: Tectonic map of Turkey and surrounding regions (modified from Okay and Tüysüz, 1999). Green and pale blue colors highlight the Laurasia and Gondawana-derived terranes, respectively. CPS: Central Pontide Supercomplex, CTF: Lower Cretaceous submarine turbidite fan. North of the Black Sea, "V" marks the Albian volcanic arc modified from Nikishin et al. (2015).

İstanbul Zone is a continental fragment in the western part of the Pontides. It has a crystalline basement consisting of gneiss, amphibolite and metavolcanic rocks that were intruded by Neo-Proterozoic granitoids (Ustaömer and Rogers, 1999; Yiğitbaş et al., 2004; Ustaömer et al., 2005). The crystalline basement is unconformably overlain by an Ordovician-Carboniferous sedimentary sequence (Görür et al., 1997; Özgül, 2012). The Paleozoic sequence is unconformably covered by Triassic red beds passing into shallow and deep marine limestones. Upper Cretaceous arc related volcanic and volcanoclastic rocks are exposed in the northern part of the İstanbul Zone along the SW shores of the Black Sea. In terms of lithology, İstanbul Zone is similar to the Moesian platform to the north and Okay et al. (1994) suggested that İstanbul Zone was rifted from the Moesian Platform during opening of the Black Sea.

The Sakarya Zone forms a ribbon along the northern Turkey and consists of slivers of Devonian granitoids (Okay et al., 1996; Aysal et al., 2012; Sunal, 2012), PermoCarboniferous high-grade metamorphic and related intrusive rocks (Okay et al., 1996, 2006b; Topuz et al., 2004, 2007), and Late Triassic subduction-accretionary
complexes with eclogite and blueschist slices (Okay and Monié, 1997; Okay, 2000; Okay et al., 2002; Okay and Göncüoğlu, 2004; Pickett and Robertson, 2004; Topuz et al., 2014). Recently, Middle Jurassic subduction-accretion complexes were also reported in the Eastern Pontides suggesting episodic accretionary construction of the Pontides (Topuz et al., 2013a). This basement is unconformably overlain by a Jurassic to Lower Cretaceous transgressive sedimentary sequence (Okay, 2008).

To the South, the Pontides are separated from the Gondwana-derived terranes, the Anatolide-Taurides and the Kırșehir Massif, along the İAES. The Anatolide-Taurides is represented by HP/LT metamorphic rocks that were metamorphosed by continental subduction in Late Cretaceous (80-60 Ma, Sherlock et al., 1999; Okay, 2002; Plunder et al., 2013; Pourteau et al., 2010, 2013). The HP/LT metamorphic rocks are widely overlain by ophiolite and accretionary complexes. Metamorphic sole ages of the ophiolites concentrate between 90-93 Ma (Dilek et al., 1999; Parlak \& Delaloye, 1999; Önen \& Hall, 2000; Önen, 2003; Çelik et al., 2006). In the central Turkey, there is a large area of Upper Cretaceous metamorphic and granitic rocks, known as the Kırşehir Massif. It is composed of Late Cretaceous LP/HT metamorphic rocks and associated widespread granitoids representing an ensiallic arc (Seymen, 1981; Whitney et al., 2003; Whitney \& Hamilton, 2004; İlbeyli et al., 2004; Köksal et al., 2004; Boztuğ et al., 2009; Lefebvre et al., 2013).

### 1.2 Geology of the Central Pontides

In the Central Pontides, the İstanbul Zone is represented by a Neo-Proterozoic crystalline basement consisting of tonalitic and granitic metagranitoids with 590-560 Ma zircon crystallization ages exposed northwest of Araç (Fig. 1.3, Chen et al., 2002). The basement rocks are stratigraphically overlain by Early Ordovician to Devonian sedimentary rocks of the İstanbul Paleozoic sequence (Boztuğ, 1992; Dean et al., 2000). The Sakarya Zone consists of Permo-Carboniferous granitic rocks (Nzegge et al., 2006; Okay et al., 2015) and Upper Triassic distal turbidites with dismembered ophiolites, known as the Küre Complex, (Ustaömer and Robertson, 1993, 1994, 1997; Okay et al., 2015).


Figure 1.3: Geological map of the Central Pontides modified from Tüysüz (1990), Uğuz et al. (2002), Okay et al. $(2013,2014)$ and this study.

The Küre Complex is intruded by Middle Jurassic granitoids representing part of a major magmatic arc (Yılmaz and Boztuğ, 1986; Okay et al., 2014). The granitoids are associated with Middle Jurassic low pressure and high temperature metamorphic rocks constituting deeper levels of the magmatic arc (Okay et al., 2014).

Upper Jurassic limestones, named as the İnaltı Formation unconformably cover the basement rocks of İstanbul and Sakarya zones indicating that these tectonic zones
were juxtaposed before the Late Jurassic (Tüysüz 1999; Okay et al., 2013). The limestones of the İnaltı Formation are unconformably overlain by the Lower Cretaceous turbidites, known as the Çağlayan Formation, which forms an arc shaped sub-marine fan (Fig. 1.3). The Çağlayan Formation consists of sandstone and dark shale and comprises a large number of blocks of Upper Jurassic-Lower Cretaceous shallow marine limestone and lesser amounts of Paleozoic sandstone derived from the underlying sequences of the İstanbul Zone. Based on its nannofossil contents, the Çağlayan Formation is assigned a Barremian to Aptian age (Hippolyte et al., 2010). A Lower Cretaceous depositional age is further supported by detrital zircons (Okay et al., 2013). The Çağlayan Formation is interpreted as syn-rift deposits related to opening of the Black Sea basin (Görür, 1988; Hippolyte et al., 2010). Detrital zircons from the turbidites, however, indicate a major source area in the East European Craton and Scythian Platform north of the Black Sea implying that the Black Sea basin opened after Early Cretaceous time (Okay et al., 2013).

South of the Lower Cretaceous turbidites, metamorphic rocks crop out over a large area. These rocks were previously considered as the pre-Jurassic basement of the Central Pontides (e.g., Yılmaz and Şengör, 1985; Tüysüz, 1990; Ustaömer and Robertson, 1994, 1999; Yılmaz et al., 1997; Uğuz et al., 2002). However, recent studies have shown that this area consists of Middle Jurassic and Early Cretaceous subduction-accretionary complexes and was collectively named as Central Pontide Supercomplex (CPS, Fig. 1.3; Okay et al., 2013). Within the CPS, Cretaceous accretionary complexes comprise Albian HP/LT metamorphic rocks of both continental and oceanic origin, forearc deposits and mélanges representing the Laurasian active continental margin. An associated Albian arc is reported north of the Black Sea (Fig. 1.2, Nikishin et al., 2015). Middle Jurassic subductionaccretionary complexes are exposed as tectonically emplaced slices within the Cretaceous wedge (Okay et al., 2013; Marroni et al., 2014; this study). They form part of a Middle Jurassic subduction system along the Laurasian margin (Çelik et al., 2011; Dilek and Thy, 2006; Okay et al., 2014; Robertson et al., 2013; Topuz et al., 2013a, 2013b). The CPS is unconformably overlain by an Upper Cretaceous volcano-sedimentary sequence representing fore-arc deposits (Okay et al., 2006a, 2013; Tüysüz and Tekin, 2007). The Central Pontides have been subjected up to $40^{\circ}$

Late Cretaceous-Paleocene counter-clockwise rotation revealed by paleomagnetic studies (Meijers et al., 2010b).

The Cretaceous accretionary wedge is characterized by presence of voluminous accreted terrigenous metaclastic rocks in the north. The accreted metaclastic rocks consist of slate/phyllite and metasandstone with blocks of marble, metabasite and serpentinite, all representing the distal parts of the Lower Cretaceous turbidite fan deposited on Laurasian active continental margin and subsequently accreted (Okay et al., 2013; this study). There are two main exposures of the metaflysch units: i) the Martin Complex exposed north of Araç with a contact with Lower Cretaceous turbidites (Fig.1.3) ii) the Esenler Unit exposed further southeast of Kastamonu (area studied). While the Martin Complex is characterized by greenschist facies metamorphism, the Esenler Unit contains metabasites showing incipient blueschist facies metamorphism. ${ }^{40} \mathrm{Ar} /{ }^{39} \mathrm{Ar}$ phengite dating on the Martin Complex constrains the regional metamorphism to Albian (112-106 Ma) (Okay et al., 2013). Clastic zircon studies on the Martin Complex reveal that it represents metamorphosed equivalents of the Lower Cretaceous turbidite fan (Okay et al., 2013).

The metaclastic sequence is underlain by a metabasite-dominated HP/LT metamorphic sequence. This Domuzdağ Complex consists mainly of eclogite to blueschist facies metabasite and micaschist with serpentinites, metagabbro, metachert and marble representing deep levels of a subduction-accretionary complex (Tüysüz, 1990; Ustaömer and Robertson, 1993, 1994, 1999; Altherr et al., 2004; Okay et al., 2006a, 2013). ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ phengite ages from the Domuzdağ Complex constrained HP/LT metamorphism to Albian (114-104 Ma; Okay et al., 2006a, 2013).

To the south of the CPS, arc-related metavolcanic rocks known as Kösdağ Formation exposes as a tectonic slice along the main Tethyan İzmir-Ankara-Erzincan suture (Tüysüz, 1990, 1993). The Kösdağ Formation consists of greenschist facies mafic and felsic lavas with pyroclastic rocks that are interbedded with Upper Cretaceous pelagic limestone without terrigenous input (Tüysüz 1990, 1993; Rice et al., 2006). Although initially described as an intra-oceanic arc sequence (Tüysüz, 1990), the Kösdag Formation was later interpreted as an ensialic arc formed above the Laurasian active margin (Tüysüz, 1993; Tüysüz et al., 1995) or an island arc that was
separated from the Laurasian margin by a marginal sea formed as a back arc basin (Rice et al., 2006).

### 1.3 Purpose of Thesis

During this study, the geology of a north-south transect across the Central Pontides between Kastamonu and Tosya towns is mapped. The transect consists of two main sections (Fig. 1.3). The first section is made across the CPS between Kastamonu and Tosya towns. The aim of the section is to characterize continental and oceanic crust derived accretionary units and their contacts, determine the age and grade of the HP/LT metamorphism. The geological, petrologic and geochronological data are interpreted in terms of the tectonic growth of the wedge. A major portion of the accretionary wedge comprises deep-seated HP/LT metamorphic sequences. Another aim of the thesis is to unveil exhumation of these deep-seated HP/LT metamorphic sequences in connection with their tectonic thickening within the wedge.

The second section is located south of Tosya town, along the northern edge of the İAES where metavolcanic rocks are widely exposed. Main purpose of this section is to understand the petrogenesis of the metavolcanic rocks and unveiling transition between the CPS and the İAES with regional tectonic implications.

### 1.4 Methodology

The study based on field and laboratory data. The field studies include geological mapping, collecting structural data and sampling. During geological mapping, 1:25.000 scaled topographic maps were used. Planar and linear structural data were collected by Brunton-type compass and geographical positions by Navitech GPS. In the field studies, around 300 samples were collected along the two sections mapped between Kastamonu and Tosya. Thin sections of the samples were prepared in thin section laboratories of both İstanbul Technical University and University of Potsdam. After petrographical investigations of the thin sections, selected thin sections ( $\sim 10$ ) were polished for detailed mineral chemistry analysis. Polished thin sections were prepared at the University of Potsdam.

### 1.4.1 PT calculations

Metamorphic conditions were constrained through conventional geothermobarometric methods based on the single reactions calculated by THERMOCALC software, PT modelling through pseudosections calculated by Theriak-Domino and Raman spectra of carbonaceous material. Necessary analytical data were obtained by electron microprobe analysis (EMPA), X-ray fluorescence (XRF) and Raman microspectrometry on carbonaceous material.

Microprobe analyses of the selected samples were performed on polished thin sections by JEOL JXA8200 electron microprobe at the University of Potsdam using natural standards. Measurement conditions were 15 keV acceleration potential, 10 nA beam current and a beam diameter of 2-5 microns for counting times of 20 s . Beside the single spot analyses, element mapping of suitable minerals like garnet were also made.

### 1.4.1.1 THERMOCALC

Single reactions were calculated by THERMOCALC (3.33) software (Powell \& Holand, 1988). The program uses the thermodynamic data set of Holland \& Powell (1998). Activities of reaction were calculated by the AX program for a known mineral chemistry. Metamorphic conditions of an incipient blueschist facies metabasite are constrained by THERMOCALC software.

### 1.4.1.2 Theriak-Domino

Metamorphic conditions of two blueschists and three chloritoid-micaschists were constrained by pseudosections produced by Theriak-Domino software. It calculates equilibrium mineral assemblages of a given bulk rock composition by the Gibbs free energy minimization method (de Capitani \& Brown, 1987; de Capitani \& Petrakakis, 2010). Bulk rock chemistry was obtained by XRF analysis of the metamorphic rocks. During calculation of the pseudosections, Mn was excluded due to lack of thermodynamic data for Mn-endmembers.

### 1.4.1.3 Raman spectra of carbonaceous material (RSCM) thermometers

Raman microspectroscopy on carbonaceous material were performed at the Raman Laboratory of the Institute of Earth and Environmental Sciences at Potsdam

University using a confocal, edge filter-based spectrometer (LabRam HR 800, HORIBA Jobin-Yvon) with a Nd:YAG laser for excitation at a wavelength of 532 nm . Measurements were performed in situ on polished thin sections. Thin sections of the samples were cut orthogonal to the foliation and parallel to lineation. Carbonaceous particles below transparent minerals like quartz, albite and white mica were measured to exclude influences of polishing-related mechanical destruction of the carbonaceous material structure (Fig. 1.4; Pasteris, 1989; Beyssac et al., 2002a, 2002b; Scharf et al., 2013). 20 to 30 points were measured for each of all samples with an acquisition time of 120 s . However, some of the analyzed points were eliminated due to the contamination with the epoxy. Measured Raman spectra of the carbonaceous material were decomposed for all Raman peaks of carbon by the PeakFit (v4.12) software. A Voight function was used for fitting of the Raman bands. Fitting procedures applied to $1000-2000 \mathrm{~cm}-1$ interval of the Raman range, which contains the first-order region of the Raman spectrum (Tuinstra and Koenig, 1970; Nemanich and Solin, 1979). The first-order region (1100-1800 cm-1) hosts the main graphite band (G band, at $1580 \mathrm{~cm}-1$ ) with defects bands D1, at $1350 \mathrm{~cm}-1$ and D2, at $1620 \mathrm{~cm}-1$. Broad defect bands of D3 (at $1500 \mathrm{~cm}-1$ ) and D4 (at $1200 \mathrm{~cm}-1$ ) occur in carbonaceous material at low temperatures with very poorly ordered atomic structure (Beyssac et al., 2002a, 2002b; Sadezky et al., 2005; Lahfid et al., 2010).


Figure 1.4 : A photomicrograph in plane polarized light showing carbonaceous material (CM) within albite and quartz. $\mathrm{Ab}=$ albite, $\mathrm{Qtz}=$ quartz, $\mathrm{Phe}=$ phengite.

Raman spectra of carbonaceous material (RSCM) is a relatively new, emprically calibrated thermometer based on the graphitization of the carbonaceous material with increasing metamorphic grade (Beyssac et al., 2002a; Rahl et al., 2005; Aoya et al.,

2010; Lahfid et al., 2010). For temperature calculations two empirical RSCM calibrations were used. Calibration of Beyssac et al. (2002a) uses R2 ratio of integral intensities of the bands $[(\mathrm{D} 1 /(\mathrm{G}+\mathrm{D} 1+\mathrm{D} 2)) \mathrm{A}]$ and is applicable for a temperature range of $330-650^{\circ} \mathrm{C}$. Rahl et al. (2005) presented a modified calibration using both $\mathrm{R} 1[(\mathrm{D} 1 / \mathrm{G}) \mathrm{H}]$ and R 2 ratios extending the temperature range between $100-700^{\circ} \mathrm{C}$. Both calibrations are reported to have $\pm 50^{\circ} \mathrm{C}$ error range. Since graphitization is an irreversible process; RSCM thermometers give the peak metamorphic temperatures (Pasteris and Wopenka, 1991; Beyssac et al., 2002a).

### 1.4.2 Geochemistry

Major, trace and rare earth element composition were analyzed to investigate the petrogenesis of metavolcanic rocks of Kösdağ Formation. Bulk rock compositions were obtained by XRF. Rocks powders were prepared at the Mineral Separation Laboratory of İTU by crushing and grinding. Glassy pellets were prepared in the University of Potsdam. XRF measurements on the prepared pellets were performed at GeoForschungsZentrum, Potsdam. Trace and rare earth elements analyses of the selected rhyolitic and basaltic volcanic rocks were made by Inductively Coupled Plasma Atomic Emission Spectrometry (ICP-AES) at the University of Potsdam. Detailed trace element analyses of three samples were made at the ACME Lab, Canada.

### 1.4.3 Geochronology

### 1.4.3.1 ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ white mica geochronology

In order to constrain the age of regional metamorphism, phengite separates were dated by ${ }^{40} \mathrm{Ar} /{ }^{39} \mathrm{Ar}$ method. The dating has been conducted at the ${ }^{40} \mathrm{Ar} /{ }^{39} \mathrm{Ar}$ geochronology laboratory at the University of Potsdam. Phengite separates were prepared in the Mineral Separation Lab, ITU. After crushing and sieving the rocks, white mica of 125 to $250 \mu \mathrm{~m}$ across is enriched by the Frantz magnetic separator. Lastly, we used paper to remove the impurities within the separates.

The ${ }^{40} \mathrm{Ar}{ }^{\beta 9} \mathrm{Ar}$ analytical system of University of Potsdam consists of (1) a New Wave Gantry Dual Wave laser ablation system with a $50 \mathrm{~W} \mathrm{CO}_{2}$ laser (wavelength 10.6 micrometer) for heating samples, (2) an ultra-high vacuum purification line with SAES getters and a cold trap kept at freezing temperature of ethanol, and (3) a high-
sensitivity Micromass 5400 noble gas mass spectrometer with an electron multiplier to conduct pulse-counting analysis, which effectively works for very minor amounts of Ar gas.

Neutron activation of phengite separates was performed at the reactor of NRG Petten, Netherlands and at the OSTR, the TRIGA Reactor in the Oregon State University, Oregon, USA (samples 1214 and 1290). All the unknown samples were wrapped by commercial Al foils, then contained in $99.999 \%$ pure two Al containers with 22.7 mm diameter and 32.5 mm height. The neutron flux monitoring mineral Fish Canyon Tuff sanidine age standard, which was prepared by Geological Survey of Japan (27.5 Ma; Uto et al., 1997; Ishizuka, 1998), the crystals of $\mathrm{K}_{2} \mathrm{SO}_{4}$ and $\mathrm{CaF}_{2}$ for correction of interference by the Ar isotopes produced by nuclear reaction on K and Ca in samples, and two biotite K-Ar age standards, SORI93 biotite $(92.6 \pm 0.6$ Ma; Sudo et al., 1998) and HD-B1 biotite ( $24.21 \pm 0.32$ Ma; Hess and Lippolt, 1994) for checking accuracy of the system, were contained together for neutron irradiation. The Al container was finally wrapped by Cd foil with 1.0 mm thickness to cut off thermal neutron fluxes in order to reduce unnecessary nuclear reactions in samples. The neutron irradiation was performed for 10 hours with fast neutron flux of $1 \times 10^{13}$ $\mathrm{n} / \mathrm{cm}^{2} / \mathrm{s}$ (4 hours with fast neutron flux of $2.5 \times 10^{13} \mathrm{n} / \mathrm{cm}^{2} / \mathrm{s}$ for the samples 1214 and 1290). Samples were cooled down for one or two month at the reactor, and then were brought back to Potsdam for ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ analysis.

Stepwise heating of the phengite separates of around 0.1 mg has been conducted with a defocused continuous $\mathrm{CO}_{2}$ laser beam for 1 minute. Then, the extracted gas has been exposed to the SAES getters and a cold trap for 10 minutes to gain a pure Arsample gas. Finally, the Ar gas was admitted to the mass spectrometer to determine the Ar-isotope ratios. The isotopic ratios of each analysis have finally been obtained after corrections of blank, mass discrimination by the analysis of atmospheric argon, interference of Ar isotopes derived from Ca and K , and the decay of the Ar isotopes $\left({ }^{37} \mathrm{Ar}\right.$ and ${ }^{39} \mathrm{Ar}$ ) produced by the irradiation. Age and error calculation followed descriptions in Uto et al. (1997). Except the samples 1214 and 1290, the ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ analyses were performed 480-520 days after the neutron irradiation, thus, Ca-derived ${ }^{37} \mathrm{Ar}$ (half-life; 35.1 days) in phengite samples had almost decayed at the analysis because of its originally Ca poor chemical composition (typical $\mathrm{Ca} / \mathrm{K}$ ratio is 0.01 ) of phengites. Therefore, in age calculation, ${ }^{37} \mathrm{Ar}$ amounts assumed as each $\mathrm{Ca} / \mathrm{K}$ ratio in
the analytical result indicates around 0.01 were given to all Ar analytical results. The used ratio of production rate to convert from ${ }^{37} \mathrm{Ar}_{\mathrm{Ca}}{ }^{\beta 9} \mathrm{Ar}_{\mathrm{K}}$ to $\mathrm{Ca} / \mathrm{K}$ was 1.7 based on the Table 3-3 in McDougall and Harrison (1999). The ${ }^{40} \mathrm{Ar}{ }^{\beta 9} \mathrm{Ar}$ ages from all the steps adopted for the plateau ages in this study do not vary beyond 0.03 Ma between the cases of zero or 0.1 for $\mathrm{Ca} / \mathrm{K}$ ratios. Therefore, ${ }^{37} \mathrm{Ar}_{\mathrm{Ca}}$ amounts adopted in this study gives enough accurate ${ }^{40} \mathrm{Ar} /{ }^{39} \mathrm{Ar}$ ages. The criterior for identifying plateau ages followed Fleck et al. (1977).

### 1.4.3.2 U/Pb zircon geochronology

Magmatic crystallization age of the metavolcanic rocks of the Kösdağ Formation was constrained by $\mathrm{U} / \mathrm{Pb}$ zircon dating. The dating was performed by LA-ICP-MS (Thermo Scientific ELEMENT 2 single collector magnetic sector ICP-MS with CETAC LSX-213 G2 laser system) at the University of Potsdam.

Zircons were separated from two metarhyolite samples (1214 and 1215) by conventional methods in the Istanbul Technical University. After crushing, sieving and magnetic separation, zircons were enriched by using heavy liquid, sodium polytungstate (SPT). Finally, 63-125 $\mu \mathrm{m}$ sized zircon crystals were embedded in to epoxy and polished. Before the measurements, cathodoluminescence images of the zircon grains were taken.

The correction for laser-induced elemental fractionation and mass discrimination by the instrument was done by normalizing to the GJ1 zircon standard (Jackson et al., 2004). Data reduction was performed with the software PepiAGE (Dunkl et al., 2009) using a linear regression line fitting over time for the measured isotope ratios of the GJ1 zircon, and a t-zero intercept for the individual $\mathrm{U} / \mathrm{Pb}$ ratios of standard and sample analysis. Spot analyses of sample zircons were also corrected for common Pb contributions using the Hg -corrected ${ }^{204} \mathrm{~Pb}$ signal.

## 2. GEOLOGY

Along the Kastamonu-Tosya transect, there are two main lithological associations below the Eocene and younger sedimentary cover. The first one comprises Mesozoic subduction-accretionary complexes including HP/LT metamorphic rocks, mélanges and forearc deposits forming part of the Central Pontide Supercomplex (CPS). The HP/LT metamorphic units consist of Cretaceous low-grade HP metaflysch sequence, the Esenler Unit and a metabasite-dominated oceanic Domuzdağ Complex. Within the Cretaceous HP/LT metamorphic rocks, Middle Jurassic subduction-accretionary complexes, known as Kunduz Metamorphics, crop out as a tectonic slice. The HP/LT metamorphic rocks are unconformably overlain by a volcanoclastic unit. In the upper levels, this Kirazbaşı Complex passes into a debris flow and a mélange consisting predominately of ophiolitic blocks. In the southern part of the CPS, alkaline volcanic rocks and turbidites are exposed as a tectonic slice. This İkiçam Formation were possibly iniatially unconforbaly covering the HP/LT metamorphic rocks.

The second lithological association is an accreted arc sequence exposed north of the İzmir-Ankara-Erzincan Suture (IAES) south of Tosya. It consists predominately of metavolcanic rocks with their pyroclastic equivalents, the Kösdağ Formation and stratigraphically overlying metasedimentary rocks, the Dikmen Formation. To the south, the arc sequence overthrust an ophiolitic mélange. To the north, the arc sequence is structurally underlain the CPS. General geological features of the acceretionary units and the arc sequence are given below.

### 2.1 Mesozoic Subduction-Accretionary Complexes

In the area studied, accretionary complexes comprise Middle Jurassic and Cretaceous subduction-accretion complexes forming part of the Central Pontide Supercomplex (CPS). The Cretaceous accretionary complexes consist of HP/LT metamorphic units derived from both continental and oceanic crust, forearc basin deposits and mélanges representing the Laurasian active continental margin. Three major metamorphic units are distinguished (Figs. 2.1, 2.2).


Figure 2.1 : Geological map of the area studied across the Central Pontide Supercomplex between Kastamonu-Tosya. For the location see Figure 1.3.

In the northwest, there is a thick metaclastic sequence, the Esenler Unit, showing incipient blueschist facies metamorphism. To the south, the Esenler Unit is underlain along an extensional shear zone by a tectonically thickened metamorphic sequence dominated by metabasic rocks. This Domuzdağ Complex has undergone a highgrade blueschist facies metamorphism. In the south of the Domuzdag Complex, there is another metabasite-dominated unit, the Kunduz Metamorphics with low-grade
blueschist facies metamorphism of Middle Jurassic age. It is separated from the Domuzdağ Complex by the North Anatolian Fault.


Figure 2.2 : Geological cross sections. See Fig. 2.1 for their locations. a) A-A' section shows the initial structural relations between the Çangaldağ Complex, the

Esenler Unit and the Domuzdağ Complex. b) B-B' section shows the postmetamorphic faulting and folding between the Esenler Unit, Domuzdağ Complex and the Kunduz Metamorphics.

### 2.1.1 Esenler Unit: Accreted distal turbidites

In the NW part of the area studied, there is a thick metasedimentary sequence, named as the Esenler Unit (Fig. 2.1). The Esenler Unit is a metamorphosed flysch sequence composed predominantly of slate/phyllite and metasandstone intercalation with marble olistoliths and tectonic slices of Na -amphibole bearing metabasite and serpentinite. Metasandstones are mostly of greywacke type and occur as sheared pods within the slates (Fig. 2.3a). Close to the contacts, however, slates transform into phyllites (Fig. 2.3b). The nature of this slate to phyllite transition is not clear. Debris flow levels consisting of clast to boulder size recrystallized limestone blocks occur within the metaflysch sequence (Fig. 2.3c). These gravity-driven recrystallized limestone blocks can reach up to 200 m across. Barkurt et al. (1990) describe Lower Cretaceous fossils from these limestones. Some of the marble blocks show a fibrous fabric suggesting that aragonite was the stable phase (Brady et al., 2004; Seaton et al., 2009). Most of the metabasites show an incipient blueschist facies metamorphism and consist of sodic- and calcic-amphibole, sodic-pyroxene, epidote, chlorite, albite and titanite with augite relicts. Sheared serpentinite slices are found along shear zones together with phyllites (Fig. 2.3d).

North of the study area, the Esenler Unit is overthrust by low-grade metavolcanic rocks, which possibly form part of the Çangaldağ Complex, a Middle Jurassic volcanic arc (Okay et al., 2014). To the south, the Esenler Unit is separated from the

Domuzdağ Complex by an extensional shear zone, partly overprinted by a postEocene strike slip fault, called as Akçataş Fault (Figs. 2.1, 2.2).


Figure 2.3 : Field photos of the Esenler Unit. a) Sheared and deformed slate and metasandstone intercalation. b) Phyllite exposed close to the contact c) A debris flow level consisting of various sized marble olistoliths. d) A serpentinite slice within the phyllite along the main Kastamonu-Tosya road.

In terms of lithology, grade and metamorphic age, the Esenler Unit resembles the metamorphosed Lower Cretaceous turbidites, the Martin Complex, exposed NW of Araç (Fig. 1.2). However, while the metabasites of Esenler Unit show an incipient blueschist facies metamorphism, in the Martin Complex the metamorphism is in greenschist facies. The Esenler Unit possibly represents distal parts of the Lower Cretaceous turbidite fan deposited on Laurasian active continental margin that was subsequently accreted and metamorphosed.

### 2.1.2 Domuzdağ Complex: Oceanic metabasalt and metasediments

The Domuzdağ Complex consists of micaschist (about $40 \%$ of the sequence) metabasite ( $40 \%$ ) and serpentinite (5\%) with minor amounts of metachert, marble and metagabbro (Fig. 2.4). Metabasites are mostly well-foliated epidote-blueschists locally with garnet. The common mineral assemblage in the blueschist metabasites is
sodic amphibole + epidote + chlorite + titanite + albite $\pm$ garnet. Retrogression to albite-chlorite fels is common with relicts of sodic-amphibole.

In terms of geochemistry, metabasites of the Domuzdağ Complex predominantly exhibit MORB (Mid-Ocean Ridge Basalt) and within-plate basalt characteristics (Ustaömer and Robertson, 1999). Dominant oceanic crustal lithologies, geochemistry of the metabasites, and blueschist to eclogite facies HP/LT metamorphism indicate that the Domuzdağ Complex was formed along an active margin and represents a subduction-accretionary complex (Tüysüz, 1990; Ustaömer and Robertson, 1994, 1999; Okay et al., 2006a, 2013).


Figure 2.4 : Field photos of the Domuzdağ Complex. a) Micaschist with its typical dark grey outcrop color. b) Well-foliated epidote-blueschist.

In the area studied, Domuzdağ Complex is overlain by the Esenler Unit along an extensional shear zone. To the south of the extensional shear zone, the footwall micaschists consist of quartz, phengite, paragonite, chlorite, rutile, apatite, tourmaline and calcite with syn-kinematic albite porphyroblast with rotational inclusion fabrics. They are characterized by absence of peak metamorphic assemblages and occurrence of syn-kinematic albite porphyroblasts suggesting that they were formed by pervasive shearing during exhumation. Peak metamorphic assemblages are partly preserved in the chloritoid-micaschists farther away from the shear zone representing zero strain domains of exhumation. Three peak metamorphic assemblages are identified: 1) garnet + chloritoid + glaucophane with pseudomorphs after lawsonite 2) chloritoid with pseudomorphs after glaucophane 3) chloritoid with pseudomorphs after jadeite in addition to phengite, paragonite, quartz, chlorite, rutile and apatite.

### 2.1.2.1 Extensional shear zone

An extensional shear zone of ca. 600 m in thickness forms the main boundary between continental and oceanic accretionary units. It is characterized by mixing of blocks from the hanging- and foot-wall within a highly sheared brittle fault cataclasite (Fig. 2.5).


Figure 2.5 : Field photos of the extensional shear zone. a) General view of the shear zone with blocks of (from S to N ) marbles and a retrogressed metabasite within a grey cataclasite. b) A micaschist block within the cataclasite with NW sense of shear (pencil // lineation).

Marble, serpentinite, metabasite, and micaschist blocks are found along the extensional shear zone. Metabasites consist of fresh lawsonite-blueschist and retrogressed albite-chlorite fels with relicts of aligned Na -amphibole, which resemble the metabasites of the Domuzdağ Complex. Lawsonite-blueschists predominantly consist of lawsonite and sodic-amphibole with additional sodic-pyroxene, chlorite, phengite, epidote, albite and titanite. Absence of lawsonite relicts in the albitechlorite fels suggests that mixing of distinct metabasites occurred late in the exhumation. NW dipping stretching lineations are observed within or around the shear zone, especially in the footwall micaschists, which are characterized by the syn-kinematic albite porphyroblasts. Blocks within the fault gouge consistently indicate top to NW sense of shear (Fig. 2.5b). The shear zone was responsible for the final exhumation of deeply buried metabasites and micaschists of the Domuzdağ Complex to shallow levels.

### 2.1.3 Kunduz Metamorphics

To the south, the Domuzdağ Complex is in contact with a metabasite-marble dominated unit along the North Anatolian Fault. The unit, which forms the western extension of the Kunduz Metamorphics of Tüysüz (1990), consists of metabasite, volcanogenic metasandstone and metatuff intercalated with pale marble horizons and minor metachert (Fig. 2.6). It is easily identified by its multicolored exposures in the field.


Figure 2.6 : Field photos of the Kunduz Metamorphics. a) General view of the multicolored exposures of marble and metabasite/metatuff intercalation. b) A closer view of the same outcrop. Marbles form strained pods within the metatuff. c) Thinly bedded sodic-amphibole-bearing metatuff and marble intercalation. d) Light colored metatuff, metachert and metabasite interbeds.

The common mineral paragenesis in the metabasites is calcic-amphibole, epidote, chlorite, albite and titanite locally with sodic-amphibole. Primary sedimentary structures of the volcanogenic metasediments like bedding are locally preserved. Marble forms sheared pods within relatively weak sodic-amphibole bearing metatuff.

In the area studied, the Kunduz Metamorphics are structurally divided into two parts by a thrust fault (Figs. 2.1, 2.2). The structurally upper part shows higher degree of
metamorphism and deformation, which are evidenced by well-foliated metabasites locally with sodic amphibole. In the lower part, however, metabasites are weakly foliated and do not have any high-pressure index minerals, and magmatic textures and mafic minerals are mostly retained. In the lower part, metamorphism is characterized by growth of tiny calcic amphibole, epidote and chlorite.

### 2.1.4 Kirazbaşı Complex: Forearc basin and mélanges

The CPS is widely overlain by an Upper Cretaceous volcano-sedimentary sequence. Most of the contacts of the volcano-sedimentary unit, called the Kirazbaşı Complex, with the metamorphic rocks are faulted. However, at a few localities a stratigraphic contact is preserved (Yiğitbaş et al., 1990, Okay et al., 2006a). In these localities, the Kirazbaşı Complex starts with Turonian pelagic limestones, which pass up into siliciclastic turbidites overlain by mass flows and ophiolitic mélange, consisting of basalt, radiolarian chert, serpentinite, pelagic shale and limestone. Tüysüz and Tekin (2007) describe Cretaceous (Barremian to Cenomanian) radiolaria from the chert blocks within the Kirazbaşı Complex. In the area studied, there are two main exposures of this unit. Around Akkaya village, reddish hemipelagic siltstone-shale alternation is overlain by slightly sheared greywacke-shale alternation with pillow basalt and chert blocks (Fig. 2.7a, b and c). Around Bürnük, a clastic sedimentary sequence consisting of sandstone and siltstone is overlain by a mélange consisting of serpentinite, radiolarian chert, basalt, diabase and grey limestone blocks. There are also sheared debris flows consisting mainly of blocks of sandstone and siltstone with serpentinite and limestone (Fig. 2.7d). The Kirazbaşı Complex is interpreted as a forearc basin developed in front of southward verging thrust slices in the accretionary prism.


Figure 2.7 : Field photos of the Kirazbaşı Complex. a) Sheared slate-greywacke type
mélange. b) Hemipelagic volcanoclastic sedimentary rocks. c) Discoidal shape pillow lavas. d) Sheared sedimentary mélange/debris flow with blocks of sandstone, siltstone, limestone and serpentinite.

### 2.1.5 İkiçam Formation: Alkaline volcanic rocks and turbidites

In the southern part of the area studied, alkaline volcanic rocks and volcanoclastic turbidites are widely exposed in tectonic contact with the Kunduz Metamorphics (Fig. 2.1). The volcanic sequence consists predominantly of pyroclastic rocks interbedded with volcanogenic sandstone and siltstone (Fig. 2.8a, b). Some of the basalts contains large leucite crystals characteristic of alkaline volcanism (Fig. 2.8c). The contact relation between the pyroclastic rocks and leucite-bearing basalts is not clear. Syenitic dykes, probably form part of the İkiçam Formation, are common and crosscut the metamorphic rocks of the Kunduz Metamorphics. The volcanic
sequence is associated with dark turbidites possibly representing distal and deeper of the basin (Fig. 2.8d).


Figure 2.8 : Field photos of the İkiçam Formation. (a) \& (b) Pyroclastic rocks of andesitic composition. c) Leucite-bearing basalt. d) Volcanoclastic dark grey turbidites. This part mainly consists of sandstone and siltstone.

The volcanic sequence is reported to be of Campanian-Maastrichtian age based on the planktonic foraminifera content (per. com. in Rice et al., 2006). This is supported by recent ${ }^{40} \mathrm{Ar} /{ }^{39} \mathrm{Ar}$ biotite age data on the alkaline volcanic rocks of $73.6 \pm 0.18$ and $76.78 \pm 0.19 \mathrm{Ma}$ (Genç et al., 2013, 2014). Although, the volcanoclastic rocks were probably initially deposited on the Kunduz Metamorphics, they now form a steeply dipping tectonic slice within the complex (Fig. 2.1, 2.2). Initial limited geochemical data from the volcanic rocks suggested a within-plate setting without any subduction influence (Rice et al., 2006), more recent data, however, indicate subduction-related volcanism (Genç et al., 2014; Gülmez et al., 2014).

### 2.2 Accreted Arc Sequence

In the Kastamonu-Tosya transect, a coherent arc sequence is exposed south of the Tosya Basin along the İAES (Figs. 1.3, 2.9). It is structurally overlain by the CPS to
the north and overthrusts an ophiolitic mélange in the south. Two main units are identified in the arc sequence: 1) a low-grade metavolcanic unit, the Kösdağ Formation, consisting predominately of basaltic andesitic and felsic rocks together with the associated pyroclastic rocks and 2 ) an overlying low grade metasedimentary sequence consisting of micritic limestone and shale, the Dikmen Formation.


Figure 2.9 : Geological map of the studied section south of Tosya, along the İzmir-Ankara-Erzincan suture with a cross section between A and A'. For location see Fig. 1.3.

### 2.2.1 Kösdağ Formation

The Kösdağ Formation consists mainly of greenschist facies rhyolitic and basaltic andesitic rocks with their pyroclastic equivalents with a structural thickness of ca. 2 km . Rhyolites and basaltic andesites are intercalated and are easily identified by their
whitish and greenish colors, respectively (Fig. 2.10). Towards the top, the metavolcanic rocks become pyroclastic and finer grained. Recrystallized red pelagic limestone and chert are interbedded with the pyroclastic rocks indicating submarine deposition (Fig. 2.10). The Kösdağ Formation is stratigraphically overlain by recrystallized limestone and slate of the Dikmen Formation (see section 2.2.2 below).

The base of the Kösdağ Formation is not exposed in the area studied.


Figure 2.10 : Field photos of the Kösdağ Formation. All rocks have undergone a low-grade metamorphism but are referred here with their protolith names. a) Basaltic andesite and rhyolite. b) Thin chert interbeds within the acidic and basic rocks. Pyroclastic rocks of acidic (c) and andesitic (d) compositions. e) Red pelagic limestone/chert interbeds with basaltic andesitic rocks. f) A close look to the same outcrop in (e). Folded bright cherts occur within the red pelagic limestone.

### 2.2.2 Dikmen Formation

The Dikmen Formation is a metasedimentary sequence consisting of yellowish and pinkish siliceous slate and recrystallized micritic limestone with light colored volcanogenic metasandstone layers (Fig. 2.11). The recrystallized limestones are thinly bedded, pink and contain deformed radiolaria indicating deposition in a deep marine environment. The majority of the metacarbonates exhibit alternation of dark relatively thickly bedded (up to 50 cm ) limestone representing calciturbidites and thin pinkish pelagic limestone (Fig. 2.11c, d). Cenomanian fossils are reported from the recrystallized limestone of the Dikmen Formation (Tüysüz, 1990, 1993), however, our paleontological samples were too recrystallized to yield any fossils.


Figure 2.11 : Field photos of the Dikmen Formation. a) Contact between rhyolite and overlying recrystallized reddish micritic limestone of Dikmen Formation in the upper limb of the overturned anticline. b) Yellowish and reddish slate and recrystallized limestone exposed along the transional section between Kösdağ and Dikmen formations. c \& d) Recrystallized calciturbidite consisting intercalation of moderately bedded dark levels consisting of transported coarse grains and thinly bedded pinkish pelagic levels.

The Dikmen Formation stratigraphically overlies the volcanic rocks of the Kösdağ Formation. The contact is transitional involving alternation of volcanic and
sedimentary beds although it is overturned. A typical section of the contact is exposed along a ca. 250 m section on the main road to the Yukarıdikmen village. In the section, mostly acidic volcanic rocks with minor pinkish micritic limestone pass into a limestone and shale dominated sedimentary sequence with volcanogenic sandstone (Fig. 2.12).


Figure 2.12 : Measured stratigraphic section illustrating the transitional character between the Kösdağ Formation and the overlying Dikmen Formation. The section is about 250 m and is measured along the main road to Yukaridikmen village. For location, see Fig. 2.9.

Sedimentary rocks of the Dikmen Formation also overlie the rhyolitic rocks in the northern part of the area studied. This suggests a structure that is characterized by a large southward overturned anticline, whereby the metasedimentary sequence lies both over and under the metavolcanic rocks (Fig. 2.9). The overturned folding probably formed during the southward thrusting of the Kösdağ and Dikmen formations over the ophiolitic mélange.

### 2.2.3 Ophiolitic mélange (İAES)

The southernmost unit of the Kastamonu-Tosya transects is an ophiolitic mélange overthrust by the accreted arc sequence (Fig. 2.13a). The ophiolitic mélange forms part of the İzmir-Ankara-Erzincan suture and consists of tectonic slices and blocks of serpentinite, pillow basalts and radiolarian cherts with red pelagic and grey neritic limestones without a clearly defined matrix (Fig. 2.13b). Compared to the Kirazbaşı Complex exposed in the north, the ophiolitic mélange is characterized by absence of continent derived detritus. The dominant ophiolitic character of the İAES is reported elsewhere in central Turkey (Dilek and Thy, 2006; Çelik et al., 2011; Dangerfield et al., 2011; Rojay, 2013, Sarifakioğlu et al., 2014).


Figure 2.13 : Field photos of the ophiolitic mélange. a) Serpentinite (green) that is overthrusted by recrystallized limestone (pinkish) of Dikmen Formation. b) Pillow basalts.

Along the contact with the overlying arc sequence, NW dipping foliation is observed in the ophiolitic mélange, which is otherwise generally free of metamorphic minerals. To the south, the ophiolitic mélange is unconformably overlain by sedimentary rocks of the Çankırı Basin, ca. 4 km thick basin of mainly Tertiary age (Kaymakcı, 2000).

### 2.3 Cover Units

In the area studied, the oldest cover unit is Maastrichtian limestones exposed as a tectonic sliver along the east-west striking post-Eocene Akçataş fault. It is separated from the Domuzdağ Complex by the Akçataş Fault and is thrust over the Eocene turbidites suggesting a positive flower structure. The limestone is reddish and contains abundant Helonocyclina and Orbitoides type benthic foraminifera indicating a Maastrichtian deposition age (Fig. 2.14, Ercan Özcan, per. com.). Absence of metamorphism and significant deformation suggest that the Maastrichtian limestone initially unconformably covered the accretionary wedge units.


Figure 2.14 : Maastrichtian limestone with Orbitoides (a) and Helonocyclina (b) type foraminifera.

In the Central Pontides, Eocene limestones crop out widely and unconformably overlie the metamorphic rocks. Benthic foraminifera studies, particularly orthophragminids, on Eocene sedimentary succession reveals an Early-Middle Eocene age (KASB section near Kastamonu and TAŞ section near Taşköprü, Özcan et al., 2007). In the area studied, Eocene sedimentary sequence consists of both limestone and turbidites with abundant benthic Eocene foraminifera including Nummulites and Alveolina. In the northern part of the area studies, Eocene sequence consists of alternation of white and yellowish sandy limestones (Fig 2.15a). To the south, the sandy limestones pass to a turbidite sequence. The turbidite sequence consists of thin to moderately bedded sand/siltstone and shale/marl alternation (Fig 2.15b). Similar turbidites are also exposed west of the studied area along the Akçataş Fault. To the east, north of the Güney town, the Eocene sequence is represented by
moderate to thickly bedded sparitic limestone (Fig. 2.1). The Eocene sedimentary sequence generally shows subhorizontal bedding.

The Eocene sedimentary succession and the underlying HP/LT metamorphic rocks are covered by Eocene basalts (Fig. 2.15c). Basalt exhibits columnar structures and vesicular textures indicating subaerial cooling (Fig 2.15d). There is no radiometric data on the basalt but it is reported to be Eocene in age (e.g. Uğuz et al., 2002). The fact that the age of the underlying sedimentary sequence is Early to Middle Eocene, a post-Lutetian age can be inferred for the basalt.


Figure 2.15 : Field photos of the Lower-Middle Eocene limestone (a) and turbidite (b). c) Eocene limestone and overlaying basalt cover. d) Vesicular basalts of probable Eocene age.

In the NW of the studied area, horizontally bedded Neogene clastic rocks crop out on top of hills and unconformably cover the Eocene succession and the post- Lutetian basalts. They consist of weakly bedded and unconsolidated fluvial conglomerate and sandstone. Around Tosya, however, Neogene sedimentary rocks were originated by activity of the North Anatolian Fault (NAF). The rocks comprise unconsolidated thick succession of conglomerates, gravelstone and sandstone that fill the NAF controlled Tosya basin as alluvial fans (Fig. 2.16). In the north, they unconformably
cover the metamorphic rocks of the Kunduz Metamorphics with slightly inclined bedding planes towards to the south. In some localities, unconsolidated Neogene sequence is faulted by the activity of the NAF.


Figure 2.16 : a) A general view of the Tosya Basin between the Kunduz Metamorphics to the north and hills made up from the metavolcanic rocks of the Kösdağ Formation to the south. b \& c) Field photos of the NAF-related unconsolidated Neogene gravelstone and sandstone of the Tosya Basin.

### 2.4 Structural Analysis of the Wedge

In the Central Pontides, general regional structures of the accretionary units trend in SW-NE direction. In the study area, this trend is represented by the extensional shear zone separating the Esenler Unit from the underlying Domuzdağ Complex. Similar contact relations are also observed in the NW of the study area where the Domuzdağ Complex is overlain by the Çangaldağ Complex (Okay et al., 2006a). Orientation of foliation planes of the Esenler Unit varies depending on their proximity to different contacts (Fig. 2.17a). In the north of the study area, Esenler Unit is overthrust by the metavolcanic rocks of the Çangaldağ Complex. Around the thrust, strong stretching lineation developed in the metaflysch and the metavolcanics trending SW-NE and slightly plunging to NE (Fig. 2.17b). Lineations with similar trends are common throughout the metaflysch suggesting a general transport in this direction.


Figure 2.17 : Lower hemisphere, equal area projections of the foliations and the lineations. a) Foliation and b) lineation data from the Esenler Unit. c) Foliation and d) lineation measurements of the northern part of the Domuzdağ Complex. Lineations are mostly orthogonal to the NW dipping foliation planes and plunge to the NW related exhumation. e) Foliation data from the southern part of the Domuzdağ Complex and f) the Kunduz Metamorphics. Contour interval is 10 .

To the south of the extensional shear zone, foliation planes of the Domuzdağ Complex strike SW-NE parallel to the extensional shear zone (Fig. 2.17c). Micaschist exhibit NW plunging stretching lineation orthogonal to the foliation (Fig.
2.17d). Similar lineations are also observed in the blocks and fault gouge of the extensional shear zone. This suggests that they developed by pervasive shearing during exhumation. An east-west striking post-Eocene fault, the Akçataş Fault (Fig. $2.1,2.2 b$ ), overprints the extensional shear zone and separates the Domuzdağ Complex into two sectors. South of the Akçataş fault, foliation planes define large synforms (Fig. 2.17e) bounded possibly by faults parallel to the Akçataş Fault. Such a folding pattern is also observed within the Kunduz Metamorphics (Fig. 2.17f). In the south of the Tosya Basin, the Kösdağ and Dikmen formations show consistent NW dipping foliations and structurally underlie the Central Pontide Supercomplex (Fig. 2.18).


Kösdağ Formation - Foliations
Figure 2.18 : Lower hemisphere, equal area projections of the foliations of the Kösdağ Formation. Contour interval is 10.

## 3. PETROGRAPHY AND MINERAL CHEMISTRY

This chapter provides detailed petrography and mineral chemistry of the metamorphic and magmatic units in the area studied. The metamorphic units comprise 1) the low-grade metaflysch sequence, the Esenler Unit, 2) underlying HP/LT metabasite dominated the Domuzdağ Complex and the Kunduz Metamorphics. Finally, petrography of the low-grade arc sequence, the Kösdağ Formation is given.

### 3.1 The Esenler Unit

The Esenler Unit is a metamorphosed flysch sequence composed predominantly of slate/phyllite and metasandstone intercalation with marble olistoliths and tectonic slices of Na-amphibole bearing metabasite and serpentinite. The majority of the metapelitic rocks of the Esenler Unit are weakly recrystallized slates. The slates consist of quartz, phengite, chlorite and albite. They exhibit continuous foliation defined by fine-grained white mica (Fig. 3.1a). Detrital quartz is partly retained in the slate (Fig. 3.1b). Close to the contacts, slate transformed to phyllites that consist of larger grain of quartz, phengite, chlorite and albite. The phyllites show spaced foliation defined by white mica and quartz (Fig. 3.1c, d).

Metasandstones are of greywacke type composed 70-75 \% of clasts and 25-30 \% of cement. The clasts consist of quartz, plagioclase, and white mica which are embedded by silt to clay sized cement (Fig. 3.2). The cement of the metasandstone is recrystallized and consists mainly of quartz and minor mica. Texturally, the metasandstones are poorly to moderately sorted. The clasts are moderately rounded and generally have irregular outlines.


Figure 3.1 : Representative photomicrographs of the metapelitic rocks from the Esenler Unit (plane polarized light). a \& b) Fine-grained slates (samples 847 and 853). Foliation is defined by white mica and partly recrystallized detrital quartz. c \& d) Coarser grained phyllites with spaced foliation (samples 31 and 879). $\mathrm{Ab}=\mathrm{albite}$, Qtz=quartz, Phe=phengite, $\mathrm{Cc}=$ calcite .


Figure 3.2 : Photomicrographs of a greywacke-type metasandstone from the Esenler Unit under plane (left) and cross (right) polarized light. Majority of the clasts are quartz (Qtz) with minor feldspar ( Pl ) and white mica (Ms).

Some of the marble blocks within the Esenler Unit show a fibrous fabric possibly representing pseudomorphs after aragonite (Fig. 3.3, Brady et al., 2004; Seaton et al., 2009).


Figure 3.3: A marble with elongated fibrous calcite.
The majority of the metabasites of the Esenler Unit show incipient blueschist facies metamorphic assemblages (Fig. 3.4). The metabasite consists of sodic- and calcicamphibole, chlorite, epidote, albite, titanite locally with pumpellyite. The metabasites are usually well foliated defined by oriented sodic-amphibole (Fig. 3.4a).


Figure 3.4 : Plane-polarized photomicrographs of incipient blueschist facies metabasites from the Esenler Unit. a) A foliated blueschist. Foliation is defined by sodic-amphibole (Na-amph). b) An unfoliated metabasite with randomly distributed sodic-amphiboles (sample 211B). Cpx= augite relicts; $\mathrm{Ep}=$ epidote; $\mathrm{Chl}=$ chlorite; $\mathrm{Ab}=$ albite.

One of these metabasite samples was analyzed by the electron microprobe. Mineral chemistry data is given in Appx (Tables A.1-A.10). This sample 211B is an unfoliated metabasite with augite relicts (Fig. 3.4b). It is taken from NW of the area studied where plenty of tectonic slices of serpentinites are also found. The metamorphic mineral assemblage of the sample 211B is sodic- and calcic-amphibole, sodic-pyroxene, epidote, chlorite, albite and titanite. Sodic-amphibole forms
randomly distributed prismatic crystals and is of crossite composition (Fig. 3.5). Crossite is rimmed by actinolite. Sodic-pyroxene overgrows and replaces magmatic augite with a compositional range of $\mathrm{Aeg}_{29-40} \mathrm{Jd}_{6-13} \mathrm{Di}_{53-58}$ (Fig. 3.6). Texturally, there are two types of epidote. The first one forms large grains with iron-rich cores ( $\mathrm{Fe} 3+/$ $(\mathrm{Fe} 3++\mathrm{Al})=0.36)($ Fig. 3.7). The Fe3+\# ratio decreases rimward to $0.23-0.24$. The second type forms small grains with pistacite-poor cores $(\mathrm{Fe} 3+/(\mathrm{Fe} 3++\mathrm{Al})=0.14-$ $0.17)$ and $\mathrm{Fe}-$ rich rims $(\mathrm{Fe} 3+/(\mathrm{Fe} 3++\mathrm{Al})=0.29-0.31)$. The $\mathrm{Mg} /(\mathrm{Mg}+\mathrm{Fe})$ ratio of chlorites ranges between $0.47-0.57$ (Fig. 3.8). Si content of the chlorite increases at the expense of Al while $\mathrm{Mg} \#$ increases. The stable Ti-bearing phase is titanite. Plagioclase is almost pure albite ( $\mathrm{An}=0-1 \%$ ).


Figure 3.5 : Compositional range of amphibole. Act= actinolite, Gln= glaucophane.


Figure 3.6: Compositional range of sodic-pyroxene.


Figure 3.7 : Compositional range of epidotes.


Figure 3.8: Compositional range of chlorites.

### 3.2 The Domuzdağ Complex

The Domuzdağ Complex mainly consists of HP/LT metabasites and micaschists with minor serpentinite, metagabbro, metachert, and marble. It represents a subductionaccretionary complex of oceanic origin. Below detailed petrography and mineral chemistry of metabasites and micaschists are given. The analytical data are shown in Appx (Tables A.1-A.10).

### 3.2.1 Metabasites

The metabasites of the Domuzdağ Complex are mostly well-foliated epidoteblueschist consisting of sodic amphibole + epidote + chlorite + titanite + albite $\pm$ garnet (Fig. 3.9a). Augite relicts are retained in some of the epidote-blueschist. The blueschist locally show strong greenschist facies metamorphic overprint and are transformed into albite-chlorite fels (Fig. 3.9b). The HP/LT origin of this albitechlorite fels is evidenced by relicts of aligned Na -amphibole within the albite domain. Fresh lawsonite-blueschists crop out within the extensional shear zone as blocks. They consist of lawsonite, sodic-amphibole, sodic-pyroxene, chlorite, phengite, epidote, albite and titanite and possibly form part of the Domuzdağ Complex. Detailed petrography and mineral chemistry of a garnet-epidote-blueschist (702A) and a lawsonite-blueschist is provided below.


Figure 3.9 : Plane-polarized photomicrographs from the Domuzdağ Complex of a) epidote-blueschist and b) albite-chlorite fels that represents a strongly retrogressed blueschist. Na -amph $=$ sodic-amphibole, $\mathrm{Chl}=$ chlorite, $\mathrm{Ab}=$ albite, $\mathrm{Ep}=$ epidote, $\mathrm{Cc}=$ calcite.

### 3.2.1.1 Lawsonite-blueschist (sample 128)

The sample 128 is an unfoliated metabasite consisting of sodic-amphibole, lawsonite, sodic-pyroxene, chlorite, phengite, epidote, albite and titanite (Fig. 3.10). Sodicamphibole has crossitic cores rimmed by glaucophane (Fig. 3.5). Glaucophane also forms single grains and is found as inclusion in lawsonite. Sodic-amphibole is partly replaced by albite and chlorite. Lawsonite is partly decomposed to epidote and albite along its rims.


Figure 3.10 : Microphotograph of a lawsonite blueschist (sample 128) from the Domuzdağ Complex. Na-amph= sodic-amphibole, Na-cpx=sodic-pyroxene, Lws= lawsonite, $\mathrm{Ttn}=$ titanite.

Sodic-pyroxene shows two compositional ranges: $\mathrm{Jd}_{4-18} \mathrm{Aeg}_{31-23} \mathrm{Di}_{65-59}$ and $\mathrm{Jd}_{25}$ ${ }_{36} \mathrm{Aeg}_{32-28} \mathrm{Di}_{43-37}$ (Fig. 3.6). Epidote has pistacite-poor cores ( $\mathrm{Fe}^{3+} \#: 0.02-0.09$ ) and $\mathrm{Fe}^{3+}$ rich rims ( $\mathrm{Fe}^{3+} \#: 0.23-0.33$ )(Fig. 3.7). Chlorite replaces the Na-clinopyroxene and has $\mathrm{Mg} /(\mathrm{Mg}+\mathrm{Fe})$ ratio ranging between $0.53-0.54$ (Fig. 3.8). Plagioclase is albite ( $\mathrm{An}=0-5 \%$ ) and Ti-bearing phase is titanite. Si content of phengites ranges between 3.41-3.54 cation per formula unit (c. p.f.u.)(Fig. 3.11).


Figure 3.11 : Compositional range of phengites. Al vs Si diagram showing the celadonite exchange. Decreasing Si content mainly depicts the effect of the retrogression.

### 3.2.1.2 Garnet-epidote-blueschist (sample 702A)

The sample 702A consists of garnet, sodic- and calcic-amphibole, epidote, chlorite, albite, phengite, titanite and magmatic augite relicts (Fig. 3.12a). Garnets grow over the foliation defined by glaucophane and epidote. They are $0.2-0.4 \mathrm{~mm}$ across and comprise glaucophane, epidote, and titanite inclusions aligned parallel to the main foliation. Generation of pressure shadows defined by glaucophane, epidote and phengite suggests a syn-kinematic origin (Fig. 3.12a). Compositional range of the garnet is $\mathrm{Alm}_{44-56} \mathrm{Py}_{2-4} \mathrm{Grs}_{39-38} \mathrm{Sps}_{15-2}$ (Fig. 3.13). From core to rim, almandine increases at the expense of spessartine but the sum of these components remains stable with little change in the grossular and pyrope contents. This can also be seen in element maps showing a typical garnet growth zoning evidenced by progressive exchange of Fe and Mn (Fig. 3.12b). The progressive garnet zonation is cut by an augite relict indicating that garnet did not grow in that direction. This suggests that there was no reaction between these two minerals.


Figure 3.12 : a) BSE image of the garnet-blueschist (702A). Garnet (Grt) exhibits pressure shadow consisting of glaucophane (Gln) + phengite (Phe) +epidote (Ep). Calcic-amphibole (Ca-amph), chlorite (Chl) and albite ( Ab ) are secondary and replace glaucophane. b) Element mapping of the garnet in Fig. 3.9a with typical growth zoning. Ca and Mg are almost stable. Mn decreases rimward while Fe increases. Ttn=titanite, Aug=augite.

Sodic-amphibole is mostly ferroglaucophane (Fig. 3.5) and replaces augite relicts along their rims or forms aligned prismatic crystals that define the foliation. Sodicamphibole is partly replaced by calcic-amphibole and secondary chlorite. Epidote has pistacite-poor core $\left(\mathrm{Fe}^{3+} \#=0.06-0.12\right)$ and pistacite-rich rim $\left(\mathrm{Fe}^{3+} \#=0.18-0.22\right)$ (Fig. 3.7). Si contents of phengite range between $3.41-3.53$ c.p.f.u. (Fig. 3.11).
Within the garnet pressure shadows, Si content of phengite decreases rimward (3.503.44 c.p.f.u). $\mathrm{Mg} /(\mathrm{Mg}+\mathrm{Fe})$ ratio of secondary chlorite ranges between $0.48-0.56$ (Fig. 3.8). The stable Ti-bearing phase is titanite. Plagioclase is albite ( $\mathrm{An}=1-6 \%$ ).


Figure 3.13 : Compositional range of garnets.

### 3.2.2 Micaschists

Micaschists of the Domuzdağ Complex consists of quartz, phengite, paragonite, albite, chlorite, rutile, apatite $\pm$ chloritoid $\pm$ garnet $\pm$ glaucophane $\pm$ tourmaline $\pm$ calcite. In the chloritoid-micaschist, there are also abundant pseudomorphs after glaucophane, lawsonite and jadeite. The micaschists exhibit two distinct types of retrogression that was mainly controlled by strain during exhumation. The first types of the micaschists are mainly exposed southeast of the extensional shear zone that separates the Domuzdağ Complex from the overlying Esenler Unit. These micaschists consist of quartz, phengite, paragonite, albite, chlorite, calcite, rutile, apatite and tourmaline (Fig. 3.14). Albite is found as syn-kinematic porphyroblasts with rotated inclusion fabrics. Brown tourmaline is a common accessory phase in this type of micaschists and is found as oriented parallel to foliation or as inclusion in albite porphyroblasts. Calcite is also found as a secondary mineral.


Figure 3.14 : Microphotograph of a retrogressed micaschists from the Domuzdağ Complex with syn-kinematic albite porphyroblasts under plane (left) and cross (right) polarized light. $\mathrm{Qtz}=$ quartz, $\mathrm{Phe}=$ phengite, $\mathrm{Chl}=$ chlorite, $\mathrm{Ab}=$ albite.

The second type constitutes chloritoid-micaschists that are exposed farther away from the shear zone (Fig. 3.15a, b). In these chloritoid-micaschists, retrogression is characterized by the pseudomorphing of the peak metamorphic mineral assemblages (Fig. 3.15ac, d). Three peak metamorphic assemblages are identified: 1) garnet-chloritoid-glaucophane with pseudomorphs after lawsonite 2) chloritoid with pseudomorphs after glaucophane and 3) relatively high-Mg chloritoid with pseudomorphs after jadeite in addition to phengite, paragonite, quartz, chlorite, rutile and apatite. Tourmaline or calcite is not found in the chloritoid-micaschists. Albite occurs as secondary tiny aggregates. Below, detailed petrography and mineral chemistry of three chloritoid-micaschists as well as a micaschist with syn-kinematic albite porphyroblasts are described.


Figure 3.15 : Photomicrographs of the chloritoid-micaschists from the Domuzdağ Complex (plane polarized light). a) Sample 775, a garnet-chloritoid micaschist. b) Sample 753A, a chloritoid-micaschist. c) Syn-kinematic pseudomorphs after lawsonite (sample 775). Former lawsonite was replaced by epidote (Ep), paragonite $(\mathrm{Pa})$ and quartz (Qtz). The syn-kinematic character of the pseudomorphs suggests that they were stable during the peak metamorphic conditions. d) Pseudomorphs after glaucophane (sample 753A). They are replaced by chlorite (Chl), phengite (Phe), and quartz (Qtz). Cld=chloritoid, Grt= garnet.

### 3.2.2.1 Garnet-chloritoid-micaschist (sample 775)

Sample 775 consists of garnet, chloritoid, glaucophane, phengite, quartz, chlorite, paragonite, albite, rutile and apatite with pseudomorphs after lawsonite (Fig. 3.15a, c). Foliation is determined by phengitic mica and chlorite. Garnet forms tiny euhedral crystals ( $<0.3 \mathrm{~mm}$ ) and contain quartz inclusions (Fig. 3.16). Absence of any pressure shadows and rotational inclusion fabric indicates a post-kinematic origin for the garnets. Chloritoids form large porphyroblasts consisting of radially grown crystals (Fig. 3.16). Foliation bends around them. Most of the glaucophanes are replaced by chlorite + phengite + quartz (Fig. 3.16) but relicts are preserved (Fig. 3.17). Rectangular epidote + paragonite + quartz pseudomorphs are found parallel to the foliation or as rotated crystals suggesting a syn-kinematic growth (Fig. 3.15c). They are interpreted as pseudomorphs after lawsonite. Paragonite is replaced by lowSi phengites. Albite is secondary usually with secondary chlorites. It also replaces glaucophane along the rims.


Figure 3.16 : BSE images of the sample 775. Epidote $(\mathrm{Ep})+$ paragonite $(\mathrm{Pa})+$ quartz (Qtz) replaced former lawsonite and chlorite (Chl) + phengite (Phe) + quartz (Qtz) replaced former glaucophane. Albite (Ab) is secondary. Quartz (Qtz) inclusions in garnet (Grt) do not show any rotational fabric. $\mathrm{Cld}=$ chloritoid.


Figure 3.17 : A photomicrograph showing the partly preserved glaucophane (Gln) relicts in the sample 775. The majority of the glaucophanes are replaced by chlorite, phengite and quartz (see Fig 3.16) $\mathrm{Qtz}=$ quartz, Phe= phengite.

Garnets of the sample 775 have two distinct compositional ranges with Alm 28 ${ }_{39} \mathrm{Py}_{1} \mathrm{Grs}_{7-10} \mathrm{Spss}_{63-49}$ in the core and $\mathrm{Alm}_{34} \mathrm{Py}_{1} \mathrm{Grs}_{16} \mathrm{Spss}_{49}$ in the outermost rim (Fig. 3.13). In the core, $\mathrm{X}_{\mathrm{Mn}}$ decreases rimward, while $\mathrm{X}_{\mathrm{Fe}}$ increases proportionally. $\mathrm{X}_{\mathrm{Ca}}$ slightly increases and is stabilized around $10 \%$. The outermost rim is characterized by a dramatic increase of $\mathrm{X}_{\mathrm{Ca}}$ to $16 \%$ and decrease of $\mathrm{X}_{\mathrm{Mn}}$. Element mapping shows that while $\mathrm{X}_{\mathrm{Ca}}$ continues to increase within the outermost rim, $\mathrm{X}_{\mathrm{Fe}}$ slightly decreases (Fig. 3.18).


Figure 3.18 : Element mapping of the post-kinematic garnet in sample 775. Dark areas within the garnet are quartz inclusions. Notice the Ca-rich outermost rim possibly depicting increasing rate of lawsonite breakdown.

Compositional range of the chloritoid is $\mathrm{Fe}_{81-83} \mathrm{Mg}_{7-10} \mathrm{Mn}_{11-8}$ (Fig. 3.19). From core to rim, Mn decreases while Fe and Mg increases. Relicts of the glaucophane in sample 775 are almost free of ferric iron component and plot on the glaucophaneferroglaucophane boundary in a Miyashiro diagram (Fig. 3.5).


Figure 3.19 : Compositional range of the chloritoids. (a) Ternary compositional diagram. While in sample 178, main exchange occur between Fe and Mg , in the samples 775 and 753A Mg mainly substitute with Mn . (b) $\mathrm{X}_{\mathrm{Mg}}$ vs $\mathrm{X}_{\mathrm{Fe}}$ diagram.

Si content of the phengite ranges between 3.50-3.04 (Fig. 3.11). Mg\# number of the phengite increases with increasing Si content. $\mathrm{Mg} /(\mathrm{Mg}+\mathrm{Fe})$ number of matrix chlorite is 0.32-0.35 (Fig. 3.8). Secondary chlorites have higher Mg\# numbers (0.40$0.41)$ and their Si content increases at the expense of $\mathrm{Al} . \mathrm{Fe}^{3+} /\left(\mathrm{Fe}^{3+}+\mathrm{Al}\right)$ ratio of the epidotes that replace lawsonite ranges between $7-11 \%$ (Fig. 3.7). Plagioclase is almost pure albite.

### 3.2.2.2 Chloritoid-micaschist (sample 753A)

Sample 753A consists of chloritoid, phengite, quartz, chlorite, paragonite, albite, allanite, rutile, and apatite (Fig. 3.15b). Chloritoid forms large porphyroblasts of radially grown crystals. There are chlorite + phengite + quartz pseudomorphs interpreted as former glaucophane (Fig. 3.15d, 3.20). Chlorite, which was once in equilibrium with glaucophane, penetrates into glaucophane as thin stripes during retrogression (Fig. 3.20). From rim to core, grain size of the pseudomorphing minerals markedly decreases indicative for increasing nucleation rate. Allanite is replaced by epidote at their rim.


Figure 3.20 : BSE images of the sample 753A. Chlorite (Chl) + phengite (Phe) + quartz (Qtz) replaced former glaucophane. The matrix chlorite penetrates into former
glaucophane as secondary thin stripes. Paragonite also possibly replaced former glaucophane along its rims. Apa= apatite.

In the sample 753A, compositional range of the chloritoids is $\mathrm{Fe}_{81-83} \mathrm{Mg}_{8-11} \mathrm{Mn}_{6-12}$ (Fig. 3.19). Fe component slightly increases rimward and then becomes relatively stable. Mg slightly increases and the Mn decreases rimward. Si content of phengite ranges between 3.47-3.23 c.p.f.u (Fig. 3.11). Mg\# number increases with increasing Si content. Mg\# number of matrix chlorites ranges between 0.34-0.39 and that of secondary chlorites between $0.39-0.47$ (Fig. 3.8). The secondary chlorites are also characterized by increasing of Si content at the expense of $\mathrm{Al} . \mathrm{Fe}^{3+} \#$ ratio of epidote that replaces allanite ranges between 14-18\% (Fig. 3.7). Plagioclase is almost pure albite.

### 3.2.2.3 Chloritoid-micaschist (sample 178)

Sample 178 consists of chloritoid, phengite, quartz, chlorite, albite, rutile and apatite (Fig. 3.21). Chloritoids are found as aggregates or single crystals parallel to the foliation. They are replaced by both paragonite and chlorite (Fig. 3.21b). There are chlorite + quartz + phengite pseudomorphs within the quartz-rich domains (Fig. 3.21 b , c). They have chloritoid-like prismatic outlines and are interpreted as
pseudomorphs after chloritoid. In sample 178, albite is found as tiny grains within rectangular pseudomorphs interpreted as former jadeite (Fig. 3.21a). Around the pseudomorph after jadeite foliation bends and tiny phengite inclusions show rotational fabric indicating a syn-kinematic origin. Paragonite is partly replaced by low-Si phengites.

In sample 178, three generations of chloritoid are identified. Cld1 is relatively rich in Mn and low in Mg with compositional range of $\mathrm{Fe}_{87-84} \mathrm{Mg}_{10-14} \mathrm{Mn}_{3-2}$ (Fig. 3.19). Cld2 is marked by its elevated Mg content with compositional range of $\mathrm{Fe}_{82-81} \mathrm{Mg}_{16-17} \mathrm{Mn}_{2}$. Finally, the high- Mg Cld2 is replaced by low- Mg and high-Mn chloritoid (Cld3) along the rims. Cld 3 are identical in composition with $\mathrm{Cld} 1\left(\mathrm{Fe}_{87} \mathrm{Mg}_{10} \mathrm{Mn}_{3}\right)$ but show inverse zoning suggesting a retrogressive origin.

Si content of phengite ranges between 3.45-3.10 c.p.f.u (Fig. 3.11). Mg\# number increases with increasing Si content. In sample 178, chlorite is a secondary phase replacing the high-Mg chloritoid (Cld2). Mg\# number of the chlorite ranges between 0.32-0.35 (Fig. 3.8). Plagioclase is almost pure albite.


Figure 3.21 : BSE images of the sample 178. a). Chloritoid (Cld) aggregates together with a rectangular pseudomorph at the center of image composed of albite (Ab). Tiny inclusions of phengites within the pseudomorph exhibit rotational fabrics. Foliation also slightly bends around the pseudomorphs indicating its syn-kinematic origin. It is interpreted as former jadeite. b) A pseudomorph after chloritoid (Cld) within the quartz rich domain. It is replaced by chlorite (Chl) + quartz (Qtz) + phengite (Phe). c) The same pseudomorph in "b" under microscope (plane polarized light). $\mathrm{Pa}=$ paragonite.

### 3.2.2.4 Micaschist with albite porphyroblasts (sample 392)

Sample 392 consists of phengite, paragonite, chlorite, quartz, albite, rutile, apatite and tourmaline. Albite forms syn-kinematic porphyroblasts with rotated phengite, paragonite, chlorite and rutile inclusions. Si content of phengites ranges between 3.48-3.33 c.p.f.u (Fig. 3.11). Although they are indistinguishable in terms of texture, in the Si vs Al diagram, phengites with relatively low Si plot above the celadonite exchange trend. A similar separation is visible with the iron contents of the phengites. Fe forms two clusters with similar $\mathrm{Mg} \#$ ratio around 0.2 and 0.1 c.p.f.u for the high- and low-Si phengites, respectively. While calculating mineral formula, iron is normally treated as ferrous. Recalculation as ferric iron resets the low-Si phengites on the muscovite-celadonite exchange line. Thus, an effect of oxygen fugacity might be inferred. Within the high-Si phengites, Si decreases from core to rim (3.44-3.41 c.p.f.u.) forming rims of low-Si phengite. Albites contain of low-Si phengite (3.353.39 Si c.p.f.u) inclusions and are mostly surrounded by similar phengites. Paragonite is replaced by low-Si phengites along the rims. $\mathrm{Mg} /(\mathrm{Mg}+\mathrm{Fe})$ ratio of chlorite ranges between $0.35-0.38$ (Fig. 3.8). There is no significant Al and Si exchange in the chlorite.

### 3.3 The Kunduz Metamorphics

The Kunduz Metamorphics consist mainly of metabasites and marble with volcanogenic metasandstone, phyllite/metatuff and metachert. Structurally, it is separated into two sectors (Fig. 2.1). In the upper part, the metabasites are mostly greenschists consisting of calcic-amphibole, chlorite, albite, epidote, phengite and titanite sometimes with augite relicts (Fig. 3.22a). There are also sodic-amphibole bearing metabasites preserved locally. These well-foliated blueschists consist of sodic amphibole, albite, epidote, chlorite, phengite, titanite (Fig. 3.22b). Foliation is defined by the sodic-amphibole. Sodic-amphibole also occurs in the metatuff interbedded with the marbles. The metatuff consists of sodic-amphibole, chlorite, epidote, albite, calcite and titanite. Phyllites form interbeds within the metavolcanic rocks and consist of quartz, phengite, albite, bluish-green amphibole, chlorite and stilpnomelane with magmatic augite relicts (Fig. 3.22c). In the lower part of the Kunduz Metamorphics, the magmatic texture and mineralogy of the metabasites are mostly retained. They are weakly metamorphosed basaltic rocks and consist of
pyroxene and altered plagioclase (Fig. 3.22d). Metamorphism is defined by secondary chlorite and albite.


Figure 3.22 : Microphotographs from the Kunduz Metamorphics. a) A greenschist consisting of calcic-amphibole (Ca-amph), chlorite (Chl), albite (Ab) and titanite (Ttn). b) A blueschist consisting of sodic amphibole (Na-amph), epidote (Ep), albite (Ab) and chlorite (Chl). c) a phyllitic metatuff consisting of phengite (Phe), quartz (Qtz), calcic-amphibole (Ca-amph), chlorite (Chl), stilpnomelane (Stp) with relict clinopyroxene ( Cpx ). d) A weakly metamorphosed metabasite form the lower part of the Kunduz Metamorphics. $\mathrm{Cpx}=$ magmatic pyroxene, $\mathrm{Chl}=$ chlorite, $\mathrm{Al}=$ albite .

Mineral chemistry of a quartz-micaschist from the Kunduz Metamorphics is investigated (Tables A.1-A.10). The sample is taken from the upper part of the Kunduz Metamorphics and is interlayered with light colored marble and metabasite. This sample 626A consists of quartz, phengitic white mica, epidote, hematite and titanite. Phengites define a weakly developed foliation. Si content of the phengite ranges between 3.37 and 3.44 c.p.f.u (Fig. 3.11). $\mathrm{Fe}^{3+} /\left(\mathrm{Fe}^{3+}+\mathrm{Al}\right)$ ratio of the epidote is $0.28-0.30$ (Fig. 3.7).

### 3.4 Kösdağ Formation

The Kösdağ Formation consists mainly of low-grade mafic and felsic volcanic rocks together with pyroclastic rocks. Red recrystallized pelagic limestone and chert are interbedded within the metavolcanic rocks. The mafic rocks are basaltic andesite and andesite consisting mainly of clinopyroxene and plagioclase phenocrysts. Felsic rocks are dacitic to rhyolitic in composition and consist of quartz and feldspar phenocrysts. Below detailed petrography of the rhyolitic and basaltic andesitic/andesitic rocks are given. Their major, trace and rare earth element compositions are provided in the geochemistry chapter (Ch. 7).

### 3.4.1 Felsic volcanic rocks

Felsic rocks of the Kösdağ Formation mainly consist of plagioclase and locally of quartz and K-feldspar phenocrysts set on a recrystallized and foliated groundmass of feldspar, quartz, chlorite, muscovite and epidote (Fig. 3.23a, b). Besides the single phenocrysts, there are also glomeroporphyritic aggregates consisting of plagioclase, clinopyroxene, opaque ( $\mathrm{Fe}-\mathrm{Ti}$ oxides) and apatite interpreted as cognate mafic xenoliths (Fig. 3.23c, d). They were possibly formed earlier in the magma chamber during differentiation of the parental magma. In more evolved samples with a higher amount of quartz phenocrysts, the xenoliths are absent or consist only of plagioclase clusters sometimes with interstitial clinopyroxene (Fig. 3.23e-h). In the xenoliths, plagioclase and clinopyroxene usually have sharp boundaries with the surrounding groundmass suggesting they are in equilibrium.


Figure 3.23 : Microphotographs (plane polarized in the left side and cross polarized in the right) of the acidic rocks of the Kösdağ Formation. a \& b) $\mathrm{SiO}_{2}$ saturated metarhyolite with quartz ( Qtz ) and feldspar ( Pl ) phenocrysts. c \& d) A cognate mafic xenolith within a metarhyolite (1213). It consists of plagioclase ( Pl ), clinopyroxene, $\mathrm{Fe}-\mathrm{Ti}$ oxide ( Ilm ), and apatite. e \& f) A cognate xenolith with cumulate-like plagioclases $(\mathrm{Pl})$ with interstitial clinopyroxene $(\mathrm{Cpx})$ in a metarhyolite. Apatite (Apa) and Fe -Ti oxides also occur in the xenolith. g \& h) A cumulate-like plagioclase cluster without mafic minerals within a metarhyolite (1304B).

### 3.4.2 Basaltic andesite/andesite

Basaltic andesite/andesites consist of plagioclase and clinopyroxene. Plagioclase forms large phenocrysts or microliths in the groundmass. Clinopyroxene is free of optical zonation. Mafic cognate xenoliths are not found in basaltic andesite though glomeroporphyritic texture occurs (Fig. 3.24a-d). In an andesite (sample 1286), however, cognate xenoliths consisting of clinopyroxene, plagioclase and apatite are observed (Fig. 3.24e, f). Apatite is also found as monomineralic aggregates in the same sample 1286. In the glomeroporphyritic aggregates, replacement by secondary calcite is common.


Figure 3.24 : Microphotographs (plane polarized in the left side and cross polarized in the right) of the basaltic andesite/andesite of the Kösdağ Formation. a \&b) Glomeroporphyritic cluster within basaltic andesites, sample $1287(\mathrm{a} \& \mathrm{~b})$ and sample 1288 (c\&d). e \& f) A cognate xenolith consisting of clinopyroxene, plagioclase and apatite in an andesite (1286). Apatite is also found as inclusion within the clinopyroxene. g \& h) Metabasite with augite relicts. Ca-amphibole replaces the augite along their rims.

Metamorphic mineral assemblages in the metabasaltic-andesites consist of chlorite, epidote, albite and minor actinolite replacing clinopyroxene indicative for greenschist facies metamorphism $(3.24 \mathrm{~g}, \mathrm{~h})$. The glassy groundmass of the metarhyolites is completely recrystallized and consists of tiny quartz, albite, white mica, chlorite and epidote grains. The preservation of monomineralic quartz phenocrysts locally with undoluse extinction in the metarhyolites (Fig. 3.23a, b) indicates that the temperatures during the metamorphism were less than $300^{\circ} \mathrm{C}$ (Stipp et al., 2002).

## 4. RAMAN MICROSPECTROSCOPY OF CARBONACEOUS MATERIAL

The majority of the pelitic rocks contain primary organic matter consisting of a mixture of $\mathrm{C}, \mathrm{H}, \mathrm{O}, \mathrm{N}$ and S known as kerogen (e.g. Tissot and Welte, 1978). With increasing metamorphic temperature, the weakly-ordered atomic structure of the kerogens transform into well-organized graphite called as graphitization (Landis, 1971; Diessel et al., 1978; Itaya, 1981; Beyssac et al., 2002b). Raman microspectrometry is possibly the most efficient method for revealing the degree of graphitization of the carbonaceous material (Pasteris and Wopenka, 1991; Wopenka and Pasteris, 1993; Yui et al., 1996; Beyssac et al., 2002a, 2002b). Recently, calibrations of Raman spectra of carbonaceous material (RSCM) allow deriving precise temperature constraints on metapelitic rocks (Beyssac et al., 2002b; Rahl et al., 2005; Aoya et al., 2010; Lahfid et al., 2010). It is particularly efficient in lowgrade metamorphism where index minerals are generally absent for conventional geothermobarometic methods (Beyssac et al., 2004, 2007).

Raman microspectroscopy on carbonaceous material was performed on metapelitic rocks of the Esenler Unit and the Domuzdağ Complex to constrain their peak metamorphic temperatures. The measurements were performed at the Raman Laboratory of the Institute of Earth and Environmental Sciences at Potsdam University. Details of the measurement procedure are given in the methodology part. Below, general parameters of the measured Raman spectra are given. The temperature calculations and the obtained values are discussed in the next chapter under the title of Metamorphic Conditions.

### 4.1 Esenler Unit

Raman microspectrometry was performed on carbonaceous material of 15 metapelitic rocks including both slates and phyllites. In accordance with the texture, the metapelitic rocks of the Esenler Unit exhibit variable Raman spectra indicating distinct degrees of graphitization and metamorphic temperatures.

The majority of the measured samples are slates which show G (graphite) and D2 defect bands as a single band near $1600 \mathrm{~cm}-1$ (Fig. 4.1). After decomposing, the positions of the bands are $1595 \mathrm{~cm}-1$ (G) and around $1615 \mathrm{~cm}-1$ (D2), respectively (Fig. 4.2a, Table 4.1). The D1 defect band occurs at $1345 \mathrm{~cm}-1$. The slates are also characterized by showing broad D3 and D4 defect bands with high intensities. These bands generally cause upward shift of the spectrum including G, D1 and D2 bands. The D3 band was located mostly between $1520 \mathrm{~cm}-1$ and $1530 \mathrm{~cm}-1$ and the D4 band between $1230 \mathrm{~cm}-1$ and $1245 \mathrm{~cm}-1$ (Fig. 4.2a). One sample (871) differs from the rest by its low intensities of D3 and D4 defect bands (Fig. 4.1).


Figure 4.1 : Representative Raman spectra of carbonaceous material of the measured samples from the Esenler Unit. The RSCM forms two clusters corresponding texturally to the slates and phyllites. The sample numbers are given at the right side of the each spectrum.

The phyllite is characterized by the differentiation of the graphite (G) and D2 defect peaks and absence of D3 and D4 defect bands (Fig. 4.1). The G band occurs at 1588$1590 \mathrm{~cm}-1$ and D 2 defect band is placed as a shoulder of $G$ band around 1618-1620 $\mathrm{cm}-1$ (Fig. 4.2b). The D1 band of these samples is located at ca. $1352 \mathrm{~cm}-1$.


Figure 4.2 : Representative examples of fitting of the measured Raman spectra. a) Low-temperature slates that are characterized by undifferentiating of graphite (G) and D2 defect bands and occurrence of broad D3 and D4 defect bands (sample 847).
b) Phyllite showing slightly pronounced G band with D2 occurring on its shoulder (sample 230).

Table 4.1 : Parameters of decomposed Raman spectra of carbonaceous material from the Esenler Unit

|  |  | G |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Sample | $\#$ | Position | $1 \sigma$ | FWHM | $1 \sigma$ | D1 |  |  |  |
| position | $1 \sigma$ | FWHM | $1 \sigma$ |  |  |  |  |  |  |
| 31 | 24 | 1587.46 | 0.63 | 35.88 | 1.72 | 1352.33 | 0.66 | 41.90 | 1.69 |
| 230 | 17 | 1588.08 | 1.13 | 37.57 | 1.77 | 1352.00 | 1.10 | 45.09 | 3.25 |
| 850 | 21 | 1588.82 | 0.78 | 44.04 | 0.99 | 1350.62 | 0.70 | 50.35 | 2.66 |
| 844 | 25 | 1587.98 | 1.00 | 41.24 | 1.29 | 1350.65 | 1.00 | 46.95 | 1.96 |
| 879 | 15 | 1590.24 | 0.93 | 43.30 | 2.19 | 1352.16 | 0.69 | 46.93 | 2.43 |
| 871 | 20 | 1590.15 | 1.88 | 47.73 | 2.38 | 1350.38 | 1.30 | 56.70 | 4.81 |
| 876 | 23 | 1594.27 | 2.35 | 46.48 | 5.55 | 1343.98 | 2.36 | 85.92 | 10.16 |
| 874 | 29 | 1596.56 | 0.97 | 44.73 | 1.91 | 1344.10 | 1.83 | 81.73 | 7.74 |
| 841 | 22 | 1595.57 | 1.10 | 42.98 | 1.91 | 1343.97 | 1.43 | 79.77 | 5.23 |
| 847 | 30 | 1595.31 | 1.11 | 46.30 | 3.41 | 1343.94 | 1.15 | 93.20 | 3.44 |
| 851 | 23 | 1595.04 | 1.13 | 45.84 | 4.07 | 1344.89 | 3.21 | 84.77 | 11.70 |
| 853 | 20 | 1596.01 | 1.35 | 45.36 | 2.64 | 1346.26 | 2.83 | 81.79 | 11.52 |
| 1126 | 26 | 1596.08 | 1.04 | 44.63 | 1.74 | 1346.44 | 0.80 | 77.49 | 3.90 |
| 859 | 19 | 1595.22 | 1.82 | 45.98 | 1.95 | 1346.80 | 1.48 | 82.82 | 5.59 |
| 861 | 23 | 1593.42 | 1.18 | 50.71 | 0.94 | 1345.82 | 0.66 | 94.46 | 2.30 |

\# is the number of the measured spectra used in calculations. $1 \sigma=$ standard deviation.

### 4.2 Domuzdağ Complex

Raman microspectroscopy on carbonaceous material was performed on four micaschists samples from the Domuzdağ Complex. The samples include three chloritoid-micaschists (775, 753A and 178) and a retrogressed micaschist with synkinematic albite porphyroblasts (392) exposed southeast of the extensional shear zone.

All of the measured samples show clear G band occurring at ca. $1580 \mathrm{~cm}-1$ in the Raman range (Table 4.2, Figure 4.3). The D1 defect band located at ca. $1552 \mathrm{~cm}-1$ and D2 defect band at ca. $1620 \mathrm{~cm}-1$. In the chloritoid-micaschists, D2 defect band placed on the shoulder of $G$ band whereas in the sample 392, it is slightly differentiated from G band and appear as a separate peak. Intensities of the D1 and D2 bands is variable. Samples 775 and 178 have D1 and D2 band with higher intensities. Sample 753A and 392, however, is characterized by low intensity defect bands.

Table 4.2 : Parameters of decomposed Raman spectra of carbonaceous material from the Domuzdağ Complex.

| Sample | $\#$ | G <br> Position | $1 \sigma$ | G <br> FWHM | $1 \sigma$ | D1 <br> position | $1 \sigma$ | D1 |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| FWHM | $1 \sigma$ |  |  |  |  |  |  |  |  |  |
| 775 | 33 | 1583.72 | 0.97 | 30.99 | 2.71 | 1352.62 | 0.73 | 43.44 | 1.26 |  |
| 178 | 32 | 1582.16 | 1.03 | 28.06 | 3.71 | 1351.97 | 1.08 | 41.53 | 2.57 |  |
| 753A | 30 | 1583.38 | 0.84 | 25.34 | 3.51 | 1354.16 | 0.84 | 45.65 | 3.46 |  |
| 392 | 25 | 1581.17 | 0.64 | 21.73 | 1.61 | 1353.38 | 1.39 | 45.26 | 1.25 |  |

\# is the number of the measured spectra used in calculations. $1 \sigma=$ standard deviation.


Figure 4.3 : Representative RSCM of the metapelitic rocks of the Domuzdağ Complex.

### 4.3 Concluding Remarks

Raman microspectroscopy is performed on carbonaceous material of the metapelitic rocks of both the Esenler Unit and tectonically underlying the Domuzdağ Complex. In the low-grade Esenler Unit, Raman spectra of carbonaceous material of the metapelitic rocks show two overlapping groups described as slates and phyllites. The first group (slates) is characterized by 1) occurrences of broad D3 and D4 defect bands 2) appearance of G and D2 defect bands as a single band. The second group (phyllites), however, does not show broad D3 and D4 defect bands. In this group, G and D 2 bands are slightly differentiated.

The Domuzdağ Complex is the deep-seated part of the subduction-accretionary wedge and undergone a higher grade blueschist facies metamorphism than the Esenler Unit. This is also confirmed by the RSCM of the micaschist showing a higher degree of graphitization. This is evidenced by the existence of a clear $G$ band and low intensities of the defect bands in the measured Raman spectra.

## 5. METAMORPHIC CONDITIONS

This chapter presents metamorphic constraints on the HP/LT metamorphism of the subduction-accretionary complexes exposed in the Central Pontides including Esenler Unit and Domuzdağ Complex. The Esenler Unit is a low-grade metaflysch sequence consisting of slate/phyllite and metasandstone intercalation with blocks of marble, sodic-amphibole bearing metabasite and serpentinites. The Domuzdağ Complex consists mainly of deep-seated HP/LT metabasite and micaschists and tectonically underlies the Esenler Unit along an extensional shear zone. Metamorphic conditions of an incipient blueschist facies metabasite lens within the Esenler Unit were constrained by reactions calculated by THERMOCALC (3.33) software (Powell and Holand, 1988). The program uses the thermodynamic data set of Holland and Powell (1998). Activities of the reactions were calculated from the mineral chemistries by the AX program (https://www.esc.cam.ac.uk/research/research-groups/holland/ax). Peak metamorphic temperatures of the metapelitic rocks of the Esenler Unit were estimated by their Raman spectra of carbonaceous material.

Peak metamorphic conditions of a lawsonite- and a garnet-blueschist as well as three chloritoid-micaschists from the Domuzdağ Complex were modelled by pseudosections. The pseudosections were produced by Theriak-Domino software. The program calculates equilibrium mineral assemblages of a given bulk rock composition by the Gibbs free energy minimization method (de Capitani \& Brown, 1987; de Capitani \& Petrakakis, 2010). In addition to the pseudosections, peak metamorphic temperatures of the chloritoid-micaschists were also obtained by their Raman spectra of carbonaceous material.

### 5.1 Esenler Unit

Below PT constraints of an incipient blueschist facies metabasite lens within the Esenler Unit are given. Peak metamorphic temperatures of the slate and the phyllites of the metaflysch sequence are also constrained using two distinct RSCM
thermometers that were calibrated by Beyssac et al. (2002a) and by Rahl et al. (2005).

### 5.1.1 Metabasite within the Esenler Unit

The sample 211B consists of sodic- and calcic-amphibole, sodic-pyroxene, epidote, chlorite, albite and titanite with augite relicts. Presences of albite within the sample 211B give an upper limit for the metamorphic pressure. A lower limit of pressure is constrained by the reaction (reaction 1 in Figure 5.1);

$$
\mathrm{Daph}_{0.012}+\mathrm{Ab}+\text { Fact }_{0.0045}=\mathrm{Fgl}_{0.0132}+\mathrm{Cz}_{0.53}+\mathrm{Qtz}+\mathrm{H}_{2} \mathrm{O}(1)
$$

A possible lower temperature limit is constrained by absence of lawsonite by the following reactions;

$$
\begin{gather*}
\mathrm{Fgl}_{0.0132}+\mathrm{Lws}=\mathrm{Ab}+\mathrm{Qtz}+\mathrm{Daph}_{0.012}+\mathrm{Cz}_{0.53}+\mathrm{H}_{2} \mathrm{O}  \tag{2}\\
\mathrm{Gl}_{0.0061}+\mathrm{Law}=\mathrm{Cz}_{0.53}+\mathrm{Ab}+\mathrm{Qtz}+\mathrm{Ames}_{0.021}+\mathrm{H}_{2} \mathrm{O}(3)
\end{gather*}
$$

Metamorphic conditions of the sample 211 B were constrained as $400 \pm 70{ }^{\circ} \mathrm{C}$ temperatures and 7-12 kbar pressures (Fig. 5.1).


Figure 5.1 : PT conditions of an incipient blueschist facies metabasite within the Esenler Unit. Stabilities of the amphiboles are from Otsuki and Banno (1990) and a petrogenetic grid in NCMASH system for low-grade metabasite is from Schiffman and Day (1999).

### 5.1.2 Peak metamorphic temperatures of the slate/phyllites

The metapelitic rocks of the Esenler Unit exhibit distinct Raman spectra of carbonaceous material indicating distinct degrees of graphitization and peak metamorphic temperatures. Two groups are identified which texturally correspond to slates and phyllites, respectively. Rahl et al. (2005) calibration constrains peak metamorphic temperatures of slates to ca. $330{ }^{\circ} \mathrm{C}$ and of phyllites to $365-380{ }^{\circ} \mathrm{C}$ (Table 5.1). Beyssac et al. (2002a)'s calibration estimates ca. $345^{\circ} \mathrm{C}$ temperature for the slates and $355-370^{\circ} \mathrm{C}$ temperature for the phyllites. A comparison diagram of the temperatures obtained by two calibrations is shown in Figure 5.2.

Table 5.1 : Raman parameters and the calculated peak metamorphic temperatures for the Esenler Unit.

| Sample | $\#$ | R 1 | $1 \sigma$ | R 2 | $1 \sigma$ | $\mathrm{~T}^{\mathrm{B}}\left({ }^{\circ} \mathrm{C}\right)$ | CI | $1 \sigma$ | $\mathrm{~T}^{\mathrm{R}}\left({ }^{\circ} \mathrm{C}\right)$ | CI | $1 \sigma$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 31 | 24 | 1.48 | 0.24 | 0.61 | 0.03 | 371 | 4 | 11 | 385 | 5 | 12 |
| 230 | 17 | 1.47 | 0.16 | 0.61 | 0.02 | 369 | 3 | 7 | 381 | 4 | 9 |
| 850 | 21 | 1.77 | 0.18 | 0.64 | 0.02 | 357 | 3 | 7 | 370 | 5 | 12 |
| 844 | 25 | 1.80 | 0.10 | 0.64 | 0.01 | 355 | 2 | 4 | 367 | 3 | 8 |
| 879 | 15 | 1.91 | 0.10 | 0.64 | 0.02 | 355 | 4 | 7 | 369 | 8 | 15 |
| 871 | 20 | 2.27 | 0.19 | 0.67 | 0.01 | 344 | 2 | 5 | 334 | 6 | 13 |
| 876 | 23 | 1.66 | 0.14 | 0.67 | 0.01 | 342 | 2 | 4 | 330 | 4 | 9 |
| 874 | 29 | 1.78 | 0.19 | 0.67 | 0.01 | 341 | 1 | 3 | 331 | 3 | 8 |
| 841 | 22 | 1.71 | 0.14 | 0.67 | 0.01 | 342 | 1 | 3 | 332 | 2 | 5 |
| 847 | 30 | 1.53 | 0.11 | 0.66 | 0.01 | 345 | 1 | 4 | 330 | 2 | 6 |
| 851 | 23 | 1.64 | 0.22 | 0.67 | 0.01 | 343 | 2 | 4 | 328 | 4 | 11 |
| 853 | 20 | 1.79 | 0.26 | 0.67 | 0.01 | 341 | 2 | 5 | 328 | 4 | 9 |
| 1126 | 26 | 1.83 | 0.15 | 0.68 | 0.01 | 340 | 1 | 2 | 331 | 2 | 5 |
| 859 | 19 | 1.55 | 0.08 | 0.67 | 0.01 | 344 | 1 | 3 | 328 | 3 | 6 |
| 861 | 23 | 1.54 | 0.07 | 0.66 | 0.01 | 346 | 1 | 3 | 332 | 2 | 6 |

\# is the amount of spectra that used during temperature calculations. $1 \sigma=$ standard deviation; $\mathrm{CI}=95 \%$ confidence interval. $\mathrm{T}^{\mathrm{B}}$ and $\mathrm{T}^{\mathrm{R}}$ are the temperature values obtained by calibrations of the Beyssac et al. (2002a) and by the Rahl et al. (2005), respectively.


Figure 5.2 : Diagram showing RSCM temperatures of the metapelitic rocks of the Esenler Unit. The temperatures are obtained using calibrations of Beyssac et al. (2002a) and Rahl et al. (2005).

### 5.2 Domuzdağ Complex

The Domuzdağ Complex is a deep-seated oceanic HP/LT subduction-accretionary complex consisting mainly of metabasites and micaschists with minor amount of metachert, marble and serpentinite. Peak metamorphic mineral assemblages of the metabasites are sodic amphibole + epidote + chlorite + titanite $\pm$ garnet $\pm$ lawsonite. Metamorphic conditions of a lawsonite- and a garnet-blueschist were modelled by pseudosections produced by Theriak-Domino. Micaschists of the Domuzdağ Complex consist of phengite, paragonite, quartz, chlorite, rutile $\pm$ chloritoid $\pm$ garnet $\pm$ glaucophane in addition to pseudomorphs after lawsonite and jadeite. Peak metamorphic conditions of three chloritoid-micaschists were constrained by pseudosections and by Raman spectra of carbonaceous material. The bulk rock chemistry of the samples is shown in Table A.11.

### 5.2.1 Lawsonite-blueschist

Peak mineral assemblages of the lawsonite-blueschist (sample 128) are sodicamphibole, lawsonite, sodic-pyroxene, chlorite, phengite and titanite with secondary epidote and albite. The PT conditions of the sample 128 are constrained by a pseudosection (Fig. 5.3). The bulk rock chemistry of the samples is shown in Table A.11. In the modelled PT space, sudden increase of jadeite and decrease of diopside contents of sodic-pyroxene $\left(\mathrm{Jd}_{4-18} \mathrm{Aeg}_{31-23} \mathrm{Di}_{65-59}\right.$ and $\left.\mathrm{Jd}_{25-36} \mathrm{Aeg}_{32-28} \mathrm{Di}_{43-37}\right)$ overlaps with the first appearance of lawsonite. The peak mineral assemblages of lawsonite + glaucophane + sodic-pyroxene + phengite + chlorite for the sample 128 cover a large P-T area in the modelled PT space. Following the $\mathrm{X}_{\mathrm{Jd}}$ isopleths of omphacite and Si contents of phengite, metamorphic conditions of the sample 128 are constrained to 12-16 kbar pressure and $370-440{ }^{\circ} \mathrm{C}$ temperature (Fig. 5.3).


Figure 5.3 : Pseudosection produced by the Theriak-Domino for the sample 128. Thick blue line limits the stability field of lawsonite, thick dashed line garnet and thin dashed line of paragonite in this specific rock composition. According to $\mathrm{X}_{\mathrm{Jd}}$ (red) of sodic-pyroxene and Si isopleths (grey) of phengite, metamorphic conditions of the sample are constrained to $14 \pm 2 \mathrm{kbar}$ and $370-440{ }^{\circ} \mathrm{C}$ (the shaded area).

### 5.2.2 Garnet-blueschist

One garnet-blueschist sample (702A) with an inferred peak metamorphic assemblage of garnet + sodic-amphibole + epidote + phengite is modelled by a pseudosection (Fig. 5.4). The bulk rock chemistry of the sample is given in Table A.11. In the modelled PT space, garnet grows at the expense of epidote. In the shaded area of Figure 5.4, $\mathrm{X}_{\text {Grs }}$ increases and $\mathrm{X}_{\text {Alm }}$ decreases with increasing pressure. In sample 702 A , grossular component is stable around $40 \%$. This suggests that garnet growth occurred along a path parallel to the $\mathrm{X}_{\text {Grs }}$ isopleths (Fig. 5.4).


Figure 5.4 : Pseudosection produced by the Theriak-Domino for the sample 702A. For this specific rock composition, thick black line limits stability field of epidote and dashed line of garnet. Stability field of lawsonite is marked by thick blue line. Isopleths; grey, Si content of phengite; green and red, XGrs and XAlm, respectively; dashed isopleth, XJd. Metamorphic conditions constrained as $17 \pm 1 \mathrm{kbar}$ and $500 \pm$ $40^{\circ} \mathrm{C}$ (the shaded area).

In the shaded area, Si isopleths of the phengite are orthogonal to the $\mathrm{X}_{\text {Grs }}$ and $\mathrm{X}_{\text {Alm }}$ isopleths and decreases along the inferred path. This is in accordance with measured phengite composition in the garnet pressure shadow. Modelled clinopyroxene composition forms hedenbergite and diopside solid solution. Jadeite component appear in the HP part of the PT area after consumption of epidote ( $\mathrm{X}_{\mathrm{Jd}}$ : $\sim 3 \%$ ).

However, there is no growth of metamorphic clinopyroxene in sample 702A but relicts of augites. Metamorphic conditions are constrained as $17 \pm 1 \mathrm{Kbar}$ and $500 \pm$ $40^{\circ} \mathrm{C}$.

### 5.2.3 Chloritoid-micaschists

Three peak metamorphic assemblages are identified in the chloritoid micaschists: 1) garnet-chloritoid-glaucophane with pseudomorphs after lawsonite 2) chloritoid with pseudomorphs after glaucophane and 3) relatively high-Mg chloritoid (17\%) with pseudomorphs after jadeite in addition to phengite, paragonite, quartz, chlorite, rutile and apatite. Peak metamorphic conditions of the chloritoid micaschists were obtained by Raman spectra of carbonaceous material and by pseudosection modelling. The bulk rock chemistry of the samples is shown in Table A.11.

### 5.2.3.1 Sample 775

An inferred peak metamorphic assemblage of this sample is garnet + chloritoid + glaucophane + lawsonite + phengite + quartz + chlorite + paragonite + rutile + apatite. In the modelled PT space, stability of chloritoid is constrained above ca. 16 kbar (Fig. 5.5). $\mathrm{X}_{\mathrm{Mg}}$ of the chloritoid is modelled as $8-9 \%$ and comparable with the measured values. Garnet growth starts around $445^{\circ} \mathrm{C}$ at the expense of lawsonite. Coexistence of lawsonite and garnet is limited in the modelled PT field. Within this area, increase of grossular component is modeled along a path with decreasing pressure without significant change in temperature (Fig. 5.5b). Due to the large and unconstrained effect of Mn in the garnet core, we only account for the grossular content of the outer rim ( $\sim 16 \%$ ) indicating pressures of around 18 kbar (Fig. 5.5). Si isopleths of phengite also indicate similar pressures. According to the pseudosections, paragonite is not a stable phase at peak metamorphic conditions and coexistence of paragonite and chloritoid is not expected in the calculated PT space. Thus paragonite was produced by reactions along a retrograde path during exhumation

Peak metamorphic temperatures are constrained to $415 \pm 9^{\circ} \mathrm{C}$ and to $407 \pm 5^{\circ} \mathrm{C}$ by RSCM thermometers using Beyssac et al. (2002a) and Rahl et al. (2005) calibrations, respectively (Table 5.2). These temperatures are slightly lower than the modelled lawsonite + garnet stability field. This possibly related to the effect of Mn
which is excluded from the pseudosection calculations. Metamorphic conditions of the sample 775 are constrained to $17.5 \pm 1 \mathrm{kbar}$ and $390-450{ }^{\circ} \mathrm{C}$.


Figure 5.5 : a) Pseudosection modelling for the sample 775. Stability field of garnet is limited by thick black dashed line, of lawsonite by thick red line, of chloritoid by thick green dashed line, of glaucophane by blue line, of paragonite by thin black dashed line for this specific bulk rock composition. b) A detailed part of the pseudosection. Isopleths; grey $=\mathrm{Si}$ content of phengite; black and orange are $\mathrm{X}_{\mathrm{Alm}}$ and $X_{\text {Grs }}$ of garnet, respectively, green $=X_{M g}$ of chloritoid. Vertical grey strip indicates temperature values obtained by the RSCM thermometer based on Rahl et al. (2005) calibration. Metamorphic conditions constrained to $17.5 \pm 1 \mathrm{kbar}$ and 390$450^{\circ} \mathrm{C}$ (the shaded area in orange).

Table 5.2 : Raman parameters and the calculated peak metamorphic temperatures for the chloritoid micaschists.

| Sample | $\#$ | R 1 | $1 \sigma$ | R 2 | $1 \sigma$ | $\mathrm{~T}^{\mathrm{B}}\left({ }^{\circ} \mathrm{C}\right)$ | CI | $1 \sigma$ | $\mathrm{~T}^{\mathrm{R}}\left({ }^{\circ} \mathrm{C}\right)$ | CI | $1 \sigma$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 775 | 33 | 0.86 | 0.25 | 0.51 | 0.06 | 415 | 9 | 25 | 407 | 5 | 15 |
| 178 | 32 | 0.60 | 0.23 | 0.43 | 0.07 | 451 | 11 | 33 | 441 | 11 | 32 |
| 753 A | 30 | 0.35 | 0.17 | 0.34 | 0.07 | 490 | 12 | 33 | 477 | 14 | 38 |

\# is the amount of spectra that used during temperature calculations. $1 \sigma=$ standard deviation; $\mathrm{CI}=95 \%$ confidence interval. TB and TR are the temperature values obtained by calibrations of the Beyssac et al. (2002a) and by the Rahl et al. (2005), respectively.

### 5.2.3.2 Sample 753A

Sample 753A has an inferred peak metamorphic assemblage of chloritoid + glaucophane + phengite + quartz + chlorite + paragonite + rutile + apatite + allanite. Peak metamorphic conditions of the sample 753A were modelled by a pseudosection (Fig. 5.6). In sample 753A, Ca is taken up by apatite and allanite. Due to the fact that there is no thermodynamic data available for these minerals, Ca is excluded from the bulk rock composition in the calculation of pseudosection (Table A.11).

For the sample 753A, stability of chloritoid is modelled above ca. 16 kbar similar to the sample 775 (Fig. 5.6). Modelled PT space for the peak metamorphic assemblage of Cld + Phe $+\mathrm{Gln}+\mathrm{Chl}+$ Qtz resembles sample 775 with larger uncertainties of temperature. In the absence of Ca-Al-silicate minerals, the garnet-in reaction is shifted to higher temperatures. Coexistence of paragonite and chloritoid is not expected in the calculated PT space. It is inferred from the pseudosection that sample 753 A experienced similar metamorphic pressure as the sample 775 but higher temperatures $\left(400-475{ }^{\circ} \mathrm{C}\right)$. This is in accordance with the temperature values obtained by RSCM thermometers (Table 5.2). The calibration of Beyssac et al. (2002a) gives $490 \pm 12{ }^{\circ} \mathrm{C}$ peak metamorphic temperatures and the calibration of Rahl et al. (2005) estimates temperatures of $477 \pm 14^{\circ} \mathrm{C}$. Metamorphic conditions of the sample 753 A were constrained to $16-18$ kbar pressures and $475 \pm 40{ }^{\circ} \mathrm{C}$ temperatures.


Figure 5.6 : Pseudosection produced by the Theriak-Domino for the sample 753A. Stability field of garnet is limited by thick black dashed line, of chloritoid by thick green dashed line, of glaucophane by blue line, of paragonite by thin black dashed line for this specific bulk rock composition. Isopleths; grey= Si content of phengite, green $=X_{\mathrm{Mg}}$ of chloritoid. Vertical grey strip indicates temperature values obtained by the RSCM thermometer based on Rahl et al. (2005) calibration. Metamorphic conditions are constrained to $16-18 \mathrm{kbar}$ and $475 \pm 40^{\circ} \mathrm{C}$ (the shaded area in orange).

### 5.2.3.3 Sample 178

The inferred peak metamorphic assemblage of the sample 178 is chloritoid + jadeite + phengite + quartz + rutile + apatite. Metamorphic conditions were constrained by pseudosection modelling (Fig. 5.7). In sample 178, the only Ca-bearing mineral is apatite whereas thermodynamic data is not available for apatite. Hence, Ca is excluded from the pseudosection modelling. In sample 178, there is a considerable amount of iron-sulfide mineral. Hence, iron content was slightly reduced in the bulk rock composition that was used in the calculation.

For the sample 178, coexistence of chloritoid and jadeite is modelled above 20 kbar (Fig. 5.7). Upper and lower limit of the temperature are given by the stabilities of garnet and carpholite, respectively. Absence of glaucophane suggests that it is totally
consumed during growth of chloritoid and jadeite. $\mathrm{X}_{\mathrm{Mg}}$ of chloritoid coexisting with jadeite increases up to $15 \%$ along an increasing pressure path and stays stable after consumption of the glaucophane. Modelled Mg content is identical with the measured composition of Cld2 (0.16-17). The Qtz/Cs transformation forms an upper boundary for metamorphic pressure. Within the modeled peak metamorphic conditions, estimated Si content of phengite is higher than our measurements. This should be related to the retrogression. Peak temperatures from the Raman spectroscopy are $451 \pm 11^{\circ} \mathrm{C}$ and $441 \pm 11^{\circ} \mathrm{C}$ by the Beyssac et al. (2002a) and Rahl et al. (2005) calibrations, respectively (Table 5.2). The peak metamorphic conditions of the sample 178 are constrained to $440 \pm 30{ }^{\circ} \mathrm{C}$ temperatures and $22-25 \mathrm{kbar}$ pressures.


Figure 5.7 : Pseudosection modelling for the sample 178. Stability field of garnet is limited by thick black dashed line, of chloritoid by thick green dashed line, of glaucophane by blue lines, of paragonite by red dashed line for the specific bulk rock composition. Isopleths; grey $=\mathrm{Si}$ content of phengite, green $=\mathrm{X}_{\mathrm{Mg}}$ of chloritoid Vertical grey strip indicates temperature values obtained by the RSCM thermometer based on Rahl et al. (2005) calibration. Metamorphic conditions constrained to 22-25 kbar and $440 \pm 30^{\circ} \mathrm{C}$ (the shaded area in orange).

### 5.3 Concluding Remarks

### 5.3.1 The Esenler Unit

Raman spectra of carbonaceous material of the metapelitic rocks of the Esenler Unit reveal that their peak metamorphic temperatures range between $330-385{ }^{\circ} \mathrm{C}$. In Figure 5.8, the RSCM temperatures from the Esenler Unit are plotted and contoured on the geological map of the studied area (for sample locations see Table A. 12 in the Appx). The map reveals the thermal structure pattern of the accreted Lower Cretaceous distal turbidites.


Figure 5.8 : Thermal structure of the accreted distal turbidites. The temperature values are from the calibration of Rahl et al. (2005) due to the fact that it is essentially calibrated for low-grade metamorphic rocks. Relatively high temperature phyllites form a sliver within the low temperature slates. A sliver of Na-amphibolebearing metabasite is possibly associated with the phyllites.

The most striking aspect of the thermal structure map is that the phyllites (370-385 ${ }^{\circ} \mathrm{C}$ ) are exposed as a NE-SW extending slice within the low temperature slates (ca. $330^{\circ} \mathrm{C}$ ) in the central part of the map (Fig. 5.8). The transition between phyllites and slate is sharp suggesting that they are juxtaposed tectonically. A strip of Naamphibole bearing metabasites with similar foliation are exposed west of the phyllites suggesting that they are associated with the phyllites. Although obtained temperature values of the sample 871 is similar to the slates, its degree of graphitization is slightly higher than the rest of the slates as revealed by Raman spectra of carbonaceous material (Fig. 4.1 in Ch. 4). This sample is interpreted as showing a gradual increase in metamorphic temperature towards to the contact with the Domuzdağ Complex. To the south of the area studied, however, the metaflysch sequence shows an inverted metamorphic grade, where phyllites structurally overlie the slates. This is related to post-metamorphic deformation along the Akçataş Fault causing underthrusting of the Esenler Unit beneath the Domuzdağ Complex.

PT conditions of a sodic-amphibole bearing metabasite were constrained to 7-12 kbar pressure and $400 \pm 70{ }^{\circ} \mathrm{C}$ temperatures through THERMOCALC. This metabasite (211B) is exposed as a lens in the northern part of the studied area (Fig. 5.8). Due to its location and the large error range in temperature, it is probably misleading to correlate the obtained PT conditions of the metabasite with RSCM temperatures.

### 5.3.2 The Domuzdağ Complex

Domuzdağ Complex is a deep-seated subduction-accretionary complex of oceanic origin. Peak mineral assemblages and PT estimates of the metabasite and micaschists of the Domuzdağ Complex reveal that it is composed of metamorphic sequences with distinct PT histories. Metabasites are mostly epidote-blueschist locally with garnet. Metamorphic conditions of a garnet-blueschist (702A) were constrained to 17 $\pm 1 \mathrm{kbar}$ and $500 \pm 40^{\circ} \mathrm{C}$. Lawsonite-blueschists, however, represent a major break in the metamorphic grade. Metamorphic conditions of a lawsonite-blueschist (128) were estimated as $14 \pm 2 \mathrm{kbar}$ and $370-440^{\circ} \mathrm{C}$.

The chloritoid-micaschists exhibit generally two peak metamorphic assemblages defined by chloritoid + glaucophane (samples 775 and 753A) and by the inferred chloritoid + jadeite (178) parageneses. The latter is one of the rarely reported HP/LT
mineral assemblages formed at the expense of chloritoid + glaucophane above 20 kbar pressures (Okay, 2002; Proyer, 2003). Metamorphic conditions of the chloritoid-glaucophane-micaschists were constrained to $17.5 \pm 1 \mathrm{kbar}$ and 390-450 ${ }^{\circ} \mathrm{C}$ for the sample 775 and $17 \pm 1 \mathrm{kbar}$ and $475 \pm 40{ }^{\circ} \mathrm{C}$ for the 753 A , respectively. PT conditions of a chloritoid-jadeite-micaschist is constrained to 22-25 kbar and 440 $\pm 30^{\circ} \mathrm{C}$. The PT estimates on the chloritoid-micaschists of the Domuzdağ Complex reveal significant depth of burial within the wedge.

Figure 5.9 shows a compilation of the available PT data from the Domuzdağ Complex. The available PT data suggest that the Domuzdağ Complex is composed of metamorphic rocks that were experienced lawsonite-blueschist, epidote $\pm$ garnetblueschist and eclogite facies conditions. This suggests that the Domuzdağ Complex is a tectonic stack of distinct metamorphic slices rather than a coherent metamorphic sequence.


Figure 5.9 : Compilation of the estimated PT data from the Domuzdağ Complex including chloritoid-micaschists, a lawsonite- and a garnet-blueschist. PT conditions of the Elekdağ eclogites are from Okay et al. (2006a). Possible PT paths for the sample 178 and 775 are indicated by thick grey lines. The dashed parts are stand for zero strain decompressional exhumation. Metamorphic facies are modified from Evans (1990). Facies: LBS: lawsonite-blueschist, $E B S$ : epidote-blueschist, $E C$ : eclogite, $A E A$ : albite-epidote amphibolite, $G S$ : greenschist facies. The compilation suggests that Domuzdağ Complex consists of metamorphic rocks that were metamorphosed under distinct PT conditions.

## 6. GEOCHRONOLOGY

## 6.1 ${ }^{40} \mathrm{Ar}{ }^{39}$ Ar Geochronology

Age of regional HP/LT metamorphism in the area studied is constrained by ${ }^{40} \mathrm{Ar} /{ }^{39} \mathrm{Ar}$ phengite dating method. Phengitic white mica was separated from the accretionary wedge units including the Esenler Unit, the Domuzdağ Complex and the Kunduz Metamorphics. Foliated muscovites from a low-grade metarhyolite of the Kösdağ Formation were also dated to constrain the accretion of the intra-oceanic arc sequence to the overlying accretionary wedge. Sample locations are shown in Figure 2.1 and Figure 2.10. The dating has been conducted at the ${ }^{40} \mathrm{Ar} /{ }^{39} \mathrm{Ar}$ geochronology laboratory of University of Potsdam. The dating system and the procedures are summarized in the methodology part. ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ analytical data are given in the Appendix (Table A.13).

### 6.1.1 Esenler Unit

Two phyllites from the accreted distal turbidites were dated. The samples 31 and 230 both have identical mineralogy and texture. They consist of white mica, quartz, chlorite, albite and calcite (Fig. 6.1). Fine grained phengite and chlorite define the foliation. Both samples show similar Raman spectra of carbonaceous material which constrains their metamorphic temperatures to $370-385^{\circ} \mathrm{C}$ (Table 5.1).

The ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ phengite ages of the samples 31 and 230 are $101.5 \pm 4.8 \mathrm{Ma}$ and 99.9 $\pm 1.7 \mathrm{Ma}$, respectively (Fig. 6.2). The ages agree within the 1 sigma error limits and indicate an Albian regional metamorphism and accretion of the distal turbidites.


Figure 6.1 : Microphotographs of the samples from the Esenler Unit dated by
${ }^{40} \mathrm{Ar}{ }^{\beta 9} \mathrm{Ar}$ phengite method. a\&b) Sample 31 and $\mathrm{c} \& \mathrm{~d}$ ) sample 230 are phyllites that consist of quartz (Qtz), phengite (Phe), albite (Ab), chlorite (Chl).


Figure 6.2 : ${ }^{40} \mathrm{Ar} /{ }^{39} \mathrm{Ar}$ age spectra of the analyzed phengites from the (a) sample 31 and (b) sample 230 from the Esenler Unit.

### 6.1.2 Domuzdağ Complex

${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ dating was performed on two micaschists from the Domuzdağ Complex (Fig 6.3). Sample 392 consists of phengite, paragonite, chlorite, quartz, rutile and syn-kinematic albite porphyroblasts with rotational inclusion fabrics. Foliation is defined by phengite (3.33-3.48 Si c.p.f.u) and chlorite. Sample 753A is a chloritoidmicaschist exposed farther away from the extensional shear zone and consists of chloritoid, phengite, chlorite, paragonite, quartz and rutile with pseudomorphs after glaucophane. Chloritoid forms porphyroblasts of radially growing crystals. Foliation, which is defined by phengite and chlorite, bends around the chloritoid porphyroblasts. Si content of phengite ranges between 3.23-3.47 c.p.f.u.


Figure 6.3 : Microphotographs of the samples from the Domuzdağ Complex dated by ${ }^{40} \mathrm{Ar} /^{39} \mathrm{Ar}$ phengite method. a\&b) Sample 392 consists of quartz (Qtz), phengite (Phe), albite (Ab), chlorite (Chl). c\&d) Sample 753A consists of chloritoid (Cld), quartz ( Qtz ), phengite (Phe), chlorite (Chl).

Sample 392 exhibits a slightly inclined plateau. A partial plateau age of $100.6 \pm 1.3$ Ma is obtained (Fig. 6.4a). Sample 753A, however, gives a pronounced plateau age of $91.8 \pm 1.8 \mathrm{Ma}$ (Turonian) (Fig. 6.4b). This age is apparently younger than all the other ages.


Figure 6.4: ${ }^{40} \mathrm{Ar} /{ }^{39} \mathrm{Ar}$ age spectra of the analyzed phengites from the Domuzdağ Complex. a) Sample 392, a micaschist. b) Sample 753A, a chloritoid-micaschist.

### 6.1.3 Kunduz Metamorphics

Two samples from the Kunduz Metamorphics were dated. The first sample (626A) is a quartz-micaschist interlayered with light colored marble and metabasite. It consists of quartz, phengitic white mica, epidote, hematite and titanite. Phengites define a weakly developed foliation (Fig. 6.5a, b). Their Si contents range between 3.37-3.44 c.p.f.u. The second sample (1290) is a well-foliated phyllitic metatuff consisting of quartz, phengite, albite, bluish-green amphibole, chlorite and stilpnomelane with magmatic augite relicts (Fig. 6.5c, d). Amphibole replaces the augite. Phengite defines the foliation. Besides, there is also late tiny white mica of secondary origin.


Figure 6.5 : Microphotographs of the samples from the Kunduz Metamorphics dated by ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ phengite method. a \& b) Sample 626A is a quartz-micaschist consisting of quartz (Qtz), phengite (Phe), hematite (Hem), epidote (Ep). c \& d) Sample 1290 is a phyllite consisting of quartz (Qtz), phengite (Phe), chlorite (Chl), calcic-amphibole (Amph).

The ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ phengite age of the sample 626 A is constrained as $159.3 \pm 1$ Ma with a pronounced plateau (Fig. 6.6a). Sample 1290 exhibits, however, a disturbed plateau. Released radiogenic Ar is concentrated in 151-161 Ma (Middle Jurassic) in the middle temperature steps and ca. 100 Ma (Albian) with a $101 \pm 0.6 \mathrm{Ma}$ in the last 5 steps making an apparent short "plateau" within the steps (Fig. 6.6b). Due to the fact that radiogenic Ar was predominantly released in third and fourth steps (Table A.13), we infer a Middle Jurassic metamorphic age for the sample, which is possibly overprinted during the Albian orogeny.


Figure 6.6: ${ }^{40} \mathrm{Ar} /{ }^{39} \mathrm{Ar}$ age spectra of the analyzed phengites from the Kunduz Metamorphics. a) Sample 626, a quartz-micaschist. b) Sample 1290, a phyllitic metatuff.

### 6.1.4 Metamorphic age of the Kösdağ Formation

The age of the metamorphism in the low-grade metavolcanic rocks of the Kösdağ Formation was constrained by ${ }^{40} \mathrm{Ar} /{ }^{39} \mathrm{Ar}$ dating on metamorphic muscovite from a metarhyolite sample. The sample 1214 consists of plagioclase and quartz phenocrysts in a recrystallized and foliated groundmass of quartz, white mica, chlorite and epidote (Fig. 6.7a). Plagioclase is partly replaced by epidote, chlorite and quartz. Quartz phenocrysts exhibit undoluse extinction related to deformation. White mica marks the foliation and bends around the phenocrysts. The ${ }^{40} \mathrm{Ar} /{ }^{39} \mathrm{Ar}$ muscovite age is constrained to $69.86 \pm 0.37 \mathrm{Ma}$ with a pronounced plateau in the age spectrum diagram (Fig. 6.7b).


Figure 6.7 : a) Microphotograph of the dated sample 1214 from the Kösdağ Formation. The rock consists of quartz (Qtz), plagioclase (Pl), epidote (Ep) and muscovite (Ms). b) ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ age spectrum diagram obtained from the muscovites of this sample by stepwise heating method.

### 6.2 U/Pb Zircon Geochronology

$\mathrm{U} / \mathrm{Pb}$ zircon dating was performed on the metarhyolites of the Kösdağ Formation to unveil the magmatic crystallization age of the accreted intra-oceanic arc-sequence. Zircons were separated from two metarhyolite samples (1214 and 1215) in the Istanbul Technical University. Measurements were performed by LA-ICP-MS at the University of Potsdam. Details of the dating procedure are given in the methodology part. Sample locations are shown in Figure 2.10.

### 6.2.1 Crystallization age of the Kösdağ Formation

Dated zircons from the two metarhyolite samples are mostly euhedral to subhedral (Fig. 6.8). Zircons from the sample 1214 show dark cores with weak or no zoning. They are surrounded by typical oscillatory zoned mantle sometimes with sector
zoning and bright outer rims. In the sample 1215, zircons show cores with weak oscillatory zoning. They are surrounded by bright rims. The core - rim boundary is regular for all zircons magmatic.


Figure 6.8 : Cathodoluminescence images of the dated zircons from the sample 1214 (a) and 1215 (b).

The dated zircons did not provide concordia ages. Their ${ }^{206} \mathrm{~Pb} /{ }^{238} \mathrm{U}$ ages scatter between 90-100 Ma with weighted mean age values of $93.8 \pm 1.9(M S W D=4.7)$ and $94.4 \pm 1.9(\mathrm{MSWD}=2.7)$ for the samples 1214 and 1215 , respectively (Fig. 6.9, Table 6.1).


Figure 6.9 : Zircon U/Pb concordia diagram for the sample 1214 (a) and 1215 (b) and their weighted mean age values ( c and d ).

Table 6.1 : LA-ICP-MS U-Pb data of the measured zircons from the metarhyolites.

| No | Isotope ratios |  |  |  |  |  | Ages |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | ${ }^{206} \mathrm{~Pb}{ }^{2 / 38} \mathrm{U}$ | 2б(\%) | ${ }^{207} \mathrm{~Pb} /{ }^{235} \mathrm{U}$ | 2б(\%) | ${ }^{207} \mathrm{~Pb} /{ }^{206} \mathrm{~Pb}$ | $2 \sigma(\%)$ | ${ }^{206} \mathrm{~Pb}{ }^{2 / 38} \mathrm{U}$ | $\pm 2 \sigma$ | ${ }^{207} \mathrm{~Pb}{ }^{235} \mathrm{U}$ | $\pm 2 \sigma$ |
| Sample 1214 |  |  |  |  |  |  |  |  |  |  |
| 01 | 0.01434 | 5.4 | 0.14037 | 19.5 | 0.05315 | 25 | 93.5 | 5 | 152.5 | 28 |
| 02 | 0.01486 | 3.5 | 0.10243 | 10.4 | 0.05054 | 9.9 | 93.1 | 3.2 | 96.4 | 9.6 |
| 03 | 0.01583 | 3.8 | 0.14917 | 16.7 | 0.07511 | 14.9 | 98 | 3.7 | 127.5 | 20.2 |
| 04 | 0.01471 | 2.8 | 0.15756 | 8 | 0.05983 | 10.1 | 95.9 | 2.7 | 193.4 | 14.2 |
| 05 | 0.01453 | 2.5 | 0.10247 | 4.9 | 0.04655 | 5.3 | 93.2 | 2.3 | 100.5 | 4.7 |
| 06 | 0.01355 | 2.9 | 0.10734 | 8 | 0.05125 | 8 | 88.7 | 2.6 | 102.3 | 7.8 |
| 07 | 0.01537 | 2.8 | 0.10876 | 9.7 | 0.04987 | 9.6 | 98.8 | 2.7 | 108.9 | 10.1 |
| 08 | 0.01545 | 4 | 0.14025 | 14.5 | 0.06394 | 15.7 | 95.7 | 3.8 | 144.8 | 19.8 |
| 09 | 0.01458 | 3.6 | 0.11633 | 8.6 | 0.0536 | 9.2 | 91.9 | 3.3 | 106.5 | 8.7 |
| 10 | 0.01509 | 2.5 | 0.10518 | 5.5 | 0.05005 | 5 | 95.8 | 2.4 | 103.3 | 5.4 |
| 11 | 0.01428 | 4.8 | 0.11484 | 13.7 | 0.05142 | 15.5 | 90.9 | 4.3 | 109.6 | 14.3 |
| 12 | 0.01425 | 3.1 | 0.08868 | 17.1 | 0.05556 | 13 | 92.4 | 2.8 | 97.9 | 16.1 |
| Sample 1215 |  |  |  |  |  |  |  |  |  |  |
| 01 | 0.01453 | 4.5 | 0.12798 | 13.5 | 0.05887 | 13.9 | 94.2 | 4.2 | 134.1 | 17.1 |
| 02 | 0.01407 | 4.1 | 0.06508 | 25.6 | 0.04529 | 18.8 | 89.9 | 3.7 | 81.9 | 20.4 |
| 03 | 0.01432 | 4.2 | 0.10757 | 10.8 | 0.05428 | 11.8 | 92.1 | 3.8 | 109.1 | 11.2 |
| 04 | 0.01445 | 4.2 | 0.13518 | 10.2 | 0.05372 | 13.5 | 91.9 | 3.8 | 120.7 | 11.7 |
| 05 | 0.01433 | 4.2 | 0.12716 | 14 | 0.05825 | 14.3 | 91.8 | 3.8 | 125.9 | 16.7 |
| 06 | 0.01544 | 5.3 | 0.11721 | 26.3 | 0.0546 | 28.7 | 98.9 | 5.2 | 115 | 29 |
| 07 | 0.0152 | 4.6 | 0.12789 | 17.9 | 0.04998 | 22.1 | 98.1 | 4.5 | 140.7 | 23.8 |
| 08 | 0.01497 | 5.3 | 0.13432 | 16.7 | 0.05782 | 17.8 | 98 | 5.2 | 139 | 21.9 |
| 09 | 0.01433 | 4.3 | 0.09352 | 17.8 | 0.04203 | 18.8 | 94.7 | 4 | 86.1 | 14.8 |
| 10 | 0.0151 | 4.1 | 0.11311 | 12.7 | 0.04938 | 14.2 | 96.3 | 3.9 | 123.7 | 14.9 |
| 11 | 0.0154 | 3.7 | 0.13715 | 9.9 | 0.04944 | 14 | 98.4 | 3.6 | 131.7 | 12.3 |
| 12 | 0.01445 | 4.8 | 0.09291 | 21.3 | 0.05343 | 18.3 | 90.9 | 4.3 | 86.2 | 17.8 |
| 13 | 0.01514 | 5.2 | 0.14237 | 15.5 | 0.05604 | 20.2 | 97.8 | 5 | 137.2 | 20.1 |
| 14 | 0.01556 | 5.5 | 0.09565 | 16.2 | 0.04375 | 16.9 | 99.1 | 5.4 | 91.2 | 14.2 |

### 6.3 Concluding Remarks

### 6.3.1 Accretionary wedge

The ${ }^{40} \mathrm{Ar} /{ }^{39} \mathrm{Ar}$ age results are in accordance with previously published data confirming the wide occurrence of Middle Jurassic and Early Cretaceous subductionaccretion complexes in the Central Pontides (Okay et al., 2006a, 2013). The $91.8 \pm$ 1.8 Ma age, the youngest ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ age obtained in Central Pontides so far, indicates that subduction-accretion continued until Turonian with a possible southward younging of the deep level accretion and HP/LT regional metamorphism.

Interpretation of the radiogenic ages depends on the isotopic closure concept accepting that the temperature is a fundamental parameter together with grain size
and cooling rate in controlling radiometric ages (Dodson, 1973). Additionally deformation and fluids have significant effect on isotopic diffusion (Villa, 1998). Harrison et al. (2009) report a closure temperature of muscovite grain $100 \mu \mathrm{~m}$ across, cooling at $10{ }^{\circ} \mathrm{C} / \mathrm{Ma}$ at 10 kbar as $425{ }^{\circ} \mathrm{C}$ for the ${ }^{40} \mathrm{Ar} /^{39} \mathrm{Ar}$ system. The relatively low temperatures of metamorphism $\left(370-385^{\circ} \mathrm{C}\right.$ for the Esenler Unit and $400-500{ }^{\circ} \mathrm{C}$ for the Domuzdağ Complex) indicate that the obtained ages represent the ages of crystallization. In sample 392, however, the obtained age should be an exhumation age related to recrystallization by pervasive shearing during exhumation. This is evidenced by i) its structural position close to the extensional shear zone with identical lineation development and ii) absence of any peak metamorphic assemblages but syn-kinematic albite porphyroblasts. The $91.8 \pm 1.8 \mathrm{Ma}$ age suggests that extensional shearing and exhumation of the Domuzdag Complex occurred during on-going subduction.

Early-Middle Eocene limestones unconformably overlie the HP/LT metamorphic rocks of the Central Pontide Supercomplex and elsewhere in the Central Pontides (Özcan et al., 2007; Okay et al., 2013). In the area studied, Eocene rocks lie unconformably over the Esenler Unit and Domuzdağ Complex (Fig. 2.1). Furthermore, Helonocyclina and Orbitoides-bearing Maastrichtian limestones are exposed as a tectonic sliver along the post-Eocene Akçataş strike-slip fault. They were most probably deposited on the metamorphic rocks prior to the faulting, which provide an upper age limit for the regional metamorphism and exhumation.

### 6.3.2 Arc sequence

Zircon $\mathrm{U} / \mathrm{Pb}$ dating of the metarhyolites reveals that the Kösdağ intra-oceanic paleoarc formed around $94 \pm 2 \mathrm{Ma}$ ago. These ages are slightly younger and partly overlap with the HP/LT metamorphism of the overlying subduction-accretion complexes suggesting a separate subduction for the generation of arc volcanism. Low-grade regional metamorphism of the Kösdağ Formation is constrained to 70 Ma by ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ muscovite dating. The fact that the overlaying Middle Jurassic and AlbianTuronian HP/LT metamorphic rocks and the Kösdağ Formation share a similar NW dipping foliation suggests that the ca. 70 Ma age represents the final accretion of the intra-oceanic arc sequence to the Laurasian active margin. The youngest ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ phengite age from the overlying subduction-accretion complexes is 92 Ma confirming southward younging of an accretionary-type orogen.

## 7. GEOCHEMISTRY

This chapter deals with petrogenesis and tectonic setting of the accreted arc sequence. Geochemistry of 12 basic and acidic metavolcanic rocks from the Kösdağ Formation was investigated in detail including major, trace and rare earth element analysis. The samples were collected along north-south sections within the Kösdağ Formation (for location of the samples see Fig. 2.10). The bulk rock compositions were obtained by XRF at the GeoForschungsZentrum-Potsdam (GFZ). Trace and rare earth element (REE) analyses were made by ICP-AES at the University of Potsdam and in ACME Lab, Canada. The analytical data are given in Table 7.1.

### 7.1 Geochemistry of the Kösdağ Formation

### 7.1.1 Major element compositions

In a total alkalis $v s$ silica diagram, acidic rocks plot in the rhyolite field (Fig. 7.1). Basic compositions are sub-alkaline and mainly plot in the basaltic andesite and andesite fields. One sample plots in the basaltic trachy-andesite field (sample 1305). Basaltic andesites are tholeiitic and calc-alkaline (Fig. 7.2). Occurrence of abundant xenoliths and low-grade metamorphism, however, affect the bulk rock chemistry of the volcanic rocks, making further discussion redundant. Trace elements are generally regarded as being immobile during metamorphism.

Table 7.1 : Major, trace and rare earth element analysis of the metavolcanic rocks.

|  | Rhyolite |  |  |  |  | BA1 |  |  | BA2 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Sample | 1212 | 1213 | 1214 | 1215 | 1304B | 1285 | 1286 | 1287 | 1284 | 1288 | 1304A | 1305 |
| Major oxides (wt\%) |  |  |  |  |  |  |  |  |  |  |  |  |
| $\mathrm{SiO}_{2}$ | 74.20 | 75.61 | 73.51 | 78.50 | 75.88 | 55.36 | 61.04 | 54.10 | 51.21 | 53.51 | 50.55 | 53.51 |
| $\mathrm{TiO}_{2}$ | 0.48 | 0.53 | 0.43 | 0.30 | 0.31 | 0.91 | 0.80 | 0.85 | 0.91 | 0.83 | 1.07 | 1.27 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 12.00 | 11.49 | 12.12 | 11.22 | 12.55 | 16.03 | 15.80 | 13.78 | 17.44 | 13.61 | 14.68 | 17.24 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 3.61 | 3.34 | 3.56 | 1.73 | 1.93 | 9.86 | 7.31 | 12.39 | 10.81 | 12.15 | 12.68 | 9.86 |
| MnO | 0.08 | 0.05 | 0.10 | 0.02 | 0.04 | 0.09 | 0.07 | 0.19 | 0.11 | 0.18 | 0.20 | 0.29 |
| MgO | 0.59 | 0.27 | 2.14 | 0.41 | 1.01 | 5.03 | 2.76 | 3.53 | 5.43 | 3.48 | 5.95 | 5.39 |
| CaO | 2.88 | 2.64 | 3.14 | 1.88 | 0.58 | 5.73 | 4.04 | 6.75 | 6.10 | 6.65 | 5.00 | 1.51 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 1.82 | 3.99 | 0.98 | 4.89 | 4.32 | 0.73 | 3.82 | 4.11 | 0.91 | 4.06 | 2.60 | 5.78 |
| $\mathrm{K}_{2} \mathrm{O}$ | 2.56 | 0.80 | 1.14 | 0.15 | 1.57 | 1.00 | 0.70 | 0.03 | 1.50 | 0.03 | 0.01 | 0.21 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | 0.11 | 0.14 | 0.07 | 0.06 | 0.05 | 0.14 | 0.20 | 0.08 | 0.13 | 0.08 | 0.13 | 0.29 |
| LOI | 1.54 | 1.04 | 2.76 | 0.78 | 1.67 | 4.93 | 3.34 | 4.00 | 5.22 | 5.32 | 6.93 | 4.51 |
| Total | 99.86 | 99.90 | 99.94 | 99.94 | 99.90 | 99.81 | 99.88 | 99.81 | 99.77 | 99.91 | 99.80 | 99.87 |
| Trace and rare earth elements (ppm) |  |  |  |  |  |  |  |  |  |  |  |  |
| Sc | 13.1 | 17.6 | 15.3 | 10.7 | 6.2 | 31.9 | 27.4 | 25.8 | 39.0 | 47.9 | 42.3 | 32.6 |
| Cr | $<10$ | 10 | <10 | <10 | 10 | 15 | 12 | 34 | 41 | 30 | 34 | 14 |
| V | 45 | 22 | 34 | 26 | 16 | 199 | 93 | 343 | 567 | 354 | 453 | 100 |
| Ni | $<10$ | $<10$ | $<10$ | $<10$ | $<10$ | $<10$ | $<10$ | 14 | 7.8 | 13.4 | 9.1 | $<10$ |
| Co |  |  |  |  |  |  |  |  | 26.4 | 32.8 | 32.8 |  |
| Cu |  |  |  |  |  |  |  |  | 61.8 | 59.9 | 77.1 |  |
| Zn | 45 | 43 | 49 | 29 | 39 | 96 | 108 | 96 | 92 | 87 | 91 | 128 |
| Ga | 12 | 13 | 16 | 10 | $<10$ | 19 | 18 | 15 | 21.7 | 14.5 | 15.6 | 16 |
| Rb | 63 | 31 | 31 | 15 | 43 | 31 | 34 | 15 | 45 | 16 | 16 | 17 |
| Sr | 162 | 247 | 226 | 126 | 150 | 349 | 150 | 72 | 227 | 73 | 145 | 93 |
| Y | 27.2 | 28.7 | 33.3 | 29.3 | 18.3 | 27.1 | 31.7 | 34.9 | 24.9 | 21.1 | 22.3 | 33.1 |
| Zr | 130 | 91 | 159 | 166 | 137 | 99 | 128 | 58 | 58 | 54 | 65 | 128 |
| Nb |  |  |  |  |  |  |  |  | 1.0 | 1.1 | 1.2 |  |
| Th |  |  |  |  |  |  |  |  | 0.6 | 0.4 | 0.2 |  |
| Cs |  |  |  |  |  |  |  |  | 0.3 | 0.4 | $<0.1$ |  |
| Ba | 426 | 139 | 256 | 70 | 424 | 304 | 285 | 69 | 316 | 81 | 74 | 134 |
| Pb |  |  |  |  |  |  |  |  | 0.2 | 0.5 | 1.4 |  |
| Ta |  |  |  |  |  |  |  |  | $<0.1$ | $<0.1$ | $<0.1$ |  |
| Hf |  |  |  |  |  |  |  |  | 1.2 | 1.4 | 1.4 |  |
| U |  |  |  |  |  |  |  |  | 0.2 | 0.1 | 0.2 |  |
| La | 10.02 | 9.54 | 10.85 | 11.39 | 10.91 | 11.03 | 12.32 | 13.97 | 4.90 | 2.90 | 3.90 | 9.22 |
| Ce | 22.33 | 19.56 | 24.94 | 21.99 | 24.24 | 24.58 | 28.63 | 31.58 | 10.30 | 7.00 | 9.50 | 23.72 |
| Pr | 2.21 | 2.13 | 2.67 | 2.73 | 2.50 | 2.75 | 3.34 | 3.80 | 1.48 | 1.00 | 1.43 | 3.16 |
| Nd | 12.97 | 11.91 | 14.23 | 13.02 | 12.64 | 14.23 | 17.03 | 17.97 | 7.90 | 5.20 | 7.10 | 15.75 |
| Sm | 3.68 | 3.55 | 4.16 | 3.60 | 3.06 | 4.05 | 4.86 | 5.01 | 2.25 | 1.79 | 2.37 | 4.92 |
| Eu | 1.32 | 1.51 | 1.35 | 1.00 | 0.98 | 1.52 | 1.88 | 1.83 | 0.96 | 0.72 | 0.85 | 1.76 |
| Gd | 4.43 | 4.62 | 5.36 | 4.14 | 3.31 | 4.73 | 5.85 | 6.00 | 3.65 | 2.73 | 3.16 | 6.00 |
| Tb | 0.78 | 0.87 | 0.96 | 0.74 | 0.57 | 0.82 | 1.05 | 1.09 | 0.66 | 0.55 | 0.59 | 1.10 |
| Dy | 5.03 | 5.29 | 6.15 | 4.89 | 3.47 | 5.11 | 6.43 | 6.61 | 4.22 | 3.48 | 3.65 | 6.73 |
| Ho | 0.91 | 1.00 | 1.23 | 1.01 | 0.63 | 0.99 | 1.26 | 1.27 | 0.90 | 0.76 | 0.83 | 1.35 |
| Er | 3.22 | 3.38 | 3.86 | 3.41 | 2.23 | 3.23 | 4.02 | 4.25 | 2.75 | 2.55 | 2.54 | 4.30 |
| Tm | 0.53 | 0.56 | 0.59 | 0.58 | 0.42 | 0.54 | 0.65 | 0.69 | 0.39 | 0.36 | 0.34 | 0.67 |
| Yb | 3.31 | 3.26 | 3.92 | 3.88 | 2.71 | 3.42 | 4.04 | 4.14 | 2.51 | 2.38 | 2.38 | 4.28 |
| Lu | 0.51 | 0.50 | 0.59 | 0.61 | 0.45 | 0.53 | 0.62 | 0.63 | 0.40 | 0.41 | 0.36 | 0.64 |
| Eu/Eu* | 1 | 1.14 | 0.87 | 0.79 | 0.94 | 1.06 | 1.08 | 1.02 | 1.02 | 1 | 0.95 | 0.99 |
| $\mathrm{La}_{\mathrm{N}} / \mathrm{Yb}_{\mathrm{N}}$ | 2.04 | 1.97 | 1.87 | 1.98 | 2.71 | 2.17 | 2.06 | 2.27 | 1.32 | 0.82 | 1.1 | 1.45 |
| $\mathrm{La}_{\mathrm{N}} / \mathrm{Sm}_{\mathrm{N}}$ | 1.71 | 1.69 | 1.64 | 1.99 | 2.24 | 1.71 | 1.59 | 1.75 | 1.37 | 1.02 | 1.04 | 1.18 |



Figure 7.1 : Total alkali $v s$ silica classification diagram (Le Bas et al., 1986) of the metavolcanic rocks of the Kösdağ Formation. Symbols are in accordance with REE diagram in Fig. 7.3a.


Figure 7.2 : a) AFM (Irvine and Barager, 1971) and b) SiO 2 vs $\mathrm{FeO} / \mathrm{MgO}$ (Miyashiro, 1974) diagrams for the basaltic andesite/andesite of the Kösdag Formation.

### 7.1.2 Trace and rare earth element compositions

Metavolcanic rocks exhibit two distinct REE patterns when normalized to chondrite (Fig. 7.3a). Both the rhyolites and some of the basaltic andesite/andesite, denoted as BA1, show relative enrichment of the LREE with respect to $\operatorname{HREE}\left(\mathrm{La}_{\mathrm{N}} / \mathrm{Yb}_{\mathrm{N}}=1.87-\right.$ 2.71 and $2.06-2.27$, respectively). While the rhyolites show both positive and
negative Eu anomalies $\left(\mathrm{Eu} / \mathrm{Eu}^{*}=0.79-1.14\right), \mathrm{BA} 1$ do not have any significant Eu anomaly ( $\mathrm{Eu} / \mathrm{Eu}^{*}=1.02-1.08$ ).

The second group of basaltic andesites (BA2) are characterized by relatively flat REE pattern $\left(\mathrm{La}_{\mathrm{N}} / \mathrm{Yb}_{\mathrm{N}}=0.82-1.45\right)$. In this group, one sample differs with its higher amounts of REE (sample 1305, dashed line in Fig. 7.3a). Eu/Eu* ratio of BA2 ranges between 0.95-1.02.


Figure 7.3 : a) Chondrite normalized REE diagram for the metavolcanic rocks according to normalization values of Boynton (1984). b) NMORB normalized multielement spider diagram of the metavolcanic rock normalized to Sun and McDonough (1989) values.

In an N-MORB normalized spider diagram, rhyolites and the first group of basaltic andesites (BA1) are characterized by i) enrichment of fluid-mobile LIL elements and ii) depletion of Ti and P with clear negative anomalies (Fig. 7.3b). BA2, although
enriched in LIL elements does not show Ti or P depletion except the one with elevated amounts of REE (sample 1305), which shows a weak Ti negative anomaly.

### 7.2 Interpretation of the Analytical Data

Enrichment of fluid-mobile LILE and depletion of $\operatorname{HFSE}$ ( $\mathrm{Ti}, \mathrm{P}$ and Nb ) of the metavolcanic rocks indicate supra-subduction magmatism (Tatsumi et al., 1986; Pearce and Peate, 1995; You et al., 1996; Münker et al., 2004). The similarity between the REE patterns of the rhyolites and the first group of basaltic andesites (BA1) suggest that they are genetically related. Their REE pattern is identical with calc-alkaline series in island arc volcanism (Gill, 1981; Fujinawa, 1992). Rhyolites with abundant gabbroic xenoliths show positive Eu anomalies. However, in the rhyolites without gabbroic xenoliths, there is a pronounced negative Eu anomaly. The absence of any Eu anomaly in the genetically related first group of basaltic andesites (BA1) indicates that plagioclase was not an important phase in the differentiation of BA1. This fits-well with the experimental data of Müntener et al. (2001) showing that increasing water content in the primitive arc magma suppresses plagioclase crystallization to later stages in favor of crystallization of pyroxene. In the cognate xenoliths, apatite and $\mathrm{Fe}-\mathrm{Ti}$ oxides are common accessory phases. Hence, fractionation of apatite and ilmenite might be inferred for the P and Ti depletion. There is again a positive correlation between the degree of depletion of Ti and P with the absence of cognate xenoliths and $\mathrm{SiO}_{2}$ saturation of the rhyolites. We infer that BA1 and rhyolites are cogenetic and have a common parental magma.

The flat REE pattern of the BA2 is identical with an island arc low-K tholeiite series (Jakeš and Gill, 1970; Fujinawa, 1992). Absence of any Eu anomaly suggests plagioclase was not an accumulated phase. In BA2, there is no Ti and P depletion suggesting there was no fractionation of ilmenite or apatite. Absence of HFSE depletion and LILE enrichment is concordant with its island arc tholeiitic character (Winter, 2001). Weak negative Ti anomaly and elevated amounts of REE of the sample 1305 suggest that this sample is more evolved than the rest of BA2.

### 7.3 Concluding Remarks

Petrography, mineralogy and geochemistry of the metavolcanic rocks indicate that they are mainly composed of basaltic andesite and rhyolite with abundant cognate
gabbroic/mafic xenoliths. Trace and rare earth element compositions of the metavolcanic rocks, depletion of HFSE and enrichment of LILE, indicate suprasubduction volcanism. Recently, Berber et al. (2014) also reported geochemical data from the Kösdağ metavolcanic rocks, which fall in the andesite/basalt, andesite and dacite fields in a $\mathrm{Nb} / \mathrm{Y}$ versus $\mathrm{Zr} / \mathrm{TiO} 2$ diagram. Trace and rare earth element compositions of the samples also indicate a supra-subduction origin, which resemble our BA1 and rhyolites. The lithological association and the geochemistry of the metavolcanic rocks indicate that the Kösdağ Formation formed in a submarine arc setting.

The low-grade arc-sequence is sandwiched between the Laurasian active margin represented by a Cretaceous accretionary wedge and the Gondwana-derived Kırşehir Massif along the İAES. Absence of continent derived detritus within the arc sequence suggests that the Kösdağ Arc was isolated from any continental domain during its formation (Rice et al., 2006). A possible tectonic scenario is that the Kösdağ Arc represents an island arc separated from the Laurasian margin from a marginal sea forming a natural barrier for continental detritus supply (Rice et al., 2006). This scenario mainly based on the assumption that the overlying Middle Jurassic and Cretaceous subduction-accretionary complexes in the north were part of the Triassic basement of the Sakarya Zone. However, the ages of the overlaying HP/LT metamorphic rocks (114-92 Ma, Okay et al., 2006, 2013 and this study) suggest that subduction was still going on in the north when the Kösdag Arc was formed (ca. $94 \pm 2 \mathrm{Ma}$, see Ch . 6). This suggests a dual subduction along the Laurasian margin rather than a slab-roll related opening of a marginal basin (Fig. 7.4a). It is also noteworthy that the overlaying HP/LT metamorphic rocks lack of any intrusive or volcanic rocks coeval to the Kösdağ Arc volcanism. Hence, we propose that the Kösdağ Arc formed initially in an intra-oceanic setting located south of the Laurasian active margin.

Late Cretaceous intra-oceanic arc volcanism of the Kösdağ Arc was relatively shortlived suggesting that the arc was an infant stage (Stern, 2010). This is based on the short time span of volcanism constrained by zircon age data, and the possible Cenomanian age of the overlying pelagic limestones of the Dikmen Formation (Tüysüz 1990, 1993). Termination of volcanism is indicated by absence of any volcanic layers in the sediments of the overlying Dikmen Formation. Extinction of
the Kösdağ Arc should have occurred before the collision of the Kırşehir Massif as indicated by the pelagic character of the overlying Dikmen Formation. This is probably related to the southward migration of the arc volcanism leaving the Kösdağ as a remnant arc (Fig. 7.4b). South of the Kösdağ Formation, unmetamorphosed Upper Cretaceous arc-related calc-alkaline to tholeiitic basalts and basaltic andesites with pyroclastic rocks interbedded with pelagic limestone crop out along the İAES (Yaylaçayı Formation in Fig. 1.3, Rice et al., 2006; Tüysüz et al., 1995). The Yaylaçayı Formation is interpreted as an intra-oceanic arc sequence (Tüysüz et al., 1995) and might represent products of the inferred southward migration of the Kösdağ arc volcanism. Such slab rollback controlled migration of the arc volcanism is also proposed for the Upper Cretaceous (ca. $68-76 \mathrm{Ma},{ }^{40} \mathrm{Ar} /{ }^{39} \mathrm{Ar}$ whole rock and biotite ages) alkaline magmatic rocks within the İAES; possibly with respect to the Yaylaçayı Formation (Sarifakioğlu et al., 2014). Arc migration is best seen at present in the Izu-Bonin-Mariana arc system. This arc system was rifted episodically from the Palau Kyushu and West Mariana ridges located to the west which represent the remnant arcs (Stern et al., 2003; Takahashi et al., 2007; Stern, 2010).


Figure 7.4 : a) Cartoon model showing the Late Cretaceous tectonic configuration of the Central Pontides. a) The Kösdağ Arc formed at ca. 95 Ma in an intra-oceanic setting located south of the Laurasian active margin. b) Slab rollback controlled arc migration leading to the termination of volcanism in the Kösdağ Arc.

An intriguing aspect of the intra-oceanic arc sequence is the occurrence of voluminous felsic volcanic rocks. Arc-related meter-sized blocks of rhyolites associated with larger basaltic rocks were also reported within the ophiolitic mélange farther southwest near Eldivan (Dangerfield et al., 2011). Previously, it was generally accepted that intra-oceanic volcanic arcs are dominated by basalts and basaltic andesites (e.g. Ewart et al., 1977; Gill, 1981; Hawkins et al., 1984; Gamble et al., 1997). However, acidic volcanic rocks have also been reported in the last decades from the modern intra-oceanic volcanic arcs (e.g. Robin et al., 1993; Monzier et al., 1994; Worthington et al., 1999). There are two models for their genesis a) differentiation of the parental magma generated in the mantle wedge through fractional crystallization and crystal accumulation (Pearce et al., 1995; Straub, 1995), b) re-melting of the andesitic (Tamura and Tatsumi, 2002) or basaltic (Leat et al., 2003; Smith et al., 2003) sub-arc crust. Our petrographic and geochemical data support the fractional crystallization and accumulation model for the formation of the rhyolites. This is evidenced by 1) identical trace and rare earth elements of both the BA1 and the rhyolites characterized by LREE enrichment and depletion of Ti and P 2) occurrences of abundant $\mathrm{Fe}-\mathrm{Ti}$ oxide and apatite-bearing mafic cognate xenoliths within the rhyolites and an andesite. Cumulate-like clusters of plagioclase with interstitial clinopyroxene and monomineralic cluster of apatite further support accumulation as a main agent in differentiation of the parental magma.

Occurrence of cognate mafic xenoliths is compatible with the exposed intra-oceanic sub-arc crustal rocks consisting mainly of gabbro (DeBari and Coleman, 1989; DeBari and Sleep, 1991; Miller and Christensen, 1994; Greene et al., 2006; Hacker et al., 2008; Jagoutz et al., 2010; Petterson, 2010). Distinct Eu/Eu* ratios of the BA1 (1.02-1.08) and the rhyolites ( $0.79-1.14$ ) suggest that plagioclase accumulation played a major role during differentiation of the felsic rocks from a common mafic parental melt. A recent experimental work showed that at the late stage of the differentiation of the primitive arc magmas, fractionation of amphibole, plagioclase and $\mathrm{Fe}-\mathrm{Ti}$ oxide (extraction of amphibole-bearing gabbroic rocks) can increase the SiO 2 content of the remaining melt from $53 \%$ up to $78 \%$ (Nandedkar et al., 2014). This is in accordance with the trace and rare earth element signature of the rhyolite and with the cognate mafic xenoliths found within the rhyolites.

## 8. GEODYNAMICS

The Central Pontides is an interesting and newly recognized accretionary-type orogenic area formed along the southern Laurasian active continental margin. The region is characterized by Mesozoic subduction-accretionary complexes called as Central Pontide Supercomplex exposed over a large area. The majority of subduction-accretionary complexes form an Albian-Turonian accretionary wedge consisting of both continental and oceanic crust derived HP/LT metamorphic rocks, mélanges and forearc deposits. The continental accretionary rocks are a low-grade HP metaflysch sequence consisting of phyllite/metasandstone intercalation with recrystallized limestone, Na-amphibole-bearing metabasite and serpentinite (the Esenler and Martin formations). It represents distal parts of a large Lower Cretaceous submarine turbidite fan deposited on the Laurasian active continental margin that was subsequently accreted and metamorphosed during Albian. The accreted continental metaclastic rocks are tectonically underlain by a metabasite dominated oceanic deep-seated HP/LT metamorphic sequence (the Domuzdağ Complex) along an extensional shear zone. Middle Jurassic subduction-accretion complexes are found within the Albian-Turonian wedge as tectonically emplaced slices. At the latest Cretaceous, a nascent Late Cretaceous ( $94 \pm 2 \mathrm{Ma}$ ) intra-oceanic arc was accreted to the Albian-Turonian wedge along İAES, the main tectonic division throughout Turkey separating Laurasia from Gondwana-derived continental blocks. In this tectonic framework, the Central Pontides consists of subduction and accretion related units which are sandwiched between the southern Laurasian active margin and the Gondwana-derived Kırşehir Block without involvement of any continental fragments. This indicates a major contribution of pre-collisional accretionary continental growth of the Laurasian active continental margin during Tethyan subduction.

Below, geodynamic implication of this new accretionary-type orogenic area is discussed in terms of tectonic thickening of the wedge and exhumation of the deep
seated HP/LT metamorphic rocks. Finally, the accretionary units are correlated with surrounding region to understand their role in regional geological studies.

### 8.1 Tectonic Thickening of the Albian-Turonian Accretionary Wedge

In the Central Pontides, the Albian-Turonian accretionary wedge is characterized by the presence of voluminous accreted terrigenous metaclastic rocks. They represent accreted distal parts of a large Lower Cretaceous submarine turbidite fan. The proximal parts of the turbidite fan is exposed farther north (Okay et al., 2013). This makes the Central Pontides one of the key areas for understanding the tectonic thickening of an accretionary wedge under high amount of sediment flux into the trench, which has a primary effect on frontal enlarging of the wedge by offscraping and shallow level underplating of the continental detritus (e.g. Platt et al., 1985; McCall, 2002).

The uniform lithology of the accreted turbidite sequence makes recognition of structural elements such as major faults difficult. Besides, due to the rheological weakness of the slate and phyllites, the primary structural framework can be easily overprinted by subsequent, younger deformation events. RSCM of the metapelitic rocks of the Esenler Unit, however, revealed that the accreted distal turbidites consist of metamorphic packets with distinct peak metamorphic temperatures. A major portion of the wedge is formed by low-temperature slates that possibly represent scraped off turbidites coming to the trench. The phyllites most likely formed during shallow level underplating of continental detritus beneath the offscraped distal turbidites. The associated incipient blueschist facies metabasite sliver was peeled off from the sinking oceanic crust and mixed tectonically with the phyllites during underplating along the basal décollement (Kimura and Ludden, 1995).

Recognition of the relatively high-temperature phyllites as a sliver within the lowtemperature slates is significant and provides new insights on the structure of the Albian-Turonian accretionary wedge. There are no kinematic data for the tectonic emplacement of the underplated phyllites into the offscraped turbidites. The youngest deformation in the studied area is represented by post-Eocene east-west striking oblique strike-slip faulting which strongly modified the southern part of the area studied. The post-Eocene deformation is characterized by northward tilting and emplacement of the higher-grade Domuzdağ Complex over the Esenler Unit.

However, this northward tilting is absent in the phyllites that are exposed as a slice within the slates in central part of studied area. These phyllites and the associated metabasite sliver show gently NW dipping foliation planes similar to the general structural trend of the Albian-Turonian wedge. This suggests that their tectonic emplacement occurred during syn-subduction stacking.

In accretionary wedges, out-of-sequence thrusting is a common phenomenon in order to maintain the critical wedge taper during tectonic thickening of the wedge (Platt, 1986; Morley, 1988). It can juxtapose distinct metamorphic sequences during accretion within the wedge. Vitrinite reflectance studies revealed a sharp thermal discontinuity in the Cretaceous Shimanto accretionary complex, Japan, which is interpreted as an out-of-sequence thrust (Ohmori et al., 1997). Tectonic emplacement of the underplated phyllites and incipient blueschist facies metabasites within the lower-temperature offscraped slates is thus regarded as controlled by out-of-sequence thrusting causing tectonic thickening and uplift of the wedge (Fig. 8.1).


Figure 8.1 : A model showing a possible initial situation of the phyllites and the slates within the Albian accretionary wedge. While the slates represent the offscraped distal turbidites, the phyllite possibly represent underplated metasediments of the turbidite fan. Out-of-sequence thrusting (dashed thick lines) is proposed for uplift and tectonic emplacement of the phyllites. The model is modified after Moore et al. (2001).

The accreted distal turbidites are tectonically underlain by oceanic crust-derived HP/LT metabasite and micaschist along an extensional shear zone. The underlying oceanic unit comprises the major proportion of the wedge and consists of a tectonically thickened eclogite to blueschist facies metamorphic sequence (Okay et al., 2006a, 2013). In the area studied, deep-seated metabasites are epidote-blueschist locally with garnet. Lawsonite-blueschist blocks within the extensional shear zone indicate a major break in the metamorphic grade. The tectonic position of the
lawsonite-blueschists suggests that they formed at the earlier stages of the accretion along the overriding plate boundary. Lawsonite stability is generally regarded as indicative for relatively cold thermal setting and fast exhumation (Zack et al., 2004; Whitney and Davis, 2006; Tsujimori et al., 2006b). There are many parameters that can affect the thermal structure of a subduction zone including age of the slab, convergence rate, geometry of subduction, shear related heating etc. (Peacock, 1996). Numerical modelling indicates that temperatures at the upper part of a subduction zone along the plate contact are primarily determined by frictional heating (van den Beukel and Wortel, 1987). Accordingly, Peacock (1992) showed that occurrence of blueschist facies low temperatures rocks are associated with low shear stress and low shear heating along the subduction zone. We infer that during the initial stage of the Albian convergence, accretion was characterized by low shear stress generating a relatively cold thermal structure where the lawsonite-blueschists formed. With southward growth of the wedge, the thermal setting is modified in favor of epidoteblueschist facies metamorphism related to increase of shear stress within the prism. Although the chloritoid-micaschist samples 775 and 753A have identical depth of burial, the latter experienced considerably higher temperatures revealed by the different degrees of graphitization. It is noteworthy that these two samples of chloritoid-micaschists are exposed close to each other without a clear tectonic boundary in between. Source of the extra heat of the sample 753A might also be originated by relatively high shear stress within the wedge. This may point significance of shear stress deviation across a developing accretionary prism and dispute a steady-state thermal setting even in a limited area.

So far, majority of the studies on deep-seated HP/LT metamorphic rocks are focused on their exhumation (e.g. Platt, 1993; Sedlock, 1999; Jolivet et al., 2003; Ganne et al., 2006; Brun and Faccenna, 2008; Agard et al., 2009). However, tectonic thickening of these metamorphic sequences requires considerable amount of space along the plate interface, which is generally disregarded. Plat (1986) proposed a general thickening mechanism of the wedge by sediment underplating in a stable subduction zone configuration. Steady state thickening of the wedge is compensated by extension of the wedge (gravity-driven collapse) which also exhumes the HP/LT metamorphic rocks. Such steady state wedge configuration is also set up for analog sandbox experiments where thickening simply compensated by uplift of the
overburden (e.g. Malavieille, 2010). Although this mechanism may cause thickening and uplift of shallow level underplated rocks (e.g. accreted distal turbidites of the Esenler Unit, see above), it is unlikely to solve the space problem along the plate interface for the tectonic thickening of deep-seated eclogite to blueschist facies metamorphic sequences (e.g. Domuzdağ Complex). Most of the subduction zones, however, are far from being in a stable configuration, and are characterized by advancing or retreating trench migration (Uyeda and Kanamori, 1979; Carlson and Melia, 1984; Lallemand et al., 2008). The latter is attributed to the slab rollback, which causes extension of the overriding plate (Garfunkel et al. 1986; Schellart, 2008). By interpreting seismic reflection data, Minelli and Faccenna (2010) reported a post-Messinian basal décollement ramp related to slab rollback in the Calabrian wedge, Italy, which caused frontal thrust propagation and underplating. It is discussed by the same authors that forward propagation of décollement associated with slab rollback can cause crustal thickening (Minelli and Faccenna, 2010). By analogy with the Calabrian wedge, slab rollback also seems the most plausible process that can create a necessary space along plate interface for progressive deep level basal accretion. Thickening of the HP/LT metamorphic sequences by continuous underplating of oceanic basalts and sediments is possibly achieved by décollement propagation along the retreating subduction interface. By extending the overriding plate, slab rollback also plays a major role in subsequent exhumation of the deep-seated metamorphic sequence suggesting that tectonic thickening and exhumation of deeply underplated rocks are connected (Figure 8.2).

Estimated arc-parallel strain rates in modern fore-arc regions suggest that extensional strain is more common than compression (McCaffrey, 1996). The extension of the Albian-Turonian accretionary wedge is structurally evidenced by the extensional shear zone and footwall micaschists with syn-kinematic albite porphyroblasts. ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ dating on phengites reveals that the extensional pervasive shearing occurred in a syn-subduction setting during the Late Cretaceous. Upper Cretaceous leucite-bearing alkaline volcanic rocks and related turbidites possibly were extruded and deposited above the thinning accretionary wedge. All these data suggest a precollisional diffusive and broad extension of the overriding plate from back arc (opening of the Black Sea) to fore-arc region. To the south, accretion of the Kösdağ
paleo-arc to the Albian-Turonian wedge is constrained to ca. 70 Ma providing an upper age limit for pre-collisional extension of the wedge.


Figure 8.2 : Model for the tectonic evolution of the Albian-Turonian wedge with a possible tectonic thickening and subsequent exhumation mechanism of the deep seated subduction-accretion complexes. During Albian, the wedge was dominated by accretion of clastic detritus. Lawsonite blueschists possibly formed during initial low shear stress Albian subduction. Late Albian to Turonian evolution of the wedge was mainly controlled by slab rollback, which creates the necessary space for deep level underplating by décollement propagation. Synchronous extension of the wedge by slab rollback, subsequently exhumes the deep-seated metamorphic sequence.

It is noteworthy that progressive material accretion and subsequent exhumation should appear with a younging pattern towards the trench. This fits with the $91.8 \pm$ 1.8 Ma age obtained from a chloritoid-micaschist exposed relatively south compared to $100.6 \pm 1.3 \mathrm{Ma}$ age in the north.

### 8.2 Exhumation of the Deep-Seated Domuzdağ Complex

Unlike the HP/LT coherent metamorphic sequences formed by continental subduction as in western Alps, NW Turkey and Oman (e.g. Chopin, 1981, 1984; ElShazly and Liou, 1991; Okay, 2002), the subduction-accretionary HP/LT metamorphic rocks are formed by tectonic accumulation of down going material by underplating mainly of oceanic crustal lithologies. These two distinct types of metamorphic sequences do not only differ in their way of formations but also in their mode and rate of exhumation (e.g. Agard et al., 2009). Buoyancy-driven exhumation plays a key role in subducted continental margins or continental blocks in a rate close to the plate velocities (Chemenda et al., 1996). Exhumation of the deeply underplated oceanic accretionary rocks, however, is rather complicated and different mechanisms may operate. One of the earliest models is the flow concept based on the upward motion of the blocks within the subduction channel or wedge (Cloos, 1982; Cloos and Shreve, 1988). Later, removal of the overburden achieved largely by extension was the main focus. Different tectonic scenarios were proposed for the generation of extension including gravitational collapse due to tectonic thickening (Platt, 1986, 1993), slab rollback triggered by subduction of small continental blocks (Brun \& Faccenna, 2008) or slab rollback itself (Jolivet et al. 2003). The number of geological, petrologic, and structural data to test the models, however, is limited. One important constraint comes from the available PT data suggesting that unlike the subducted continental margins that can experienced UHP conditions, exhumation of oceanic subduction-accretionary complexes is limited to rocks buried to ca. 80 km (Agard et al. 2009; Whitney et al. 2014). As already pointed by Whitney et al. (2014), this depth intriguingly overlaps with the inferred maximum decoupling depth (MDD) below which the sinking slab is coupled with viscously flowing mantle wedge (Wada and Wang, 2009; Wada et al., 2008). The MDD or "point of no return" most probably forms also a natural limit for tectonic accumulation along the sinking
slab or ultimate tip of the subduction-accretionary wedges controlled by forearc lithospheric thickness.

In many HP/LT metamorphic areas, extension-related structures have been reported suggesting extension plays a key role in their exhumation (for example Avigad \& Garfunkel 1991; Wallis et al. 1993; Sedlock 1999, Agard et al. 2001; Jolivet et al. 2003). Similarly, in the Central Pontides, the most pronounced structural feature of the Albian-Turonian accretionary wedge is the extensional shear zone, which separates the deep-seated oceanic Domuzdağ Complex from overlying low-grade accreted distal turbidites. It forms the main boundary between the accreted continental and oceanic rocks and was probably worked as the basal decollement during initial stages of Albian convergence. Southeast of the extensional shear zone, the footwall micaschists were formed by pervasive ductile shearing during extensioninduced exhumation. In these micaschists, calcite and tourmaline are common suggesting high fluid flux during ductile pervasive shearing accompanying the exhumation. Increasing ferric iron content of the low-Si phengites is also consistent with the increasing fluid activity. Similar retrogressive textures are also observed in the eclogites of the Domuzdağ Complex. Large tourmaline crystals together with secondary chlorites are interpreted as retrogression associated with infiltration of oceanic crust derived B- and Na-rich fluids during exhumation (Altherr et al., 2004).

Farther away from the shear zone, the chloritoid-micaschists represent zero strain domains bounded possibly by localized shear zones (Fig. 8.3). Although the chloritoid micaschist were formed under varying metamorphic conditions, their retrograde nature are identical characterized by static replacement of the peak metamorphic assemblages, which now occur as pseudomorphs. Absence of tourmaline suggests that the chloritoid-micaschists did not interact with B-rich fluids during zero strain exhumation. Decomposition of hydrous minerals by decompression can release significant amount of fluid which could be source of fluids during exhumation of deep-seated rock (Zheng et al., 2007). Breakdown of lawsonite to garnet and finally to epidote by decompression is most likely fluid source for the sample 775. In the other two samples, however, product minerals like chlorite have more $\mathrm{H}_{2} \mathrm{O}$ than the reactant minerals chloritoid or glaucophane. The source of the required fluid is ambiguous. An explanation could be that the chloritoid micaschists were never completely dry. These fluids within the rocks may form the
source of water needed during zero strain exhumation and static retrogression. This may suggest that the chloritoid-micaschists possibly represent closed-systems in terms of fluid flux/infiltration during their exhumation and were retrograded by contained fluids. Grossular-rich outer rim of the post kinematic garnets and the nature of the pseudomorphs support decompression without additional heat.


Figure 8.3 : An exhumation model for the deeply underplated oceanic HP/LT metamorphic sequence by decompression of the wedge along a retreating slab. Zero strain domains are characterized by partly preservation of the peak metamorphic assemblages. Retrogression is manifested by pseudomorphs formed through trapped fluids. B-rich fluids are generated from the newly underplating oceanic metabasalts and metasediments through dehydration reactions. The fluids then were channelized and migrated upward along the high-strain shear zones.

The Domuzdağ Complex was exhumed as a whole to shallow levels along the brittle extensional shear zone. Coupling of the chloritoid-micaschists with different depth of burial (sample 178 with samples 775 and 753A) within the wedge, however, is ambiguous. An upper limit of the exhumation of the Domuzdağ Complex is provided by the Maastrichtian limestone exposed as a tectonic sliver along the Akçataş Fault. The limestone probably unconformably covered the Domuzdağ Complex. This enables calculation of an average exhumation rate for the Domuzdağ Complex, particularly for the chloritoid-micashist sample 753A. The ${ }^{40} \mathrm{Ar} /{ }^{39} \mathrm{Ar}$ phengite dating on the sample 753A constrained the HP/LT metamorphism to ca. 92 Ma . This suggests ca. 50 km uplift (slightly less than 17 kbar ) in 22 Ma (92-70 Ma). These parameters give a $2.3 \mathrm{~mm} / \mathrm{y}$ average exhumation rate for the sample 753 A .

As already discussed in section 8.1, syn-subduction decompression of the AlbianTuronian wedge is most likely triggered by the slab retreat caused tectonic thickening and subsequent exhumation of the deep seated metamorphic sequence. Brun \& Faccenna (2008) pointed out that exhumation of HP metamorphic rocks is associated with subduction of continental blocks triggering slab rollback. In the case of the Central Pontides where orogeny was of accretionary-type, however, there is no evidence of a subduction of a crustal fragment that may trigger the slab retreat.

### 8.3 Comparative Regional Geology

The Cretaceous accretionary complexes in western Turkey are poor in trench-fill turbidites, unlike those in the Central Pontides. The absence of trench-fill turbidites suggests insufficient sediment supply which has a major effect on frontal enlargement of an accretionary wedge. Another difference is that in NW Turkey the accretionary complexes are shallow level and therefore constitute slightly metamorphosed tectonic mélanges of Late Cretaceous age (Okay, 2002; Topuz et al., 2008; Göncüoğlu et al., 2008, 2014; Aygül et al., 2012; Akbayram et al., 2013, Plunder et al., 2013). Related deep-seated HP/LT metamorphic rocks are generally of continental origin representing subducted passive continental margins. These include Çamlıca-Kemer metamorphic complexes (Okay and Satır, 2000; Aygül et al., 2012; Şengün et al., 2012) and Tavşanlı-Afyon zones (Sherlock et al., 1999; Okay, 2002; Okay and Whitney, 2010; Plunder et al., 2013; Pourteau et al., 2010, 2013). Absence of tectonically thickened deep-seated subduction-accretionary complexes may be related to the along strike variation of the slab-rollback. It is noteworthy that Upper Cretaceous alkaline volcanic sequence, which lies unconformably over the accretionary complexes in the Central Pontides, is also not reported in NW Turkey.

To the east, the Tokat Massif, consists of metamorphic rocks similar to those from the Central Pontides. Bozkurt et al. (1997) reported a low grade metaclastic unit with abundant recrystallized limestone, serpentinite and radiolarian chert blocks which is comparable with Esenler Unit. Radiolaria contents give a Late Jurassic age for the cherts (Bozkurt et al., 1997). There are also Na-amphibole-bearing metabasite, metapelites with ophiolitic rocks representing an oceanic accretionary unit (Kuytul Formation, Yılmaz et al., 1997b). Age of the regional metamorphism in the Tokat Massif, however, is accepted as Triassic though there is no absolute age data.

Along strike variations of the subduction-accretion complexes within the Anatolian scale depict the diversity of tectonic settings. In western Turkey, convergence was Alpine-type dominated by continental subduction and collisions. To the east, however, it was of accretionary type causing tectonic continental growth. The distinct tectonic settings are most likely controlled by physical behavior of the sinking slab.

In the Pontides, Middle Jurassic subduction-accretion complexes (Çelik et al., 2011; Topuz et al., 2012; Okay et al., 2013; Marroni et al., 2014) or arc sequences consisting of intrusive and volcanic rocks with associated LP/HT metamorphic rocks (Genç and Tüysüz, 2010; Göçmengil et al., 2013; Okay et al., 2014) are increasingly reported in the last several years. This suggests that Middle Jurassic represents a period of convergence in Tethyan realm. In the CPS, the Middle Jurassic accretionary complexes lie south of the Albian-Turonian ones, contrary to what is expected in an evolving subduction-accretion system. The possible Albian metamorphic overprint in the Middle Jurassic metamorphic rocks suggests that tectonic emplacement occurred during Albian orogeny. Subduction obliquity can cause significant along strike tectonic transport by using zones of weakness (e.g., Fitch 1972; Beck, 1983; Jarrard, 1986; McCaffrey, 1992). Emplacement of the Middle Jurassic metamorphic rocks is most probably related to oblique convergence during the Albian subduction. E-W trending post-Eocene strike slip faults (e.g. Akçataş fault; Figure 1.2,2.1) are widespread in the Central Pontides and might have also caused further eastward tectonic transport.

### 8.3.1 Late Cretaceous intra-oceanic Kösdağ arc and supra-subduction ophiolites

Although intra-oceanic arcs comprise about $40 \%$ of the present volcanic arcs by length (Leat and Larter, 2003), they are rarely reported in geological record. Recognition of former intra-oceanic arcs exposed within older orogenic belts is important. They can play a key role in regional tectonic studies due to i) their ultimate collision/accretion along convergent margins causing tectonic growth of the continental crust (Taylor, 1967; Şengör et al., 1993; Rudnick, 1995; Clift and Vannucchi, 2004) and ii) possible linkage with supra-subduction ophiolite obduction (Dilek, 2003; Pearce, 2003). Most of the obducted ophiolitic bodies are underlain by thin metamorphic soles consisting of strongly sheared high-grade metabasic rocks (Williams and Smyth, 1973; Malpas et al., 1973; Spray, 1984; Jamieson, 1986;

Hacker, 1990; Wakabayashi and Dilek, 2000). These metamorphic soles are generally interpreted as marking the initiation of intra-oceanic subduction, although associated intra-oceanic arcs are generally not recognized. Despite the well-described exposures of Tethyan supra-subduction ophiolites found as obducted on Gondwanaderived continental fragments (Thuizat et al., 1981; Dilek et al., 1999; Searle and Cox, 1999; Robertson, 2002; Malpas et al., 2003; Önen, 2003; Çakır, 2009; Hébert et al., 2012; Hässig et al., 2013a, 2013b; Yalınız et al., 1996, 2000a, 2000b), presumed arcs are generally not well defined within the Alpine-Himalayan range.

The intra-oceanic Kösdağ paleo-arc is exposed along the İAES, which is the main division between the former Laurasia and Gondwana. In the northeastern part of the İAES, greenschist facies basaltic andesites, andesite and associated pyroclastic rocks are exposed, which are known as Karadağ Formation (Rice et al., 2006, 2009). The rocks show supra-subduction geochemical signature and interpreted as products of an Upper Cretaceous oceanic arc. It is proposed that Karadağ Formation forms the lateral continuation of the Kösdağ Formation. In western and central Anatolia, the age of the supra-subduction ophiolite obduction on Gondwana-derived terranes, as revealed by the dating of the metamorphic soles, is similar to that of the Kösdağ Arc. The ${ }^{40} \mathrm{Ar} r^{39} \mathrm{Ar}$ ages from the amphibolites of the metamorphic sole range generally between 90-94 Ma (Dilek et al., 1999; Parlak and Delaloye, 1999; Önen and Hall, 2000; Önen, 2003; Çelik et al., 2006) and predate the HP/LT metamorphism of the subducted Anatolide-Tauride continental margin (ca. 80 Ma ; Sherlock et al., 1999). Hence, we infer that the Kösdağ Arc forms part of the presumed intra-oceanic arc associated with the supra-subduction ophiolite generation that is obducted on the Gondwana-derived the Anatolide-Taurides.

The linkage of Tethyan closure from Turkey to Iran is evidenced by Late Cretaceous blueschists exposed within the Tethyan subduction-accretionary complexes in Armenia (90-95 Ma, Rolland et al., 2009) and in Iran (e.g. 85-95 Ma, Agard et al., 2006). Furthermore, Rolland et al. (2009) also reported a metamorphic event at ca. 71-73 Ma. It is interpreted as accretion of South Armenian Block and southward subduction jump (Rolland et al., 2012). These ages are similar to the accretion of the Kösdağ Arc at ca. 70 Ma suggesting that the final amalgamation of the Gondwanaderived continental blocks, Laurasian active margin and the intra-oceanic arc occurred at the latest Cretaceous along the İAES. Initiation of a Late Cretaceous
intra-oceanic subduction in the Tethyan realm is mainly evidenced by formation and obduction of the supra-subduction ophiolites. Along the Sevan-Akera suture, Amasia ophiolite was obducted over the South Armenian continental block, eastward extension of the Anatolide-Tauride Block (Galoyan et al., 2009; Rolland et al., 2010; Sosson et al., 2010). Garnet amphibolites with ca. 90 Ma isotopic age are interpreted as the metamorphic sole rocks (Hässig et al., 2011). It is proposed that NE Anatolian and Armenian ophiolite obduction occurred as a single event (Hässig et al., 2013a, 2013b). Ophiolitic fragments within the Zagros orogeny widely show suprasubduction affinity (Babaie et al., 2006; Desmons and Beccaluva, 1983; Moghadam et al., 2010). ${ }^{40} \mathrm{Ar} /{ }^{39} \mathrm{Ar}$ dating on hornblende and $\mathrm{U} / \mathrm{Pb}$ dating on zircons from the plagiogranites give ca. 92-93 Ma (Babaie et al., 2006) and ca. 101-103 Ma (Moghadam et al., 2013) ages, respectively. An associated Cretaceous arc sequence consisting of island arc volcanic and volcanoclastic rocks with pelagic sediments structurally overlie the ophiolitic rocks (Babaie et al., 2001). The geology of Oman is strikingly similar to the geology of NW Turkey characterized by ophiolite obduction and continental subduction. $\mathrm{U} / \mathrm{Pb}$ zircon geochronology of the plagiogranites/trondhjemites revealed that the Semail ophiolite formed ca. 95 Ma ago (Tilton et al., 1981; Warren et al., 2005). ${ }^{40} \mathrm{Ar} /{ }^{39} \mathrm{Ar}$ ages from the metamorphic soles are identical with ophiolite formation (Hacker et al., 1996; Searle and Cox, 1999; Warren et al., 2005). U/Pb zircon ages of the eclogite underlying the ophiolite constrain the continental subduction in Oman as ca. 80 Ma (Warren et al., 2003).

Remnants of Late Cretaceous intra-oceanic paleo-arcs are also reported farther east along the Indus suture of Himalayas. In north Pakistan, sub-arc crustal rocks and metamorphosed basic to intermediate volcanic rocks known as Kohistan intraoceanic paleo-arc crop out over large areas (Coward et al., 1987; Petterson, 2010). Ages of the zircons and garnets from the lower parts of the arc ranges between 90100 Ma (Yamamoto \& Nakamura, 1996; Anczkiewicz \& Vance, 2000 Schaltegger et al., 2002). There are also gabbroic rocks with ca. $85 \mathrm{Ma} \mathrm{U} / \mathrm{Pb}$ zircon age (Schaltegger et al., 2002) which were associated with intra-arc extension-related mantle diapirs (Burg et al., 1998). Further east along the Indus suture, Dras Volcanics are accepted as continuation of the Kohistan paleo-arc (Honegger et al., 1982; Radhakrishna et al., 1984; Clift et al., 2002; Robertson \& Collins, 2002). Associated possible metamorphic soles are exposed at the base of the ophiolite
fragments within the Indus Suture and were dated as $80-90 \mathrm{Ma}$ (Malpas et al., 2003). It is proposed that ophiolite obduction occurred as a single event from Oman to eastern Himalaya along the northern margin of India during Late CretaceousPaleocene before the ca. 55 Ma India-Asia collision (Searle et al., 1997).

All these data reveal existence of connected Late Cretaceous intra-oceanic subduction along the sinking Tethyan slab (e.g. Aitchison et al., 2000) which is also supported by mantle tomographic imaging (Fig. 8.4; Van der Voo et al., 1999). It generated ensimatic arcs ultimately accreted to Laurasian active margin (such as in the Central Pontides and in the Himalayas) and associated supra-subduction ophiolite obduction on Gondwana-derived terranes (e.g. in NW Turkey and Oman).


Figure 8.4 : Paleogeographic sketch map at ca. 95 Ma showing the dual subduction during Tethyan consumption (modified from Van der Voo et al. (1999)). Dashed line indicates location of the $\mathrm{N}-\mathrm{S}$ section along the Central Pontides in Figure 7.4. ATB:

Anatolide-Tauride Block.

## 9. CONCLUSIONS

The Central Pontides is a new accretionary-type orogenic area within the AlpineHimalayan orogenic belt characterized by pre-collisional tectonic continental growth of Laurasian margin. The region comprises Mesozoic subduction-accretion complexes and an accreted intra-oceanic arc that are sandwiched between Laurasian active continental margin and Gondwana-derived the Kırşehir Block. The subduction-accretion complexes mainly consist of an Albian-Turonian accretionary wedge consisting of HP/LT metamorphic rocks of continental and oceanic origin, forearc deposits as well as mélange and cover an area of 50 km by 100 km in the Central Pontides. In the north, the wedge is characterized by voluminous accreted metaclastic rocks consisting of slate/phyllite and metasandstone intercalation with recrystallized limestone, Na-amphibole-bearing metabasite ( $\mathrm{PT}=7-12$ kbar and 400 $\pm 70^{\circ} \mathrm{C}$ ) and serpentinite blocks. This metaflysch sequence represents distal parts of a large Lower Cretaceous submarine turbidite fan deposited on active Laurasian continental margin and subsequently accreted. RSCM of the metapelitic rocks of the metaflysch sequence reveals that the accreted distal turbidites have undergone low temperature (ca. $330{ }^{\circ} \mathrm{C}$ ) metamorphism. Phyllites with higher metamorphic temperatures (ca. $380{ }^{\circ} \mathrm{C}$ ) occur as a slice within the low-temperature slates. The phyllites are associated with a strip of Na-amphibole bearing metabasites and possibly represent shallow level underplated terrigenous metaclastic rocks along the basal decollement. The metabasites were possibly peeled off from the sinking oceanic crust and underplated. The phyllites and the Na- amphibole bearing metabasites were tectonically emplaced within the offscraped part of the wedge by possibly out-of-sequence thrusting causing tectonic thickening and uplift of the wedge. ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ phengite ages from the phyllites are ca. 100 Ma , indicating Early Cretaceous subduction.

The accreted continental metaclastic rocks are underlain by HP/LT metamorphic rocks of oceanic origin along an extensional shear zone. The oceanic metamorphic sequence mainly comprises tectonically thickened deep-seated eclogite to blueschist
facies metabasites and micaschists with minor serpentinite, metagabbro, metachert and marble. In the studied area, metabasites are epidote-blueschists locally with garnet ( $\mathrm{PT}=17 \pm 1 \mathrm{kbar}$ and $500 \pm 40^{\circ} \mathrm{C}$ ). Lawsonite-blueschists are exposed along the extensional shear zone ( $\mathrm{PT}=14 \pm 2 \mathrm{kbar}$ and $370-440{ }^{\circ} \mathrm{C}$ ). They are possibly associated with low shear stress regime of the initial stage of convergence. South of the extensional shear zone, the footwall micaschists consist of quartz, phengite, paragonite, chlorite, rutile with syn-kinematic albite porphyroblast formed by pervasive shearing during exhumation. These type of micaschists are tourmalinebearing and their retrograde nature suggests high-fluid flux along shear zones. Peak metamorphic assemblages are partly preserved in the chloritoid-micaschist farther away from the shear zone representing the zero strain domains during exhumation. Three peak metamorphic assemblages are identified and their PT conditions were constrained by pseudosections produced by Theriak-Domino and by Raman spectra of carbonaceous material: 1) garnet-chloritoid-glaucophane with pseudomorphs after lawsonite (P: $17.5 \pm 1 \mathrm{Kbar}, \mathrm{T}: 390-450{ }^{\circ} \mathrm{C}$ ) 2) chloritoid with pseudomorphs after glaucophane ( $\mathrm{P}: 16-18 \mathrm{Kbar}, \mathrm{T}: 475 \pm 40{ }^{\circ} \mathrm{C}$ ) and 3) relatively high-Mg chloritoid ( $17 \%$ ) with pseudomorphs after jadeite ( $\mathrm{P}: 22-25 \mathrm{Kbar}$; $\mathrm{T}: 440 \pm 30^{\circ} \mathrm{C}$ ) in addition to phengite, paragonite, quartz, chlorite, rutile and apatite. The last mineral assemblage is interpreted as the transformation of the chloritoid + glaucophane assemblage to chloritoid + jadeite paragenesis with increasing pressure. Absence of tourmaline suggests that the chloritoid-micaschist did not interact with B-rich fluids during zero strain exhumation. ${ }^{40} \mathrm{Ar} /{ }^{39} \mathrm{Ar}$ phengite age of a pervasively sheared footwall micaschist is constrained to $100.6 \pm 1.3 \mathrm{Ma}$ and that of a chloritoidmicaschist is constrained to $91.8 \pm 1.8 \mathrm{Ma}$ suggesting exhumation during on-going subduction with a southward younging of the basal accretion and regional metamorphism. To the south, the subduction-accretionary complexes consist of blueschist and greenschist facies metabasite, marble and volcanogenic metasediment intercalation. ${ }^{40} \mathrm{Ar} /{ }^{39} \mathrm{Ar}$ phengite dating reveals that this part of the wedge is of Middle Jurassic age which is partly overprinted during Albian. Emplacement of the Middle Jurassic subduction-accretion complexes is possibly associated with obliquity of the Albian convergence.

Peak metamorphic assemblages and PT estimates suggest that the Domuzdağ Complex is a metamorphic stack of tectonic slices rather than a coherent
metamorphic sequence. Coupling and exhumation of the distinct metamorphic slices are controlled by decompression of the wedge possibly along a retreating subduction. Structurally, decompression of the wedge is evident by the extensional shear zone and the footwall micaschists with syn-kinematic albite porphyroblasts. Mineral chemistry and textures of the chloritoid-micaschists (post-kinematic garnets with increasing grossular content and pseudomorphing minerals) also support decompression model without an extra heating.

Thickening of subduction-accretion complexes is attributed to i) significant amount of clastic sediment supply from the overriding continental domain and ii) deep level basal underplating by propagation of the décollement along a retreating subduction. Underplating by basal décollement propagation and subsequent exhumation of the deep-seated subduction-accretion complexes are connected and controlled by slab rollback creating a necessary space for progressive basal accretion along the plate interface and extension of the wedge above for exhumation of the tectonically thickened metamorphic sequences. This might be the most common mechanism of the tectonic thickening and subsequent exhumation of deep-seated HP/LT subduction-accretion complexes.

To the south, the Albian-Turonian accretionary wedge structurally overlies a lowgrade arc-sequence which is thrusted over the ophiolitic rocks of the İzmir-AnkaraErzincan suture (İAES), separating the Laurasian active margin from Gondwanaderived terranes. The arc sequence consists of low-grade metavolcanic rocks and stratigraphically overlying recrystallized micritic limestone with rare volcanogenic metasediments. The metavolcanic rocks mainly consist of rhyolite and basaltic andesite with pyroclastic equivalents that are interbedded with pelagic limestone and chert. Geochemistry of the metavolcanic rocks indicates supra-subduction volcanism evidenced by depletion of HFSE and enrichment of LILE. The tectonic setting and absence of continent derived detritus within the arc sequence indicate an intraoceanic setting. Abundant cognate gabbroic xenoliths and identical trace and rare earth elements compositions show that rhyolites and some of the basaltic andesites/andesites (BA1) are cogenetic and felsic rocks of the Kösdağ Formation were derived from a common mafic parental magma by fractional crystallization and accumulation processes.

Zircons from two metarhyolite samples give $93.8 \pm 1.9 \mathrm{Ma}$ and $94.4 \pm 1.9 \mathrm{Ma} \mathrm{U} / \mathrm{Pb}$ ages indicating Cenomanian-Turonian arc magmatism. These ages are slightly younger and partly overlap with the age of the HP/LT metamorphism of the overlying subduction-accretionary complexes (114-92 Ma). The tectonic setting and age data favor a sepearate subduction for the arc sequence that is formed in an intraoceanic setting at the southern offshore of the Laurasian active continental margin. This is supported by the absence of terrigenous detritus within arc sequence and by basaltic andesites with island arc tholeiite REE pattern (BA2). The low-grade regional metamorphism of the arc sequence is constrained to $69.9 \pm 0.4 \mathrm{Ma}$ by ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ muscovite dating indicating that the arc sequence was part of a wide Cretaceous subduction-accretionary complex by the latest Cretaceous; confirming southward younging of an accretionary-type orogenic area. Abrupt non-collisional termination of arc volcanism is best explained by southward migration of the arc volcanism similar to the Izu-Bonin-Mariana arc system.

Age of the intra-oceanic arc volcanism is coeval with the metamorphic sole ages of the obducted supra-subduction ophiolites in NW Turkey. This suggests that the Kösdağ paleo-arc forms part of the presumed but missing intra-oceanic arc associated with the initial subduction stage leading regional ophiolite obduction. Remnants of a Late Cretaceous intra-oceanic paleo-arc and supra-subduction ophiolites can be traced eastward within the Alp-Himalayan orogenic belt. This reveals that Late Cretaceous intra-oceanic subduction was a widespread event. It resulted in arc accretion to Laurasian active margin and supra-subduction ophiolite obduction on Gondwana-derived terranes.

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## APPENDICES

APPENDIX A: Geological maps and cross sections.


Figure A.1 : Geological map of the area studied across the Central Pontide Supercomplex between Kastamonu-Tosya with cross sections.


Figure A. 2 : Geological map of the studied section south of Tosya, along the İzmir-Ankara-Erzincan suture with a cross section between A and A'.
Table A. 1 : Mineral chemistry of the measured amphiboles. Mineral formula calculations are based on 23 oxygens.

| Na-Amph | $211 \mathrm{~B}^{\text {a }}$ | 211B | 211B | 211B | 211B | 211B | 211B | 211B | 211B | 211B | 211B | 211B | 211B | 211B | 211B |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 252 | 258 | 259 | 260 | 262 | 269 | 270 | 271 | 299 | 300 | 313 | 316 | 317 | 319 | 323 |
| $\mathrm{SiO}_{2}$ | 53.93 | 54.83 | 55.16 | 54.56 | 53.66 | 54.03 | 54.43 | 54.78 | 53.41 | 53.90 | 53.43 | 53.32 | 53.78 | 53.10 | 54.14 |
| $\mathrm{TiO}_{2}$ | 0.05 | 0.26 | 0.02 | 0.05 | 0.05 | 0.02 | 0.06 | 0.02 | 0.02 | 0.05 | 0.02 | 0.09 | 0.08 | 0.04 | 0.04 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 4.98 | 5.38 | 5.19 | 4.98 | 4.46 | 5.11 | 5.90 | 5.83 | 4.65 | 5.07 | 5.63 | 4.85 | 5.98 | 5.30 | 5.89 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.01 | 0.00 | 0.02 | 0.00 | 0.01 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 |
| FeO | 23.43 | 18.69 | 19.63 | 20.55 | 24.14 | 21.97 | 20.33 | 19.42 | 24.36 | 22.17 | 22.76 | 23.06 | 20.94 | 22.60 | 21.33 |
| MnO | 0.17 | 0.15 | 0.13 | 0.16 | 0.15 | 0.19 | 0.14 | 0.11 | 0.11 | 0.17 | 0.14 | 0.15 | 0.15 | 0.16 | 0.15 |
| MgO | 6.52 | 8.73 | 8.84 | 8.41 | 6.56 | 7.27 | 7.53 | 8.35 | 5.96 | 7.18 | 6.14 | 6.71 | 7.29 | 6.49 | 7.08 |
| CaO | 0.75 | 0.82 | 0.53 | 1.63 | 1.45 | 1.46 | 0.42 | 0.53 | 0.74 | 0.89 | 0.91 | 1.36 | 0.81 | 1.00 | 1.00 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 6.45 | 6.66 | 6.56 | 5.87 | 5.97 | 6.01 | 6.47 | 6.53 | 6.37 | 6.35 | 6.41 | 6.19 | 6.52 | 6.31 | 6.34 |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.09 | 0.01 | 0.01 | 0.03 | 0.02 | 0.01 | 0.01 | 0.01 | 0.01 | 0.01 | 0.02 | 0.02 | 0.01 | 0.02 | 0.02 |
| Total | 96.37 | 95.52 | 96.08 | 96.23 | 96.47 | 96.06 | 95.31 | 95.57 | 95.63 | 95.78 | 95.45 | 95.73 | 95.57 | 95.02 | 95.98 |
| Si | 7.884 | 7.914 | 7.933 | 7.911 | 7.877 | 7.895 | 7.944 | 7.926 | 7.901 | 7.883 | 7.874 | 7.844 | 7.839 | 7.856 | 7.885 |
| Ti | 0.006 | 0.028 | 0.002 | 0.005 | 0.005 | 0.002 | 0.006 | 0.003 | 0.003 | 0.005 | 0.002 | 0.010 | 0.009 | 0.005 | 0.004 |
| A1 | 0.858 | 0.914 | 0.880 | 0.851 | 0.771 | 0.881 | 1.015 | 0.994 | 0.811 | 0.873 | 0.978 | 0.840 | 1.027 | 0.924 | 1.011 |
| Cr | 0.001 | 0.000 | 0.002 | 0.000 | 0.001 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{3+}$ | 1.172 | 1.063 | 1.074 | 0.962 | 1.158 | 1.027 | 0.913 | 0.980 | 1.208 | 1.150 | 1.098 | 1.212 | 1.120 | 1.161 | 0.999 |
| $\mathrm{Fe}^{2+}$ | 1.692 | 1.194 | 1.288 | 1.530 | 1.806 | 1.658 | 1.568 | 1.370 | 1.805 | 1.562 | 1.707 | 1.625 | 1.433 | 1.635 | 1.598 |
| Mn | 0.021 | 0.018 | 0.016 | 0.020 | 0.019 | 0.023 | 0.018 | 0.013 | 0.014 | 0.021 | 0.018 | 0.019 | 0.019 | 0.020 | 0.018 |
| Mg | 1.421 | 1.879 | 1.896 | 1.818 | 1.436 | 1.584 | 1.638 | 1.801 | 1.314 | 1.566 | 1.348 | 1.471 | 1.584 | 1.431 | 1.537 |
| Ca | 0.117 | 0.127 | 0.082 | 0.253 | 0.227 | 0.228 | 0.066 | 0.081 | 0.117 | 0.139 | 0.143 | 0.214 | 0.127 | 0.159 | 0.156 |
| Na | 1.828 | 1.864 | 1.828 | 1.650 | 1.699 | 1.703 | 1.832 | 1.832 | 1.827 | 1.801 | 1.832 | 1.765 | 1.844 | 1.809 | 1.791 |
| K | 0.016 | 0.002 | 0.001 | 0.005 | 0.004 | 0.002 | 0.003 | 0.002 | 0.002 | 0.001 | 0.004 | 0.004 | 0.002 | 0.004 | 0.003 |
| Total | 15.016 | 15.002 | 15.001 | 15.005 | 15.004 | 15.002 | 15.003 | 15.002 | 15.002 | 15.001 | 15.004 | 15.004 | 15.002 | 15.004 | 15.003 |

Table A. 1 (continued) : Mineral chemistry of the measured amphiboles. Mineral formula calculations are based on 23 oxygens.

| Na-Amph | $128^{\text {a }}$ | 128 | 128 | 128 | 128 | 128 | 128 | 128 | 128 | 128 | 128 | 128 | 128 | 128 | 128 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 145 | 146 | 148 | 151 | 152 | 159 | 165 | 167 | 168 | 169 | 171 | 176 | 178 | 181 | 188 |
| $\mathrm{SiO}_{2}$ | 57.08 | 56.84 | 56.60 | 56.25 | 56.21 | 56.49 | 56.91 | 56.80 | 56.13 | 57.04 | 56.97 | 57.90 | 56.57 | 56.49 | 56.31 |
| $\mathrm{TiO}_{2}$ | 0.09 | 0.21 | 0.02 | 0.00 | 0.11 | 0.08 | 0.07 | 0.31 | 0.09 | 0.08 | 0.08 | 0.03 | 0.15 | 0.38 | 0.05 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 10.52 | 9.57 | 8.64 | 5.79 | 9.04 | 8.37 | 10.16 | 9.81 | 6.39 | 8.84 | 9.65 | 10.22 | 5.96 | 9.82 | 7.80 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.00 | 0.04 | 0.00 | 0.02 | 0.00 | 0.00 | 0.00 | 0.00 | 0.01 | 0.05 | 0.02 | 0.00 | 0.00 | 0.03 | 0.00 |
| FeO | 14.45 | 14.32 | 15.45 | 16.39 | 15.82 | 15.13 | 14.45 | 14.86 | 18.09 | 16.42 | 15.46 | 15.00 | 17.77 | 15.19 | 17.33 |
| MnO | 0.34 | 0.36 | 0.47 | 0.46 | 0.42 | 0.53 | 0.35 | 0.32 | 0.14 | 0.20 | 0.12 | 0.25 | 0.29 | 0.28 | 0.13 |
| MgO | 8.31 | 8.81 | 9.11 | 10.30 | 8.43 | 9.20 | 8.30 | 8.70 | 8.94 | 8.22 | 8.23 | 8.24 | 9.39 | 8.15 | 8.38 |
| CaO | 0.56 | 1.00 | 1.67 | 2.70 | 0.99 | 1.31 | 0.58 | 0.68 | 1.49 | 0.60 | 0.50 | 0.37 | 1.36 | 0.80 | 1.15 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 7.40 | 6.93 | 6.76 | 5.96 | 7.19 | 6.91 | 7.26 | 7.12 | 6.62 | 7.12 | 7.49 | 7.31 | 6.81 | 7.15 | 6.82 |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.03 | 0.03 | 0.03 | 0.06 | 0.03 | 0.05 | 0.01 | 0.02 | 0.01 | 0.02 | 0.02 | 0.01 | 0.01 | 0.02 | 0.04 |
| Total | 98.78 | 98.10 | 98.75 | 97.92 | 98.23 | 98.07 | 98.09 | 98.62 | 97.91 | 98.58 | 98.54 | 99.33 | 98.32 | 98.31 | 98.01 |
| Si | 7.868 | 7.885 | 7.853 | 7.903 | 7.849 | 7.877 | 7.896 | 7.832 | 7.905 | 7.913 | 7.904 | 7.923 | 7.923 | 7.857 | 7.901 |
| Ti | 0.009 | 0.022 | 0.002 | 0.000 | 0.011 | 0.008 | 0.008 | 0.032 | 0.010 | 0.008 | 0.008 | 0.003 | 0.016 | 0.039 | 0.005 |
| A1 | 1.710 | 1.564 | 1.413 | 0.959 | 1.487 | 1.376 | 1.661 | 1.595 | 1.061 | 1.446 | 1.577 | 1.648 | 0.984 | 1.610 | 1.290 |
| Cr | 0.000 | 0.004 | 0.000 | 0.002 | 0.000 | 0.000 | 0.000 | 0.000 | 0.001 | 0.005 | 0.002 | 0.000 | 0.000 | 0.003 | 0.000 |
| $\mathrm{Fe}^{3+}$ | 0.389 | 0.452 | 0.556 | 0.789 | 0.545 | 0.585 | 0.406 | 0.570 | 0.849 | 0.611 | 0.428 | 0.450 | 0.880 | 0.425 | 0.690 |
| $\mathrm{Fe}^{2+}$ | 1.276 | 1.210 | 1.236 | 1.137 | 1.302 | 1.180 | 1.271 | 1.144 | 1.281 | 1.295 | 1.365 | 1.267 | 1.201 | 1.342 | 1.344 |
| Mn | 0.040 | 0.042 | 0.056 | 0.054 | 0.050 | 0.063 | 0.041 | 0.038 | 0.016 | 0.023 | 0.014 | 0.029 | 0.035 | 0.033 | 0.016 |
| Mg | 1.708 | 1.821 | 1.884 | 2.157 | 1.756 | 1.912 | 1.717 | 1.789 | 1.877 | 1.700 | 1.701 | 1.680 | 1.961 | 1.691 | 1.753 |
| Ca | 0.083 | 0.148 | 0.249 | 0.406 | 0.148 | 0.196 | 0.086 | 0.100 | 0.224 | 0.089 | 0.075 | 0.055 | 0.204 | 0.119 | 0.172 |
| Na | 1.978 | 1.864 | 1.819 | 1.624 | 1.946 | 1.869 | 1.953 | 1.903 | 1.809 | 1.916 | 2.015 | 1.939 | 1.849 | 1.927 | 1.856 |
| K | 0.005 | 0.005 | 0.005 | 0.011 | 0.006 | 0.008 | 0.001 | 0.003 | 0.002 | 0.003 | 0.004 | 0.002 | 0.002 | 0.004 | 0.006 |
| Total | 15.065 | 15.018 | 15.073 | 15.040 | 15.100 | 15.073 | 15.041 | 15.019 | 15.035 | 15.007 | 15.094 | 15.006 | 15.055 | 15.050 | 15.035 |

Table A. 1 (continued) : Mineral chemistry of the measured amphiboles. Mineral formula calculations are based on 23 oxygens.

| Na-Amph | $128^{\text {a }}$ | 128 | 128 | 128 | 128 | 128 | 128 | 128 | 128 | 128 | 128 | 128 | 128 | 128 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 189 | 191 | 192 | 193 | 194 | 195 | 196 | 198 | 199 | 200 | 201 | 202 | 204 | 205 |
| $\mathrm{SiO}_{2}$ | 56.21 | 56.41 | 56.50 | 56.82 | 56.01 | 56.97 | 55.94 | 56.36 | 55.91 | 56.57 | 57.47 | 56.82 | 57.41 | 56.90 |
| $\mathrm{TiO}_{2}$ | 0.20 | 0.03 | 0.00 | 0.09 | 0.08 | 0.09 | 0.10 | 0.11 | 0.09 | 0.12 | 0.01 | 0.04 | 0.00 | 0.19 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 8.93 | 8.53 | 9.97 | 8.80 | 8.95 | 8.41 | 7.03 | 9.22 | 6.54 | 9.90 | 10.08 | 6.89 | 10.57 | 9.77 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.01 | 0.00 | 0.00 | 0.00 | 0.01 | 0.03 | 0.00 | 0.00 | 0.00 | 0.00 | 0.01 | 0.01 | 0.00 | 0.03 |
| FeO | 15.58 | 16.64 | 16.55 | 16.30 | 16.98 | 16.67 | 18.22 | 16.04 | 19.82 | 16.07 | 14.38 | 16.44 | 14.34 | 14.19 |
| MnO | 0.29 | 0.13 | 0.22 | 0.24 | 0.31 | 0.21 | 0.17 | 0.23 | 0.11 | 0.24 | 0.26 | 0.34 | 0.28 | 0.26 |
| MgO | 8.57 | 8.23 | 7.62 | 8.16 | 7.77 | 8.21 | 7.91 | 8.03 | 7.57 | 8.02 | 8.41 | 9.46 | 8.30 | 8.83 |
| CaO | 1.07 | 0.89 | 1.12 | 0.63 | 1.15 | 0.79 | 1.07 | 0.89 | 0.43 | 0.63 | 0.29 | 0.97 | 0.23 | 0.65 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 7.05 | 6.83 | 7.06 | 6.83 | 6.72 | 7.12 | 6.81 | 6.97 | 6.87 | 7.56 | 7.49 | 7.10 | 7.22 | 7.13 |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.02 | 0.02 | 0.03 | 0.01 | 0.02 | 0.01 | 0.03 | 0.02 | 0.01 | 0.02 | 0.01 | 0.03 | 0.01 | 0.02 |
| Total | 97.93 | 97.73 | 99.06 | 97.88 | 98.01 | 98.51 | 97.27 | 97.87 | 97.33 | 99.11 | 98.41 | 98.09 | 98.35 | 97.96 |
| Si | 7.861 | 7.903 | 7.844 | 7.920 | 7.857 | 7.935 | 7.950 | 7.888 | 7.932 | 7.831 | 7.935 | 7.933 | 7.900 | 7.883 |
| Ti | 0.021 | 0.003 | 0.000 | 0.010 | 0.009 | 0.009 | 0.011 | 0.012 | 0.009 | 0.012 | 0.001 | 0.005 | 0.000 | 0.020 |
| A1 | 1.473 | 1.409 | 1.631 | 1.445 | 1.480 | 1.381 | 1.178 | 1.520 | 1.094 | 1.615 | 1.641 | 1.134 | 1.714 | 1.595 |
| Cr | 0.001 | 0.000 | 0.000 | 0.000 | 0.001 | 0.003 | 0.000 | 0.000 |  | 0.000 | 0.001 | 0.001 | 0.000 | 0.004 |
| $\mathrm{Fe}^{3+}$ | 0.526 | 0.652 | 0.442 | 0.661 | 0.609 | 0.569 | 0.696 | 0.518 | 1.000 | 0.483 | 0.392 | 0.774 | 0.490 | 0.486 |
| $\mathrm{Fe}^{2+}$ | 1.296 | 1.297 | 1.480 | 1.240 | 1.383 | 1.374 | 1.469 | 1.359 | 1.351 | 1.377 | 1.268 | 1.144 | 1.160 | 1.158 |
| Mn | 0.034 | 0.016 | 0.026 | 0.028 | 0.037 | 0.025 | 0.020 | 0.027 | 0.013 | 0.028 | 0.030 | 0.040 | 0.033 | 0.030 |
| Mg | 1.787 | 1.719 | 1.577 | 1.697 | 1.625 | 1.705 | 1.676 | 1.676 | 1.601 | 1.655 | 1.732 | 1.969 | 1.702 | 1.825 |
| Ca | 0.160 | 0.133 | 0.167 | 0.094 | 0.173 | 0.117 | 0.162 | 0.134 | 0.066 | 0.093 | 0.043 | 0.145 | 0.034 | 0.096 |
| Na | 1.912 | 1.856 | 1.900 | 1.845 | 1.829 | 1.922 | 1.876 | 1.891 | 1.889 | 2.028 | 2.005 | 1.922 | 1.926 | 1.914 |
| K | 0.004 | 0.004 | 0.005 | 0.002 | 0.004 | 0.002 | 0.005 | 0.004 | 0.002 | 0.003 | 0.002 | 0.004 | 0.002 | 0.003 |
| Total | 15.076 | 15.004 | 15.072 | 15.002 | 15.006 | 15.042 | 15.043 | 15.029 | 15.002 | 15.124 | 15.050 | 15.071 | 15.005 | 15.014 |

Table A. 1 (continued) : Mineral chemistry of the measured amphiboles. Mineral formula calculations are based on 23 oxygens.

| Na-Amph | $702 \mathrm{~A}^{\text {a }}$ | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 1 | 20 | 22 | 25 | 26 | 30 | 36 | 38 | 41 | 43 | 54 | 55 | 60 | 63 | 64 |
| $\mathrm{SiO}_{2}$ | 55.96 | 56.34 | 54.97 | 55.64 | 56.14 | 55.80 | 55.65 | 56.12 | 55.72 | 56.48 | 55.71 | 56.84 | 55.79 | 56.28 | 56.32 |
| $\mathrm{TiO}_{2}$ | 0.03 | 0.07 | 0.02 | 0.05 | 0.09 | 0.05 | 0.07 | 0.16 | 0.11 | 0.01 | 0.10 | 0.25 | 0.10 | 0.05 | 0.12 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 12.05 | 10.29 | 10.26 | 7.96 | 9.12 | 9.27 | 9.77 | 10.07 | 9.95 | 11.01 | 9.48 | 11.37 | 10.52 | 10.29 | 9.63 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.01 | 0.02 | 0.00 | 0.02 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.03 | 0.01 | 0.01 | 0.00 | 0.00 | 0.03 |
| FeO | 15.92 | 15.58 | 18.10 | 20.18 | 18.88 | 18.44 | 17.05 | 17.92 | 18.53 | 14.56 | 19.26 | 15.09 | 16.66 | 17.05 | 16.74 |
| MnO | 0.12 | 0.11 | 0.26 | 0.09 | 0.08 | 0.09 | 0.13 | 0.12 | 0.15 | 0.03 | 0.08 | 0.08 | 0.04 | 0.11 | 0.07 |
| MgO | 6.28 | 7.55 | 6.26 | 6.00 | 6.29 | 6.55 | 6.84 | 6.29 | 6.28 | 8.50 | 6.09 | 7.12 | 6.84 | 6.70 | 7.36 |
| CaO | 0.52 | 1.40 | 1.79 | 0.85 | 0.62 | 1.14 | 1.21 | 0.81 | 0.84 | 1.41 | 0.66 | 0.65 | 1.02 | 0.76 | 0.90 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 7.66 | 6.72 | 6.58 | 6.83 | 7.09 | 6.90 | 6.90 | 7.18 | 6.90 | 6.88 | 7.20 | 7.39 | 6.84 | 7.10 | 6.99 |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.06 | 0.02 | 0.03 | 0.02 | 0.01 | 0.03 | 0.02 | 0.03 | 0.04 | 0.05 | 0.03 | 0.01 | 0.01 | 0.03 | 0.02 |
| Total | 98.60 | 98.10 | 98.27 | 97.65 | 98.32 | 98.26 | 97.63 | 98.68 | 98.50 | 98.95 | 98.59 | 98.80 | 97.81 | 98.37 | 98.16 |
| Si | 7.825 | 7.883 | 7.787 | 7.951 | 7.916 | 7.883 | 7.882 | 7.882 | 7.830 | 7.781 | 7.854 | 7.884 | 7.848 | 7.885 | 7.889 |
| Ti | 0.003 | 0.007 | 0.002 | 0.005 | 0.010 | 0.005 | 0.007 | 0.017 | 0.011 | 0.001 | 0.010 | 0.026 | 0.010 | 0.006 | 0.012 |
| Al | 1.986 | 1.697 | 1.712 | 1.341 | 1.516 | 1.543 | 1.631 | 1.666 | 1.648 | 1.788 | 1.575 | 1.858 | 1.744 | 1.699 | 1.590 |
| Cr | 0.001 | 0.002 | 0.000 | 0.003 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.003 | 0.001 | 0.001 | 0.000 | 0.000 | 0.004 |
| $\mathrm{Fe}^{3+}$ | 0.115 | 0.274 | 0.352 | 0.588 | 0.505 | 0.441 | 0.325 | 0.332 | 0.533 | 0.386 | 0.525 | 0.139 | 0.367 | 0.355 | 0.431 |
| $\mathrm{Fe}^{2+}$ | 1.747 | 1.549 | 1.792 | 1.824 | 1.721 | 1.738 | 1.694 | 1.773 | 1.645 | 1.292 | 1.746 | 1.612 | 1.593 | 1.643 | 1.530 |
| Mn | 0.014 | 0.013 | 0.031 | 0.011 | 0.010 | 0.011 | 0.015 | 0.014 | 0.017 | 0.004 | 0.009 | 0.009 | 0.004 | 0.013 | 0.008 |
| Mg | 1.309 | 1.575 | 1.323 | 1.278 | 1.322 | 1.380 | 1.445 | 1.316 | 1.316 | 1.745 | 1.280 | 1.471 | 1.434 | 1.399 | 1.536 |
| Ca | 0.078 | 0.210 | 0.272 | 0.130 | 0.093 | 0.173 | 0.183 | 0.122 | 0.126 | 0.207 | 0.099 | 0.096 | 0.154 | 0.114 | 0.135 |
| Na | 2.077 | 1.824 | 1.806 | 1.893 | 1.938 | 1.890 | 1.895 | 1.955 | 1.879 | 1.837 | 1.967 | 1.988 | 1.864 | 1.929 | 1.899 |
| K | 0.010 | 0.004 | 0.006 | 0.003 | 0.002 | 0.005 | 0.004 | 0.005 | 0.007 | 0.008 | 0.005 | 0.002 | 0.002 | 0.005 | 0.003 |
| Total | 15.165 | 15.038 | 15.084 | 15.027 | 15.033 | 15.068 | 15.082 | 15.082 | 15.011 | 15.052 | 15.071 | 15.086 | 15.020 | 15.049 | 15.037 |

Table A. 1 (continued) : Mineral chemistry of the measured amphiboles. Mineral formula calculations are based on 23 oxygens.

| Na-Amph | $702 \mathrm{~A}^{\text {a }}$ | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 65 | 72 | 74 | 77 | 79 | 83 | 95 | 101 | 105 | 106 | 109 | 110 | 116 | 122 | 123 |
| $\mathrm{SiO}_{2}$ | 56.95 | 56.89 | 55.91 | 56.87 | 56.82 | 55.76 | 56.23 | 56.72 | 57.14 | 56.71 | 55.67 | 57.37 | 55.14 | 55.93 | 56.49 |
| $\mathrm{TiO}_{2}$ | 0.00 | 0.08 | 0.06 | 0.02 | 0.01 | 0.42 | 0.07 | 0.05 | 0.01 | 0.06 | 0.15 | 0.03 | 0.09 | 0.04 | 0.00 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 11.07 | 11.98 | 10.16 | 11.13 | 11.41 | 7.97 | 11.08 | 11.23 | 11.15 | 10.74 | 8.10 | 11.72 | 11.59 | 11.31 | 11.06 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.03 | 0.00 | 0.02 | 0.03 | 0.01 | 0.12 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.03 | 0.01 | 0.00 | 0.02 |
| FeO | 14.83 | 14.61 | 17.72 | 14.83 | 14.84 | 17.28 | 15.85 | 16.09 | 14.45 | 15.81 | 19.97 | 14.82 | 16.19 | 18.07 | 16.86 |
| MnO | 0.01 | 0.10 | 0.07 | 0.04 | 0.02 | 0.03 | 0.06 | 0.11 | 0.02 | 0.10 | 0.13 | 0.10 | 0.20 | 0.18 | 0.21 |
| MgO | 7.73 | 7.01 | 6.16 | 7.81 | 7.59 | 7.84 | 7.01 | 6.53 | 7.77 | 7.28 | 6.21 | 7.04 | 6.63 | 5.49 | 6.48 |
| CaO | 0.87 | 0.39 | 0.87 | 0.84 | 0.64 | 1.55 | 0.97 | 0.33 | 0.98 | 1.24 | 0.96 | 0.24 | 1.43 | 0.82 | 0.71 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 7.13 | 7.17 | 6.90 | 7.17 | 7.33 | 6.75 | 7.05 | 7.53 | 7.05 | 6.41 | 6.88 | 7.57 | 6.89 | 7.02 | 7.04 |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.02 | 0.02 | 0.03 | 0.05 | 0.03 | 0.02 | 0.03 | 0.02 | 0.01 | 0.03 | 0.02 | 0.02 | 0.03 | 0.03 | 0.03 |
| Total | 98.63 | 98.22 | 97.90 | 98.77 | 98.71 | 97.74 | 98.34 | 98.59 | 98.58 | 98.37 | 98.11 | 98.92 | 98.20 | 98.90 | 98.86 |
| Si | 7.881 | 7.883 | 7.900 | 7.857 | 7.856 | 7.902 | 7.852 | 7.906 | 7.904 | 7.879 | 7.917 | 7.918 | 7.752 | 7.840 | 7.858 |
| Ti | 0.000 | 0.008 | 0.007 | 0.002 | 0.001 | 0.045 | 0.008 | 0.005 | 0.001 | 0.006 | 0.016 | 0.003 | 0.009 | 0.005 | 0.000 |
| Al | 1.806 | 1.956 | 1.691 | 1.812 | 1.859 | 1.331 | 1.824 | 1.844 | 1.818 | 1.759 | 1.358 | 1.907 | 1.920 | 1.869 | 1.813 |
| Cr | 0.003 | 0.000 | 0.003 | 0.003 | 0.001 | 0.013 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.003 | 0.002 | 0.000 | 0.000 |
| $\mathrm{Fe}^{3+}$ | 0.254 | 0.218 | 0.336 | 0.288 | 0.264 | 0.434 | 0.254 | 0.199 | 0.188 | 0.370 | 0.581 | 0.152 | 0.243 | 0.281 | 0.356 |
| $\mathrm{Fe}^{2+}$ | 1.461 | 1.475 | 1.757 | 1.426 | 1.452 | 1.615 | 1.597 | 1.676 | 1.483 | 1.467 | 1.794 | 1.559 | 1.661 | 1.837 | 1.605 |
| Mn | 0.001 | 0.011 | 0.009 | 0.004 | 0.003 | 0.003 | 0.007 | 0.012 | 0.002 | 0.012 | 0.016 | 0.012 | 0.024 | 0.021 | 0.025 |
| Mg | 1.594 | 1.449 | 1.298 | 1.608 | 1.564 | 1.657 | 1.459 | 1.357 | 1.603 | 1.507 | 1.317 | 1.448 | 1.390 | 1.148 | 1.343 |
| Ca | 0.129 | 0.057 | 0.132 | 0.125 | 0.095 | 0.235 | 0.144 | 0.049 | 0.145 | 0.185 | 0.146 | 0.035 | 0.215 | 0.123 | 0.106 |
| Na | 1.914 | 1.925 | 1.889 | 1.920 | 1.965 | 1.854 | 1.910 | 2.034 | 1.891 | 1.726 | 1.898 | 2.025 | 1.879 | 1.909 | 1.898 |
| K | 0.004 | 0.004 | 0.005 | 0.008 | 0.005 | 0.004 | 0.005 | 0.003 | 0.002 | 0.005 | 0.004 | 0.004 | 0.005 | 0.005 | 0.005 |
| Total | 15.047 | 14.986 | 15.026 | 15.053 | 15.066 | 15.093 | 15.059 | 15.086 | 15.039 | 14.916 | 15.048 | 15.064 | 15.099 | 15.037 | 15.009 |

Table A. 1 (continued) : Mineral chemistry of the measured amphiboles. Mineral formula calculations are based on 23 oxygens.

| Na-Amph | $775^{\text {a }}$ | 775 | 775 |
| :---: | :---: | :---: | :---: |
|  | 6 | 7 | 17 |
| $\mathrm{SiO}_{2}$ | 57.69 | 57.46 | 57.12 |
| $\mathrm{TiO}_{2}$ | 0.03 | 0.05 | 0.13 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 12.43 | 12.18 | 11.79 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.00 | 0.00 | 0.00 |
| FeO | 12.88 | 13.14 | 13.92 |
| MnO | 0.12 | 0.15 | 0.13 |
| MgO | 7.71 | 7.56 | 7.15 |
| CaO | 0.04 | 0.06 | 0.07 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 7.39 | 7.32 | 7.33 |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.01 | 0.01 | 0.01 |
| Total | 98.29 | 97.91 | 97.64 |
| Si | 7.933 | 7.945 | 7.947 |
| Ti | 0.003 | 0.005 | 0.014 |
| A1 | 2.015 | 1.985 | 1.934 |
| Cr | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{3+}$ | 0.083 | 0.075 | 0.120 |
| $\mathrm{Fe}^{2+}$ | 1.398 | 1.444 | 1.499 |
| Mn | 0.013 | 0.017 | 0.015 |
| Mg | 1.581 | 1.558 | 1.482 |
| Ca | 0.005 | 0.009 | 0.011 |
| Na | 1.970 | 1.962 | 1.977 |
| K | 0.002 | 0.001 | 0.001 |
| Total | 15.002 | 15.001 | 15.001 |

Table A. 1 (continued) : Mineral chemistry of the measured amphiboles. Mineral formula calculations are based on 23 oxygens.

| Ca-Amph | $211 \mathrm{~B}^{\text {a }}$ | 211B | 211B | 211B | 211B | 211B | 702A | 702A | 702A | 702A | 702A | 702A |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 244 | 249 | 261 | 285 | 291 | 324 | 126 | 133 | 21 | 39 | 44 | 56 |
| $\mathrm{SiO}_{2}$ | 52.13 | 52.63 | 52.26 | 52.27 | 52.67 | 52.29 | 51.71 | 54.84 | 50.38 | 49.30 | 51.63 | 53.64 |
| $\mathrm{TiO}_{2}$ | 0.02 | 0.04 | 0.03 | 0.01 | 0.03 | 0.48 | 0.06 | 0.02 | 0.04 | 0.01 | 0.02 | 0.03 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 1.80 | 1.37 | 1.79 | 1.30 | 1.04 | 1.51 | 4.47 | 1.19 | 5.50 | 6.71 | 4.20 | 1.52 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.02 | 0.01 | 0.00 | 0.03 | 0.00 | 0.00 | 0.01 |
| FeO | 19.42 | 17.87 | 19.07 | 17.28 | 17.25 | 17.11 | 16.63 | 13.44 | 19.85 | 19.11 | 16.26 | 16.38 |
| MnO | 0.28 | 0.26 | 0.30 | 0.28 | 0.25 | 0.27 | 0.08 | 0.17 | 0.15 | 0.09 | 0.06 | 0.09 |
| MgO | 11.57 | 12.82 | 12.15 | 12.84 | 12.91 | 12.78 | 12.14 | 15.75 | 10.19 | 10.22 | 14.58 | 13.47 |
| CaO | 9.93 | 10.77 | 9.76 | 10.65 | 11.08 | 10.40 | 11.54 | 11.84 | 10.83 | 11.15 | 9.89 | 11.97 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 1.59 | 1.00 | 1.36 | 0.97 | 0.86 | 1.24 | 1.08 | 0.45 | 1.56 | 1.41 | 0.85 | 0.46 |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.13 | 0.10 | 0.09 | 0.10 | 0.06 | 0.11 | 0.19 | 0.04 | 0.24 | 0.32 | 0.07 | 0.05 |
| Total | 96.87 | 96.86 | 96.81 | 95.72 | 96.15 | 96.18 | 97.90 | 97.72 | 98.78 | 98.31 | 97.55 | 97.61 |
| Si | 7.702 | 7.726 | 7.658 | 7.753 | 7.797 | 7.723 | 7.545 | 7.824 | 7.373 | 7.242 | 7.297 | 7.808 |
| Ti | 0.002 | 0.004 | 0.003 | 0.001 | 0.004 | 0.054 | 0.006 | 0.002 | 0.004 | 0.001 | 0.002 | 0.004 |
| A1 | 0.313 | 0.237 | 0.309 | 0.228 | 0.182 | 0.262 | 0.768 | 0.200 | 0.949 | 1.161 | 0.699 | 0.260 |
| Cr | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.001 | 0.000 | 0.003 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{3+}$ | 0.654 | 0.609 | 0.900 | 0.580 | 0.443 | 0.518 | 0.182 | 0.398 | 0.410 | 0.383 | 1.462 | 0.240 |
| $\mathrm{Fe}^{2+}$ | 1.745 | 1.585 | 1.438 | 1.563 | 1.693 | 1.596 | 1.848 | 1.205 | 2.019 | 1.965 | 0.461 | 1.754 |
| Mn | 0.035 | 0.033 | 0.037 | 0.035 | 0.032 | 0.033 | 0.010 | 0.020 | 0.019 | 0.011 | 0.008 | 0.011 |
| Mg | 2.548 | 2.806 | 2.655 | 2.840 | 2.850 | 2.814 | 2.641 | 3.350 | 2.223 | 2.238 | 3.072 | 2.922 |
| Ca | 1.572 | 1.695 | 1.532 | 1.693 | 1.757 | 1.645 | 1.803 | 1.809 | 1.698 | 1.754 | 1.498 | 1.868 |
| Na | 0.455 | 0.285 | 0.387 | 0.280 | 0.247 | 0.354 | 0.305 | 0.124 | 0.444 | 0.402 | 0.233 | 0.130 |
| K | 0.024 | 0.019 | 0.018 | 0.018 | 0.012 | 0.021 | 0.035 | 0.006 | 0.045 | 0.059 | 0.012 | 0.010 |
| Total | 15.052 | 14.999 | 14.937 | 14.991 | 15.016 | 15.021 | 15.144 | 14.940 | 15.187 | 15.216 | 14.743 | 15.008 |

Table A. 2 : Mineral chemistry of the measured phengitic white micas. Mineral formula calculations are based on 11 oxygens.

| Phengite | 775 | 775 | 775 | 775 | 775 | 775 | 775 | 775 | 775 | 775 | 775 | 775 | 775 | 775 | 775 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 10 | 11 | 23 | 213 | 216 | 218 | 222 | 225 | 228 | 233 | 239 | 240 | 248 | 252 | 255 |
| $\mathrm{SiO}_{2}$ | 52.53 | 53.71 | 47.68 | 51.18 | 50.43 | 49.91 | 51.14 | 49.02 | 51.67 | 50.47 | 52.09 | 50.75 | 51.41 | 47.12 | 47.08 |
| $\mathrm{TiO}_{2}$ | 0.08 | 0.05 | 0.10 | 0.07 | 0.18 | 0.15 | 0.27 | 0.63 | 0.08 | 0.20 | 0.15 | 0.10 | 0.24 | 0.10 | 0.03 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 25.90 | 25.39 | 34.58 | 26.73 | 27.64 | 28.61 | 30.10 | 30.13 | 26.22 | 32.51 | 27.77 | 27.43 | 32.57 | 35.91 | 35.65 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.02 | 0.00 | 0.01 | 0.04 | 0.02 | 0.01 | 0.02 | 0.03 | 0.02 | 0.06 | 0.03 | 0.02 | 0.00 | 0.03 | 0.00 |
| FeO | 2.44 | 2.33 | 1.93 | 2.75 | 3.91 | 3.83 | 2.93 | 2.25 | 2.58 | 3.12 | 3.17 | 3.12 | 1.85 | 0.82 | 1.32 |
| MnO | 0.04 | 0.06 | 0.03 | 0.10 | 0.08 | 0.04 | 0.10 | 0.04 | 0.05 | 0.14 | 0.07 | 0.04 | 0.06 | 0.02 | 0.06 |
| MgO | 3.85 | 4.17 | 1.17 | 3.45 | 2.33 | 2.27 | 2.45 | 2.08 | 3.44 | 1.32 | 2.67 | 2.77 | 1.46 | 0.50 | 0.59 |
| CaO | 0.01 | 0.00 | 0.07 | 0.02 | 0.00 | 0.01 | 0.02 | 0.01 | 0.01 | 0.07 | 0.03 | 0.00 | 0.09 | 0.00 | 0.09 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.11 | 0.10 | 0.53 | 0.22 | 0.25 | 0.29 | 0.34 | 0.41 | 0.12 | 0.45 | 0.23 | 0.29 | 0.32 | 0.73 | 1.75 |
| $\mathrm{K}_{2} \mathrm{O}$ | 10.35 | 10.75 | 10.09 | 10.60 | 10.47 | 10.23 | 9.32 | 10.34 | 10.54 | 9.58 | 8.29 | 10.63 | 8.83 | 9.81 | 8.41 |
| Total | 95.33 | 96.57 | 96.18 | 95.16 | 95.30 | 95.34 | 96.68 | 94.93 | 94.73 | 97.91 | 94.50 | 95.14 | 96.81 | 95.05 | 94.99 |
| Si | 3.491 | 3.526 | 3.139 | 3.427 | 3.388 | 3.346 | 3.340 | 3.282 | 3.466 | 3.264 | 3.456 | 3.404 | 3.316 | 3.116 | 3.112 |
| Ti | 0.004 | 0.003 | 0.005 | 0.004 | 0.009 | 0.007 | 0.013 | 0.032 | 0.004 | 0.010 | 0.007 | 0.005 | 0.011 | 0.005 | 0.002 |
| Al | 2.028 | 1.964 | 2.683 | 2.109 | 2.188 | 2.261 | 2.317 | 2.377 | 2.072 | 2.478 | 2.172 | 2.168 | 2.476 | 2.799 | 2.777 |
| Cr | 0.001 | 0.000 | 0.001 | 0.002 | 0.001 | 0.000 | 0.001 | 0.002 | 0.001 | 0.003 | 0.002 | 0.001 | 0.000 | 0.002 | 0.000 |
| $\mathrm{Fe}^{3+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{2+}$ | 0.136 | 0.128 | 0.106 | 0.154 | 0.220 | 0.214 | 0.160 | 0.126 | 0.145 | 0.169 | 0.176 | 0.175 | 0.100 | 0.046 | 0.073 |
| Mn | 0.002 | 0.003 | 0.002 | 0.006 | 0.005 | 0.002 | 0.006 | 0.002 | 0.003 | 0.008 | 0.004 | 0.002 | 0.003 | 0.001 | 0.003 |
| Mg | 0.381 | 0.408 | 0.115 | 0.345 | 0.233 | 0.227 | 0.238 | 0.208 | 0.344 | 0.127 | 0.265 | 0.277 | 0.140 | 0.049 | 0.058 |
| Ca | 0.001 | 0.000 | 0.005 | 0.001 | 0.000 | 0.001 | 0.001 | 0.001 | 0.001 | 0.005 | 0.002 | 0.000 | 0.006 | 0.000 | 0.007 |
| Na | 0.015 | 0.012 | 0.067 | 0.028 | 0.032 | 0.037 | 0.044 | 0.053 | 0.016 | 0.056 | 0.029 | 0.037 | 0.040 | 0.094 | 0.224 |
| K | 0.877 | 0.900 | 0.848 | 0.905 | 0.897 | 0.875 | 0.776 | 0.883 | 0.901 | 0.790 | 0.701 | 0.910 | 0.726 | 0.827 | 0.709 |
| Total | 6.936 | 6.945 | 6.971 | 6.981 | 6.973 | 6.972 | 6.897 | 6.965 | 6.953 | 6.909 | 6.815 | 6.980 | 6.818 | 6.939 | 6.964 |


| Phengite | 775 | 775 | 775 | 775 | 775 | 775 | 775 | 775 | 775 | 753A | 753A | 753A | 753A | 753A |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 257 | 260 | 264 | 267 | 270 | 273 | 278 | 281 | 282 | 75 | 76 | 78 | 79 | 81 |
| $\mathrm{SiO}_{2}$ | 48.10 | 48.20 | 50.29 | 46.48 | 50.82 | 51.79 | 51.46 | 51.60 | 52.31 | 51.04 | 51.30 | 50.60 | 50.01 | 48.55 |
| $\mathrm{TiO}_{2}$ | 0.26 | 0.23 | 0.16 | 0.12 | 0.09 | 0.07 | 0.09 | 0.05 | 0.07 | 0.13 | 0.09 | 0.10 | 0.17 | 0.16 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 31.02 | 32.37 | 28.19 | 37.15 | 27.34 | 26.66 | 26.59 | 27.07 | 25.46 | 27.72 | 27.58 | 27.34 | 28.13 | 30.04 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.00 | 0.01 | 0.04 | 0.02 | 0.04 | 0.03 | 0.00 | 0.00 | 0.00 | 0.00 | 0.03 | 0.04 | 0.00 | 0.04 |
| FeO | 2.59 | 2.29 | 2.77 | 1.04 | 2.92 | 3.02 | 2.91 | 2.51 | 2.26 | 2.90 | 2.44 | 2.64 | 3.33 | 3.20 |
| MnO | 0.06 | 0.12 | 0.10 | 0.02 | 0.03 | 0.08 | 0.08 | 0.04 | 0.04 | 0.07 | 0.03 | 0.03 | 0.06 | 0.09 |
| MgO | 1.48 | 1.23 | 2.73 | 0.49 | 3.14 | 3.46 | 3.41 | 3.39 | 3.89 | 2.82 | 3.10 | 2.97 | 2.50 | 1.83 |
| CaO | 0.05 | 0.05 | 0.05 | 0.16 | 0.01 | 0.02 | 0.01 | 0.02 | 0.00 | 0.00 | 0.00 | 0.02 | 0.00 | 0.00 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.58 | 0.59 | 0.26 | 3.16 | 0.20 | 0.19 | 0.13 | 0.18 | 0.10 | 0.20 | 0.13 | 0.19 | 0.24 | 0.34 |
| $\mathrm{K}_{2} \mathrm{O}$ | 10.24 | 10.15 | 10.34 | 6.67 | 10.55 | 10.29 | 10.29 | 10.48 | 10.76 | 10.31 | 10.41 | 10.29 | 10.34 | 10.18 |
| Total | 94.37 | 95.23 | 94.92 | 95.31 | 95.14 | 95.60 | 94.97 | 95.33 | 94.88 | 95.19 | 95.09 | 94.21 | 94.76 | 94.42 |
| Si | 3.246 | 3.215 | 3.372 | 3.044 | 3.404 | 3.445 | 3.444 | 3.436 | 3.500 | 3.408 | 3.421 | 3.411 | 3.367 | 3.281 |
| Ti | 0.013 | 0.012 | 0.008 | 0.006 | 0.004 | 0.003 | 0.005 | 0.002 | 0.003 | 0.006 | 0.005 | 0.005 | 0.009 | 0.008 |
| A1 | 2.467 | 2.544 | 2.228 | 2.867 | 2.158 | 2.090 | 2.097 | 2.124 | 2.008 | 2.182 | 2.167 | 2.172 | 2.232 | 2.392 |
| Cr | 0.000 | 0.001 | 0.002 | 0.001 | 0.002 | 0.001 | 0.000 | 0.000 | 0.000 | 0.000 | 0.001 | 0.002 | 0.000 | 0.002 |
| $\mathrm{Fe}^{3+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{2+}$ | 0.146 | 0.128 | 0.156 | 0.057 | 0.163 | 0.168 | 0.163 | 0.140 | 0.126 | 0.162 | 0.136 | 0.149 | 0.187 | 0.181 |
| Mn | 0.003 | 0.007 | 0.005 | 0.001 | 0.002 | 0.005 | 0.005 | 0.002 | 0.002 | 0.004 | 0.002 | 0.002 | 0.003 | 0.005 |
| Mg | 0.149 | 0.122 | 0.273 | 0.048 | 0.313 | 0.343 | 0.340 | 0.336 | 0.388 | 0.281 | 0.308 | 0.298 | 0.251 | 0.185 |
| Ca | 0.004 | 0.004 | 0.003 | 0.011 | 0.001 | 0.002 | 0.001 | 0.001 | 0.000 | 0.000 | 0.000 | 0.002 | 0.000 | 0.000 |
| Na | 0.076 | 0.076 | 0.034 | 0.401 | 0.026 | 0.024 | 0.017 | 0.023 | 0.012 | 0.026 | 0.016 | 0.025 | 0.031 | 0.045 |
| K | 0.881 | 0.863 | 0.885 | 0.557 | 0.902 | 0.873 | 0.879 | 0.890 | 0.918 | 0.878 | 0.886 | 0.885 | 0.888 | 0.877 |
| Total | 6.986 | 6.971 | 6.965 | 6.995 | 6.975 | 6.954 | 6.950 | 6.956 | 6.958 | 6.947 | 6.941 | 6.951 | 6.968 | 6.975 |

Table A. 2 (continued) : Mineral chemistry of the measured phengitic white micas. Mineral formula calculations are based on 11 oxygens.

| Phengite | 753 A | 753 A | 753 A | 753 A | 753 A | 753 A | 753 A | 753 A | 753 A | 753 A | 753 A | 753 A | 753 A |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 83 | 92 | 93 | 108 | 112 | 123 | 127 | 130 | 131 | 139 | 143 | 146 | 148 |
| $\mathrm{SiO}_{2}$ | 50.66 | 50.93 | 50.53 | 50.41 | 48.96 | 51.15 | 48.18 | 50.28 | 50.85 | 49.46 | 48.69 | 50.11 | 49.62 |
| $\mathrm{TiO}_{2}$ | 0.08 | 0.06 | 0.17 | 0.14 | 0.07 | 0.06 | 0.14 | 0.15 | 0.07 | 0.10 | 0.05 | 0.13 | 0.18 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 27.21 | 27.34 | 27.98 | 27.51 | 30.49 | 25.94 | 31.40 | 28.12 | 26.92 | 29.19 | 32.64 | 28.73 | 28.71 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.01 | 0.05 | 0.01 | 0.04 | 0.04 | 0.01 | 0.07 | 0.00 | 0.01 | 0.04 | 0.01 | 0.00 | 0.01 |
| FeO | 2.43 | 2.46 | 3.70 | 2.90 | 2.62 | 2.99 | 3.12 | 3.26 | 2.72 | 3.15 | 1.96 | 3.01 | 2.75 |
| MnO | 0.06 | 0.05 | 0.10 | 0.06 | 0.08 | 0.06 | 0.15 | 0.07 | 0.05 | 0.07 | 0.06 | 0.05 | 0.03 |
| MgO | 3.05 | 2.94 | 2.25 | 2.59 | 1.98 | 3.25 | 1.49 | 2.41 | 3.33 | 2.07 | 1.48 | 2.15 | 2.43 |
| CaO | 0.00 | 0.02 | 0.00 | 0.01 | 0.00 | 0.00 | 0.01 | 0.00 | 0.00 | 0.00 | 0.00 | 0.01 | 0.00 |
| Na 2 O | 0.12 | 0.18 | 0.16 | 0.16 | 0.34 | 0.10 | 0.33 | 0.24 | 0.16 | 0.23 | 0.42 | 0.33 | 0.22 |
| $\mathrm{~K}_{2} \mathrm{O}$ | 10.48 | 10.37 | 10.36 | 10.26 | 10.14 | 10.24 | 9.90 | 10.41 | 10.33 | 10.30 | 10.02 | 10.07 | 10.14 |
| Total | 94.10 | 94.40 | 95.25 | 94.06 | 94.71 | 93.79 | 94.77 | 94.93 | 94.44 | 94.61 | 95.33 | 94.59 | 94.09 |
|  |  |  |  |  |  |  |  |  |  | 94.32 |  |  |  |
| Si | 3.419 | 3.424 | 3.388 | 3.407 | 3.285 | 3.467 | 3.236 | 3.378 | 3.423 | 3.332 | 3.229 | 3.367 | 3.351 |
| Ti | 0.004 | 0.003 | 0.009 | 0.007 | 0.003 | 0.003 | 0.007 | 0.007 | 0.003 | 0.005 | 0.003 | 0.007 | 0.009 |
| Al | 2.164 | 2.166 | 2.211 | 2.191 | 2.411 | 2.072 | 2.486 | 2.226 | 2.136 | 2.317 | 2.551 | 2.275 | 2.285 |
| Cr | 0.000 | 0.003 | 0.000 | 0.002 | 0.002 | 0.001 | 0.004 | 0.000 | 0.001 | 0.002 | 0.000 | 0.000 | 0.001 |
| Cl | 0.056 |  |  |  |  |  |  |  |  |  |  |  |  |
| $\mathrm{Fe}^{3+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}{ }^{2+}$ | 0.137 | 0.138 | 0.207 | 0.164 | 0.147 | 0.170 | 0.175 | 0.183 | 0.153 | 0.177 | 0.109 | 0.169 | 0.155 |
| Mn | 0.003 | 0.003 | 0.005 | 0.003 | 0.004 | 0.003 | 0.008 | 0.004 | 0.003 | 0.004 | 0.003 | 0.003 | 0.002 |
| Mg | 0.307 | 0.294 | 0.224 | 0.261 | 0.198 | 0.328 | 0.149 | 0.241 | 0.334 | 0.208 | 0.146 | 0.215 | 0.245 |
| Ca | 0.000 | 0.001 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.001 | 0.000 |
| Na | 0.016 | 0.024 | 0.021 | 0.021 | 0.045 | 0.013 | 0.043 | 0.031 | 0.021 | 0.030 | 0.054 | 0.043 | 0.029 |
| K | 0.902 | 0.889 | 0.886 | 0.885 | 0.868 | 0.886 | 0.849 | 0.892 | 0.887 | 0.885 | 0.848 | 0.863 | 0.874 |
| K | 0.029 |  |  |  |  |  |  |  |  |  |  |  |  |
| Total | 6.954 | 6.946 | 6.952 | 6.942 | 6.962 | 6.943 | 6.957 | 6.963 | 6.960 | 6.961 | 6.943 | 6.942 | 6.949 |

Table A. 2 (continued) : Mineral chemistry of the measured phengitic white micas. Mineral formula calculations are based on 11 oxygens.

| Phengite | 178 | 178 | 178 | 178 | 178 | 178 | 178 | 178 | 178 | 178 | 178 | 178 | 178 | 178 |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 22 | 23 | 24 | 29 | 30 | 33 | 37 | 42 | 45 | 46 | 47 | 52 | 56 | 57 |
| $\mathrm{SiO}_{2}$ | 46.70 | 50.75 | 50.58 | 48.06 | 51.07 | 51.64 | 50.79 | 51.05 | 46.06 | 50.58 | 46.98 | 50.10 | 50.22 | 46.70 |
| $\mathrm{TiO}_{2}$ | 0.16 | 0.11 | 0.17 | 0.20 | 0.15 | 0.10 | 0.12 | 0.22 | 0.13 | 0.16 | 0.18 | 0.14 | 0.16 | 0.09 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 35.05 | 27.40 | 27.23 | 31.78 | 27.51 | 27.10 | 27.50 | 27.10 | 34.85 | 27.91 | 34.19 | 27.78 | 28.08 | 34.99 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.00 | 0.03 | 0.05 | 0.02 | 0.00 | 0.02 | 0.00 | 0.02 | 0.03 | 0.04 | 0.02 | 0.05 | 0.02 | 0.09 |
| FeO | 1.81 | 2.58 | 3.26 | 3.14 | 2.50 | 1.96 | 2.35 | 3.05 | 2.32 | 3.10 | 2.06 | 2.56 | 3.36 | 1.77 |
| MnO | 0.03 | 0.02 | 0.03 | 0.02 | 0.03 | 0.01 | 0.00 | 0.03 | 0.03 | 0.01 | 0.01 | 0.03 | 0.01 | 0.01 |
| MgO | 0.72 | 2.97 | 2.59 | 1.36 | 2.99 | 3.24 | 3.11 | 3.01 | 0.73 | 2.57 | 0.89 | 2.79 | 2.53 | 0.84 |
| CaO | 0.00 | 0.00 | 0.00 | 0.01 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 |
| $\mathrm{Na} \mathrm{O}_{2} \mathrm{O}$ | 0.70 | 0.22 | 0.27 | 0.39 | 0.24 | 0.13 | 0.32 | 0.22 | 0.56 | 0.27 | 0.68 | 0.29 | 0.21 | 0.63 |
| $\mathrm{~K}_{2} \mathrm{O}$ | 9.44 | 10.18 | 10.22 | 9.91 | 10.19 | 10.49 | 10.22 | 10.11 | 9.52 | 10.40 | 9.75 | 10.08 | 10.17 | 9.79 |
| Total | 94.61 | 94.26 | 94.39 | 94.89 | 94.67 | 94.68 | 94.41 | 94.81 | 94.23 | 95.03 | 94.75 | 93.82 | 94.77 | 94.90 |
|  |  |  |  |  |  |  | 93.78 |  |  |  |  |  |  |  |
| Si | 3.115 | 3.415 | 3.413 | 3.223 | 3.420 | 3.450 | 3.411 | 3.423 | 3.096 | 3.390 | 3.139 | 3.389 | 3.376 | 3.112 |
| Ti | 0.008 | 0.006 | 0.009 | 0.010 | 0.007 | 0.005 | 0.006 | 0.011 | 0.006 | 0.008 | 0.009 | 0.007 | 0.008 | 0.005 |
| Al | 2.756 | 2.173 | 2.166 | 2.512 | 2.171 | 2.134 | 2.177 | 2.141 | 2.761 | 2.205 | 2.692 | 2.215 | 2.225 | 2.748 |
| Cr | 0.000 | 0.002 | 0.002 | 0.001 | 0.000 | 0.001 | 0.000 | 0.001 | 0.002 | 0.002 | 0.001 | 0.003 | 0.001 | 0.004 |
| Cr | 0.001 |  |  |  |  |  |  |  |  |  |  |  |  |  |
| $\mathrm{Fe}^{3+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{2+}$ | 0.101 | 0.145 | 0.184 | 0.176 | 0.140 | 0.110 | 0.132 | 0.171 | 0.131 | 0.174 | 0.115 | 0.145 | 0.189 | 0.098 |
| Mn | 0.002 | 0.001 | 0.002 | 0.001 | 0.001 | 0.001 | 0.000 | 0.001 | 0.002 | 0.001 | 0.001 | 0.002 | 0.001 | 0.000 |
| Mg | 0.072 | 0.298 | 0.260 | 0.136 | 0.298 | 0.322 | 0.311 | 0.301 | 0.073 | 0.257 | 0.089 | 0.281 | 0.253 | 0.083 |
| Ca | 0.000 | 0.000 | 0.000 | 0.001 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| Na | 0.091 | 0.028 | 0.036 | 0.051 | 0.031 | 0.017 | 0.041 | 0.029 | 0.073 | 0.035 | 0.088 | 0.038 | 0.028 | 0.082 |
| Na | 0.031 |  |  |  |  |  |  |  |  |  |  |  |  |  |
| K | 0.803 | 0.874 | 0.880 | 0.848 | 0.871 | 0.894 | 0.876 | 0.865 | 0.817 | 0.889 | 0.831 | 0.869 | 0.872 | 0.832 |
| Total | 6.947 | 6.943 | 6.952 | 6.960 | 6.939 | 6.933 | 6.953 | 6.943 | 6.961 | 6.961 | 6.965 | 6.949 | 6.953 | 6.964 |
| 6.954 |  |  |  |  |  |  |  |  |  |  |  |  |  |  |

Table A. 2 (continued) : Mineral chemistry of the measured phengitic white micas. Mineral formula calculations are based on 11 oxygens.

| Phengite | 178 | 178 | 178 | 392 | 392 | 392 | 392 | 392 | 392 | 392 | 392 | 392 | 392 | 392 | 392 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 64 | 95 | 102 | 1 | 3 | 5 | 6 | 9 | 11 | 14 | 15 | 18 | 20 | 21 | 23 |
| $\mathrm{SiO}_{2}$ | 49.77 | 51.02 | 50.77 | 51.84 | 51.76 | 51.51 | 52.25 | 51.09 | 51.76 | 51.92 | 51.28 | 51.56 | 51.35 | 51.82 | 50.86 |
| $\mathrm{TiO}_{2}$ | 0.16 | 0.07 | 0.16 | 0.18 | 0.15 | 0.09 | 0.19 | 0.19 | 0.25 | 0.18 | 0.15 | 0.07 | 0.16 | 0.19 | 0.15 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 27.91 | 27.11 | 26.92 | 30.59 | 30.94 | 31.94 | 28.48 | 28.83 | 31.67 | 27.90 | 31.75 | 31.30 | 28.08 | 30.87 | 31.31 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.00 | 0.00 | 0.01 |  |  |  |  |  |  |  |  |  |  |  |  |
| FeO | 3.45 | 2.27 | 2.95 | 1.88 | 1.62 | 1.40 | 4.03 | 3.81 | 1.84 | 3.52 | 1.76 | 1.60 | 3.48 | 2.28 | 2.11 |
| MnO | 0.01 | 0.00 | 0.00 | 0.01 | 0.01 | 0.00 | 0.03 | 0.01 | 0.02 | 0.02 | 0.00 | 0.00 | 0.01 | 0.00 | 0.00 |
| MgO | 2.55 | 3.26 | 2.79 | 1.41 | 1.27 | 1.28 | 2.45 | 2.72 | 1.35 | 2.68 | 1.47 | 1.40 | 2.69 | 1.42 | 1.46 |
| CaO | 0.00 | 0.01 | 0.00 | 0.03 | 0.02 | 0.02 | 0.09 | 0.02 | 0.02 | 0.02 | 0.04 | 0.02 | 0.00 | 0.00 | 0.01 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.24 | 0.17 | 0.20 | 0.49 | 0.45 | 0.54 | 0.25 | 0.29 | 0.45 | 0.38 | 0.46 | 0.51 | 0.29 | 0.50 | 0.41 |
| $\mathrm{K}_{2} \mathrm{O}$ | 9.87 | 10.31 | 10.21 | 9.72 | 9.85 | 9.64 | 8.18 | 9.82 | 9.68 | 10.06 | 9.53 | 9.76 | 9.96 | 9.76 | 9.68 |
| Total | 93.97 | 94.22 | 94.01 | 96.15 | 96.10 | 96.43 | 96.01 | 96.79 | 97.04 | 96.68 | 96.44 | 96.22 | 96.03 | 96.85 | 95.99 |
| Si | 3.372 | 3.431 | 3.432 | 3.386 | 3.380 | 3.345 | 3.426 | 3.360 | 3.347 | 3.416 | 3.335 | 3.362 | 3.400 | 3.368 | 3.334 |
| Ti | 0.008 | 0.004 | 0.008 | 0.009 | 0.007 | 0.004 | 0.009 | 0.009 | 0.012 | 0.009 | 0.007 | 0.003 | 0.008 | 0.009 | 0.007 |
| Al | 2.229 | 2.149 | 2.145 | 2.355 | 2.381 | 2.444 | 2.201 | 2.235 | 2.414 | 2.163 | 2.434 | 2.405 | 2.191 | 2.365 | 2.419 |
| Cr | 0.000 | 0.000 | 0.001 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{3+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{2+}$ | 0.196 | 0.128 | 0.167 | 0.103 | 0.088 | 0.076 | 0.221 | 0.210 | 0.100 | 0.194 | 0.096 | 0.087 | 0.193 | 0.124 | 0.116 |
| Mn | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.002 | 0.000 | 0.001 | 0.001 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| Mg | 0.258 | 0.327 | 0.281 | 0.137 | 0.123 | 0.124 | 0.240 | 0.267 | 0.130 | 0.263 | 0.142 | 0.136 | 0.266 | 0.138 | 0.142 |
| Ca | 0.000 | 0.001 | 0.000 | 0.002 | 0.002 | 0.001 | 0.007 | 0.002 | 0.002 | 0.002 | 0.002 | 0.001 | 0.000 | 0.000 | 0.000 |
| Na | 0.031 | 0.022 | 0.026 | 0.062 | 0.057 | 0.068 | 0.032 | 0.036 | 0.056 | 0.049 | 0.058 | 0.064 | 0.037 | 0.063 | 0.052 |
| K | 0.853 | 0.884 | 0.880 | 0.810 | 0.821 | 0.799 | 0.684 | 0.824 | 0.799 | 0.844 | 0.791 | 0.812 | 0.841 | 0.809 | 0.809 |
| Total | 6.947 | 6.945 | 6.940 | 6.864 | 6.860 | 6.862 | 6.822 | 6.943 | 6.861 | 6.940 | 6.865 | 6.870 | 6.936 | 6.877 | 6.880 |

Table A. 2 (continued) : Mineral chemistry of the measured phengitic white micas. Mineral formula calculations are based on 11 oxygens.

| Phengite | 392 | 392 | 392 | 392 | 392 | 392 | 392 | 392 | 392 | 392 | 392 | 392 | 392 | 392 |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 24 | 26 | 27 | 29 | 33 | 34 | 35 | 36 | 37 | 40 | 41 | 43 | 44 | 46 |
| $\mathrm{SiO}_{2}$ | 51.38 | 51.34 | 51.20 | 51.22 | 53.10 | 52.47 | 51.45 | 52.54 | 52.24 | 51.20 | 51.49 | 51.07 | 51.90 | 51.51 |
| $\mathrm{TiO}_{2}$ | 0.31 | 0.24 | 0.19 | 0.13 | 0.17 | 0.17 | 0.16 | 0.18 | 0.23 | 0.20 | 0.14 | 0.20 | 0.15 | 0.15 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 29.31 | 31.87 | 31.31 | 31.83 | 27.05 | 27.30 | 27.76 | 27.65 | 30.29 | 30.96 | 31.11 | 27.77 | 27.37 | 26.54 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ |  |  |  |  |  |  |  |  |  |  | 28.01 |  |  |  |
| FeO | 3.95 | 1.73 | 1.91 | 1.57 | 3.27 | 3.33 | 3.38 | 3.20 | 1.92 | 2.23 | 1.89 | 3.64 | 3.19 | 3.42 |
| MnO | 0.02 | 0.01 | 0.01 | 0.02 | 0.00 | 0.00 | 0.00 | 0.01 | 0.00 | 0.00 | 0.00 | 0.01 | 0.00 | 0.02 |
| MgO | 1.98 | 1.19 | 1.39 | 1.13 | 2.88 | 2.63 | 2.81 | 2.93 | 1.60 | 1.39 | 1.52 | 2.39 | 2.94 | 3.00 |
| CaO | 0.03 | 0.02 | 0.03 | 0.01 | 0.06 | 0.02 | 0.00 | 0.01 | 0.00 | 0.01 | 0.03 | 0.00 | 0.01 | 0.15 |
| Na 2 O | 0.29 | 0.46 | 0.44 | 0.57 | 0.30 | 0.39 | 0.28 | 0.34 | 0.35 | 0.52 | 0.35 | 0.32 | 0.37 | 0.23 |
| $\mathrm{~K}_{2} \mathrm{O}$ | 9.35 | 9.68 | 9.59 | 9.68 | 9.93 | 9.91 | 10.03 | 9.86 | 9.71 | 9.43 | 9.69 | 9.95 | 10.17 | 11.18 |
| Total | 96.63 | 96.54 | 96.08 | 96.17 | 96.85 | 96.22 | 95.88 | 96.72 | 96.35 | 95.96 | 96.23 | 95.36 | 96.11 | 96.23 |
|  |  |  |  |  |  |  |  |  |  |  | 96.55 |  |  |  |
| Si | 3.372 | 3.337 | 3.347 | 3.340 | 3.477 | 3.459 | 3.411 | 3.442 | 3.402 | 3.354 | 3.360 | 3.409 | 3.433 | 3.433 |
| Ti | 0.015 | 0.012 | 0.010 | 0.006 | 0.008 | 0.008 | 0.008 | 0.009 | 0.011 | 0.010 | 0.007 | 0.010 | 0.007 | 0.007 |
| Al | 2.267 | 2.441 | 2.412 | 2.447 | 2.087 | 2.121 | 2.169 | 2.135 | 2.325 | 2.390 | 2.392 | 2.184 | 2.133 | 2.085 |
| $\mathrm{Al}^{2}$ | 0.009 |  |  |  |  |  |  |  |  |  |  |  |  |  |
| Cr | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{3+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{2+}$ | 0.217 | 0.094 | 0.104 | 0.086 | 0.179 | 0.184 | 0.187 | 0.175 | 0.105 | 0.122 | 0.103 | 0.203 | 0.176 | 0.191 |
| Mn | 0.001 | 0.001 | 0.001 | 0.001 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.001 |
| Mg | 0.194 | 0.115 | 0.136 | 0.110 | 0.281 | 0.258 | 0.278 | 0.286 | 0.156 | 0.136 | 0.148 | 0.238 | 0.290 | 0.298 |
| Ca | 0.002 | 0.001 | 0.002 | 0.001 | 0.004 | 0.001 | 0.000 | 0.001 | 0.000 | 0.001 | 0.002 | 0.000 | 0.001 | 0.011 |
| Na | 0.037 | 0.058 | 0.055 | 0.072 | 0.038 | 0.049 | 0.036 | 0.043 | 0.044 | 0.066 | 0.045 | 0.041 | 0.048 | 0.030 |
| K | 0.783 | 0.803 | 0.800 | 0.805 | 0.829 | 0.833 | 0.848 | 0.824 | 0.807 | 0.788 | 0.807 | 0.847 | 0.858 | 0.951 |
| K | 0.860 |  |  |  |  |  |  |  |  |  |  |  |  |  |
| Total | 6.889 | 6.861 | 6.865 | 6.869 | 6.905 | 6.914 | 6.938 | 6.915 | 6.849 | 6.868 | 6.863 | 6.933 | 6.946 | 7.007 |

Table A. 2 (continued) : Mineral chemistry of the measured phengitic white micas. Mineral formula calculations are based on 11 oxygens.

| Phengite | 626A | 626A | 626A | 626A | 626A | 626A | 626A | 626A | 626A | 626A | 626A | 626A | 626A | 626A | 626A |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 47 | 48 | 51 | 52 | 53 | 58 | 60 | 61 | 64 | 65 | 67 | 75 | 77 | 79 | 83 |
| $\mathrm{SiO}_{2}$ | 49.78 | 50.36 | 49.98 | 50.30 | 50.34 | 49.91 | 50.43 | 50.15 | 50.57 | 51.01 | 50.62 | 50.97 | 50.35 | 50.96 | 51.40 |
| $\mathrm{TiO}_{2}$ | 0.11 | 0.10 | 0.11 | 0.12 | 0.08 | 0.12 | 0.16 | 0.10 | 0.12 | 0.07 | 0.16 | 0.13 | 0.14 | 0.10 | 0.09 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 26.39 | 25.85 | 26.22 | 25.68 | 25.73 | 25.69 | 25.96 | 25.71 | 26.14 | 26.37 | 26.35 | 25.94 | 25.92 | 25.68 | 25.39 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.00 | 0.00 | 0.02 | 0.03 | 0.02 | 0.00 | 0.00 | 0.02 | 0.03 | 0.00 | 0.01 | 0.01 | 0.01 | 0.04 | 0.01 |
| FeO | 4.03 | 4.39 | 4.13 | 4.26 | 4.17 | 4.23 | 4.39 | 4.25 | 4.31 | 4.13 | 4.40 | 4.03 | 4.30 | 4.78 | 4.18 |
| MnO | 0.05 | 0.05 | 0.03 | 0.04 | 0.01 | 0.05 | 0.06 | 0.08 | 0.06 | 0.04 | 0.04 | 0.04 | 0.08 | 0.06 | 0.07 |
| MgO | 3.61 | 3.73 | 3.55 | 3.54 | 3.61 | 3.46 | 3.59 | 3.72 | 3.61 | 3.69 | 3.45 | 3.70 | 3.37 | 3.87 | 3.73 |
| CaO | 0.01 | 0.02 | 0.00 | 0.00 | 0.01 | 0.00 | 0.00 | 0.00 | 0.01 | 0.01 | 0.00 | 0.00 | 0.02 | 0.01 | 0.00 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.39 | 0.28 | 0.41 | 0.26 | 0.23 | 0.34 | 0.31 | 0.22 | 0.23 | 0.18 | 0.32 | 0.33 | 0.36 | 0.25 | 0.18 |
| $\mathrm{K}_{2} \mathrm{O}$ | 10.62 | 10.83 | 10.57 | 10.76 | 10.91 | 10.64 | 10.79 | 10.98 | 10.86 | 11.12 | 10.86 | 10.59 | 10.62 | 10.84 | 11.00 |
| Total | 94.99 | 95.61 | 95.00 | 95.00 | 95.12 | 94.42 | 95.69 | 95.23 | 95.94 | 96.63 | 96.20 | 95.74 | 95.16 | 96.59 | 96.04 |
| Si | 3.373 | 3.398 | 3.385 | 3.411 | 3.410 | 3.405 | 3.398 | 3.398 | 3.397 | 3.400 | 3.392 | 3.419 | 3.407 | 3.407 | 3.445 |
| Ti | 0.006 | 0.005 | 0.005 | 0.006 | 0.004 | 0.006 | 0.008 | 0.005 | 0.006 | 0.004 | 0.008 | 0.006 | 0.007 | 0.005 | 0.005 |
| Al | 2.108 | 2.056 | 2.094 | 2.052 | 2.054 | 2.065 | 2.062 | 2.054 | 2.070 | 2.072 | 2.081 | 2.050 | 2.067 | 2.024 | 2.005 |
| Cr | 0.000 | 0.000 | 0.001 | 0.002 | 0.001 | 0.000 | 0.000 | 0.001 | 0.002 | 0.000 | 0.001 | 0.000 | 0.000 | 0.002 | 0.001 |
| $\mathrm{Fe}^{3+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{2+}$ | 0.229 | 0.248 | 0.234 | 0.242 | 0.236 | 0.241 | 0.247 | 0.241 | 0.242 | 0.230 | 0.246 | 0.226 | 0.243 | 0.267 | 0.234 |
| Mn | 0.003 | 0.003 | 0.001 | 0.002 | 0.001 | 0.003 | 0.003 | 0.005 | 0.003 | 0.002 | 0.002 | 0.002 | 0.004 | 0.004 | 0.004 |
| Mg | 0.364 | 0.375 | 0.358 | 0.358 | 0.365 | 0.352 | 0.361 | 0.376 | 0.362 | 0.367 | 0.345 | 0.370 | 0.340 | 0.386 | 0.372 |
| Ca | 0.001 | 0.001 | 0.000 | 0.000 | 0.001 | 0.000 | 0.000 | 0.000 | 0.001 | 0.001 | 0.000 | 0.000 | 0.002 | 0.000 | 0.000 |
| Na | 0.051 | 0.036 | 0.054 | 0.035 | 0.030 | 0.044 | 0.041 | 0.029 | 0.030 | 0.023 | 0.042 | 0.043 | 0.047 | 0.032 | 0.023 |
| K | 0.918 | 0.932 | 0.913 | 0.931 | 0.943 | 0.926 | 0.928 | 0.950 | 0.931 | 0.946 | 0.928 | 0.906 | 0.916 | 0.924 | 0.940 |
| Total | 7.052 | 7.054 | 7.046 | 7.038 | 7.045 | 7.042 | 7.047 | 7.058 | 7.042 | 7.045 | 7.045 | 7.024 | 7.034 | 7.053 | 7.030 |

Table A. 2 (continued) : Mineral chemistry of the measured phengitic white micas. Mineral formula calculations are based on 11 oxygens.

| Phengite | 626A | 626A | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 84 | 86 | 18 | 27 | 28 | 45 | 46 | 61 | 92 | 96 | 128 | 131 |
| $\mathrm{SiO}_{2}$ | 50.49 | 50.94 | 52.31 | 53.47 | 50.94 | 51.13 | 50.27 | 51.96 | 52.47 | 51.68 | 51.25 | 51.97 |
| $\mathrm{TiO}_{2}$ | 0.14 | 0.15 | 0.12 | 0.11 | 0.12 | 0.05 | 0.08 | 0.09 | 0.04 | 0.08 | 0.12 | 0.07 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 25.90 | 25.99 | 25.83 | 25.70 | 25.11 | 24.48 | 25.96 | 23.36 | 23.96 | 24.34 | 25.47 | 25.13 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.03 | 0.03 | 0.00 | 0.00 | 0.00 | 0.03 | 0.00 | 0.07 | 0.01 | 0.02 | 0.03 | 0.01 |
| FeO | 4.43 | 4.70 | 3.58 | 4.05 | 4.22 | 3.98 | 3.76 | 4.00 | 3.33 | 3.77 | 3.57 | 3.71 |
| MnO | 0.02 | 0.04 | 0.03 | 0.04 | 0.04 | 0.02 | 0.05 | 0.02 | 0.00 | 0.00 | 0.00 | 0.04 |
| MgO | 3.67 | 3.64 | 3.81 | 3.85 | 3.62 | 4.04 | 3.50 | 4.27 | 4.03 | 4.09 | 3.69 | 3.62 |
| CaO | 0.00 | 0.00 | 0.10 | 0.04 | 0.01 | 0.38 | 0.02 | 0.19 | 0.05 | 0.04 | 0.03 | 0.05 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.35 | 0.30 | 0.10 | 0.11 | 0.14 | 0.13 | 0.10 | 0.13 | 0.07 | 0.11 | 0.23 | 0.10 |
| $\mathrm{K}_{2} \mathrm{O}$ | 10.77 | 10.82 | 11.05 | 9.73 | 11.30 | 10.77 | 11.14 | 10.88 | 11.33 | 11.31 | 11.23 | 11.46 |
| Total | 95.79 | 96.60 | 96.93 | 97.11 | 95.51 | 95.01 | 94.87 | 94.96 | 95.29 | 95.44 | 95.61 | 96.16 |
| Si | 3.399 | 3.403 | 3.458 | 3.500 | 3.442 | 3.463 | 3.410 | 3.519 | 3.530 | 3.484 | 3.445 | 3.476 |
| Ti | 0.007 | 0.008 | 0.006 | 0.005 | 0.006 | 0.003 | 0.004 | 0.004 | 0.002 | 0.004 | 0.006 | 0.004 |
| A1 | 2.055 | 2.046 | 2.012 | 1.983 | 2.000 | 1.954 | 2.075 | 1.865 | 1.900 | 1.934 | 2.018 | 1.981 |
| Cr | 0.002 | 0.002 | 0.000 | 0.000 | 0.000 | 0.002 | 0.000 | 0.004 | 0.001 | 0.001 | 0.002 | 0.000 |
| $\mathrm{Fe}^{3+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{2+}$ | 0.249 | 0.263 | 0.198 | 0.222 | 0.239 | 0.226 | 0.213 | 0.227 | 0.187 | 0.212 | 0.201 | 0.207 |
| Mn | 0.001 | 0.002 | 0.002 | 0.002 | 0.002 | 0.001 | 0.003 | 0.001 | 0.000 | 0.000 | 0.000 | 0.002 |
| Mg | 0.368 | 0.362 | 0.375 | 0.376 | 0.365 | 0.408 | 0.354 | 0.431 | 0.404 | 0.411 | 0.370 | 0.361 |
| Ca | 0.000 | 0.000 | 0.007 | 0.003 | 0.001 | 0.027 | 0.001 | 0.014 | 0.004 | 0.003 | 0.002 | 0.004 |
| Na | 0.046 | 0.039 | 0.012 | 0.013 | 0.018 | 0.017 | 0.013 | 0.018 | 0.009 | 0.014 | 0.030 | 0.013 |
| K | 0.925 | 0.922 | 0.932 | 0.813 | 0.974 | 0.930 | 0.964 | 0.940 | 0.973 | 0.973 | 0.963 | 0.978 |
| Total | 7.051 | 7.046 | 7.002 | 6.917 | 7.048 | 7.030 | 7.037 | 7.022 | 7.009 | 7.037 | 7.036 | 7.026 |

Table A. 2 (continued) : Mineral chemistry of the measured phengitic white micas. Mineral formula calculations are based on 11 oxygens.

| Phengite | 128 | 128 | 128 | 128 |
| :--- | :---: | :---: | :---: | :---: |
|  | 143 | 144 | 160 | 173 |
| $\mathrm{SiO}_{2}$ | 53.93 | 51.74 | 51.82 | 50.59 |
| $\mathrm{TiO}_{2}$ | 0.07 | 0.04 | 0.07 | 0.05 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 24.52 | 23.60 | 23.54 | 26.50 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.01 | 0.00 | 0.04 | 0.00 |
| FeO | 4.61 | 4.03 | 4.11 | 5.30 |
| MnO | 0.09 | 0.05 | 0.06 | 0.03 |
| MgO | 4.03 | 4.08 | 4.18 | 2.87 |
| CaO | 0.21 | 0.03 | 0.01 | 0.08 |
| $\mathrm{Na} \mathrm{O}_{2} \mathrm{O}$ | 0.26 | 0.08 | 0.06 | 0.12 |
| $\mathrm{~K}_{2} \mathrm{O}$ | 9.35 | 11.45 | 11.36 | 9.36 |
| Total | 97.08 | 95.11 | 95.26 | 94.91 |
|  |  |  |  |  |
| Si | 3.537 | 3.509 | 3.509 | 3.414 |
| Ti | 0.004 | 0.002 | 0.004 | 0.002 |
| Al | 1.896 | 1.887 | 1.879 | 2.107 |
| Cr | 0.001 | 0.000 | 0.002 | 0.000 |
| Fe | 0.000 | 0.000 | 0.000 | 0.000 |
| Fe |  | 0.253 | 0.228 | 0.233 |
| Mn | 0.005 | 0.003 | 0.004 | 0.099 |
| Mg | 0.394 | 0.412 | 0.422 | 0.289 |
| Ca | 0.015 | 0.002 | 0.001 | 0.006 |
| Na | 0.033 | 0.011 | 0.007 | 0.016 |
| K | 0.782 | 0.991 | 0.982 | 0.805 |
| Total | 6.919 | 7.046 | 7.042 | 6.941 |

Table A. 3 : Mineral chemistry of the measured paragonites. Mineral formula calculations are based on 11 oxygens.

| Paragonite | 775 | 775 | 775 | 775 | 775 | 775 | 775 | 775 | 775 | 753A | 753A | 753A | 753A |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 16 | 217 | 237 | 238 | 258 | 259 | 262 | 263 | 279 | 38 | 90 | 104 | 109 |
| $\mathrm{SiO}_{2}$ | 47.91 | 47.41 | 48.66 | 48.91 | 46.71 | 46.50 | 49.19 | 49.03 | 47.53 | 47.07 | 46.27 | 46.70 | 46.65 |
| $\mathrm{TiO}_{2}$ | 0.01 | 0.01 | 0.03 | 0.05 | 0.02 | 0.04 | 0.06 | 0.02 | 0.05 | 0.03 | 0.03 | 0.03 | 0.02 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 40.05 | 38.83 | 39.96 | 39.55 | 40.21 | 40.17 | 40.38 | 39.97 | 38.87 | 39.47 | 40.13 | 39.41 | 40.17 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.02 | 0.01 | 0.01 | 0.02 | 0.07 | 0.04 | 0.03 | 0.01 | 0.01 | 0.02 | 0.02 | 0.00 | 0.00 |
| FeO | 0.98 | 0.79 | 0.31 | 0.43 | 0.27 | 0.19 | 0.13 | 0.28 | 0.45 | 0.81 | 0.62 | 0.55 | 0.30 |
| MnO | 0.06 | 0.01 | 0.02 | 0.00 | 0.02 | 0.01 | 0.02 | 0.00 | 0.01 | 0.02 | 0.02 | 0.04 | 0.00 |
| MgO | 0.16 | 0.24 | 0.11 | 0.10 | 0.03 | 0.02 | 0.05 | 0.14 | 0.23 | 0.25 | 0.14 | 0.14 | 0.14 |
| CaO | 0.05 | 0.03 | 0.05 | 0.03 | 0.36 | 0.38 | 0.42 | 0.05 | 0.04 | 0.11 | 0.02 | 0.04 | 0.05 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 6.89 | 7.08 | 7.36 | 7.25 | 7.04 | 7.12 | 6.66 | 6.77 | 7.47 | 6.64 | 6.84 | 6.20 | 6.57 |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.88 | 1.37 | 0.43 | 0.64 | 0.91 | 0.88 | 0.51 | 0.48 | 0.98 | 0.86 | 0.18 | 0.62 | 0.26 |
| Total | 97.02 | 95.77 | 96.93 | 96.98 | 95.62 | 95.35 | 97.44 | 96.74 | 95.64 | 95.27 | 94.25 | 93.73 | 94.15 |
| Si | 3.010 | 3.027 | 3.043 | 3.060 | 2.975 | 2.971 | 3.052 | 3.062 | 3.031 | 3.008 | 2.975 | 3.017 | 2.995 |
| Ti | 0.001 | 0.001 | 0.002 | 0.002 | 0.001 | 0.002 | 0.003 | 0.001 | 0.003 | 0.002 | 0.001 | 0.002 | 0.001 |
| A1 | 2.966 | 2.922 | 2.945 | 2.917 | 3.019 | 3.024 | 2.952 | 2.942 | 2.922 | 2.972 | 3.042 | 3.001 | 3.038 |
| Cr | 0.001 | 0.000 | 0.001 | 0.001 | 0.003 | 0.002 | 0.002 | 0.000 | 0.000 | 0.001 | 0.001 | 0.000 | 0.000 |
| $\mathrm{Fe}^{3+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{2+}$ | 0.052 | 0.042 | 0.016 | 0.022 | 0.014 | 0.010 | 0.007 | 0.014 | 0.024 | 0.043 | 0.033 | 0.029 | 0.016 |
| Mn | 0.003 | 0.000 | 0.001 | 0.000 | 0.001 | 0.001 | 0.001 | 0.000 | 0.000 | 0.001 | 0.001 | 0.002 | 0.000 |
| Mg | 0.015 | 0.022 | 0.010 | 0.010 | 0.003 | 0.002 | 0.005 | 0.013 | 0.022 | 0.024 | 0.014 | 0.014 | 0.013 |
| Ca | 0.004 | 0.002 | 0.004 | 0.002 | 0.024 | 0.026 | 0.028 | 0.003 | 0.003 | 0.007 | 0.001 | 0.003 | 0.003 |
| Na | 0.839 | 0.877 | 0.893 | 0.879 | 0.869 | 0.881 | 0.800 | 0.820 | 0.923 | 0.823 | 0.852 | 0.777 | 0.817 |
| K | 0.070 | 0.112 | 0.034 | 0.051 | 0.074 | 0.072 | 0.040 | 0.038 | 0.080 | 0.070 | 0.014 | 0.051 | 0.021 |
| Total | 6.960 | 7.005 | 6.947 | 6.944 | 6.984 | 6.991 | 6.889 | 6.894 | 7.007 | 6.951 | 6.935 | 6.895 | 6.905 |

Table A. 3 (continued): Mineral chemistry of the measured paragonites. Mineral formula calculations are based on 11 oxygens.

| Paragonite | 178 | 178 | 178 | 178 | 178 | 392 | 392 | 392 | 392 | 392 | 392 | 392 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 26 | 38 | 44 | 51 | 99 | 4 | 7 | 10 | 13 | 25 | 38 | 39 |
| $\mathrm{SiO}_{2}$ | 47.02 | 46.77 | 46.90 | 46.86 | 46.81 | 47.69 | 47.94 | 49.91 | 47.65 | 47.05 | 47.46 | 47.23 |
| $\mathrm{TiO}_{2}$ | 0.00 | 0.00 | 0.03 | 0.02 | 0.00 | 0.01 | 0.02 | 0.00 | 0.05 | 0.05 | 0.03 | 0.04 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 39.51 | 39.95 | 39.55 | 39.62 | 40.01 | 40.69 | 40.29 | 37.88 | 40.22 | 40.28 | 40.35 | 40.48 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.03 | 0.04 | 0.02 | 0.01 | 0.01 |  |  |  |  |  |  |  |
| FeO | 0.36 | 0.42 | 0.49 | 0.50 | 0.31 | 0.30 | 0.23 | 0.26 | 0.41 | 0.58 | 0.35 | 0.34 |
| MnO | 0.00 | 0.00 | 0.01 | 0.00 | 0.00 | 0.01 | 0.01 | 0.00 | 0.01 | 0.00 | 0.03 | 0.00 |
| MgO | 0.18 | 0.02 | 0.11 | 0.12 | 0.09 | 0.05 | 0.05 | 0.08 | 0.16 | 0.42 | 0.12 | 0.10 |
| CaO | 0.04 | 0.04 | 0.02 | 0.04 | 0.05 | 0.06 | 0.05 | 0.05 | 0.10 | 0.05 | 0.05 | 0.05 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 6.44 | 6.62 | 6.47 | 6.55 | 7.27 | 7.41 | 7.51 | 8.28 | 7.07 | 6.88 | 7.27 | 7.34 |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.70 | 0.32 | 0.63 | 0.80 | 0.41 | 0.47 | 0.34 | 0.50 | 0.73 | 1.02 | 0.41 | 0.52 |
| Total | 94.28 | 94.17 | 94.23 | 94.51 | 94.96 | 96.69 | 96.45 | 96.95 | 96.39 | 96.33 | 96.07 | 96.10 |
| Si | 3.021 | 3.004 | 3.016 | 3.009 | 2.992 | 2.994 | 3.014 | 3.128 | 3.004 | 2.977 | 2.998 | 2.986 |
| Ti | 0.000 | 0.000 | 0.002 | 0.001 | 0.000 | 0.001 | 0.001 | 0.000 | 0.002 | 0.002 | 0.001 | 0.002 |
| Al | 2.992 | 3.024 | 2.997 | 2.998 | 3.014 | 3.011 | 2.985 | 2.798 | 2.988 | 3.004 | 3.004 | 3.016 |
| Cr | 0.001 | 0.002 | 0.001 | 0.001 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{3+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{2+}$ | 0.019 | 0.022 | 0.026 | 0.027 | 0.017 | 0.016 | 0.012 | 0.014 | 0.021 | 0.031 | 0.019 | 0.018 |
| Mn | 0.000 | 0.000 | 0.001 | 0.000 | 0.000 | 0.000 | 0.001 | 0.000 | 0.001 | 0.000 | 0.002 | 0.000 |
| Mg | 0.017 | 0.002 | 0.011 | 0.011 | 0.008 | 0.004 | 0.004 | 0.007 | 0.015 | 0.039 | 0.011 | 0.010 |
| Ca | 0.003 | 0.003 | 0.002 | 0.003 | 0.003 | 0.004 | 0.004 | 0.003 | 0.007 | 0.003 | 0.003 | 0.003 |
| Na | 0.802 | 0.824 | 0.806 | 0.815 | 0.901 | 0.902 | 0.915 | 1.006 | 0.864 | 0.844 | 0.890 | 0.900 |
| K | 0.057 | 0.026 | 0.052 | 0.065 | 0.033 | 0.037 | 0.027 | 0.040 | 0.058 | 0.082 | 0.033 | 0.042 |
| Total | 6.913 | 6.908 | 6.913 | 6.931 | 6.968 | 6.969 | 6.964 | 6.996 | 6.961 | 6.982 | 6.960 | 6.976 |

Table A. 4 : Mineral chemistry of the measured garnets. Mineral formula calculations are based on 12 oxygens.

| Garnet | 775 | 775 | 775 | 775 | 775 | 775 | 775 | 775 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 9 | 18 | 19 | 20 | 21 | 26 | 27 | 206 |
| $\mathrm{SiO}_{2}$ | 36.36 | 36.01 | 36.53 | 36.09 | 36.52 | 36.13 | 36.65 | 37.00 |
| $\mathrm{TiO}_{2}$ | 0.07 | 0.11 | 0.11 | 0.13 | 0.04 | 0.10 | 0.08 | 0.11 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 21.70 | 21.63 | 21.62 | 21.70 | 21.70 | 21.34 | 21.56 | 21.63 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.00 | 0.02 | 0.01 | 0.05 | 0.00 | 0.01 | 0.02 | 0.03 |
| FeO | 13.00 | 12.34 | 14.65 | 12.55 | 16.87 | 12.78 | 15.12 | 12.12 |
| MnO | 25.10 | 26.25 | 24.41 | 25.93 | 21.85 | 26.10 | 21.25 | 26.84 |
| MgO | 0.31 | 0.34 | 0.33 | 0.34 | 0.33 | 0.31 | 0.19 | 0.28 |
| CaO | 3.49 | 2.50 | 3.26 | 3.14 | 3.12 | 2.95 | 5.52 | 2.66 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.02 | 0.04 | 0.00 | 0.02 | 0.03 | 0.02 | 0.05 | 0.07 |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.00 | 0.01 | 0.02 | 0.04 | 0.06 | 0.00 | 0.05 | 0.01 |
| Total | 100.04 | 99.24 | 100.94 | 99.99 | 100.53 | 99.75 | 100.48 | 100.74 |
| Si | 2.957 | 2.955 | 2.954 | 2.943 | 2.961 | 2.958 | 2.962 | 2.987 |
| Ti | 0.005 | 0.007 | 0.006 | 0.008 | 0.003 | 0.006 | 0.005 | 0.006 |
| Al | 2.080 | 2.092 | 2.061 | 2.085 | 2.074 | 2.059 | 2.054 | 2.058 |
| Cr | 0.000 | 0.001 | 0.001 | 0.003 | 0.000 | 0.001 | 0.001 | 0.002 |
| $\mathrm{Fe}^{3+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{2+}$ | 0.884 | 0.847 | 0.991 | 0.856 | 1.144 | 0.875 | 1.022 | 0.818 |
| Mn | 1.729 | 1.824 | 1.672 | 1.791 | 1.501 | 1.810 | 1.455 | 1.836 |
| Mg | 0.037 | 0.042 | 0.039 | 0.042 | 0.040 | 0.038 | 0.023 | 0.034 |
| Ca | 0.304 | 0.219 | 0.283 | 0.274 | 0.271 | 0.259 | 0.478 | 0.230 |
| Na | 0.003 | 0.006 | 0.000 | 0.003 | 0.005 | 0.003 | 0.007 | 0.010 |
| K | 0.000 | 0.001 | 0.002 | 0.004 | 0.006 | 0.000 | 0.005 | 0.001 |
| Total | 8.000 | 7.995 | 8.010 | 8.009 | 8.005 | 8.008 | 8.012 | 7.982 |
| $\mathrm{X}_{\text {Sps }}$ | 59 | 62 | 56 | 60 | 51 | 61 | 49 | 63 |
| $\mathrm{X}_{\text {Gris }}$ | 10 | 7 | 9 | 9 | 9 | 9 | 16 | 8 |
| $\mathrm{X}_{\text {Alm }}$ | 30 | 29 | 33 | 29 | 39 | 29 | 34 | 28 |
| $\mathrm{X}_{\text {Py }}$ | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 |
| Tot | 100 | 100 | 100 | 100 | 100 | 100 | 100 | 100 |

Table A. 4 (continued) : Mineral chemistry of the measured garnets. Mineral formula calculations are based on 12 oxygens.

| Garnet | 775 | 775 | 702A | 702A | 702A | 702A | 702A | 702A |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 209 | 275 | 34 | 35 | grtla | grt1b | grt2a | grt2b |
| $\mathrm{SiO}_{2}$ | 37.28 | 37.30 | 37.27 | 37.14 | 38.21 | 38.23 | 38.21 | 39.79 |
| $\mathrm{TiO}_{2}$ | 0.14 | 0.06 | 0.15 | 0.19 | 0.12 | 0.18 | 0.08 | 0.11 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 21.40 | 21.95 | 21.31 | 20.87 | 21.29 | 21.24 | 21.34 | 21.06 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.00 | 0.02 | 0.00 | 0.00 | 0.01 | 0.01 | 0.00 | 0.01 |
| FeO | 14.78 | 16.91 | 25.43 | 22.12 | 24.59 | 21.01 | 25.26 | 23.83 |
| MnO | 24.15 | 20.93 | 1.04 | 5.06 | 1.59 | 6.65 | 1.36 | 1.04 |
| MgO | 0.28 | 0.31 | 0.57 | 0.45 | 0.71 | 0.47 | 0.69 | 0.86 |
| CaO | 3.50 | 3.39 | 13.97 | 14.11 | 14.11 | 12.66 | 14.00 | 12.97 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.04 | 0.04 | 0.04 | 0.02 | 0.00 | 0.00 | 0.04 | 0.83 |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.01 | 0.26 | 0.01 | 0.01 |  |  |  |  |
| Total | 101.58 | 101.17 | 99.79 | 99.96 | 100.63 | 100.46 | 100.98 | 100.49 |
| Si | 2.990 | 2.990 | 2.973 | 2.970 | 6.020 | 6.003 | 6.008 | 6.210 |
| Ti | 0.008 | 0.004 | 0.009 | 0.011 | 0.014 | 0.022 | 0.010 | 0.012 |
| A1 | 2.022 | 2.073 | 2.003 | 1.967 | 3.953 | 3.931 | 3.955 | 3.874 |
| Cr | 0.000 | 0.001 | 0.000 | 0.000 | 0.001 | 0.002 | 0.000 | 0.001 |
| $\mathrm{Fe}^{3+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{2+}$ | 0.991 | 1.133 | 1.696 | 1.479 | 3.240 | 2.759 | 3.323 | 3.110 |
| Mn | 1.641 | 1.421 | 0.070 | 0.343 | 0.212 | 0.884 | 0.181 | 0.137 |
| Mg | 0.033 | 0.037 | 0.068 | 0.053 | 0.167 | 0.109 | 0.162 | 0.200 |
| Ca | 0.301 | 0.291 | 1.193 | 1.209 | 2.382 | 2.299 | 2.359 | 2.169 |
| Na | 0.006 | 0.006 | 0.006 | 0.002 | 0.000 | 0.001 | 0.012 | 0.250 |
| K | 0.001 | 0.027 | 0.001 | 0.001 | 0.000 | 0.000 | 0.000 | 0.000 |
| Total | 7.994 | 7.985 | 8.020 | 8.037 | 15.989 | 16.010 | 16.010 | 15.963 |
| $\mathrm{X}_{\text {Sps }}$ | 55 | 49 | 2 | 11 | 4 | 15 | 3 | 2 |
| $\mathrm{X}_{\text {Grs }}$ | 10 | 10 | 39 | 39 | 40 | 38 | 39 | 39 |
| $\mathrm{X}_{\text {Alm }}$ | 33 | 39 | 56 | 48 | 54 | 46 | 55 | 55 |
| $\mathrm{X}_{\text {Py }}$ | 1 | 1 | 2 | 2 | 3 | 2 | 3 | 4 |
| Tot | 100 | 100 | 100 | 100 | 100 | 100 | 100 | 100 |

Table A. 4 (continued) : Mineral chemistry of the measured garnets. Mineral formula calculations are based on 12 oxygens.

| Garnet | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | grt3a | grt3b | grt4a | grt4b | grt5a | grt5b | grtba | grt6b |
| $\mathrm{SiO}_{2}$ | 38.05 | 37.08 | 37.79 | 38.00 | 38.16 | 38.44 | 37.85 | 38.44 |
| $\mathrm{TiO}_{2}$ | 0.23 | 0.10 | 0.21 | 0.11 | 0.19 | 0.14 | 0.14 | 0.13 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 21.28 | 19.65 | 21.27 | 21.27 | 21.29 | 21.29 | 21.29 | 21.49 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 |
| FeO | 21.36 | 24.09 | 21.36 | 25.02 | 20.46 | 23.93 | 22.70 | 25.11 |
| MnO | 5.31 | 1.01 | 5.51 | 1.23 | 6.58 | 2.48 | 4.91 | 1.32 |
| MgO | 0.44 | 2.73 | 0.49 | 0.61 | 0.43 | 0.56 | 0.54 | 0.65 |
| CaO | 13.42 | 12.88 | 13.96 | 14.17 | 13.82 | 14.09 | 13.51 | 12.98 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.03 | 0.09 | 0.01 | 0.03 | 0.01 | 0.00 | 0.01 | 0.00 |
| $\mathrm{K}_{2} \mathrm{O}$ |  |  |  |  |  |  |  |  |
| Total | 100.12 | 97.64 | 100.60 | 100.44 | 100.94 | 100.93 | 100.94 | 100.12 |
| Si | 5.986 | 6.012 | 5.978 | 6.006 | 6.010 | 6.039 | 5.978 | 6.024 |
| Ti | 0.027 | 0.013 | 0.025 | 0.013 | 0.022 | 0.017 | 0.016 | 0.016 |
| A1 | 3.946 | 3.755 | 3.965 | 3.962 | 3.952 | 3.942 | 3.962 | 3.970 |
| Cr | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{3+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{2+}$ | 2.810 | 3.267 | 2.826 | 3.307 | 2.694 | 3.143 | 2.999 | 3.291 |
| Mn | 0.707 | 0.139 | 0.738 | 0.164 | 0.877 | 0.330 | 0.657 | 0.176 |
| Mg | 0.104 | 0.660 | 0.116 | 0.145 | 0.102 | 0.131 | 0.127 | 0.151 |
| Ca | 2.430 | 2.238 | 2.366 | 2.399 | 2.331 | 2.372 | 2.286 | 2.348 |
| Na | 0.008 | 0.029 | 0.002 | 0.008 | 0.002 | 0.001 | 0.002 | 0.000 |
| K | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| Total | 16.018 | 16.113 | 16.016 | 16.005 | 15.990 | 15.974 | 16.027 | 15.976 |
| $\mathrm{X}_{\text {Sps }}$ | 12 | 2 | 12 | 3 | 15 | 6 | 11 | 3 |
| $\mathrm{X}_{\text {Gris }}$ | 40 | 36 | 39 | 40 | 39 | 40 | 38 | 39 |
| $\mathrm{X}_{\text {Alm }}$ | 46 | 52 | 47 | 55 | 45 | 53 | 49 | 55 |
| $\mathrm{X}_{\text {Py }}$ | 2 | 10 | 2 | 2 | 2 | 2 | 2 | 3 |
| Tot | 100 | 100 | 100 | 100 | 100 | 100 | 100 | 100 |

Table A. 4 (continued) : Mineral chemistry of the measured garnets. Mineral formula calculations are based on 12 oxygens.

| Garnet | 702A | 702A | 702A | 702A |
| :---: | :---: | :---: | :---: | :---: |
|  | grt7a | grt7b | grt8a | grt8b |
| $\mathrm{SiO}_{2}$ | 38.10 | 50.84 | 38.20 | 38.26 |
| $\mathrm{TiO}_{2}$ | 0.18 | 0.04 | 0.19 | 0.14 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 21.25 | 19.26 | 21.28 | 21.27 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.02 | 0.01 | 0.01 | 0.00 |
| FeO | 20.41 | 12.01 | 21.24 | 23.67 |
| MnO | 6.76 | 0.68 | 5.86 | 2.93 |
| MgO | 0.46 | 0.72 | 0.49 | 0.58 |
| CaO | 12.95 | 6.67 | 13.01 | 13.21 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.01 | 7.73 | 0.00 | 0.01 |
| $\mathrm{K}_{2} \mathrm{O}$ |  |  |  |  |
| Total | 100.14 | 97.96 | 100.28 | 100.07 |
| Si | 5.996 | 7.525 | 6.001 | 6.012 |
| Ti | 0.022 | 0.005 | 0.022 | 0.016 |
| Al | 3.942 | 3.360 | 3.940 | 3.941 |
| Cr | 0.002 | 0.001 | 0.001 | 0.000 |
| $\mathrm{Fe}^{3+}$ | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{2+}$ | 2.686 | 1.486 | 2.790 | 3.111 |
| Mn | 0.902 | 0.085 | 0.780 | 0.390 |
| Mg | 0.108 | 0.160 | 0.116 | 0.137 |
| Ca | 2.352 | 1.058 | 2.358 | 2.392 |
| Na | 0.002 | 2.219 | 0.000 | 0.003 |
| K | 0.000 | 0.000 | 0.000 | 0.000 |
| Total | 16.012 | 15.898 | 16.008 | 16.002 |
| $\mathrm{X}_{\text {sps }}$ | 15 | 3 | 13 | 6 |
| $\mathrm{X}_{\text {Grs }}$ | 39 | 38 | 39 | 40 |
| $\mathrm{X}_{\text {Alm }}$ | 44 | 53 | 46 | 52 |
| $\mathrm{X}_{\mathrm{Py}}$ | 2 | 6 | 2 | 2 |
| Tot | 100 | 100 | 100 | 100 |

Table A. 5 : Mineral chemistry of the measured chloritoids. Mineral formula calculations are based on 12 oxygens.

| Chloritoid | 775 | 775 | 775 | 775 | 775 | 775 | 775 | 775 |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 210 | 211 | 212 | 214 | 215 | 274 | 277 | 280 |
| $\mathrm{SiO}_{2}$ | 24.45 | 24.34 | 24.77 | 24.38 | 24.62 | 24.52 | 24.59 | 24.93 |
| $\mathrm{TiO}_{2}$ | 0.02 | 0.02 | 0.02 | 0.03 | 0.02 | 0.01 | 0.00 | 0.04 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 41.15 | 40.82 | 40.27 | 41.10 | 40.93 | 41.25 | 41.42 | 41.47 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.02 | 0.05 | 0.01 | 0.04 | 0.04 | 0.05 | 0.02 | 0.04 |
| FeO | 24.18 | 24.39 | 24.77 | 24.14 | 24.56 | 23.94 | 23.94 | 23.75 |
| MnO | 2.96 | 2.54 | 2.32 | 2.71 | 2.57 | 3.32 | 2.72 | 2.25 |
| MgO | 1.24 | 1.29 | 1.60 | 1.27 | 1.30 | 1.11 | 1.18 | 1.66 |
| CaO | 0.02 | 0.01 | 0.01 | 0.01 | 0.02 | 0.01 | 0.01 | 0.00 |
| Na 2 O | 0.01 | 0.00 | 0.01 | 0.01 | 0.03 | 0.00 | 0.00 | 0.00 |
| $\mathrm{~K}_{2} \mathrm{O}$ | 0.00 | 0.01 | 0.02 | 0.03 | 0.03 | 0.03 | 0.04 | 0.03 |
| Total | 94.04 | 93.45 | 93.80 | 93.73 | 94.12 | 94.24 | 93.92 | 94.18 |
| Si | 2.005 | 2.009 | 2.038 | 2.005 | 2.018 | 2.008 | 2.013 | 2.028 |
| Ti | 0.001 | 0.001 | 0.001 | 0.002 | 0.001 | 0.001 | 0.000 | 0.003 |
| Al | 3.978 | 3.970 | 3.905 | 3.983 | 3.954 | 3.981 | 3.998 | 3.975 |
| Cr | 0.001 | 0.003 | 0.001 | 0.003 | 0.003 | 0.003 | 0.001 | 0.003 |
| $\mathrm{Fe}^{3+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{2+}$ | 1.659 | 1.683 | 1.704 | 1.660 | 1.683 | 1.639 | 1.639 | 1.615 |
| Mn | 0.205 | 0.177 | 0.162 | 0.189 | 0.179 | 0.230 | 0.189 | 0.155 |
| Mg | 0.152 | 0.158 | 0.196 | 0.155 | 0.158 | 0.135 | 0.144 | 0.201 |
| Ca | 0.001 | 0.001 | 0.001 | 0.001 | 0.001 | 0.001 | 0.001 | 0.000 |
| Na | 0.002 | 0.000 | 0.001 | 0.002 | 0.005 | 0.000 | 0.000 | 0.000 |
| K | 0.000 | 0.001 | 0.003 | 0.003 | 0.003 | 0.003 | 0.004 | 0.003 |
| Total | 8.005 | 8.004 | 8.010 | 8.003 | 8.006 | 8.001 | 7.989 | 7.982 |
| $\mathrm{X}_{\mathrm{Mn}}$ | 10 | 9 | 8 | 9 | 9 | 11 | 10 | 8 |
| $\mathrm{X}_{\mathrm{Mg}}$ | 8 | 8 | 10 | 8 | 8 | 7 | 7 | 10 |
| $\mathrm{X}_{\mathrm{Fe}}$ | 82 | 83 | 83 | 83 | 83 | 82 | 83 | 82 |
| Tot | 100 | 100 | 100 | 100 | 100 | 100 | 100 | 100 |
| $\mathrm{X}_{\mathrm{Mn}}=\mathrm{Mn} /\left(\mathrm{Mn}+\mathrm{Fe}^{2+}+\mathrm{Mg}\right), \mathrm{X}_{\mathrm{Mg}}=\mathrm{Mg} /\left(\mathrm{Mg}+\mathrm{Mn}^{2+} \mathrm{Fe}^{2+}\right), \mathrm{X}_{\mathrm{Fe}}=\mathrm{Fe}^{2+} /\left(\mathrm{Fe}^{2+}+\mathrm{Mn}+\mathrm{Mg}\right)$. |  |  |  |  |  |  |  |  |
|  |  |  |  |  |  |  |  |  |

Table A. 5 (continued) : Mineral chemistry of the measured chloritoids. Mineral formula calculations are based on 12 oxygens.

| Chloritoid | 775 | 775 | 775 | 775 | 775 | 753A | 753A | 753A |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 14 | 15 | 28 | 29 | 30 | 70 | 71 | 72 |
| $\mathrm{SiO}_{2}$ | 24.20 | 24.04 | 24.10 | 23.94 | 23.96 | 24.34 | 24.47 | 24.49 |
| $\mathrm{TiO}_{2}$ | 0.02 | 0.01 | 0.74 | 0.00 | 0.00 | 0.02 | 0.08 | 0.05 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 41.52 | 40.88 | 41.37 | 41.50 | 41.82 | 41.88 | 41.82 | 41.85 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.02 | 0.04 | 0.03 | 0.03 | 0.04 | 0.01 | 0.06 | 0.02 |
| FeO | 23.45 | 24.06 | 23.57 | 23.41 | 22.61 | 23.89 | 24.49 | 24.37 |
| MnO | 2.33 | 2.95 | 2.79 | 2.99 | 2.60 | 3.42 | 2.79 | 2.30 |
| MgO | 1.54 | 1.38 | 1.22 | 1.26 | 1.40 | 1.25 | 1.35 | 1.62 |
| CaO | 0.01 | 0.04 | 0.01 | 0.00 | 0.01 | 0.02 | 0.03 | 0.01 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.00 | 0.02 | 0.01 | 0.02 | 0.01 | 0.00 | 0.01 | 0.00 |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.02 | 0.07 | 0.01 | 0.01 | 0.02 | 0.00 | 0.01 | 0.00 |
| Total | 93.10 | 93.47 | 93.84 | 93.16 | 92.47 | 94.84 | 95.12 | 94.71 |
| Si | 1.993 | 1.987 | 1.976 | 1.977 | 1.981 | 1.980 | 1.984 | 1.988 |
| Ti | 0.001 | 0.000 | 0.045 | 0.000 | 0.000 | 0.001 | 0.005 | 0.003 |
| Al | 4.029 | 3.982 | 3.996 | 4.040 | 4.075 | 4.015 | 3.997 | 4.004 |
| Cr | 0.001 | 0.002 | 0.002 | 0.002 | 0.002 | 0.001 | 0.004 | 0.001 |
| $\mathrm{Fe}^{3+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{2+}$ | 1.615 | 1.663 | 1.616 | 1.617 | 1.563 | 1.625 | 1.661 | 1.655 |
| Mn | 0.162 | 0.206 | 0.194 | 0.209 | 0.182 | 0.236 | 0.192 | 0.158 |
| Mg | 0.188 | 0.171 | 0.149 | 0.155 | 0.173 | 0.151 | 0.163 | 0.196 |
| Ca | 0.001 | 0.003 | 0.001 | 0.000 | 0.001 | 0.002 | 0.003 | 0.001 |
| Na | 0.000 | 0.004 | 0.001 | 0.004 | 0.002 | 0.000 | 0.002 | 0.000 |
| K | 0.002 | 0.007 | 0.001 | 0.001 | 0.002 | 0.000 | 0.001 | 0.000 |
| Total | 7.992 | 8.026 | 7.981 | 8.004 | 7.982 | 8.011 | 8.012 | 8.006 |
| $\mathrm{X}_{\mathrm{Mn}}$ | 8 | 10 | 10 | 11 | 9 | 12 | 10 | 8 |
| $\mathrm{X}_{\mathrm{Mg}}$ | 10 | 8 | 8 | 8 | 9 | 8 | 8 | 10 |
| $\mathrm{X}_{\mathrm{Fe}}$ | 82 | 82 | 82 | 82 | 81 | 81 | 82 | 82 |
| Tot | 100 | 100 | 100 | 100 | 100 | 100 | 100 | 100 |

Table A. 5 (continued) : Mineral chemistry of the measured chloritoids. Mineral formula calculations are based on 12 oxygens.

| Chloritoid | 753 A | 753 A | 753 A | 753 A | 753 A | 753 A | 753 A | 753 A |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 73 | 74 | 82 | 128 | 132 | 134 | 135 | 137 |
| $\mathrm{SiO}_{2}$ | 24.39 | 24.40 | 24.61 | 24.41 | 24.51 | 24.16 | 24.09 | 24.46 |
| $\mathrm{TiO}_{2}$ | 0.00 | 0.02 | 0.02 | 0.00 | 0.03 | 0.02 | 0.03 | 0.01 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 41.99 | 42.27 | 42.17 | 41.90 | 42.33 | 41.74 | 41.99 | 42.03 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.05 | 0.03 | 0.00 | 0.03 | 0.03 | 0.00 | 0.00 | 0.00 |
| FeO | 23.81 | 23.89 | 23.93 | 24.01 | 24.10 | 23.93 | 23.82 | 23.97 |
| MnO | 1.86 | 1.69 | 3.00 | 2.65 | 2.26 | 2.60 | 1.91 | 2.77 |
| MgO | 1.82 | 1.82 | 1.48 | 1.35 | 1.51 | 1.24 | 1.72 | 1.37 |
| CaO | 0.01 | 0.00 | 0.01 | 0.01 | 0.02 | 0.01 | 0.02 | 0.08 |
| Na 2 O | 0.00 | 0.01 | 0.00 | 0.00 | 0.00 | 0.01 | 0.00 | 0.00 |
| $\mathrm{~K}_{2} \mathrm{O}$ | 0.04 | 0.04 | 0.03 | 0.05 | 0.01 | 0.01 | 0.02 | 0.01 |
| Total | 93.97 | 94.16 | 95.24 | 94.41 | 94.80 | 93.71 | 93.60 | 94.69 |
| Si | 1.987 | 1.982 | 1.988 | 1.988 | 1.983 | 1.983 | 1.972 | 1.986 |
| Ti | 0.000 | 0.001 | 0.001 | 0.000 | 0.002 | 0.001 | 0.002 | 0.000 |
| Al | 4.031 | 4.046 | 4.014 | 4.023 | 4.037 | 4.037 | 4.051 | 4.023 |
| Cr | 0.003 | 0.002 | 0.000 | 0.002 | 0.002 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{3+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{2+}$ | 1.622 | 1.623 | 1.617 | 1.636 | 1.631 | 1.643 | 1.630 | 1.628 |
| Mn | 0.129 | 0.117 | 0.205 | 0.183 | 0.155 | 0.180 | 0.132 | 0.190 |
| Mg | 0.221 | 0.220 | 0.178 | 0.164 | 0.182 | 0.151 | 0.210 | 0.166 |
| Ca | 0.001 | 0.000 | 0.001 | 0.001 | 0.002 | 0.001 | 0.002 | 0.007 |
| Na | 0.000 | 0.001 | 0.000 | 0.000 | 0.000 | 0.001 | 0.000 | 0.000 |
| K | 0.004 | 0.004 | 0.003 | 0.005 | 0.001 | 0.001 | 0.002 | 0.001 |
| Total | 7.998 | 7.995 | 8.006 | 8.002 | 7.996 | 7.999 | 8.002 | 8.002 |
| $\mathrm{X}_{\mathrm{Mn}}$ | 7 | 6 | 10 | 9 | 8 | 9 | 7 | 10 |
| $\mathrm{X}_{\mathrm{Mg}}$ | 11 | 11 | 9 | 8 | 9 | 8 | 11 | 8 |
| $\mathrm{X}_{\mathrm{Fe}}$ | 82 | 83 | 81 | 82 | 83 | 83 | 83 | 82 |
| Tot | 100 | 100 | 100 | 100 | 100 | 100 | 100 | 100 |
| $\mathrm{X}_{\mathrm{Mn}}=\mathrm{Mn} /\left(\mathrm{Mn}+\mathrm{Fe}^{2+}+\mathrm{Mg}\right), \mathrm{X}_{\mathrm{Mg}}=\mathrm{Mg} /\left(\mathrm{Mg}+\mathrm{Mn}^{2+} \mathrm{Fe}^{2+}\right), \mathrm{X}_{\mathrm{Fe}}=\mathrm{Fe}^{2+} /\left(\mathrm{Fe}^{2+}+\mathrm{Mn}+\mathrm{Mg}\right)$. |  |  |  |  |  |  |  |  |
|  |  |  |  |  |  |  |  |  |

Table A. 5 (continued) : Mineral chemistry of the measured chloritoids. Mineral formula calculations are based on 12 oxygens.

| Chloritoid | 753A | 753A | 753A | 753A | 178 | 178 | 178 | 178 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 138 | 140 | 141 | 142 | 19 | 20 | 28 | 31 |
| $\mathrm{SiO}_{2}$ | 24.48 | 24.28 | 24.53 | 24.30 | 24.63 | 24.68 | 24.59 | 24.76 |
| $\mathrm{TiO}_{2}$ | 0.00 | 0.01 | 0.00 | 0.00 | 0.00 | 0.02 | 0.01 | 0.02 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 42.15 | 41.75 | 41.64 | 41.85 | 42.59 | 42.23 | 42.50 | 42.48 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.00 | 0.02 | 0.05 | 0.04 | 0.04 | 0.02 | 0.01 | 0.00 |
| FeO | 24.11 | 23.32 | 22.88 | 24.26 | 24.43 | 25.61 | 24.30 | 24.31 |
| MnO | 2.44 | 2.86 | 2.92 | 2.26 | 0.57 | 0.80 | 0.57 | 0.53 |
| MgO | 1.50 | 1.31 | 1.38 | 1.56 | 2.73 | 1.65 | 2.59 | 2.70 |
| CaO | 0.01 | 0.02 | 0.01 | 0.02 | 0.00 | 0.00 | 0.01 | 0.00 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.01 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.02 | 0.00 |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.01 | 0.00 | 0.02 | 0.04 | 0.01 | 0.03 | 0.03 | 0.02 |
| Total | 94.70 | 93.57 | 93.42 | 94.33 | 94.99 | 95.05 | 94.63 | 94.81 |
| Si | 1.985 | 1.991 | 2.011 | 1.981 | 1.976 | 1.992 | 1.980 | 1.988 |
| Ti | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.001 | 0.000 | 0.001 |
| Al | 4.029 | 4.036 | 4.022 | 4.021 | 4.026 | 4.018 | 4.032 | 4.020 |
| Cr | 0.000 | 0.001 | 0.003 | 0.003 | 0.002 | 0.001 | 0.001 | 0.000 |
| $\mathrm{Fe}^{3+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{2+}$ | 1.635 | 1.600 | 1.568 | 1.654 | 1.639 | 1.729 | 1.636 | 1.632 |
| Mn | 0.168 | 0.199 | 0.202 | 0.156 | 0.039 | 0.055 | 0.039 | 0.036 |
| Mg | 0.181 | 0.161 | 0.168 | 0.189 | 0.326 | 0.199 | 0.310 | 0.323 |
| Ca | 0.001 | 0.001 | 0.001 | 0.001 | 0.000 | 0.000 | 0.001 | 0.000 |
| Na | 0.001 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.003 | 0.000 |
| K | 0.001 | 0.000 | 0.002 | 0.004 | 0.001 | 0.003 | 0.003 | 0.002 |
| Total | 8.001 | 7.990 | 7.978 | 8.009 | 8.010 | 7.999 | 8.006 | 8.002 |
| $\mathrm{X}_{\text {Mn }}$ | 8 | 10 | 10 | 8 | 2 | 3 | 2 | 2 |
| $\mathrm{X}_{\mathrm{Mg}}$ | 9 | 8 | 9 | 9 | 16 | 10 | 16 | 16 |
| $\mathrm{X}_{\mathrm{Fe}}$ | 82 | 82 | 81 | 83 | 82 | 87 | 82 | 82 |
| Tot | 100 | 100 | 100 | 100 | 100 | 100 | 100 | 100 |

Table A. 5 (continued) : Mineral chemistry of the measured chloritoids. Mineral formula calculations are based on 12 oxygens.

| Chloritoid | 178 | 178 | 178 | 178 | 178 | 178 | 178 | 178 |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 32 | 34 | 35 | 36 | 39 | 40 | 41 | 48 |
| $\mathrm{SiO}_{2}$ | 24.71 | 24.41 | 24.40 | 24.13 | 24.64 | 24.95 | 24.90 | 24.42 |
| $\mathrm{TiO}_{2}$ | 0.06 | 0.00 | 0.01 | 0.03 | 0.02 | 0.00 | 0.02 | 0.10 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 42.21 | 42.37 | 41.91 | 41.73 | 42.15 | 42.53 | 42.77 | 41.70 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.00 | 0.03 | 0.03 | 0.04 | 0.07 | 0.02 | 0.06 | 0.04 |
| FeO | 25.71 | 25.72 | 25.29 | 25.21 | 25.44 | 24.64 | 24.33 | 25.89 |
| MnO | 0.80 | 0.47 | 0.78 | 0.87 | 0.73 | 0.56 | 0.56 | 0.78 |
| MgO | 1.72 | 1.86 | 1.91 | 1.75 | 1.70 | 2.61 | 2.71 | 1.65 |
| CaO | 0.01 | 0.00 | 0.01 | 0.01 | 0.00 | 0.00 | 0.00 | 0.01 |
| Na 2 O | 0.00 | 0.00 | 0.01 | 0.00 | 0.01 | 0.01 | 0.00 | 0.03 |
| $\mathrm{~K}_{2} \mathrm{O}$ | 0.04 | 0.02 | 0.01 | 0.01 | 0.04 | 0.01 | 0.01 | 0.04 |
| Total | 95.25 | 94.87 | 94.36 | 93.76 | 94.79 | 95.34 | 95.36 | 94.64 |
| Si | 1.992 | 1.973 | 1.984 | 1.976 | 1.993 | 1.995 | 1.987 | 1.985 |
| Ti | 0.003 | 0.000 | 0.000 | 0.002 | 0.001 | 0.000 | 0.001 | 0.006 |
| Al | 4.009 | 4.036 | 4.015 | 4.027 | 4.018 | 4.007 | 4.022 | 3.997 |
| Cr | 0.000 | 0.002 | 0.002 | 0.003 | 0.004 | 0.001 | 0.004 | 0.003 |
| $\mathrm{Fe}^{3+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{2+}$ | 1.733 | 1.739 | 1.719 | 1.726 | 1.721 | 1.647 | 1.624 | 1.761 |
| Mn | 0.055 | 0.032 | 0.054 | 0.060 | 0.050 | 0.038 | 0.038 | 0.053 |
| Mg | 0.206 | 0.224 | 0.232 | 0.213 | 0.205 | 0.311 | 0.322 | 0.200 |
| Ca | 0.001 | 0.000 | 0.001 | 0.001 | 0.000 | 0.000 | 0.000 | 0.001 |
| Na | 0.000 | 0.000 | 0.001 | 0.000 | 0.001 | 0.001 | 0.000 | 0.004 |
| K | 0.004 | 0.002 | 0.001 | 0.001 | 0.004 | 0.001 | 0.001 | 0.004 |
| Total | 8.002 | 8.009 | 8.008 | 8.008 | 7.997 | 8.002 | 8.000 | 8.013 |
| $\mathrm{X}_{\mathrm{Mn}}$ | 3 | 2 | 3 | 3 | 3 | 2 | 2 | 3 |
| $\mathrm{X}_{\mathrm{Mg}}$ | 10 | 11 | 12 | 11 | 10 | 16 | 16 | 10 |
| $\mathrm{X}_{\mathrm{Fe}}$ | 87 | 87 | 86 | 86 | 87 | 83 | 82 | 87 |
| Tot | 100 | 100 | 100 | 100 | 100 | 100 | 100 | 100 |
| $\mathrm{X}_{\mathrm{Mn}}=\mathrm{Mn} /\left(\mathrm{Mn}+\mathrm{Fe}^{2+}+\mathrm{Mg}\right), \mathrm{X}_{\mathrm{Mg}}=\mathrm{Mg} /\left(\mathrm{Mg}+\mathrm{Mn}^{2+} \mathrm{Fe}^{2+}\right), \mathrm{X}_{\mathrm{Fe}}=\mathrm{Fe}^{2+} /\left(\mathrm{Fe}^{2+}+\mathrm{Mn}+\mathrm{Mg}\right)$. |  |  |  |  |  |  |  |  |
|  |  |  |  |  |  |  |  |  |

Table A. 5 (continued) : Mineral chemistry of the measured chloritoids. Mineral formula calculations are based on 12 oxygens.

| Chloritoid | 178 | 178 | 178 | 178 | 178 | 178 | 178 | 178 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 49 | 50 | 54 | 55 | 96 | 97 | 98 | 178-75 |
| $\mathrm{SiO}_{2}$ | 24.53 | 24.76 | 24.31 | 24.44 | 23.96 | 23.48 | 23.94 | 25.42 |
| $\mathrm{TiO}_{2}$ | 0.05 | 0.00 | 0.01 | 0.00 | 0.08 | 0.07 | 0.01 | 0.00 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 42.71 | 42.57 | 42.32 | 42.36 | 41.41 | 40.23 | 41.76 | 40.49 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.03 | 0.00 | 0.04 | 0.02 | 0.06 | 0.05 | 0.06 |  |
| FeO | 24.54 | 24.52 | 25.62 | 24.76 | 24.88 | 25.02 | 25.06 | 24.27 |
| MnO | 0.54 | 0.57 | 0.82 | 0.59 | 0.80 | 0.88 | 0.46 | 0.56 |
| MgO | 2.65 | 2.60 | 1.69 | 2.40 | 1.93 | 1.62 | 2.11 | 2.64 |
| CaO | 0.01 | 0.00 | 0.00 | 0.01 | 0.01 | 0.02 | 0.01 | 0.02 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.01 | 0.01 | 0.00 | 0.00 | 0.03 | 0.00 | 0.03 | 0.03 |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.02 | 0.01 | 0.03 | 0.01 | 0.05 | 0.03 | 0.04 | 0.00 |
| Total | 95.11 | 95.04 | 94.85 | 94.58 | 93.20 | 91.40 | 93.48 | 93.43 |
| Si | 1.967 | 1.985 | 1.969 | 1.974 | 1.973 | 1.979 | 1.963 | 2.073 |
| Ti | 0.003 | 0.000 | 0.001 | 0.000 | 0.005 | 0.004 | 0.000 | 0.000 |
| A1 | 4.036 | 4.023 | 4.040 | 4.032 | 4.019 | 3.996 | 4.035 | 3.891 |
| Cr | 0.002 | 0.000 | 0.003 | 0.001 | 0.004 | 0.003 | 0.004 | 0.000 |
| $\mathrm{Fe}^{3+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{2+}$ | 1.646 | 1.644 | 1.735 | 1.672 | 1.713 | 1.764 | 1.719 | 1.655 |
| Mn | 0.037 | 0.039 | 0.056 | 0.040 | 0.056 | 0.063 | 0.032 | 0.039 |
| Mg | 0.317 | 0.311 | 0.204 | 0.289 | 0.236 | 0.203 | 0.258 | 0.321 |
| Ca | 0.001 | 0.000 | 0.000 | 0.000 | 0.001 | 0.001 | 0.000 | 0.002 |
| Na | 0.002 | 0.001 | 0.001 | 0.000 | 0.005 | 0.000 | 0.004 | 0.005 |
| K | 0.002 | 0.001 | 0.003 | 0.001 | 0.005 | 0.003 | 0.004 | 0.000 |
| Total | 8.013 | 8.004 | 8.011 | 8.010 | 8.016 | 8.018 | 8.021 | 7.984 |
| $\mathrm{X}_{\text {Mn }}$ | 2 | 2 | 3 | 2 | 3 | 3 | 2 | 2 |
| $\mathrm{X}_{\mathrm{Mg}}$ | 16 | 16 | 10 | 14 | 12 | 10 | 13 | 16 |
| $\mathrm{X}_{\mathrm{Fe}}$ | 82 | 82 | 87 | 84 | 85 | 87 | 86 | 82 |
| Tot | 100 | 100 | 100 | 100 | 100 | 100 | 100 | 100 |

Table A. 5 (continued) : Mineral chemistry of the measured chloritoids. Mineral formula calculations are based on 12 oxygens.

| Chloritoid | 178 | 178 | 178 | 178 | 178 | 178 | 178 | 178 |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $178-76$ | $178-77$ | $178-78$ | $178-79$ | $178-83$ | $178-84$ | $178-85$ | $178-86$ |
| $\mathrm{SiO}_{2}$ | 25.49 | 25.43 | 25.37 | 25.13 | 25.31 | 25.17 | 25.18 | 25.20 |
| $\mathrm{TiO}_{2}$ | 0.01 | 0.04 | 0.00 | 0.00 | 0.02 | 0.03 | 0.04 | 0.03 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 40.03 | 40.56 | 40.50 | 40.45 | 40.18 | 40.26 | 39.47 | 40.06 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ |  |  |  |  |  |  |  |  |
| FeO | 25.26 | 23.97 | 23.92 | 24.11 | 25.52 | 25.44 | 25.32 | 25.22 |
| MnO | 0.50 | 0.53 | 0.56 | 0.64 | 0.80 | 0.76 | 0.66 | 0.66 |
| MgO | 2.13 | 2.77 | 2.75 | 2.57 | 1.76 | 1.99 | 1.96 | 2.01 |
| CaO | 0.01 | 0.02 | 0.02 | 0.01 | 0.00 | 0.01 | 0.01 | 0.00 |
| Na 2 O | 0.00 | 0.01 | 0.01 | 0.00 | 0.02 | 0.00 | 0.02 | 0.04 |
| $\mathrm{~K}_{2} \mathrm{O}$ | 0.02 | 0.04 | 0.02 | 0.05 | 0.00 | 0.01 | 0.08 | 0.00 |
| Total | 93.45 | 93.37 | 93.16 | 92.96 | 93.61 | 93.67 | 92.74 | 93.21 |
|  |  |  |  |  |  |  |  |  |
| Si | 2.088 | 2.072 | 2.071 | 2.061 | 2.075 | 2.062 | 2.085 | 2.072 |
| Ti | 0.001 | 0.002 | 0.000 | 0.000 | 0.001 | 0.002 | 0.003 | 0.002 |
| Al | 3.864 | 3.894 | 3.897 | 3.909 | 3.883 | 3.888 | 3.851 | 3.882 |
| Cr | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{3+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{2+}$ | 1.730 | 1.633 | 1.633 | 1.653 | 1.750 | 1.743 | 1.753 | 1.734 |
| Mn | 0.035 | 0.037 | 0.039 | 0.045 | 0.055 | 0.053 | 0.047 | 0.046 |
| Mg | 0.260 | 0.336 | 0.335 | 0.314 | 0.215 | 0.243 | 0.242 | 0.246 |
| Ca | 0.001 | 0.002 | 0.002 | 0.001 | 0.000 | 0.000 | 0.001 | 0.000 |
| Na | 0.000 | 0.002 | 0.002 | 0.000 | 0.002 | 0.000 | 0.003 | 0.006 |
| K | 0.002 | 0.004 | 0.002 | 0.005 | 0.000 | 0.001 | 0.009 | 0.000 |
| Total | 7.980 | 7.982 | 7.982 | 7.988 | 7.983 | 7.992 | 7.993 | 7.988 |
| $\mathrm{X}_{\mathrm{Mn}}$ | 2 | 2 | 2 | 2 | 3 | 3 | 2 | 2 |
| $\mathrm{X}_{\mathrm{Mg}}$ | 13 | 17 | 17 | 16 | 11 | 12 | 12 | 12 |
| $\mathrm{X}_{\mathrm{Fe}}$ | 85 | 81 | 81 | 82 | 87 | 85 | 86 | 86 |
| Tot | 100 | 100 | 100 | 100 | 100 | 100 | 100 | 100 |
| $\mathrm{X}_{\mathrm{Mn}}=\mathrm{Mn} /\left(\mathrm{Mn}+\mathrm{Fe}^{2+}+\mathrm{Mg}\right), \mathrm{X}_{\mathrm{Mg}}=\mathrm{Mg} /\left(\mathrm{Mg}+\mathrm{Mn}^{2+} \mathrm{Fe}^{2+}\right), \mathrm{X}_{\mathrm{Fe}}=\mathrm{Fe}^{2+} /\left(\mathrm{Fe}^{2+}+\mathrm{Mn}+\mathrm{Mg}\right)$. |  |  |  |  |  |  |  |  |
|  |  |  |  |  |  |  |  |  |

Table A. 5 (continued) : Mineral chemistry of the measured chloritoids. Mineral formula calculations are based on 12 oxygens.

| Chloritoid | 178 | 178 | 178 | 178 | 178 | 178 | 178 | 178 |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $178-87$ | $178-88$ | $178-89$ | $178-90$ | $178-91$ | $178-92$ | $178-93$ | $178-94$ |
| $\mathrm{SiO}_{2}$ | 25.51 | 25.12 | 25.36 | 25.24 | 25.53 | 25.34 | 25.10 | 25.11 |
| $\mathrm{TiO}_{2}$ | 0.00 | 0.00 | 0.00 | 0.01 | 0.03 | 0.02 | 0.00 | 0.00 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 40.33 | 40.14 | 40.00 | 40.15 | 40.28 | 40.11 | 40.09 | 39.98 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ |  |  |  |  |  |  |  |  |
| FeO | 25.17 | 25.43 | 25.36 | 25.14 | 25.68 | 25.31 | 25.57 | 25.34 |
| MnO | 0.59 | 0.75 | 0.89 | 0.81 | 0.93 | 0.79 | 0.80 | 0.89 |
| MgO | 2.12 | 1.93 | 1.94 | 2.00 | 1.73 | 2.16 | 1.95 | 1.71 |
| CaO | 0.00 | 0.00 | 0.00 | 0.01 | 0.01 | 0.00 | 0.01 | 0.01 |
| $\mathrm{Na} \mathrm{O}_{2} \mathrm{O}$ | 0.02 | 0.00 | 0.00 | 0.00 | 0.00 | 0.02 | 0.01 | 0.00 |
| $\mathrm{~K}_{2} \mathrm{O}$ | 0.02 | 0.00 | 0.03 | 0.02 | 0.00 | 0.02 | 0.00 | 0.06 |
| Total | 93.76 | 93.38 | 93.58 | 93.38 | 94.19 | 93.77 | 93.53 | 93.09 |
| Si | 2.082 | 2.065 | 2.080 | 2.072 | 2.082 | 2.073 | 2.062 | 2.072 |
| Ti | 0.000 | 0.000 | 0.000 | 0.001 | 0.002 | 0.002 | 0.000 | 0.000 |
| Al | 3.879 | 3.889 | 3.867 | 3.884 | 3.871 | 3.867 | 3.882 | 3.888 |
| Cr | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{3+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{2+}$ | 1.718 | 1.748 | 1.739 | 1.726 | 1.751 | 1.731 | 1.757 | 1.749 |
| Mn | 0.041 | 0.052 | 0.062 | 0.057 | 0.064 | 0.055 | 0.056 | 0.062 |
| Mg | 0.258 | 0.237 | 0.237 | 0.245 | 0.210 | 0.263 | 0.239 | 0.210 |
| Ca | 0.000 | 0.000 | 0.000 | 0.001 | 0.001 | 0.000 | 0.001 | 0.001 |
| Na | 0.003 | 0.001 | 0.000 | 0.000 | 0.000 | 0.002 | 0.001 | 0.000 |
| K | 0.002 | 0.000 | 0.003 | 0.002 | 0.000 | 0.002 | 0.000 | 0.006 |
| Total | 7.981 | 7.991 | 7.989 | 7.986 | 7.981 | 7.995 | 7.998 | 7.987 |
| $\mathrm{X}_{\mathrm{Mn}}$ | 2 | 3 | 3 | 3 | 3 | 3 | 3 | 3 |
| $\mathrm{X}_{\mathrm{Mg}}$ | 13 | 12 | 12 | 12 | 10 | 13 | 12 | 10 |
| $\mathrm{X}_{\mathrm{Fe}}$ | 85 | 86 | 85 | 85 | 86 | 84 | 86 | 87 |
| Tot | 100 | 100 | 100 | 100 | 100 | 100 | 100 | 100 |
| $\mathrm{X}_{\mathrm{Mn}}=\mathrm{Mn} /\left(\mathrm{Mn}+\mathrm{Fe}^{2+}+\mathrm{Mg}\right), \mathrm{X}_{\mathrm{Mg}}=\mathrm{Mg} /\left(\mathrm{Mg}+{\left.\mathrm{Mn}+\mathrm{Fe}^{2+}\right), \mathrm{X}_{\mathrm{Fe}}=\mathrm{Fe}^{2+} /\left(\mathrm{Fe}^{2+}+\mathrm{Mn}+\mathrm{Mg}\right) .}\right.$ |  |  |  |  |  |  |  |  |
|  |  |  |  |  |  |  |  |  |

Table A. 5 (continued) : Mineral chemistry of the measured chloritoids. Mineral formula calculations are based on 12 oxygens.

| Chloritoid | 178 | 178 | 178 | 178 | 178 | 178 | 178 | 178 |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $178-95$ | $178-96$ | $178-97$ | $178-98$ | $178-99$ | $178-100$ | $178-101$ | $178-102$ |
| $\mathrm{SiO}_{2}$ | 25.32 | 25.53 | 25.45 | 25.46 | 25.56 | 25.35 | 25.31 | 25.31 |
| $\mathrm{TiO}_{2}$ | 0.04 | 0.01 | 0.00 | 0.07 | 0.01 | 0.00 | 0.04 | 0.03 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 39.76 | 39.86 | 39.82 | 39.32 | 40.02 | 40.34 | 40.05 | 40.04 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ |  |  |  |  |  |  |  |  |
| FeO | 25.04 | 25.50 | 25.90 | 25.39 | 25.37 | 25.10 | 25.57 | 24.99 |
| MnO | 0.79 | 0.74 | 0.60 | 0.65 | 0.89 | 0.76 | 0.91 | 0.73 |
| MgO | 2.18 | 2.01 | 1.99 | 1.90 | 2.01 | 2.24 | 1.67 | 2.04 |
| CaO | 0.01 | 0.02 | 0.01 | 0.01 | 0.02 | 0.01 | 0.00 | 0.00 |
| Na 2 O | 0.04 | 0.00 | 0.00 | 0.04 | 0.00 | 0.00 | 0.00 | 0.00 |
| $\mathrm{~K}_{2} \mathrm{O}$ | 0.03 | 0.03 | 0.04 | 0.30 | 0.02 | 0.02 | 0.01 | 0.05 |
| Total | 93.21 | 93.70 | 93.81 | 93.13 | 93.89 | 93.83 | 93.57 | 93.19 |
| Si | 2.082 | 2.091 | 2.085 | 2.101 | 2.088 | 2.069 | 2.078 | 2.080 |
| Ti | 0.003 | 0.001 | 0.000 | 0.004 | 0.000 | 0.000 | 0.003 | 0.002 |
| Al | 3.854 | 3.847 | 3.845 | 3.825 | 3.854 | 3.880 | 3.876 | 3.877 |
| Cr | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Se}^{3+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{2+}$ | 1.722 | 1.746 | 1.775 | 1.753 | 1.733 | 1.713 | 1.756 | 1.717 |
| Mn | 0.055 | 0.051 | 0.042 | 0.045 | 0.062 | 0.053 | 0.064 | 0.051 |
| Mg | 0.267 | 0.245 | 0.243 | 0.234 | 0.245 | 0.273 | 0.204 | 0.250 |
| Ca | 0.001 | 0.002 | 0.001 | 0.000 | 0.001 | 0.001 | 0.000 | 0.000 |
| Na | 0.006 | 0.000 | 0.000 | 0.007 | 0.000 | 0.000 | 0.000 | 0.000 |
| K | 0.003 | 0.003 | 0.004 | 0.032 | 0.002 | 0.002 | 0.001 | 0.005 |
| Total | 7.993 | 7.987 | 7.995 | 8.001 | 7.985 | 7.992 | 7.982 | 7.982 |
| $\mathrm{X}_{\mathrm{Mn}}$ | 3 | 2 | 2 | 2 | 3 | 3 | 3 | 3 |
| $\mathrm{X}_{\mathrm{Mg}}$ | 13 | 12 | 12 | 12 | 12 | 13 | 10 | 12 |
| $\mathrm{X}_{\mathrm{Fe}}$ | 84 | 85 | 86 | 86 | 85 | 84 | 87 | 85 |
| Tot | 100 | 100 | 100 | 100 | 100 | 100 | 100 | 100 |
| $\mathrm{X}_{\mathrm{Mn}}=\mathrm{Mn} /\left(\mathrm{Mn}+\mathrm{Fe}^{2+}+\mathrm{Mg}\right), \mathrm{X}_{\mathrm{Mg}}=\mathrm{Mg} /\left(\mathrm{Mg}+{\left.\mathrm{Mn}+\mathrm{Fe}^{2+}\right), \mathrm{X}_{\mathrm{Fe}}=\mathrm{Fe}^{2+} /\left(\mathrm{Fe}^{2+}+\mathrm{Mn}+\mathrm{Mg}\right) .}\right.$ |  |  |  |  |  |  |  |  |

Table A. 5 (continued) : Mineral chemistry of the measured chloritoids. Mineral formula calculations are based on 12 oxygens.

| Chloritoid | 178 | 178 | 178 |
| :---: | :---: | :---: | :---: |
|  | 178-103 | 178-104 | 178-105 |
| $\mathrm{SiO}_{2}$ | 25.34 | 25.44 | 25.41 |
| $\mathrm{TiO}_{2}$ | 0.04 | 0.02 | 0.07 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 39.51 | 40.54 | 39.83 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ |  |  |  |
| FeO | 25.52 | 24.59 | 25.00 |
| MnO | 1.01 | 0.59 | 0.75 |
| MgO | 1.72 | 2.48 | 2.17 |
| CaO | 0.00 | 0.01 | 0.01 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.02 | 0.00 | 0.02 |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.02 | 0.03 | 0.05 |
| Total | 93.19 | 93.70 | 93.30 |
| Si | 2.091 | 2.072 | 2.086 |
| Ti | 0.002 | 0.001 | 0.004 |
| A1 | 3.843 | 3.891 | 3.854 |
| Cr | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{3+}$ | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{2+}$ | 1.762 | 1.675 | 1.716 |
| Mn | 0.071 | 0.040 | 0.052 |
| Mg | 0.212 | 0.301 | 0.266 |
| Ca | 0.000 | 0.001 | 0.001 |
| Na | 0.004 | 0.000 | 0.003 |
| K | 0.002 | 0.003 | 0.005 |
| Total | 7.987 | 7.984 | 7.987 |
| $\mathrm{X}_{\mathrm{Mn}}$ | 3 | 2 | 3 |
| $\mathrm{X}_{\mathrm{Mg}}$ | 10 | 15 | 13 |
| $\mathrm{X}_{\mathrm{Fe}}$ | 86 | 83 | 84 |
| Tot | 100 | 100 | 100 |

Table A.6 : Mineral chemistry of the measured lawsonites. Mineral formula calculations are based on eight oxygens.

| Lawsonite | 128 | 128 | 128 | 128 | 128 |
| :--- | :---: | :---: | :---: | :---: | :---: |
|  | 155 | 157 | 161 | 170 | 185 |
| $\mathrm{SiO}_{2}$ | 38.65 | 38.19 | 41.83 | 38.50 | 38.49 |
| $\mathrm{TiO}_{2}$ | 0.40 | 0.24 | 0.07 | 0.04 | 0.37 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 31.46 | 31.08 | 29.51 | 31.89 | 31.17 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.02 | 0.01 | 0.00 | 0.02 | 0.00 |
| FeO | 0.59 | 0.98 | 0.71 | 0.83 | 0.60 |
| MnO | 0.06 | 0.01 | 0.00 | 0.00 | 0.00 |
| MgO | 0.03 | 0.00 | 0.00 | 0.01 | 0.00 |
| CaO | 16.85 | 17.45 | 16.18 | 16.95 | 17.05 |
| $\mathrm{Na}{ }_{2} \mathrm{O}$ | 0.01 | 0.00 | 0.03 | 0.01 | 0.00 |
| $\mathrm{~K}_{2} \mathrm{O}$ | 0.01 | 0.01 | 0.01 | 0.01 | 0.01 |
| Total | 88.07 | 87.97 | 88.35 | 88.25 | 87.71 |
|  |  |  |  |  |  |
| Si | 2.031 | 2.020 | 2.174 | 2.022 | 2.033 |
| Ti | 0.016 | 0.010 | 0.003 | 0.002 | 0.015 |
| Al | 1.949 | 1.938 | 1.808 | 1.974 | 1.941 |
| Cr | 0.001 | 0.001 | 0.000 | 0.001 | 0.000 |
| Fe |  | 0.000 | 0.000 | 0.000 | 0.000 |
| Fe | 0.000 |  |  |  |  |
| Mn | 0.026 | 0.043 | 0.031 | 0.036 | 0.026 |
| Mg | 0.002 | 0.000 | 0.000 | 0.000 | 0.000 |
| Ca | 0.003 | 0.000 | 0.000 | 0.000 | 0.000 |
| Na | 0.949 | 0.989 | 0.901 | 0.954 | 0.965 |
| K | 0.001 | 0.000 | 0.003 | 0.001 | 0.000 |
| Total | 0.001 | 0.001 | 0.001 | 0.000 | 0.001 |
|  | 4.979 | 5.001 | 4.921 | 4.990 | 4.982 |

Table A. 7 : Mineral chemistry of the measured sodic-pyroxenes. Mineral formula calculations are based on four cations.

| Sodicpyroxene | 128 | 128 | 128 | 128 | 128 | 128 | 128 | 211B | 211B | 211B |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 140 | 141 | 142 | 147 | 149 | 172 | 203 | 289 | 318 | 321 |
| $\mathrm{SiO}_{2}$ | 53.88 | 55.02 | 53.52 | 55.43 | 54.14 | 54.30 | 54.85 | 51.82 | 51.58 | 52.02 |
| $\mathrm{TiO}_{2}$ | 0.00 | 0.08 | 0.01 | 0.08 | 0.02 | 0.01 | 0.03 | 0.03 | 0.09 | 0.08 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 2.80 | 8.71 | 2.20 | 9.17 | 2.34 | 4.62 | 6.53 | 1.49 | 1.73 | 2.55 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.00 | 0.01 | 0.01 | 0.00 | 0.00 | 0.01 | 0.02 | 0.00 | 0.00 | 0.02 |
| FeO | 13.43 | 12.11 | 13.10 | 12.08 | 12.67 | 11.81 | 13.42 | 20.07 | 19.47 | 15.99 |
| MnO | 0.61 | 0.33 | 0.58 | 0.27 | 0.61 | 0.89 | 0.29 | 0.24 | 0.21 | 0.22 |
| MgO | 8.44 | 5.20 | 9.26 | 5.13 | 9.44 | 8.19 | 5.96 | 5.56 | 5.83 | 7.25 |
| CaO | 15.12 | 9.61 | 16.34 | 9.41 | 16.44 | 14.68 | 11.02 | 12.83 | 12.34 | 13.11 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 5.50 | 8.96 | 4.94 | 9.03 | 4.88 | 5.66 | 8.14 | 5.84 | 5.93 | 5.33 |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.01 | 0.01 | 0.02 | 0.01 | 0.01 | 0.02 | 0.01 | 0.02 | 0.02 | 0.04 |
| Total | 99.77 | 100.03 | 99.96 | 100.60 | 100.55 | 100.20 | 100.28 | 97.90 | 97.19 | 96.63 |
| Si | 1.989 | 1.979 | 1.973 | 1.982 | 1.983 | 1.987 | 1.985 | 1.994 | 1.992 | 2.002 |
| Ti | 0.000 | 0.002 | 0.000 | 0.002 | 0.000 | 0.000 | 0.001 | 0.001 | 0.002 | 0.002 |
| Al | 0.122 | 0.369 | 0.095 | 0.386 | 0.101 | 0.199 | 0.278 | 0.068 | 0.079 | 0.116 |
| Cr | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.001 |
| $\mathrm{Fe}^{3+}$ | 0.294 | 0.293 | 0.311 | 0.271 | 0.278 | 0.229 | 0.322 | 0.380 | 0.378 | 0.274 |
| $\mathrm{Fe}^{2+}$ | 0.120 | 0.071 | 0.093 | 0.090 | 0.110 | 0.132 | 0.085 | 0.266 | 0.251 | 0.240 |
| Mn | 0.019 | 0.010 | 0.018 | 0.008 | 0.019 | 0.028 | 0.009 | 0.008 | 0.007 | 0.007 |
| Mg | 0.464 | 0.279 | 0.509 | 0.274 | 0.516 | 0.447 | 0.322 | 0.319 | 0.336 | 0.416 |
| Ca | 0.598 | 0.371 | 0.646 | 0.360 | 0.645 | 0.576 | 0.427 | 0.529 | 0.510 | 0.541 |
| Na | 0.393 | 0.625 | 0.353 | 0.626 | 0.346 | 0.402 | 0.571 | 0.435 | 0.444 | 0.398 |
| K | 0.000 | 0.000 | 0.001 | 0.000 | 0.000 | 0.001 | 0.001 | 0.001 | 0.001 | 0.002 |
| Total | 4.000 | 4.000 | 4.000 | 4.000 | 4.000 | 4.000 | 4.000 | 4.000 | 4.000 | 4.000 |
| Aegerine | 30 | 29 | 31 | 28 | 28 | 23 | 32 | 39 | 40 | 29 |
| Jadeite | 10 | 33 | 4 | 36 | 7 | 18 | 25 | 6 | 7 | 13 |
| Diopside | 60 | 37 | 65 | 37 | 65 | 59 | 43 | 55 | 53 | 58 |
| Total | 100 | 100 | 100 | 100 | 100 | 100 | 100 | 100 | 100 | 100 |

Table A.8: Mineral chemistry of the measured epidotes. Mineral formula calculations are based on 12.5 oxygens.

| Epidote | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 4 | 6 | 8 | 12 | 14 | 15 | 16 | 17 | 19 | 20 | 21 | 22 | 24 |
| $\mathrm{SiO}_{2}$ | 38.98 | 39.23 | 39.65 | 40.16 | 40.53 | 39.96 | 40.11 | 40.93 | 40.43 | 40.31 | 39.64 | 40.24 | 39.74 |
| $\mathrm{TiO}_{2}$ | 0.09 | 0.08 | 0.06 | 0.06 | 0.07 | 0.08 | 0.04 | 0.06 | 0.06 | 0.08 | 0.03 | 0.07 | 0.08 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 23.01 | 29.20 | 24.90 | 30.27 | 30.38 | 26.26 | 24.96 | 32.25 | 30.62 | 29.88 | 26.00 | 30.43 | 26.28 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ |  |  |  |  |  |  |  |  |  |  |  |  |  |
| FeO | 19.04 | 5.32 | 10.18 | 3.13 | 3.65 | 7.97 | 9.52 | 2.39 | 2.38 | 4.33 | 8.57 | 3.77 | 8.14 |
| MnO | 2.71 | 0.09 | 0.22 | 0.26 | 0.16 | 0.09 | 0.03 | 0.01 | 0.04 | 0.02 | 0.05 | 0.15 | 0.07 |
| MgO | 0.41 | 0.65 | 0.03 | 0.68 | 0.02 | 0.03 | 0.13 | 0.04 | 0.00 | 0.12 | 0.03 | 0.03 | 0.29 |
| CaO | 14.14 | 22.65 | 23.05 | 22.60 | 23.82 | 23.58 | 22.68 | 21.95 | 24.12 | 23.85 | 23.53 | 23.92 | 22.94 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.00 | 0.00 | 0.00 | 0.00 | 0.03 | 0.00 | 0.00 | 0.19 | 0.00 | 0.00 | 0.02 | 0.00 | 0.01 |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.00 | 0.00 | 0.01 | 0.03 | 0.01 | 0.05 | 0.49 | 0.74 | 0.00 | 0.00 | 0.03 | 0.00 | 0.03 |
| Total | 98.39 | 97.23 | 98.09 | 97.19 | 98.66 | 98.02 | 97.98 | 98.57 | 97.66 | 98.58 | 97.89 | 98.62 | 97.59 |
| Si | 3.032 | 3.023 | 3.076 | 3.074 | 3.067 | 3.086 | 3.110 | 3.078 | 3.083 | 3.059 | 3.071 | 3.050 | 3.078 |
| Ti | 0.005 | 0.005 | 0.004 | 0.003 | 0.004 | 0.005 | 0.003 | 0.004 | 0.004 | 0.004 | 0.002 | 0.004 | 0.005 |
| Al | 2.109 | 2.652 | 2.277 | 2.731 | 2.710 | 2.390 | 2.281 | 2.859 | 2.752 | 2.672 | 2.374 | 2.718 | 2.399 |
| Cr | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{3+}$ | 1.238 | 0.343 | 0.661 | 0.200 | 0.231 | 0.515 | 0.617 | 0.150 | 0.152 | 0.275 | 0.555 | 0.239 | 0.527 |
| $\mathrm{Fe}^{2+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| Mn | 0.179 | 0.006 | 0.014 | 0.017 | 0.010 | 0.006 | 0.002 | 0.001 | 0.002 | 0.001 | 0.003 | 0.010 | 0.005 |
| Mg | 0.048 | 0.075 | 0.003 | 0.078 | 0.002 | 0.003 | 0.015 | 0.004 | 0.000 | 0.013 | 0.003 | 0.003 | 0.034 |
| Ca | 1.178 | 1.870 | 1.916 | 1.853 | 1.932 | 1.951 | 1.884 | 1.769 | 1.970 | 1.939 | 1.953 | 1.943 | 1.904 |
| Na | 0.000 | 0.000 | 0.000 | 0.000 | 0.004 | 0.000 | 0.000 | 0.028 | 0.000 | 0.000 | 0.003 | 0.000 | 0.002 |
| K | 0.000 | 0.000 | 0.001 | 0.003 | 0.001 | 0.005 | 0.048 | 0.071 | 0.000 | 0.000 | 0.003 | 0.000 | 0.003 |
| Total | 7.789 | 7.975 | 7.952 | 7.959 | 7.961 | 7.960 | 7.962 | 7.963 | 7.962 | 7.963 | 7.966 | 7.967 | 7.956 |
| $\mathrm{Fe}^{3+} /\left(\mathrm{Fe}^{3}\right.$ | 0.37 | 0.11 | 0.22 | 0.07 | 0.08 | 0.18 | 0.21 | 0.05 | 0.05 | 0.09 | 0.19 | 0.08 | 0.18 |

Table A. 8 (continued) : Mineral chemistry of the measured epidotes. Mineral formula calculations are based on 12.5 oxygens.

| Epidote | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 31 | 32 | 49 | 50 | 69 | 82 | 93 | 97 | 102 | 104 | 111 | 119 | 130 |
| $\mathrm{SiO}_{2}$ | 38.96 | 38.10 | 39.10 | 38.37 | 39.08 | 38.30 | 38.59 | 38.29 | 39.08 | 38.76 | 38.37 | 38.97 | 39.53 |
| $\mathrm{TiO}_{2}$ | 0.08 | 0.17 | 0.04 | 0.07 | 0.04 | 0.06 | 0.06 | 0.08 | 0.07 | 0.06 | 0.07 | 0.06 | 0.06 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 31.11 | 25.96 | 32.00 | 26.22 | 29.35 | 25.93 | 25.57 | 26.37 | 26.36 | 30.04 | 26.07 | 30.78 | 31.26 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.03 | 0.00 | 0.00 | 0.02 | 0.01 | 0.01 | 0.04 | 0.01 | 0.01 | 0.01 | 0.00 | 0.00 | 0.00 |
| FeO | 3.46 | 9.63 | 3.06 | 9.30 | 5.67 | 9.86 | 10.39 | 9.46 | 8.88 | 4.65 | 10.04 | 4.44 | 3.54 |
| MnO | 0.02 | 0.03 | 0.08 | 0.03 | 0.09 | 0.11 | 0.06 | 0.05 | 0.06 | 0.07 | 0.32 | 0.07 | 0.07 |
| MgO | 0.00 | 0.00 | 0.02 | 0.01 | 0.01 | 0.02 | 0.00 | 0.01 | 0.28 | 0.02 | 0.03 | 0.20 | 0.03 |
| CaO | 24.20 | 23.35 | 24.12 | 23.35 | 23.80 | 23.26 | 23.38 | 23.37 | 23.33 | 24.13 | 22.91 | 23.66 | 23.87 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.04 | 0.02 | 0.02 | 0.00 | 0.01 | 0.03 | 0.00 | 0.02 | 0.07 | 0.01 | 0.00 | 0.02 | 0.00 |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.01 | 0.01 | 0.00 | 0.00 | 0.00 | 0.00 | 0.03 | 0.01 | 0.01 | 0.04 | 0.00 | 0.01 | 0.03 |
| Total | 97.90 | 97.26 | 98.44 | 97.37 | 98.05 | 97.57 | 98.12 | 97.65 | 98.15 | 97.80 | 97.80 | 98.19 | 98.39 |
| Si | 2.981 | 2.986 | 2.969 | 2.999 | 3.000 | 2.993 | 3.002 | 2.986 | 3.024 | 2.981 | 2.990 | 2.975 | 3.003 |
| Ti | 0.005 | 0.010 | 0.003 | 0.004 | 0.002 | 0.003 | 0.003 | 0.004 | 0.004 | 0.003 | 0.004 | 0.003 | 0.004 |
| Al | 2.805 | 2.398 | 2.863 | 2.415 | 2.656 | 2.388 | 2.344 | 2.424 | 2.404 | 2.723 | 2.395 | 2.769 | 2.799 |
| Cr | 0.002 | 0.000 | 0.000 | 0.001 | 0.001 | 0.000 | 0.003 | 0.000 | 0.001 | 0.001 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{3+}$ | 0.221 | 0.631 | 0.194 | 0.608 | 0.364 | 0.644 | 0.676 | 0.617 | 0.575 | 0.299 | 0.655 | 0.284 | 0.225 |
| $\mathrm{Fe}^{2+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| Mn | 0.001 | 0.002 | 0.005 | 0.002 | 0.006 | 0.007 | 0.004 | 0.003 | 0.004 | 0.005 | 0.021 | 0.004 | 0.004 |
| Mg | 0.000 | 0.000 | 0.002 | 0.002 | 0.001 | 0.002 | 0.000 | 0.001 | 0.032 | 0.002 | 0.003 | 0.022 | 0.003 |
| Ca | 1.983 | 1.961 | 1.962 | 1.955 | 1.958 | 1.947 | 1.949 | 1.952 | 1.934 | 1.988 | 1.913 | 1.935 | 1.943 |
| Na | 0.006 | 0.004 | 0.004 | 0.000 | 0.001 | 0.005 | 0.000 | 0.003 | 0.011 | 0.001 | 0.000 | 0.003 | 0.000 |
| K | 0.001 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.003 | 0.001 | 0.001 | 0.004 | 0.000 | 0.001 | 0.003 |
| Total | 8.005 | 7.992 | 8.002 | 7.985 | 7.988 | 7.990 | 7.984 | 7.991 | 7.989 | 8.007 | 7.981 | 7.997 | 7.984 |
| $\mathrm{Fe}^{3+} /\left(\mathrm{Fe}^{3+}+\mathrm{Al}\right)$ | 0.07 | 0.21 | 0.06 | 0.20 | 0.12 | 0.21 | 0.22 | 0.20 | 0.19 | 0.10 | 0.21 | 0.09 | 0.07 |

Table A. 8 (continued) : Mineral chemistry of the measured epidotes. Mineral formula calculations are based on 12.5 oxygens.

| Epidote | 128 | 128 | 128 | 128 | 128 | 211B | 211B | 211B | 211B | 211B | 211B |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 156 | 158 | 164 | 184 | 197 | 240 | 241 | 247 | 248 | 251 | 264 |
| $\mathrm{SiO}_{2}$ | 37.92 | 39.33 | 37.54 | 39.51 | 38.40 | 38.13 | 37.28 | 37.42 | 37.30 | 38.22 | 37.80 |
| $\mathrm{TiO}_{2}$ | 0.06 | 0.06 | 0.16 | 0.05 | 0.03 | 0.13 | 0.04 | 0.07 | 0.09 | 0.04 | 0.16 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 24.11 | 30.58 | 21.53 | 32.71 | 25.13 | 28.55 | 22.69 | 22.43 | 23.00 | 22.91 | 25.76 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.01 | 0.00 | 0.01 | 0.00 | 0.00 | 0.00 | 0.01 | 0.00 | 0.01 | 0.01 | 0.01 |
| FeO | 12.25 | 4.39 | 15.03 | 1.17 | 10.80 | 6.74 | 13.91 | 14.18 | 13.50 | 13.39 | 10.84 |
| MnO | 0.41 | 0.20 | 0.05 | 0.12 | 0.12 | 0.15 | 0.13 | 0.25 | 0.18 | 0.18 | 0.48 |
| MgO | 0.02 | 0.00 | 0.01 | 0.01 | 0.02 | 0.02 | 0.00 | 0.00 | 0.01 | 0.00 | 0.05 |
| CaO | 22.70 | 24.05 | 22.93 | 23.96 | 23.60 | 23.06 | 22.40 | 22.27 | 22.50 | 22.15 | 22.17 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.00 | 0.01 | 0.00 | 0.01 | 0.00 | 0.02 | 0.00 | 0.01 | 0.00 | 0.02 | 0.00 |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.00 | 0.05 | 0.01 | 0.00 | 0.01 | 0.01 | 0.01 | 0.00 | 0.01 | 0.01 | 0.01 |
| Total | 97.48 | 98.66 | 97.25 | 97.55 | 98.09 | 96.79 | 96.47 | 96.62 | 96.60 | 96.93 | 97.29 |
| Si | 2.988 | 2.992 | 2.992 | 3.008 | 2.996 | 2.974 | 2.981 | 2.990 | 2.977 | 3.029 | 2.967 |
| Ti | 0.004 | 0.004 | 0.010 | 0.003 | 0.002 | 0.008 | 0.002 | 0.004 | 0.006 | 0.003 | 0.010 |
| A1 | 2.239 | 2.742 | 2.022 | 2.935 | 2.310 | 2.624 | 2.139 | 2.112 | 2.163 | 2.140 | 2.383 |
| Cr | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.001 | 0.001 | 0.001 |
| $\mathrm{Fe}^{3+}$ | 0.807 | 0.279 | 1.000 | 0.075 | 0.704 | 0.439 | 0.930 | 0.947 | 0.901 | 0.887 | 0.712 |
| $\mathrm{Fe}^{2+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| Mn | 0.027 | 0.013 | 0.003 | 0.008 | 0.008 | 0.010 | 0.009 | 0.017 | 0.012 | 0.012 | 0.032 |
| Mg | 0.003 | 0.000 | 0.001 | 0.001 | 0.003 | 0.002 | 0.000 | 0.000 | 0.001 | 0.000 | 0.006 |
| Ca | 1.916 | 1.960 | 1.957 | 1.954 | 1.972 | 1.927 | 1.919 | 1.906 | 1.924 | 1.881 | 1.865 |
| Na | 0.000 | 0.002 | 0.000 | 0.002 | 0.000 | 0.002 | 0.000 | 0.001 | 0.000 | 0.002 | 0.000 |
| K | 0.000 | 0.005 | 0.001 | 0.000 | 0.001 | 0.001 | 0.001 | 0.000 | 0.001 | 0.001 | 0.001 |
| Total | 7.985 | 7.997 | 7.986 | 7.986 | 7.996 | 7.988 | 7.982 | 7.977 | 7.985 | 7.956 | 7.976 |
| $\mathrm{Fe}^{3+} /\left(\mathrm{Fe}^{3+}+\mathrm{Al}\right)$ | 0.26 | 0.09 | 0.33 | 0.02 | 0.23 | 0.14 | 0.30 | 0.31 | 0.29 | 0.29 | 0.23 |

Table A. 8 (continued) : Mineral chemistry of the measured epidotes. Mineral formula calculations are based on 12.5 oxygens.

| Epidote | 211B | 211B | 211B | 211B | 211B | 211B | 211B | 211B | 211B | 211B | 211B |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 265 | 266 | 287 | 294 | 296 | 303 | 304 | 306 | 307 | 308 | 314 |
| $\mathrm{SiO}_{2}$ | 38.62 | 37.42 | 38.30 | 37.94 | 39.20 | 37.32 | 36.63 | 37.40 | 37.74 | 37.31 | 37.31 |
| $\mathrm{TiO}_{2}$ | 0.05 | 0.04 | 0.05 | 0.02 | 0.05 | 0.16 | 0.04 | 0.08 | 0.05 | 0.01 | 0.04 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 28.23 | 22.72 | 22.69 | 22.76 | 27.50 | 20.79 | 20.43 | 25.17 | 24.75 | 22.73 | 22.75 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.00 | 0.00 | 0.01 | 0.00 | 0.01 | 0.01 | 0.00 | 0.00 | 0.02 | 0.00 | 0.04 |
| FeO | 7.31 | 14.03 | 13.28 | 13.55 | 7.87 | 16.23 | 16.22 | 11.08 | 10.94 | 13.82 | 13.39 |
| MnO | 0.12 | 0.20 | 0.15 | 0.10 | 0.16 | 0.10 | 0.26 | 0.12 | 0.10 | 0.17 | 0.16 |
| MgO | 0.01 | 0.00 | 0.00 | 0.01 | 0.02 | 0.01 | 0.00 | 0.05 | 0.02 | 0.00 | 0.12 |
| CaO | 23.20 | 22.47 | 21.74 | 22.40 | 22.40 | 22.34 | 21.79 | 22.79 | 22.75 | 22.46 | 22.44 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.01 | 0.02 | 0.18 | 0.09 | 0.32 | 0.01 | 0.00 | 0.01 | 0.00 | 0.02 | 0.01 |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.01 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.01 | 0.01 | 0.00 | 0.01 | 0.00 |
| Total | 97.56 | 96.89 | 96.40 | 96.87 | 97.53 | 96.96 | 95.38 | 96.71 | 96.37 | 96.52 | 96.27 |
| Si | 2.991 | 2.981 | 3.048 | 3.015 | 3.035 | 2.990 | 2.986 | 2.960 | 2.994 | 2.982 | 2.987 |
| Ti | 0.003 | 0.002 | 0.003 | 0.001 | 0.003 | 0.010 | 0.003 | 0.005 | 0.003 | 0.001 | 0.003 |
| A1 | 2.577 | 2.133 | 2.128 | 2.131 | 2.509 | 1.963 | 1.962 | 2.348 | 2.314 | 2.142 | 2.147 |
| Cr | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.001 | 0.000 | 0.000 | 0.001 | 0.000 | 0.003 |
| $\mathrm{Fe}^{3+}$ | 0.474 | 0.935 | 0.884 | 0.900 | 0.509 | 1.087 | 1.106 | 0.734 | 0.726 | 0.924 | 0.897 |
| $\mathrm{Fe}^{2+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| Mn | 0.008 | 0.013 | 0.010 | 0.007 | 0.011 | 0.007 | 0.018 | 0.008 | 0.007 | 0.011 | 0.011 |
| Mg | 0.002 | 0.000 | 0.000 | 0.001 | 0.002 | 0.001 | 0.000 | 0.006 | 0.003 | 0.000 | 0.015 |
| Ca | 1.925 | 1.918 | 1.854 | 1.906 | 1.858 | 1.918 | 1.903 | 1.933 | 1.934 | 1.924 | 1.925 |
| Na | 0.001 | 0.002 | 0.028 | 0.014 | 0.048 | 0.001 | 0.000 | 0.002 | 0.000 | 0.003 | 0.001 |
| K | 0.001 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.001 | 0.001 | 0.000 | 0.001 | 0.000 |
| Total | 7.982 | 7.984 | 7.957 | 7.976 | 7.977 | 7.976 | 7.978 | 7.995 | 7.982 | 7.987 | 7.988 |
| $\mathrm{Fe}^{3+} /\left(\mathrm{Fe}^{3+}+\mathrm{Al}\right)$ | 0.16 | 0.30 | 0.29 | 0.30 | 0.17 | 0.36 | 0.36 | 0.24 | 0.24 | 0.30 | 0.29 |

Table A. 8 (continued) : Mineral chemistry of the measured epidotes. Mineral formula calculations are based on 12.5 oxygens.

| Epidote | 775 | 775 | 775 | 775 | 775 | 775 | 753A | 753A | 753A |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 12 | 25 | 236 | 242 | 253 | 254 | 37 | 41 | 42 |
| $\mathrm{SiO}_{2}$ | 38.95 | 37.93 | 39.30 | 39.35 | 38.07 | 39.28 | 38.42 | 37.90 | 38.40 |
| $\mathrm{TiO}_{2}$ | 0.35 | 0.14 | 0.04 | 0.28 | 1.76 | 0.06 | 0.10 | 0.13 | 0.08 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 30.38 | 28.09 | 31.42 | 30.75 | 30.39 | 31.22 | 28.87 | 27.31 | 28.34 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.02 | 0.01 | 0.00 | 0.00 | 0.05 | 0.00 | 0.03 | 0.00 | 0.00 |
| FeO | 5.49 | 7.55 | 3.41 | 4.14 | 3.92 | 3.45 | 6.67 | 8.46 | 7.38 |
| MnO | 0.41 | 0.53 | 0.34 | 0.43 | 0.55 | 0.62 | 0.58 | 0.66 | 0.47 |
| MgO | 0.01 | 0.01 | 0.00 | 0.01 | 0.00 | 0.01 | 0.01 | 0.00 | 0.03 |
| CaO | 22.12 | 22.30 | 23.39 | 23.45 | 22.78 | 23.00 | 22.50 | 22.04 | 22.73 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.13 | 0.01 | 0.01 | 0.09 | 0.00 | 0.00 | 0.00 | 0.01 | 0.01 |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.22 | 0.09 | 0.10 | 0.07 | 0.04 | 0.03 | 0.02 | 0.05 | 0.03 |
| Total | 98.09 | 96.65 | 98.01 | 98.58 | 97.54 | 97.66 | 97.19 | 96.56 | 97.48 |
| Si | 2.980 | 2.970 | 2.998 | 2.993 | 2.931 | 3.006 | 2.982 | 2.978 | 2.979 |
| Ti | 0.020 | 0.008 | 0.002 | 0.016 | 0.102 | 0.003 | 0.006 | 0.008 | 0.005 |
| Al | 2.739 | 2.592 | 2.824 | 2.757 | 2.757 | 2.816 | 2.641 | 2.529 | 2.591 |
| Cr | 0.001 | 0.001 | 0.000 | 0.000 | 0.003 | 0.000 | 0.002 | 0.000 | 0.000 |
| $\mathrm{Fe}^{3+}$ | 0.351 | 0.494 | 0.217 | 0.264 | 0.252 | 0.221 | 0.433 | 0.556 | 0.479 |
| $\mathrm{Fe}^{2+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |  |  |  |
| Mn | 0.027 | 0.035 | 0.022 | 0.028 | 0.036 | 0.040 | 0.038 | 0.044 | 0.031 |
| Mg | 0.001 | 0.001 | 0.000 | 0.001 | 0.000 | 0.001 | 0.001 | 0.000 | 0.004 |
| Ca | 1.813 | 1.871 | 1.911 | 1.911 | 1.879 | 1.885 | 1.871 | 1.855 | 1.890 |
| Na | 0.019 | 0.002 | 0.002 | 0.014 | 0.000 | 0.000 | 0.000 | 0.001 | 0.001 |
| K | 0.021 | 0.009 | 0.010 | 0.007 | 0.004 | 0.003 | 0.002 | 0.005 | 0.003 |
| Total | 7.974 | 7.984 | 7.986 | 7.991 | 7.963 | 7.974 | 7.975 | 7.975 | 7.983 |
| $\mathrm{Fe}^{3+} /\left(\mathrm{Fe}^{3+}+\mathrm{Al}\right)$ | 0.11 | 0.16 | 0.07 | 0.09 | 0.08 | 0.07 | 0.14 | 0.18 | 0.16 |

Table A.9 : Mineral chemistry of the measured chlorites. Mineral formula calculations are based on 14 oxygens.

Table A. 9 (continued) : Mineral chemistry of the measured chlorites. Mineral formula calculations are based on 14 oxygens.

| Chlorite | 211B | 211B | 211B | 211B | 211B | 211B | 211B | 211B | 128 | 128 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 242 | 245 | 267 | 268 | 276 | 281 | 292 | 315 | 139 | 153 |
| $\mathrm{SiO}_{2}$ | 25.83 | 27.22 | 29.92 | 26.67 | 26.02 | 27.11 | 27.08 | 27.05 | 28.19 | 29.08 |
| $\mathrm{TiO}_{2}$ | 0.05 | 0.00 | 0.04 | 0.05 | 0.48 | 0.01 | 0.01 | 0.00 | 0.01 | 0.03 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 19.06 | 18.15 | 16.13 | 19.39 | 20.27 | 17.62 | 17.42 | 18.87 | 18.87 | 19.67 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.01 | 0.00 | 0.02 | 0.02 | 0.01 |
| FeO | 28.14 | 25.89 | 23.00 | 26.35 | 27.01 | 27.11 | 27.77 | 26.41 | 23.94 | 23.51 |
| MnO | 0.50 | 0.35 | 0.29 | 0.33 | 0.43 | 0.43 | 0.38 | 0.36 | 0.36 | 0.36 |
| MgO | 13.88 | 16.00 | 17.24 | 15.26 | 14.40 | 15.93 | 15.64 | 15.14 | 15.47 | 15.13 |
| CaO | 0.10 | 0.06 | 0.23 | 0.13 | 0.36 | 0.08 | 0.10 | 0.05 | 0.31 | 0.12 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.03 | 0.03 | 0.03 | 0.03 | 0.04 | 0.04 | 0.02 | 0.05 | 0.16 | 0.16 |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.01 | 0.02 | 0.48 | 0.02 | 0.03 | 0.02 | 0.02 | 0.07 | 0.07 | 0.17 |
| Total | 87.60 | 87.73 | 87.35 | 88.22 | 89.04 | 88.35 | 88.44 | 88.02 | 87.40 | 88.24 |
| Si | 2.777 | 2.877 | 3.122 | 2.809 | 2.730 | 2.868 | 2.873 | 2.856 | 2.952 | 2.995 |
| Ti | 0.004 | 0.000 | 0.003 | 0.004 | 0.038 | 0.001 | 0.001 | 0.000 | 0.001 | 0.002 |
| Al | 2.415 | 2.261 | 1.984 | 2.406 | 2.506 | 2.198 | 2.178 | 2.348 | 2.329 | 2.389 |
| Cr | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.001 | 0.000 | 0.002 | 0.001 | 0.001 |
| $\mathrm{Fe}^{3+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{2+}$ | 2.530 | 2.289 | 2.007 | 2.320 | 2.370 | 2.399 | 2.464 | 2.332 | 2.097 | 2.026 |
| Mn | 0.046 | 0.031 | 0.025 | 0.029 | 0.038 | 0.038 | 0.034 | 0.033 | 0.032 | 0.031 |
| Mg | 2.224 | 2.522 | 2.681 | 2.396 | 2.251 | 2.513 | 2.473 | 2.384 | 2.415 | 2.323 |
| Ca | 0.012 | 0.006 | 0.025 | 0.015 | 0.040 | 0.009 | 0.012 | 0.006 | 0.034 | 0.013 |
| Na | 0.007 | 0.007 | 0.006 | 0.006 | 0.007 | 0.007 | 0.003 | 0.009 | 0.033 | 0.033 |
| K | 0.001 | 0.003 | 0.064 | 0.003 | 0.004 | 0.002 | 0.003 | 0.009 | 0.009 | 0.023 |
| Total | 10.015 | 9.997 | 9.918 | 9.988 | 9.985 | 10.036 | 10.041 | 9.979 | 9.903 | 9.836 |
| $\mathrm{Mg} /\left(\mathrm{Mg}+\mathrm{Fe}^{2+}\right)$ | 0.47 | 0.52 | 0.57 | 0.51 | 0.49 | 0.51 | 0.50 | 0.51 | 0.54 | 0.53 |

Table A. 9 (continued) : Mineral chemistry of the measured chlorites. Mineral formula calculations are based on 14 oxygens.

Table A. 9 (continued) : Mineral chemistry of the measured chlorites. Mineral formula calculations are based on 14 oxygens.

Table A. 9 (continued) : Mineral chemistry of the measured chlorites. Mineral formula calculations are based on 14 oxygens.

Table A. 9 (continued) : Mineral chemistry of the measured chlorites. Mineral formula calculations are based on 14 oxygens.

| Chlorite | 392 | 392 | 392 | 392 | 392 | 392 | 392 | 392 | 392 | 392 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 12 | 16 | 17 | 19 | 22 | 28 | 30 | 31 | 42 | 45 |
| $\mathrm{SiO}_{2}$ | 25.51 | 25.76 | 25.83 | 25.44 | 25.73 | 26.01 | 25.46 | 25.62 | 25.97 | 25.81 |
| $\mathrm{TiO}_{2}$ | 0.03 | 0.02 | 0.03 | 0.02 | 0.03 | 0.04 | 0.05 | 0.03 | 0.05 | 0.07 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 22.46 | 22.68 | 22.99 | 22.92 | 22.63 | 22.71 | 22.40 | 22.81 | 22.05 | 20.27 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ |  |  |  |  |  |  |  |  |  |  |
| FeO | 30.50 | 30.54 | 29.51 | 31.64 | 29.97 | 29.96 | 31.29 | 29.74 | 29.53 | 31.77 |
| MnO | 0.18 | 0.16 | 0.15 | 0.17 | 0.19 | 0.14 | 0.19 | 0.20 | 0.33 | 0.23 |
| MgO | 9.84 | 9.75 | 10.31 | 10.06 | 9.47 | 9.68 | 9.45 | 9.50 | 10.29 | 10.01 |
| CaO | 0.03 | 0.01 | 0.03 | 0.03 | 0.05 | 0.05 | 0.01 | 0.02 | 0.02 | 0.04 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.01 | 0.02 | 0.00 | 0.01 | 0.07 | 0.04 | 0.01 | 0.05 | 0.00 | 0.04 |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.03 | 0.01 | 0.06 | 0.07 | 0.06 | 0.13 | 0.04 | 0.04 | 0.02 | 0.04 |
| Total | 88.58 | 88.95 | 88.93 | 90.38 | 88.21 | 88.76 | 88.90 | 88.02 | 88.26 | 88.29 |
| Si | 2.723 | 2.734 | 2.727 | 2.675 | 2.750 | 2.759 | 2.720 | 2.740 | 2.768 | 2.793 |
| Ti | 0.002 | 0.002 | 0.002 | 0.002 | 0.003 | 0.003 | 0.004 | 0.002 | 0.004 | 0.006 |
| Al | 2.826 | 2.837 | 2.861 | 2.841 | 2.850 | 2.839 | 2.820 | 2.875 | 2.770 | 2.586 |
| Cr | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{3+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{2+}$ | 2.723 | 2.711 | 2.606 | 2.783 | 2.678 | 2.657 | 2.795 | 2.660 | 2.632 | 2.876 |
| Mn | 0.016 | 0.014 | 0.013 | 0.015 | 0.017 | 0.013 | 0.017 | 0.019 | 0.030 | 0.021 |
| Mg | 1.566 | 1.543 | 1.623 | 1.577 | 1.509 | 1.531 | 1.505 | 1.515 | 1.635 | 1.615 |
| Ca | 0.003 | 0.001 | 0.004 | 0.003 | 0.005 | 0.006 | 0.001 | 0.003 | 0.002 | 0.005 |
| Na | 0.002 | 0.003 | 0.000 | 0.002 | 0.015 | 0.007 | 0.002 | 0.009 | 0.000 | 0.008 |
| K | 0.004 | 0.001 | 0.008 | 0.010 | 0.008 | 0.018 | 0.005 | 0.005 | 0.003 | 0.005 |
| Total | 9.865 | 9.847 | 9.844 | 9.908 | 9.834 | 9.832 | 9.869 | 9.828 | 9.844 | 9.914 |
| $\mathrm{Mg} /\left(\mathrm{Mg}+\mathrm{Fe}^{2+}\right)$ | 0.37 | 0.36 | 0.38 | 0.36 | 0.36 | 0.37 | 0.35 | 0.36 | 0.38 | 0.36 |

Table A.10 : Mineral chemistry of the measured plagioclases. Mineral formula calculations are based on eight oxygens.

| Plagioclase | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 702A | 128 | 128 | 128 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 48 | 57 | 67 | 76 | 98 | 100 | 103 | 127 | 154 | 187 | 190 |
| $\mathrm{SiO}_{2}$ | 68.47 | 68.55 | 68.76 | 68.91 | 68.59 | 67.84 | 67.54 | 67.56 | 68.35 | 69.21 | 69.06 |
| $\mathrm{TiO}_{2}$ | 0.00 | 0.00 | 0.01 | 0.02 | 0.02 | 0.02 | 0.03 | 0.03 | 0.02 | 0.00 | 0.00 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 19.71 | 19.48 | 19.39 | 19.44 | 19.52 | 19.38 | 20.14 | 18.10 | 20.42 | 19.46 | 19.47 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.01 | 0.00 | 0.00 |
| FeO | 0.51 | 0.17 | 0.31 | 0.13 | 0.18 | 0.18 | 0.26 | 0.98 | 0.22 | 0.15 | 0.13 |
| MnO | 0.00 | 0.01 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.02 | 0.00 |
| MgO | 0.11 | 0.00 | 0.00 | 0.00 | 0.00 | 0.01 | 0.02 | 1.11 | 0.00 | 0.01 | 0.00 |
| CaO | 0.45 | 0.13 | 0.13 | 0.03 | 0.18 | 0.27 | 0.95 | 1.27 | 1.08 | 0.14 | 0.08 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 11.41 | 11.83 | 11.70 | 12.04 | 11.69 | 11.36 | 11.12 | 11.32 | 11.15 | 11.86 | 11.98 |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.04 | 0.02 | 0.02 | 0.02 | 0.06 | 0.04 | 0.05 | 0.05 | 0.04 | 0.03 | 0.02 |
| Total | 100.70 | 100.19 | 100.32 | 100.60 | 100.24 | 99.11 | 100.10 | 100.42 | 101.29 | 100.89 | 100.73 |
| Si | 2.978 | 2.992 | 2.997 | 2.995 | 2.992 | 2.991 | 2.956 | 2.970 | 2.955 | 2.999 | 2.997 |
| Ti | 0.000 | 0.000 | 0.000 | 0.001 | 0.001 | 0.001 | 0.001 | 0.001 | 0.001 | 0.000 | 0.000 |
| Al | 1.010 | 1.002 | 0.996 | 0.996 | 1.003 | 1.007 | 1.039 | 0.938 | 1.041 | 0.993 | 0.996 |
| Cr | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{3+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{2+}$ | 0.019 | 0.006 | 0.011 | 0.005 | 0.006 | 0.007 | 0.009 | 0.036 | 0.008 | 0.005 | 0.005 |
| Mn | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.001 | 0.000 |
| Mg | 0.007 | 0.000 | 0.000 | 0.000 | 0.000 | 0.001 | 0.001 | 0.073 | 0.000 | 0.001 | 0.000 |
| Ca | 0.021 | 0.006 | 0.006 | 0.002 | 0.008 | 0.013 | 0.044 | 0.060 | 0.050 | 0.007 | 0.004 |
| Na | 0.962 | 1.001 | 0.989 | 1.014 | 0.989 | 0.971 | 0.944 | 0.965 | 0.935 | 0.996 | 1.008 |
| K | 0.002 | 0.001 | 0.001 | 0.001 | 0.003 | 0.002 | 0.003 | 0.003 | 0.002 | 0.002 | 0.001 |
| Total | 4.999 | 5.008 | 5.000 | 5.014 | 5.002 | 4.992 | 4.997 | 5.045 | 4.992 | 5.004 | 5.010 |
| $\mathrm{X}_{\mathrm{An}}(\mathrm{Ca} /(\mathrm{Ca}+\mathrm{Na})$ | 0.02 | 0.01 | 0.01 | 0.00 | 0.01 | 0.01 | 0.04 | 0.06 | 0.05 | 0.01 | 0.00 |

Table A. 10 (continued) : Mineral chemistry of the measured plagioclases. Mineral formula calculations are based on eight oxygens.

Table A. 10 (continued) : Mineral chemistry of the measured plagioclases. Mineral formula calculations are based on eight oxygens.

| Plagioclase | 178 | 178 | 178 | 178 | 178 | 178 | 392 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 53 | 81 | 82 | 91 | 92 | 101 | 1 |
| $\mathrm{SiO}_{2}$ | 68.49 | 68.30 | 68.07 | 67.02 | 66.19 | 66.68 | 67.41 |
| $\mathrm{TiO}_{2}$ | 0.00 | 0.00 | 0.01 | 0.24 | 0.04 | 0.07 | 0.00 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 20.13 | 19.14 | 19.53 | 19.75 | 19.53 | 20.63 | 20.56 |
| $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | 0.00 |  |  | 0.00 | 0.00 | 0.00 |  |
| FeO | 0.21 | 0.05 | 0.14 | 0.10 | 0.04 | 0.31 | 0.02 |
| MnO | 0.00 | 0.00 | 0.00 | 0.01 | 0.00 | 0.00 | 0.00 |
| MgO | 0.00 | 0.00 | 0.07 | 0.00 | 0.00 | 0.12 | 0.00 |
| CaO | 0.05 | 0.07 | 0.08 | 0.10 | 0.09 | 0.05 | 0.06 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 10.47 | 11.66 | 11.42 | 11.61 | 11.30 | 10.91 | 11.96 |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.05 | 0.04 | 0.36 | 0.04 | 0.06 | 0.81 | 0.04 |
| Total | 99.39 | 99.27 | 99.68 | 98.87 | 97.25 | 99.57 | 100.05 |
| Si | 2.995 | 3.004 | 2.987 | 2.965 | 2.973 | 2.939 | 2.949 |
| Ti | 0.000 | 0.000 | 0.000 | 0.008 | 0.001 | 0.002 | 0.000 |
| Al | 1.037 | 0.992 | 1.010 | 1.030 | 1.034 | 1.071 | 1.060 |
| Cr | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{3+}$ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| $\mathrm{Fe}^{2+}$ | 0.008 | 0.002 | 0.005 | 0.004 | 0.001 | 0.011 | 0.001 |
| Mn | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| Mg | 0.000 | 0.000 | 0.004 | 0.000 | 0.000 | 0.008 | 0.000 |
| Ca | 0.002 | 0.003 | 0.004 | 0.005 | 0.004 | 0.002 | 0.003 |
| Na | 0.887 | 0.994 | 0.972 | 0.996 | 0.984 | 0.932 | 1.014 |
| K | 0.003 | 0.002 | 0.020 | 0.002 | 0.003 | 0.045 | 0.002 |
| Total | 4.932 | 4.998 | 5.003 | 5.011 | 5.002 | 5.012 | 5.029 |
| $\mathrm{X}_{\mathrm{An}}(\mathrm{Ca} /(\mathrm{Ca}+\mathrm{Na})$ | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 | 0.00 |

Table A.11 : Bulk rock compositions and calculated element mol percentages used in the pseudosection calculations.

|  | 128 | 702 A | 775 | 753 A | 178 |
| :--- | :---: | :---: | :---: | :---: | :---: |
| Oxide (w.t. \%) |  |  |  |  |  |
| $\mathrm{SiO}_{2}$ | 50.21 | 46.39 | 62.71 | 56.11 | 62.07 |
| $\mathrm{TiO}_{2}$ | 1.56 | 2.01 | 0.70 | 0.70 | 0.81 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 14.91 | 14.25 | 17.41 | 19.30 | 18.20 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3}$ | 11.99 | 13.71 | 6.95 | 10.04 | 6.92 |
| MnO | 0.17 | 0.23 | 0.45 | 0.49 | 0.14 |
| MgO | 5.28 | 6.51 | 1.98 | 2.77 | 1.52 |
| CaO | 6.09 | 8.98 | 0.34 | 0.47 | 0.22 |
| $\mathrm{Na} \mathrm{O}_{2} \mathrm{O}$ | 4.40 | 2.85 | 0.66 | 0.79 | 1.06 |
| $\mathrm{~K}_{2} \mathrm{O}$ | 0.66 | 0.57 | 3.61 | 3.36 | 3.45 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | 0.16 | 0.24 | 0.12 | 0.31 | 0.12 |
| LOI | 4.38 | 3.95 | 4.89 | 5.36 | 5.21 |
| Total | 99.80 | 99.70 | 99.82 | 99.70 | 99.72 |
|  |  |  |  |  |  |
| Element (mol\%) |  |  |  |  |  |
| Si | 22.23 | 22.28 | 28.96 | 25.67 | 28.90 |
| Al | 7.89 | 7.42 | 8.98 | 9.33 | 9.41 |
| Ti |  | 1.21 |  |  |  |
| Fe | 8.39 | 9.59 | 4.86 | 6.40 | 2.54 |
| Mg | 3.18 | 3.93 | 1.19 | 1.67 | 0.92 |
| Ca | 3.18 | 6.52 | 0.16 |  |  |
| Na | 3.26 | 2.11 | 0.49 | 0.59 | 0.79 |
| K | 0.55 | 0.47 | 3.00 | 2.79 | 2.86 |
| $\mathrm{H} \mathrm{H}_{2} \mathrm{O}$ | Saturated | Saturated | Saturated | Saturated | Saturated |

Table A.12 : UTM coordinates (European 1979 datum) of the metapelitic samples of the Esenler Unit that were analyzed by Raman Spectroscopy.

| Sample | x | y |
| :---: | :---: | :---: |
| 31 | 36T 05 86007 | 4561629 |
| 230 | 36T 05 88842 | 4574615 |
| 850 | 36T 05 89035 | 4573738 |
| 844 | 36T 05 87373 | 4572916 |
| 879 | 36T 05 88350 | 4563831 |
| 871 | 36T 05 87976 | 4568095 |
| 876 | 36T 05 85922 | 4565761 |
| 874 | 36T 05 85489 | 4567304 |
| 841 | 36T 05 84538 | 4569821 |
| 847 | 36T 05 88172 | 4571385 |
| 851 | 36T 05 90262 | 4574020 |
| 853 | 36T 05 86097 | 4576405 |
| 1126 | 36T 05 83817 | 4577178 |
| 859 | 36T 05 83350 | 4577637 |
| 861 | 36T 05 82426 | 4579377 |

Table A. 13 : ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ analytical data of the dated samples.

| Laser output | ${ }^{40} \mathrm{Ar}{ }^{\text {/ }} \mathrm{Ar}$ |  |  | ${ }^{37} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ |  |  | $\begin{gathered} { }^{36} \mathrm{Ar}^{39} \mathrm{Ar} \\ \left(\mathrm{x} 10^{-3}\right) \end{gathered}$ |  |  | K/Ca | $\begin{gathered} \hline{ }^{40} \mathrm{Ar}^{*} \\ (\%) \end{gathered}$ | ${ }^{39} \mathrm{ArK}$ <br> fraction (\%) | ${ }^{40} \mathrm{Ar} *{ }^{39} \mathrm{ArK}$ |  |  | Age(Ma) |  | $\pm 1 \sigma$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Sample 31, Laboratory ID: C13038 |  |  |  | Neutron Irradiation ID: PP-3 |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| $J=0.00241$ |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| 1.2\% | 92.61 | $\pm$ | 4.00 | 0.028 | $\pm$ | 0.001 | 254.25 | $\pm$ | 75.12 | 21.39 | 18.88 | 2.21 | 17.49 | $\pm$ | 22.14 | 74.47 | $\pm$ | 92.37 |
| 1.4\% | 29.14 | $\pm$ | 1.29 | 0.047 | $\pm$ | 0.002 | 28.19 | $\pm$ | 20.15 | 12.65 | 71.42 | 13.10 | 20.81 | $\pm$ | 6.03 | 88.28 | $\pm$ | 24.96 |
| 1.6\% | 30.56 | $\pm$ | 0.28 | 0.011 | $\pm$ | 0.000 | 19.54 | $\pm$ | 4.36 | 52.33 | 81.11 | 54.28 | 24.79 | $\pm$ | 1.31 | 104.69 | $\pm$ | 5.40 |
| 1.8\% | 33.58 | $\pm$ | 1.09 | 0.025 | $\pm$ | 0.001 | 42.89 | $\pm$ | 9.60 | 23.98 | 62.27 | 24.89 | 20.91 | $\pm$ | 2.90 | 88.71 | $\pm$ | 11.99 |
| 2.0\% | 37.26 | $\pm$ | 3.17 | 0.137 | $\pm$ | 0.011 | 29.16 | $\pm$ | 41.44 | 4.28 | 76.90 | 4.45 | 28.65 | $\pm$ | 12.48 | 120.46 | $\pm$ | 50.78 |
| 2.2\% | 49.24 | $\pm$ | 8.50 | 0.057 | $\pm$ | 0.009 | 615.16 | $\pm$ | 240.11 | 10.31 | -269.15 | 1.07 | -132.54 | $\pm$ | -68.38 | -694.22 | $\pm$ | -436.83 |
|  |  |  |  |  |  |  |  |  | Plateau age (Plateau: steps from $1.4 \%$ to 1.8\%) |  |  |  |  |  |  | 101.50 | $\pm$ | 4.80 |
|  |  |  |  |  |  |  |  |  | Total | age : |  |  |  |  |  | 91.70 | $\pm$ | 6.90 |

Table A. 13 (continued) : ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ analytical data of the dated samples.

| Laser output | ${ }^{40} \mathrm{Ar}{ }^{3 / 9} \mathrm{Ar}$ |  |  | ${ }^{37} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ |  |  | $\begin{aligned} & { }^{36} \mathrm{Ar}^{39} \mathrm{Ar} \\ & \left(\times 10^{-3}\right) \end{aligned}$ |  |  | K/Ca | $\begin{aligned} & { }^{40} \mathrm{Ar}^{*} \\ & (\%) \\ & \hline \end{aligned}$ | $\begin{aligned} & { }^{39} \mathrm{ArK} \quad{ }^{40} \mathrm{Ar} * \beta^{39} \mathrm{ArK} \\ & \text { fraction (\%) } \end{aligned}$ |  |  |  | Age(Ma) | $\pm 1 \sigma$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Sample 230, Laboratory ID: C13039 |  |  |  | Neutron Irradiation ID: PP-3 |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| $J=0.002402$ |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| 1.2\% | 1078.74 | $\pm$ | 37.86 | 0.017 | $\pm$ | 0.001 | 3916.51 | $\pm$ | 193.88 | 34.14 | -7.28 | 0.56 | -78.59 | $\pm$ | -40.65 | -377.41 | $\pm$ | -217.15 |
| 1.4\% | 66.18 | $\pm$ | 2.03 | 0.020 | $\pm$ | 0.001 | 174.16 | $\pm$ | 11.23 | 29.40 | 22.24 | 4.84 | 14.72 | $\pm$ | 2.97 | 62.68 | $\pm$ | 12.43 |
| 1.5\% | 40.77 | $\pm$ | 0.65 | 0.016 | $\pm$ | 0.000 | 59.30 | $\pm$ | 7.44 | 37.39 | 57.02 | 6.15 | 23.25 | $\pm$ | 2.22 | 98.04 | $\pm$ | 9.11 |
| 1.6\% | 31.50 | $\pm$ | 0.71 | 0.016 | $\pm$ | 0.000 | 43.34 | $\pm$ | 6.32 | 37.70 | 59.35 | 6.21 | 18.70 | $\pm$ | 1.91 | 79.26 | $\pm$ | 7.91 |
| 1.7\% | 28.63 | $\pm$ | 0.50 | 0.010 | $\pm$ | 0.000 | 21.85 | $\pm$ | 4.40 | 61.52 | 77.44 | 10.14 | 22.17 | $\pm$ | 1.35 | 93.60 | $\pm$ | 5.58 |
| 1.8\% | 27.43 | $\pm$ | 0.35 | 0.008 | $\pm$ | 0.000 | 14.23 | $\pm$ | 2.88 | 77.61 | 84.67 | 12.81 | 23.22 | $\pm$ | 0.90 | 97.92 | $\pm$ | 3.72 |
| 1.9\% | 27.92 | $\pm$ | 0.36 | 0.006 | $\pm$ | 0.000 | 14.99 | $\pm$ | 2.17 | 95.37 | 84.13 | 15.75 | 23.49 | $\pm$ | 0.72 | 99.01 | $\pm$ | 2.96 |
| 2.0\% | 27.52 | $\pm$ | 0.58 | 0.009 | $\pm$ | 0.000 | 7.31 | $\pm$ | 4.19 | 63.75 | 92.15 | 10.54 | 25.36 | $\pm$ | 1.35 | 106.67 | $\pm$ | 5.53 |
| 2.1\% | 27.51 | $\pm$ | 0.51 | 0.007 | $\pm$ | 0.000 | 10.22 | $\pm$ | 2.93 | 84.37 | 89.02 | 13.96 | 24.49 | $\pm$ | 0.98 | 103.11 | $\pm$ | 4.01 |
| 2.2\% | 26.66 | $\pm$ | 0.41 | 0.018 | $\pm$ | 0.000 | 10.44 | $\pm$ | 8.41 | 31.90 | 88.43 | 5.28 | 23.58 | $\pm$ | 2.51 | 99.37 | $\pm$ | 10.31 |
| 2.4\% | 26.14 | $\pm$ | 0.65 | 0.010 | $\pm$ | 0.000 | 12.59 | $\pm$ | 5.15 | 56.33 | 85.77 | 9.33 | 22.42 | $\pm$ | 1.62 | 94.63 | $\pm$ | 6.67 |
| 2.8\% | 33.14 | $\pm$ | 0.70 | 0.022 | $\pm$ | 0.000 | 29.28 | $\pm$ | 9.04 | 26.70 | 73.89 | 4.43 | 24.49 | $\pm$ | 2.72 | 103.10 | $\pm$ | 11.13 |
|  |  |  |  |  |  |  |  |  |  | Plateau age (Plateau: steps from $1.7 \%$ to $2.8 \%$ ) Total gas age : |  |  |  |  |  | 99.60 | $\pm$ | 1.70 |
|  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  | 94.1 | $\pm$ | 2.1 |

Table A. 13 (continued) : ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ analytical data of the dated samples.

| Laser output | ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ |  |  | ${ }^{37} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ |  |  | $\begin{aligned} & { }^{36} \mathrm{Ar} \mathrm{r}^{39} \mathrm{Ar} \\ & \left(\mathrm{x} 10^{-3}\right) \end{aligned}$ |  |  | K/Ca | $\begin{aligned} & \hline{ }^{40} \mathrm{Ar}^{*} \\ & (\%) \end{aligned}$ | $\begin{aligned} & \hline{ }^{39} \mathrm{ArK} \quad{ }^{40} \mathrm{Ar}^{* \beta 9} \mathrm{ArK} \\ & \text { fraction (\%) } \end{aligned}$ |  |  |  | Age(Ma) | $\pm 1 \sigma$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Sample 392, Laboratory ID: C13040 |  |  |  | Neutron Irradiation ID: PP-3 |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| $J=0.002455$ |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| 1.2\% | 555.83 | $\pm$ | 39.49 | 0.124 | $\pm$ | 0.009 | 2082.87 | $\pm$ | 162.84 | 4.74 | -10.73 | 0.58 | -59.65 | $\pm$ | -20.75 | -285.66 | $\pm$ | -107.69 |
| 1.4\% | 60.10 | $\pm$ | 1.82 | 0.022 | $\pm$ | 0.001 | 135.05 | $\pm$ | 13.21 | 27.33 | 33.60 | 3.32 | 20.19 | $\pm$ | 3.80 | 87.29 | $\pm$ | 16.02 |
| 1.6\% | 31.72 | $\pm$ | 0.32 | 0.039 | $\pm$ | 0.000 | 23.97 | $\pm$ | 1.85 | 15.00 | 77.68 | 18.24 | 24.64 | $\pm$ | 0.60 | 105.98 | $\pm$ | 2.54 |
| 1.7\% | 27.89 | $\pm$ | 0.38 | 0.041 | $\pm$ | 0.001 | 16.21 | $\pm$ | 1.23 | 14.31 | 82.83 | 17.41 | 23.10 | $\pm$ | 0.48 | 99.51 | $\pm$ | 2.04 |
| 1.8\% | 25.03 | $\pm$ | 0.21 | 0.030 | $\pm$ | 0.000 | 7.72 | $\pm$ | 1.49 | 19.74 | 90.90 | 24.03 | 22.75 | $\pm$ | 0.48 | 98.06 | $\pm$ | 2.06 |
| 1.9\% | 23.54 | $\pm$ | 0.18 | 0.042 | $\pm$ | 0.000 | 7.47 | $\pm$ | 1.56 | 13.87 | 90.63 | 16.91 | 21.33 | $\pm$ | 0.49 | 92.09 | $\pm$ | 2.09 |
| 2.0\% | 22.63 | $\pm$ | 0.32 | 0.082 | $\pm$ | 0.001 | 4.02 | $\pm$ | 2.80 | 7.21 | 94.77 | 8.79 | 21.45 | $\pm$ | 0.88 | 92.59 | $\pm$ | 3.72 |
| 2.2\% | 22.41 | $\pm$ | 0.28 | 0.010 | $\pm$ | 0.000 | 9.23 | $\pm$ | 3.82 | 60.00 | 87.83 | 7.33 | 19.68 | $\pm$ | 1.16 | 85.12 | $\pm$ | 4.89 |
| 2.4\% | 22.87 | $\pm$ | 0.53 | 0.021 | $\pm$ | 0.000 | 9.98 | $\pm$ | 8.58 | 27.75 | 87.11 | 3.39 | 19.92 | $\pm$ | 2.58 | 86.16 | $\pm$ | 10.89 |
|  |  |  |  |  |  |  |  |  |  | Plateau age (Plateau: steps from $1.6 \%$ to 1.8\%) |  |  |  |  |  | 100.60 | $\pm$ | 1.30 |
|  |  |  |  |  |  |  |  |  |  | Total gas age : |  |  |  |  |  | 94.60 | $\pm$ | 1.30 |

Table $\mathbf{A . ~} 13$ (continued) : ${ }^{40} \mathrm{Ar}{ }^{\beta 9} \mathrm{Ar}$ analytical data of the dated samples.

| Laser output | ${ }^{40} \mathrm{Ar}{ }^{3 / 3} \mathrm{Ar}$ |  |  | ${ }^{37} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ |  |  | $\begin{aligned} & { }^{36} \mathrm{Ar} \mathrm{r}^{39} \mathrm{Ar} \\ & \left(\times 10^{-3}\right) \end{aligned}$ |  |  | K/Ca | $\begin{aligned} & { }^{40} \mathrm{Ar}^{*} \\ & (\%) \end{aligned}$ | $\begin{aligned} & { }^{39} \mathrm{ArK} \quad{ }^{40} \mathrm{Ar} *{ }^{* 9} \mathrm{ArK} \\ & \text { fraction (\%) } \end{aligned}$ |  |  |  | Age(Ma) | $\pm 1 \sigma$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Sample 753A, Laboratory ID: Cl3054 |  |  |  | Neutron Irradiation ID: PP-3 |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| $J=0.002459$ |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| 1.6\% | 92.30 | $\pm$ | 40.92 | 0.280 | $\pm$ | 0.122 | 89.53 | $\pm$ | 1489.41 | 2.10 | 71.36 | 0.03 | 65.88 | $\pm$ | 441.06 | 270.85 | $\pm$ | 1683.88 |
| 1.8\% | 23.88 | $\pm$ | 1.42 | 0.002 | $\pm$ | 0.000 | 5.26 | $\pm$ | 12.91 | 235.63 | 93.49 | 2.93 | 22.32 | $\pm$ | 4.05 | 96.40 | $\pm$ | 17.02 |
| 2.0\% | 20.05 | $\pm$ | 0.59 | 0.009 | $\pm$ | 0.000 | 0.29 | $\pm$ | 5.25 | 64.34 | 99.57 | 8.02 | 19.97 | $\pm$ | 1.66 | 86.47 | $\pm$ | 7.03 |
| 2.2\% | 20.67 | $\pm$ | 0.96 | 0.012 | $\pm$ | 0.001 | 0.87 | $\pm$ | 13.14 | 50.88 | 98.76 | 6.34 | 20.41 | $\pm$ | 4.00 | 88.35 | $\pm$ | 16.89 |
| 2.4\% | 20.85 | $\pm$ | 0.59 | 0.006 | $\pm$ | 0.000 | 0.19 | $\pm$ | 3.06 | 98.56 | 99.74 | 12.47 | 20.80 | $\pm$ | 1.08 | 89.98 | $\pm$ | 4.56 |
| 2.6\% | 22.36 | $\pm$ | 1.09 | 0.079 | $\pm$ | 0.004 | 0.25 | $\pm$ | 3.50 | 7.43 | 99.70 | 9.41 | 22.29 | $\pm$ | 1.50 | 96.28 | $\pm$ | 6.31 |
| 2.8\% | 20.85 | $\pm$ | 0.60 | 0.088 | $\pm$ | 0.002 | 0.28 | $\pm$ | 3.75 | 6.71 | 99.64 | 8.49 | 20.78 | $\pm$ | 1.26 | 89.89 | $\pm$ | 5.32 |
| 3.0\% | 21.30 | $\pm$ | 0.62 | 0.064 | $\pm$ | 0.002 | 0.20 | $\pm$ | 3.15 | 9.24 | 99.75 | 11.70 | 21.24 | $\pm$ | 1.12 | 91.86 | $\pm$ | 4.74 |
| 3.2\% | 21.02 | $\pm$ | 0.79 | 0.055 | $\pm$ | 0.002 | 0.17 | $\pm$ | 1.81 | 10.76 | 99.78 | 13.64 | 20.97 | $\pm$ | 0.96 | 90.73 | $\pm$ | 4.05 |
| 3.4\% | 22.71 | $\pm$ | 1.27 | 0.010 | $\pm$ | 0.001 | 0.32 | $\pm$ | 3.65 | 58.05 | 99.59 | 7.36 | 22.62 | $\pm$ | 1.66 | 97.65 | $\pm$ | 6.99 |
| 3.6\% | 21.85 | $\pm$ | 0.75 | 0.010 | $\pm$ | 0.000 | 0.31 | $\pm$ | 4.29 | 59.46 | 99.58 | 7.55 | 21.76 | $\pm$ | 1.47 | 94.04 | $\pm$ | 6.20 |
| 3.8\% | 22.07 | $\pm$ | 0.98 | 0.014 | $\pm$ | 0.001 | 0.45 | $\pm$ | 4.90 | 41.05 | 99.40 | 5.21 | 21.94 | $\pm$ | 1.75 | 94.78 | $\pm$ | 7.36 |
| 4.2\% | 18.37 | $\pm$ | 1.08 | 0.020 | $\pm$ | 0.001 | 0.63 | $\pm$ | 6.20 | 29.33 | 99.00 | 3.73 | 18.18 | $\pm$ | 2.12 | 78.91 | $\pm$ | 9.02 |
| 4.6\% | 19.11 | $\pm$ | 1.92 | 0.043 | $\pm$ | 0.003 | 1.34 | $\pm$ | 14.98 | 13.73 | 97.95 | 1.75 | 18.72 | $\pm$ | 4.81 | 81.18 | $\pm$ | 20.42 |
| 5.0\% | 22.25 | $\pm$ | 1.98 | 0.054 | $\pm$ | 0.004 | 1.70 | $\pm$ | 21.01 | 10.83 | 97.76 | 1.38 | 21.75 | $\pm$ | 6.50 | 94.01 | $\pm$ | 27.39 |
|  |  |  |  |  |  |  |  |  |  | Plateau age (Plateau: steps from $2.0 \%$ to $3.8 \%$ ) Total gas age : |  |  |  |  |  | 91.80 | $\pm$ | 1.80 |
|  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  | 91.40 | $\pm$ | 2.20 |

Table A. 13 (continued) : ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ analytical data of the dated samples.

Table A. 13 (continued) : ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ analytical data of the dated samples.

| Laser output | ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ |  |  | ${ }^{37} \mathrm{Ar}{ }^{3 / 9} \mathrm{Ar}$ |  |  | $\begin{aligned} & { }^{36} \mathrm{Ar}{ }^{39} \mathrm{Ar} \\ & \left(\times 10^{-3}\right) \end{aligned}$ |  |  | K/Ca | $\begin{aligned} & \hline{ }^{40} \mathrm{Ar}^{*} \\ & (\%) \\ & \hline \end{aligned}$ | $\begin{aligned} & { }^{39} \mathrm{ArK} \quad{ }^{40} \mathrm{Ar} * \beta{ }^{39} \mathrm{ArK} \\ & \text { fraction (\%) } \end{aligned}$ |  |  |  | Age(Ma) | $\pm 1 \sigma$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Sample 1290, Laboratory ID: C14088 |  |  |  | Neutron Irradiation ID: PO-2 |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| $J=0.000968$ |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| 1.4\% | 57.63 | $\pm$ | 0.89 | 0.211 | $\pm$ | 0.021 | 78.29 | $\pm$ | 1.98 | 2.79 | 59.88 | 4.80 | 34.51 | $\pm$ | 0.71 | 59.29 | $\pm$ | 1.23 |
| 1.6\% | 72.18 | $\pm$ | 0.56 | 0.111 | $\pm$ | 0.008 | 7.34 | $\pm$ | 0.87 | 5.32 | 97.01 | 8.15 | 70.02 | $\pm$ | 0.60 | 118.31 | $\pm$ | 1.09 |
| 1.8\% | 99.79 | $\pm$ | 0.67 | 0.048 | $\pm$ | 0.002 | 30.80 | $\pm$ | 3.41 | 12.24 | 90.88 | 33.01 | 90.69 | $\pm$ | 1.18 | 151.81 | $\pm$ | 1.99 |
| 2.0\% | 97.23 | $\pm$ | 0.26 | 0.076 | $\pm$ | 0.005 | 2.09 | $\pm$ | 0.49 | 7.70 | 99.37 | 21.46 | 96.63 | $\pm$ | 0.30 | 161.31 | $\pm$ | 0.77 |
| 2.2\% | 81.53 | $\pm$ | 0.54 | 0.169 | $\pm$ | 0.018 | 5.66 | $\pm$ | 1.53 | 3.48 | 97.96 | 7.64 | 79.88 | $\pm$ | 0.69 | 134.37 | $\pm$ | 1.24 |
| 2.4\% | 62.72 | $\pm$ | 0.64 | 0.222 | $\pm$ | 0.021 | 6.42 | $\pm$ | 2.15 | 2.65 | 97.00 | 5.18 | 60.85 | $\pm$ | 0.89 | 103.26 | $\pm$ | 1.52 |
| 2.5\% | 60.96 | $\pm$ | 0.39 | 0.303 | $\pm$ | 0.035 | 13.00 | $\pm$ | 3.34 | 1.94 | 93.73 | 3.50 | 57.15 | $\pm$ | 1.05 | 97.14 | $\pm$ | 1.78 |
| 2.7\% | 61.49 | $\pm$ | 0.19 | 0.383 | $\pm$ | 0.027 | 10.02 | $\pm$ | 1.69 | 1.53 | 95.24 | 5.09 | 58.58 | $\pm$ | 0.53 | 99.51 | $\pm$ | 0.96 |
| 3.0\% | 64.75 | $\pm$ | 1.31 | 0.351 | $\pm$ | 0.032 | 9.85 | $\pm$ | 4.67 | 1.67 | 95.55 | 5.19 | 61.88 | $\pm$ | 1.88 | 104.95 | $\pm$ | 3.13 |
| 3.4\% | 66.14 | $\pm$ | 0.41 | 0.251 | $\pm$ | 0.014 | 18.61 | $\pm$ | 1.79 | 2.34 | 91.71 | 6.00 | 60.67 | $\pm$ | 0.65 | 102.95 | $\pm$ | 1.14 |
|  |  |  |  |  |  |  |  |  |  | Weighted average age (steps from $2.4 \%$ to $3.4 \%$ ) Total gas age : |  |  |  |  |  | 101.00 | $\pm$ | 0.60 |
|  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  | 133.10 | $\pm$ | 0.70 |

Table A. 13 (continued) : ${ }^{40} \mathrm{Ar}{ }^{\beta 9} \mathrm{Ar}$ analytical data of the dated samples.

| Laser output | ${ }^{40} \mathrm{Ar}{ }^{39} \mathrm{Ar}$ |  |  | ${ }^{37} \mathrm{Ar}{ }^{\text {/9 }} \mathrm{Ar}$ |  |  | ${ }^{36} \mathrm{Ar}^{/{ }^{9}} \mathrm{Ar}\left(\times 10^{-3}\right)$ |  |  | K/Ca | $\begin{aligned} & \hline{ }^{40} \mathrm{~A}^{*} * \\ & (\%) \end{aligned}$ | ${ }^{39} \mathrm{Ar}_{\mathrm{K}}$ fraction(\%) | ${ }^{40} \mathrm{Ar}^{*} \beta^{99} \mathrm{~A}$ |  |  | Age (Ma) |  | $\pm 1 \sigma$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Sample 1214, Laboratory ID: C14043 |  |  |  | Neutron Irradiation ID: PO-1 |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| $J=0.001001$ |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| 1.4\% | 65.62 | $\pm$ | 0.53 | 0.038 | $\pm$ | 0.098 | 168.78 | $\pm$ | 3.31 | 15.41 | 24.00 | 5.38 | 15.75 | $\pm$ | 0.92 | 28.22 | $\pm$ | 1.64 |
| 1.6\% | 41.35 | $\pm$ | 0.45 | 0.081 | $\pm$ | 0.039 | 32.69 | $\pm$ | 1.39 | 7.26 | 76.66 | 13.69 | 31.71 | $\pm$ | 0.55 | 56.37 | $\pm$ | 0.99 |
| 1.8\% | 41.47 | $\pm$ | 0.30 | 0.029 | $\pm$ | 0.031 | 9.05 | $\pm$ | 1.06 | 20.44 | 93.56 | 15.88 | 38.80 | $\pm$ | 0.43 | 68.75 | $\pm$ | 0.79 |
| 2.0\% | 41.40 | $\pm$ | 0.38 | 0.008 | $\pm$ | 0.027 | 5.64 | $\pm$ | 0.97 | 73.08 | 95.98 | 15.20 | 39.74 | $\pm$ | 0.47 | 70.37 | $\pm$ | 0.86 |
| 2.2\% | 42.33 | $\pm$ | 0.47 | 0.005 | $\pm$ | 0.020 | 6.47 | $\pm$ | 0.93 | 125.55 | 95.49 | 13.89 | 40.42 | $\pm$ | 0.53 | 71.55 | $\pm$ | 0.96 |
| 2.4\% | 42.45 | $\pm$ | 0.44 | 0.133 | $\pm$ | 0.032 | 8.85 | $\pm$ | 1.24 | 4.42 | 93.86 | 11.24 | 39.85 | $\pm$ | 0.56 | 70.57 | $\pm$ | 1.01 |
| 2.6\% | 41.26 | $\pm$ | 0.47 | 0.061 | $\pm$ | 0.026 | 7.39 | $\pm$ | 0.79 | 9.61 | 94.72 | 12.91 | 39.08 | $\pm$ | 0.50 | 69.23 | $\pm$ | 0.92 |
| 2.8\% | 43.18 | $\pm$ | 0.14 | 0.151 | $\pm$ | 0.044 | 14.85 | $\pm$ | 1.72 | 3.89 | 89.86 | 7.44 | 38.81 | $\pm$ | 0.52 | 68.75 | $\pm$ | 0.95 |
| 3.0\% | 48.26 | $\pm$ | 1.30 | 0.228 | $\pm$ | 0.100 | 21.03 | $\pm$ | 4.12 | 2.58 | 87.16 | 2.74 | 42.07 | $\pm$ | 1.66 | 74.41 | $\pm$ | 2.89 |
| 3.4\% | 73.51 | $\pm$ | 1.92 | 0.584 | $\pm$ | 0.206 | 124.94 | $\pm$ | 8.43 | 1.01 | 49.84 | 1.41 | 36.65 | $\pm$ | 2.50 | 65.00 | $\pm$ | 4.37 |
| 4.2\% | 568.28 | $\pm$ | 53.54 | 1.801 | $\pm$ | 1.563 | 1678.28 | $\pm$ | 163.67 | 0.33 | 12.76 | 0.21 | 72.60 | $\pm$ | 14.52 | 126.57 | $\pm$ | 24.45 |
|  |  |  |  | Plateau age (Plateau: steps from $1.8 \%$ to $3.4 \%$ ) . Total gas age : |  |  |  |  |  |  |  |  |  |  |  | 69.86 | $\pm$ | 0.37 |
|  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  | 66.03 | $\pm$ | 0.34 |

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## List of Publications and Patents:

## Articles

Okay, A.I., Sunal, G., Sherlock, S., Altıner, D., Tüysüz, T., Kylander-Clark, A.R.C., Aygül, M., 2013. Early Cretaceous sedimentation and orogeny on the active margin of Eurasia: southern Central Pontides, Turkey. Tectonics, 32, 1247-1271.

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